

Figure S19:

In this attached pdf file, all the results for the 84 OTPs found in wastewater treatment are visually shown. The following information is given for each OTP:

- Top left: OTP formation in ozonation batch experiments
- Top right: Fate of parent MP in wastewater treatment
- Middle: (left) Fate of OTP in post-treatment (% formed or abated) and (right) OTP peak area in all wastewater samples in the four WWTPs. The score and evaluation of MS² spectra match of WWTP OZO samples to batch samples is given below.
- Bottom: Information on OTP identification (exact mass, formula, atomic modification, proposed structure, confidence level, Massbank identifier, MS² spectra and interpretation of it)

This Figure is part of the *Supporting Information* document belonging to the following publication (citations of references can be found in the *SI* document):

Formation of transformation products during ozonation of secondary wastewater effluent and their fate in post-treatment: From laboratory- to full-scale

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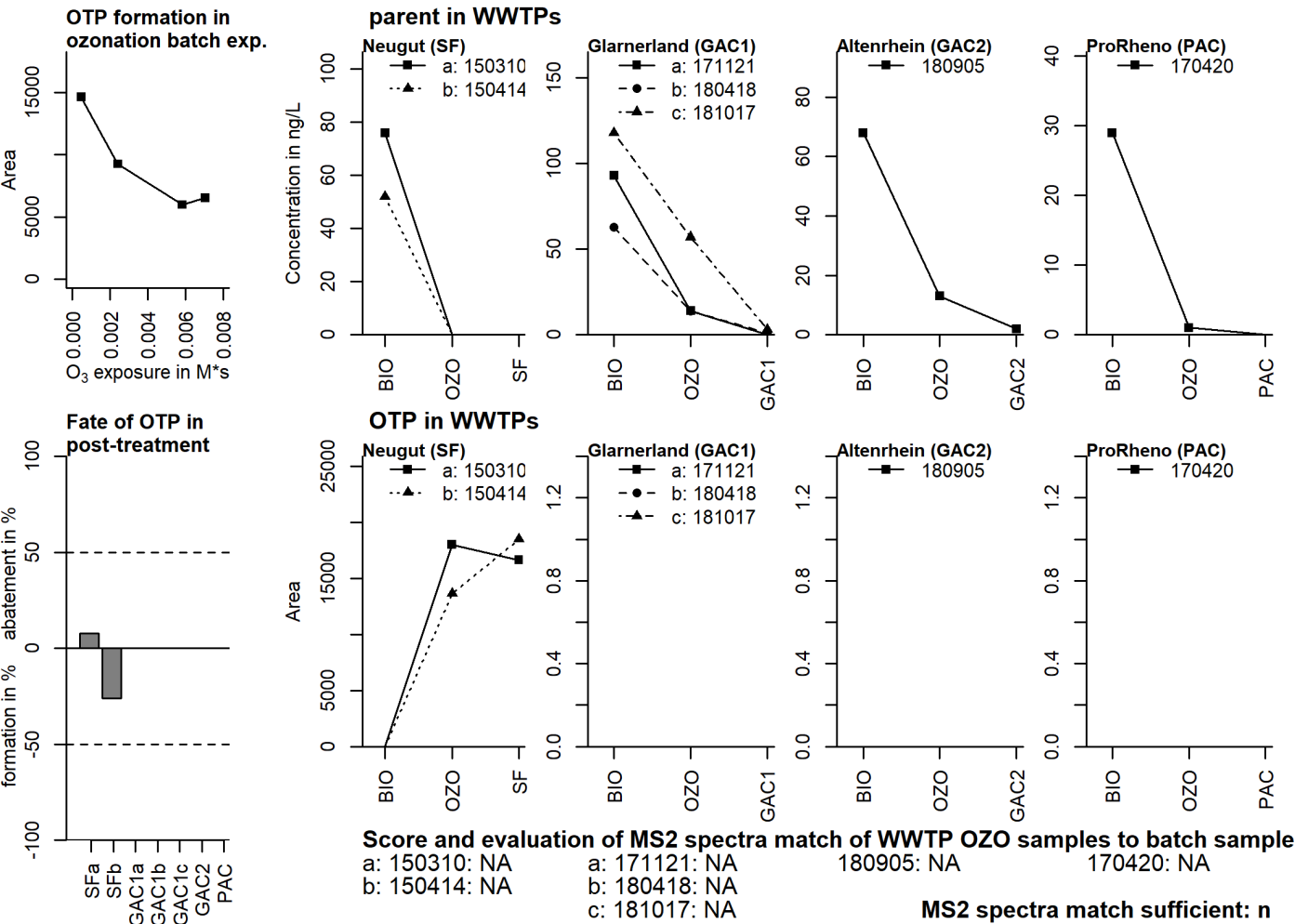
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<https://doi.org/10.1016/j.watres.2021.117200>

The atomic modification from the elemental formula of the parent compound to this OTP is -C18H31NO3. This modification fits to a cleavage of the parent molecule at a C-C bond in the middle of the molecule, followed by an oxidation. Two hydrogen atoms need to be abstracted from the drawn structure to fit the chemical formula. This was probably done by oxidation of the primary alcohol to a carbonyl but there is no indication for this. The exact type and position of the modification(s) remain unknown. The structures of the MS² fragments are drawn exemplarily.

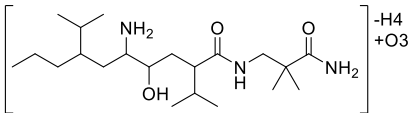


MS Spectra
Neg 428.2762 [m-H]-

Formula
C₂₁H₃₉O₆N₃

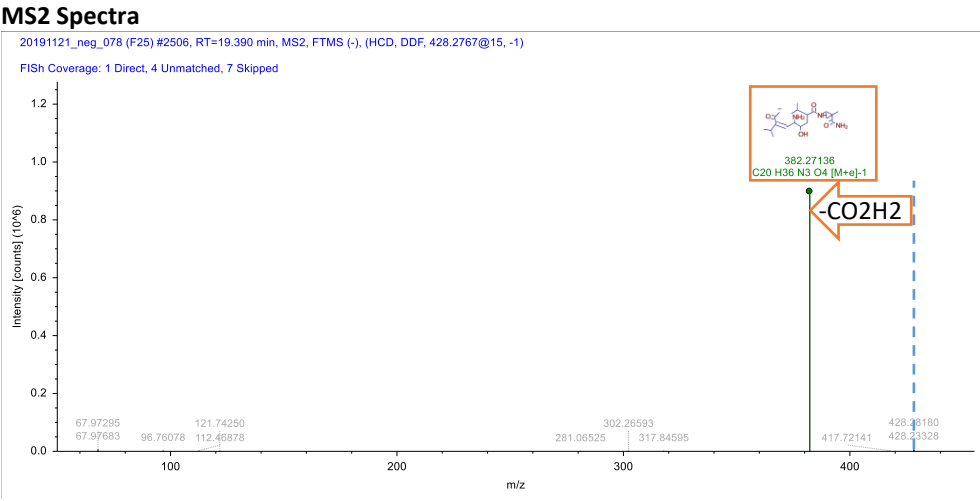
Atomic modification
-C₉H₁₄

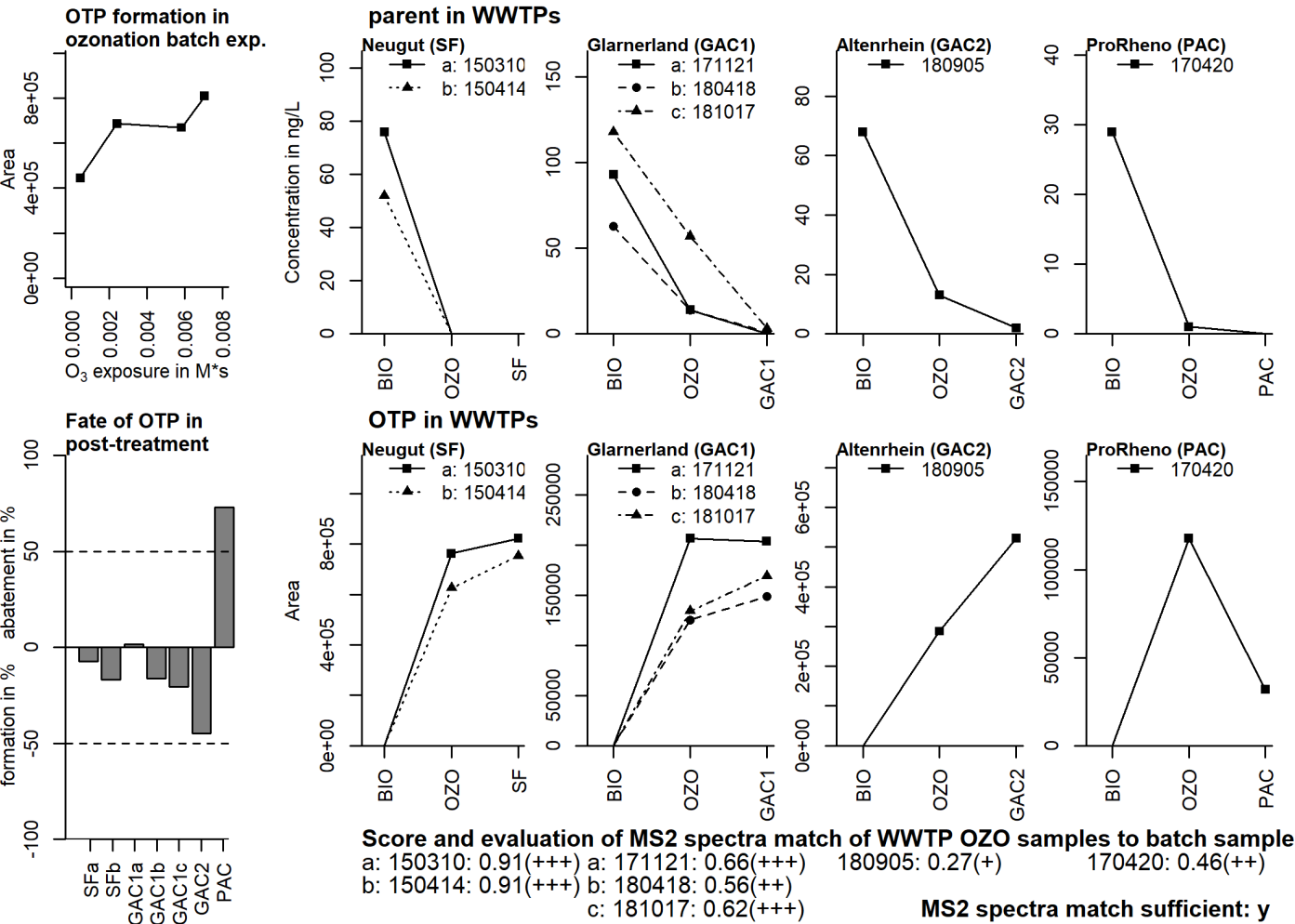
Proposed Structure



Confidence Level
Level 3

Massbank ID
ET404701



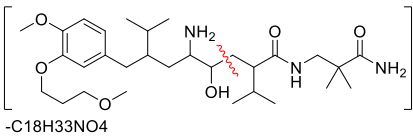


MS Spectra
Pos 225.1597 [m+H]⁺

Formula
C₁₂H₂₀O₂N₂

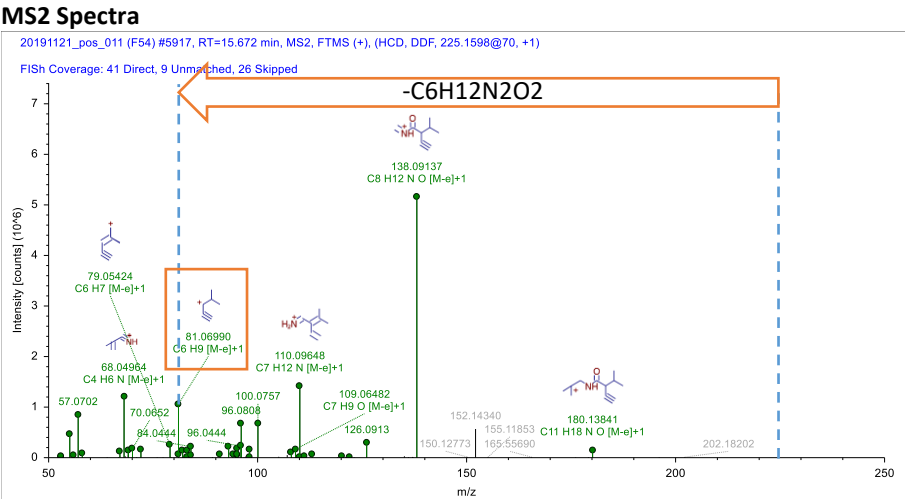
Atomic modification
-C₁₈H₃₃NO₄

Proposed Structure



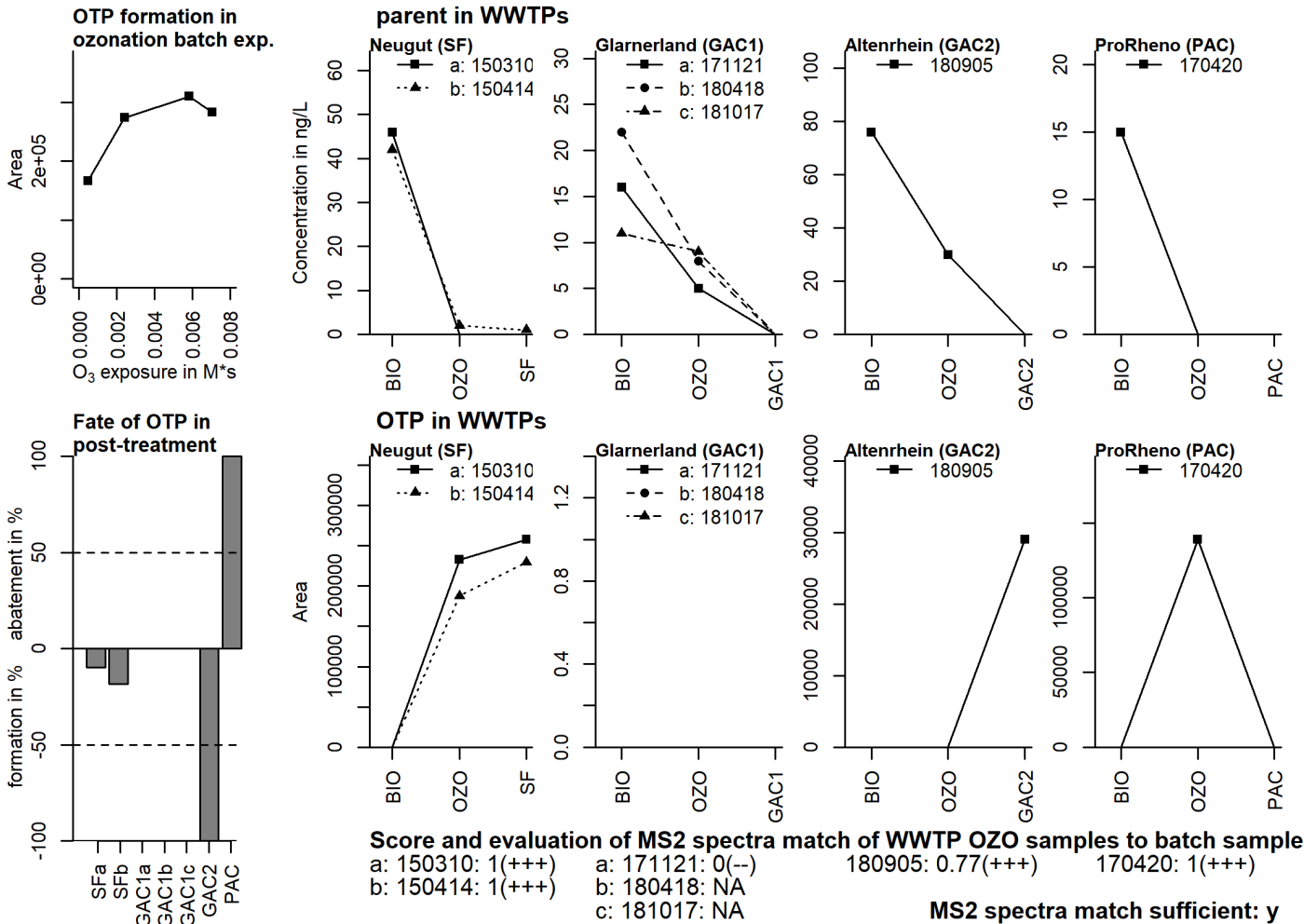
Confidence Level
Level 3

Massbank ID
ET400001



Additional Evidence for Structure Interpretation

The atomic modification from the elemental formula of the parent compound to this OTP is -C₁₈H₃₃NO₄. This modification fits to a cleavage of the parent molecule at a C-C bond in the middle of the molecule, as shown by the red wavy line. The part of the parent structure right of the line remained. To fit the molecular formula, four hydrogen atoms need to be abstracted. The exact type and position of the modification(s) remain unknown. The structures of the MS² fragments are drawn exemplarily.



MS Spectra

Pos 416.1486 [m+H]⁺

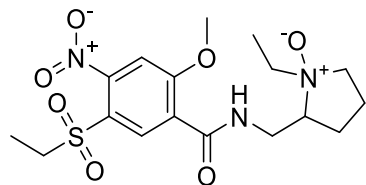
Formula

C₁₇H₂₅O₇N₃S

Atomic modification

-H₂ +O₃

Proposed Structure



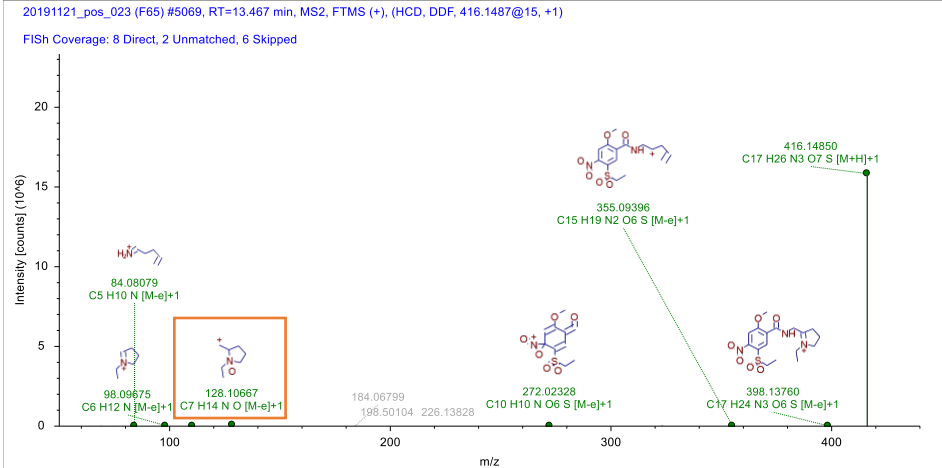
Confidence Level

Level 3

Massbank ID

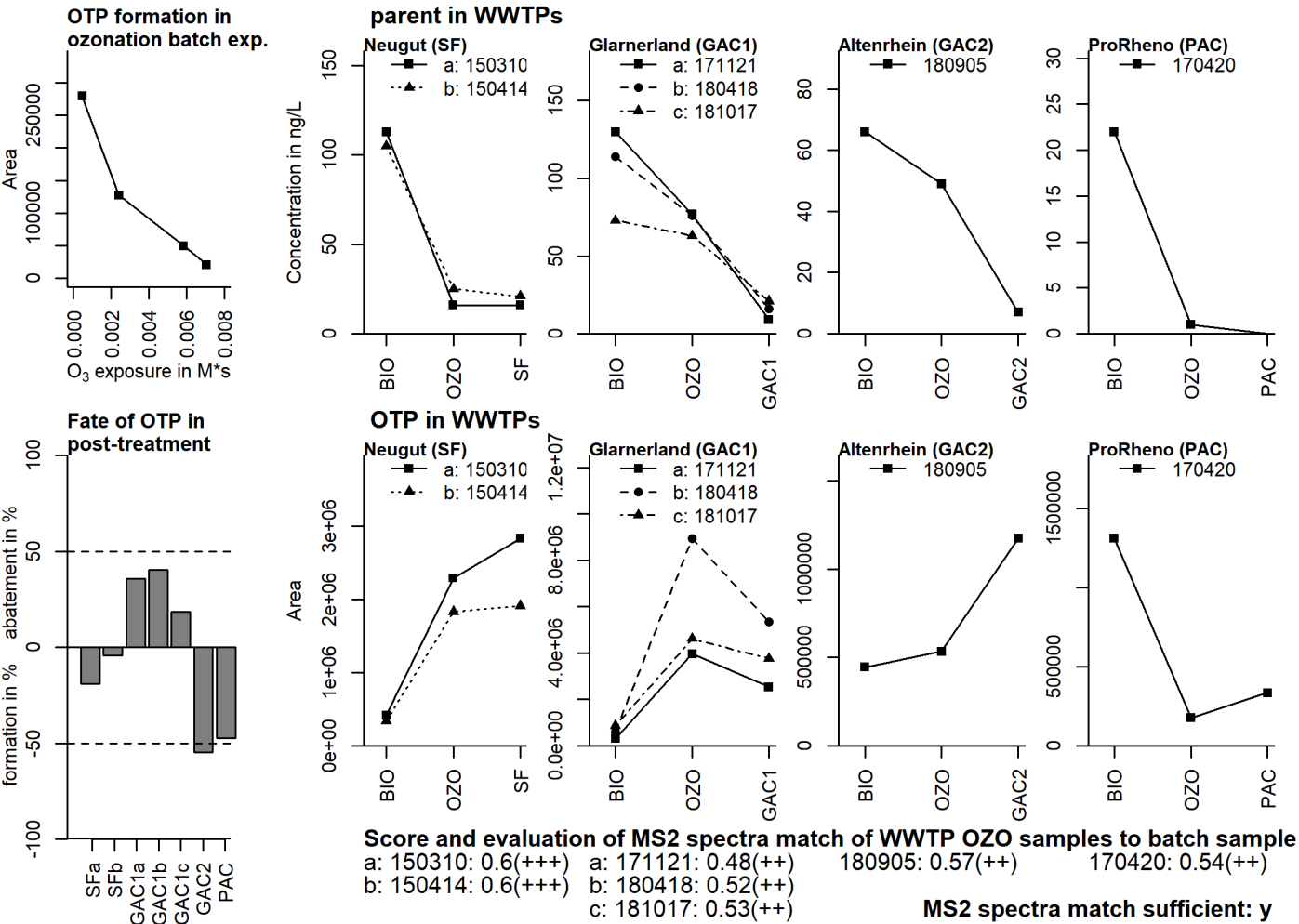
ET400101

MS2 Spectra



Additional Evidence for Structure Interpretation

A MS² fragment at the nominal mass of 112 was observed for the parent compound and corresponds to the ethylpyrrolidine moiety. The MS² fragment here at the nominal mass 128 corresponds to the fragment 112 of the parent with an addition of an oxygen atom. Tertiary amines are very reactive to form *N*-oxides during ozonation (von Sonntag & von Gunten 2012). Therefore, the formation of an *N*-oxide at the ethylpyrrolidine is very likely. A MS² fragment at the nominal mass 242 was observed for the parent compound and corresponds to the aromatic moiety. The fragment 272 corresponds to the fragment 242 at the parent moiety with an atomic modification of -H₂ +O₂. Aniline-like moieties are known to partially form nitro groups during ozonation (von Sonntag & von Gunten 2012), which would also fit here). Another possibility for an -H₂ +O₂ modification, namely the formation of a carboxylic acid of a terminal carbon, seems unlikely due to the absence of the characteristic neutral loss of water and carbon dioxide. Also, no matching signal was observed in negative mode.

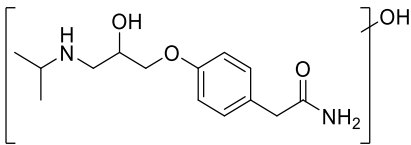


MS Spectra
Pos 283.1666 [m+H]⁺

Formula
C₁₄H₂₂O₄N₂

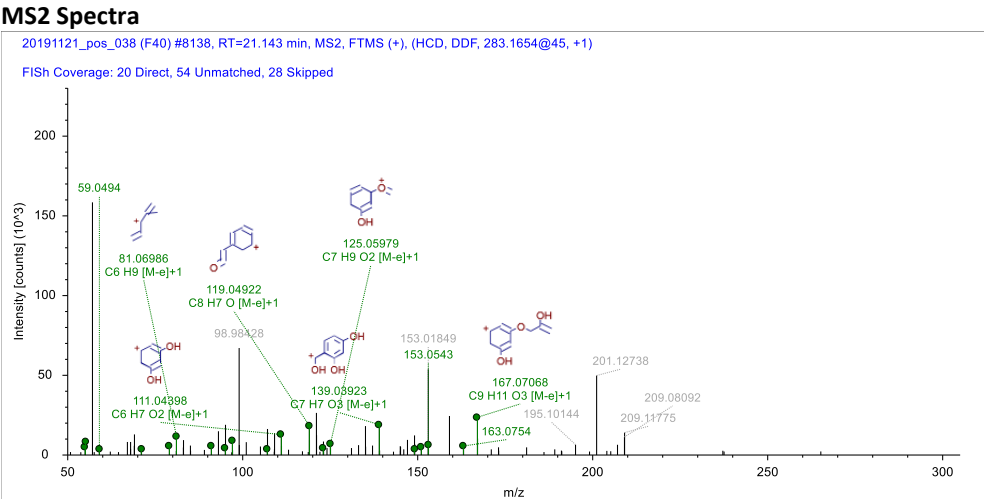
Atomic modification
+O

Proposed Structure



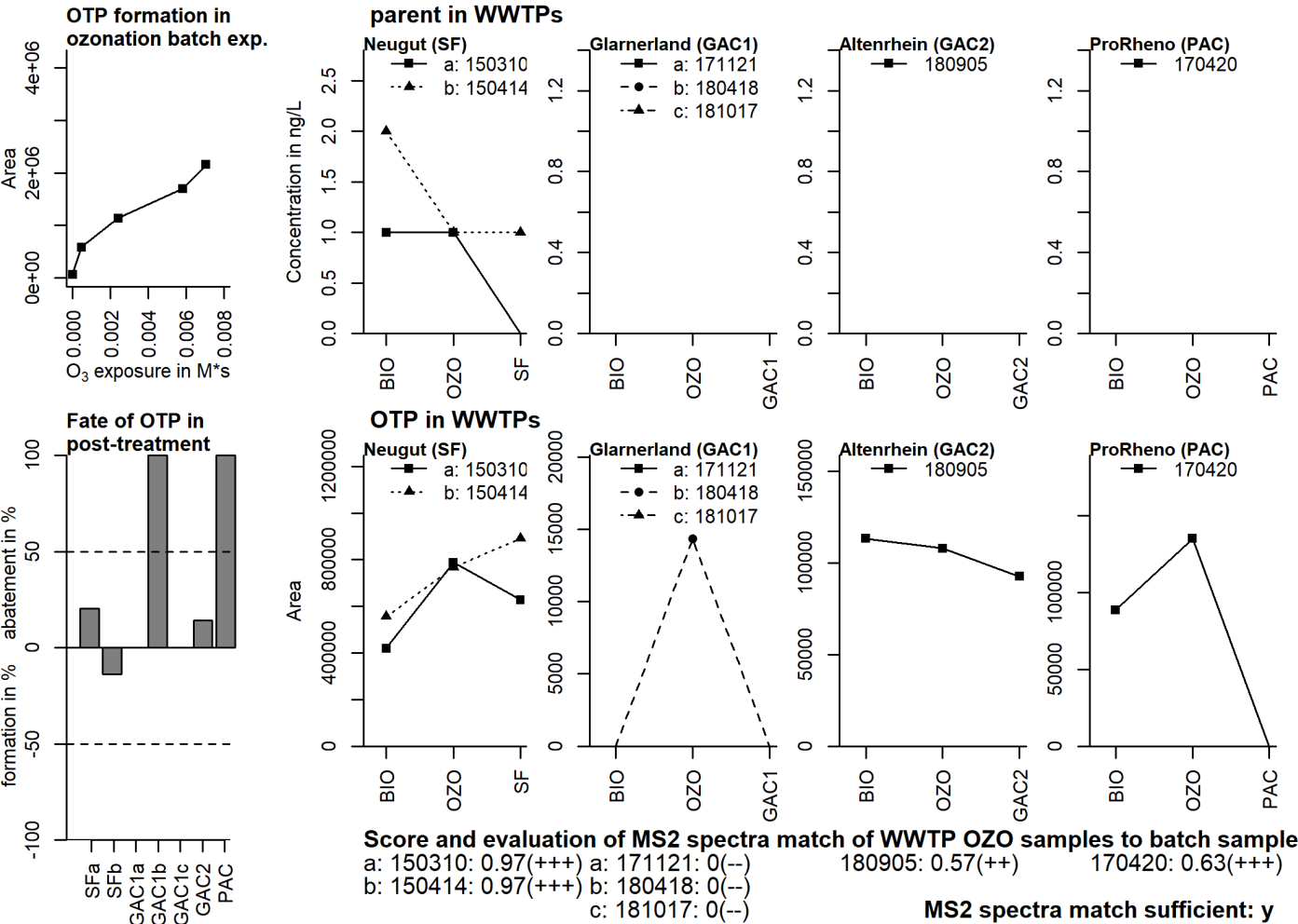
Confidence Level
Level 3

Massbank ID
ET400201



Additional Evidence for Structure Interpretation

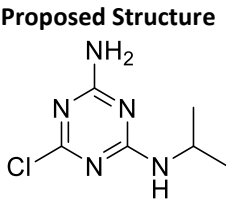
The atomic modification from the elemental formula of the parent compound to this TP is + O. This suggests a hydroxylation, which is known to happen during ozonation (von Sonntag & von Gunten 2012), likely by an OH radical. The hydroxylation likely occurred at an aliphatic C-atom, because once formed, phenol moieties react further quickly during ozonation. The exact location of the hydroxyl moiety remains unknown. The structures of the MS² fragments are drawn exemplarily. Hydroxylated atenolol was suggested as OTP of atenolol in different studies (Quaresma et al. 2019, Tay et al. 2011, Xu et al. 2019).



MS Spectra
Pos 188.0697 [m+H]⁺

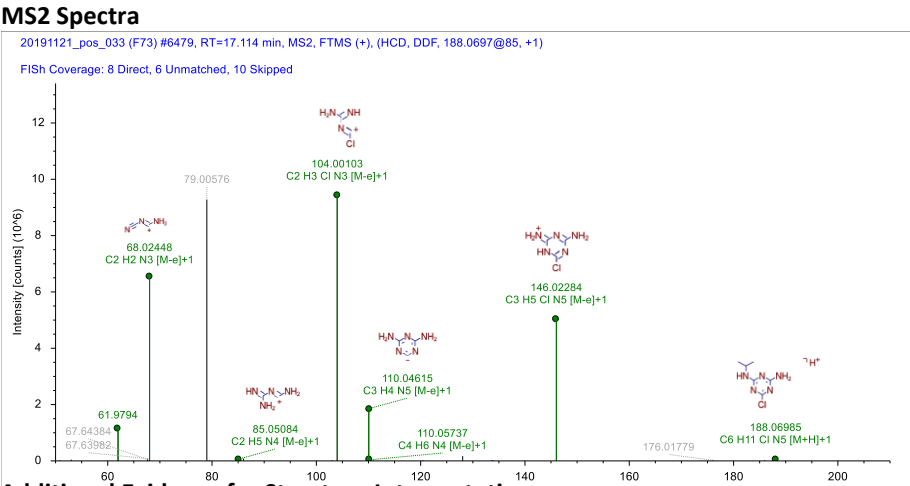
Formula
C₆H₁₀N₅Cl

Atomic modification
-C₂H₄



Confidence Level
Level 3

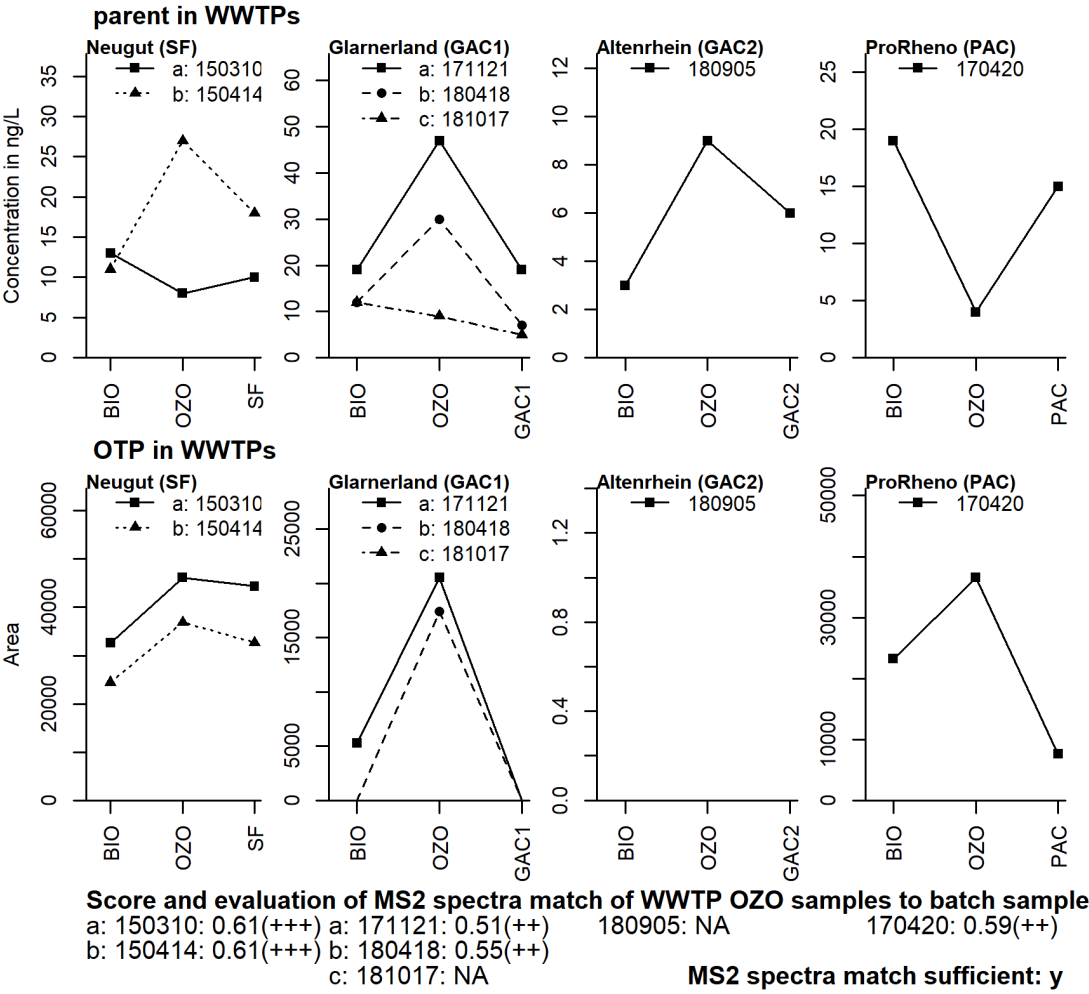
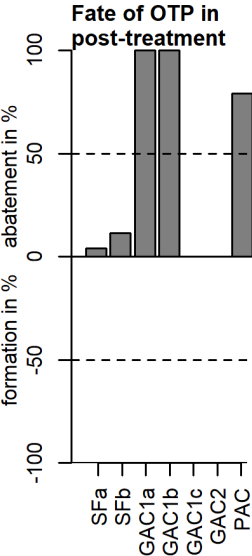
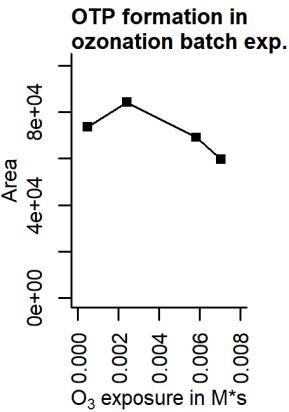
Massbank ID
ET400301



Additional Evidence for Structure Interpretation

The MS² fragments at the nominal masses 146, 104, 79, 68, 61 were also observed for the parent compound and indicate that the chlorotriazineamine moiety is also part of this TP structure.

The atomic modification from the elemental formula of the parent compound to this TP is -C₂H₄, which suggests an *N*-deethylation, which is known to happen during ozonation (von Sonntag & von Gunten 2012). Deethylatrazine was suggested as ozone/hydroxyl radical transformation product in different studies (Acero et al. 2000, Beltran et al. 1998, Barletta et al. 2003).

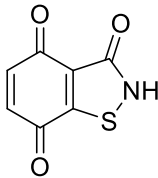


MS Spectra
Neg 215.9525 [m+Cl]-

Formula
C7H3O3NS

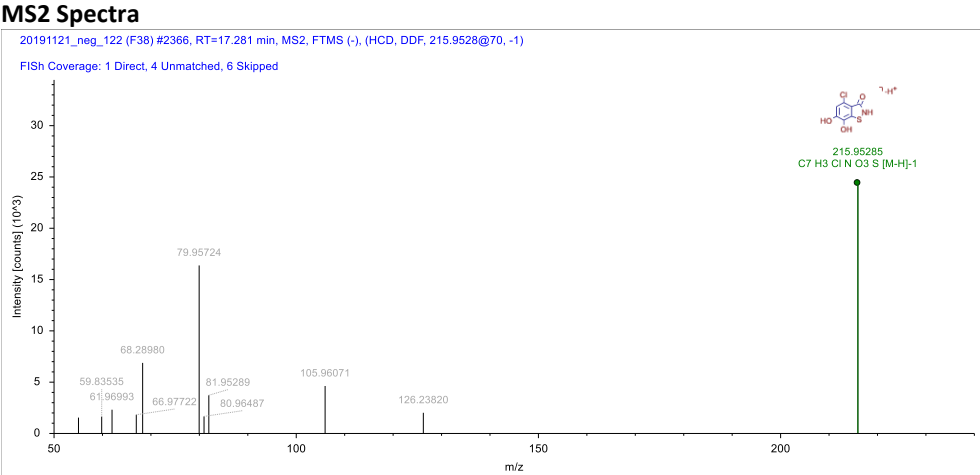
Atomic modification
-H2 + O2

Proposed Structure



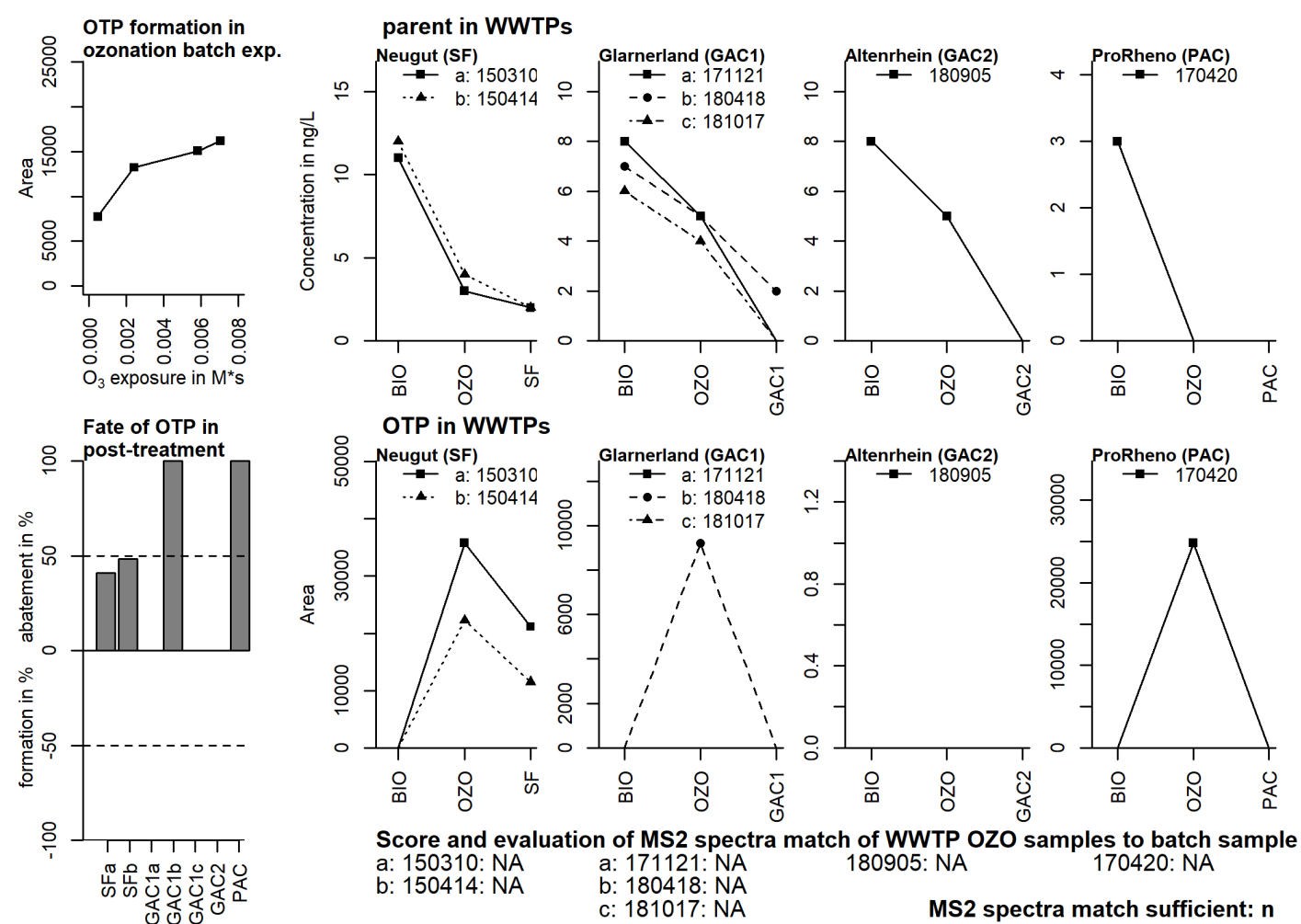
Confidence Level
Level 3

Massbank ID
ET404801



Additional Evidence for Structure Interpretation

The atomic modification from the elemental formula of the parent compound to this TP is -H2 +O2, which fits to the formation of a benzoquinone moiety after attack of OH radicals at the phenyl moiety to form a phenol intermediate, which can the be transformed to a benzoquinone (Tentscher et al. 2018, Ramseier and von Gunten 2009).



MS Spectra

Pos 212.0469 [m+H]⁺

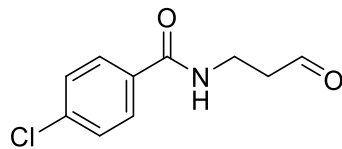
Formula

C₁₀H₁₀O₂NC_l

Atomic modification

-C₉H₁₀O₂

Proposed Structure



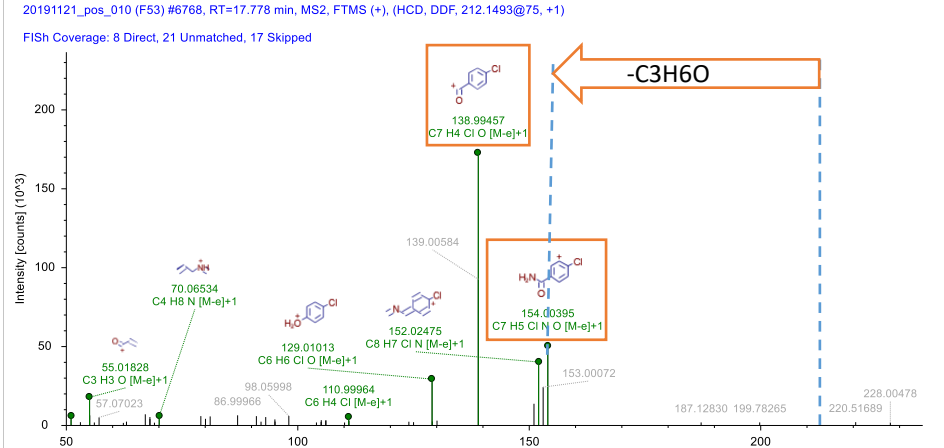
Confidence Level

Level 3

Massbank ID

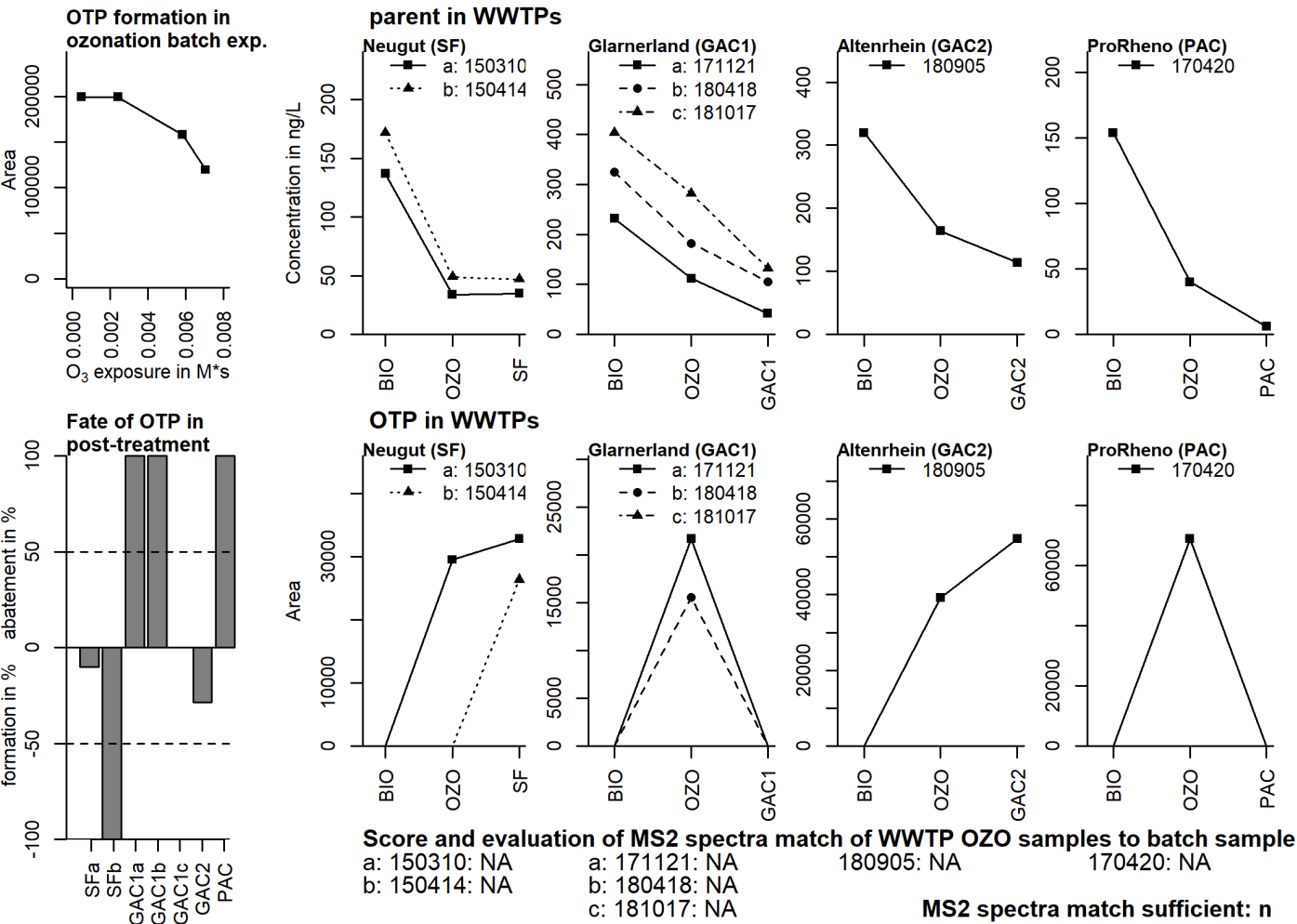
ET400501

MS2 Spectra



Additional Evidence for Structure Interpretation

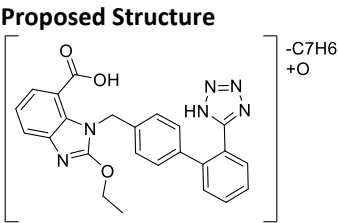
The MS² fragment at the nominal mass 138 was also observed for the parent compound and for BZF_p_156.0209_16.9. It indicates that the structure drawn for the fragment 138 is a substructure of this TP. The fragment 55, which corresponds to a C₃H₃O moiety, indicates that neutral loss of –C₃H₆O between the precursor and the fragment 154 were one single moiety. In the structure of the parent compound a phenylethyl moiety is connected to the chlorobenzamide substructure. Therefore, it is likely that the C₃H₆O moiety is also connected to the nitrogen of the chlorobenzamide substructure. The formula of this TP fits to a Criegee mechanism on the phenoxy moiety of the parent compound, followed by a C–C bond cleavage, which is known to occur during ozonation (von Sonntag & von Gunten 2012).



MS Spectra
Neg 365.1000 [m-H]-

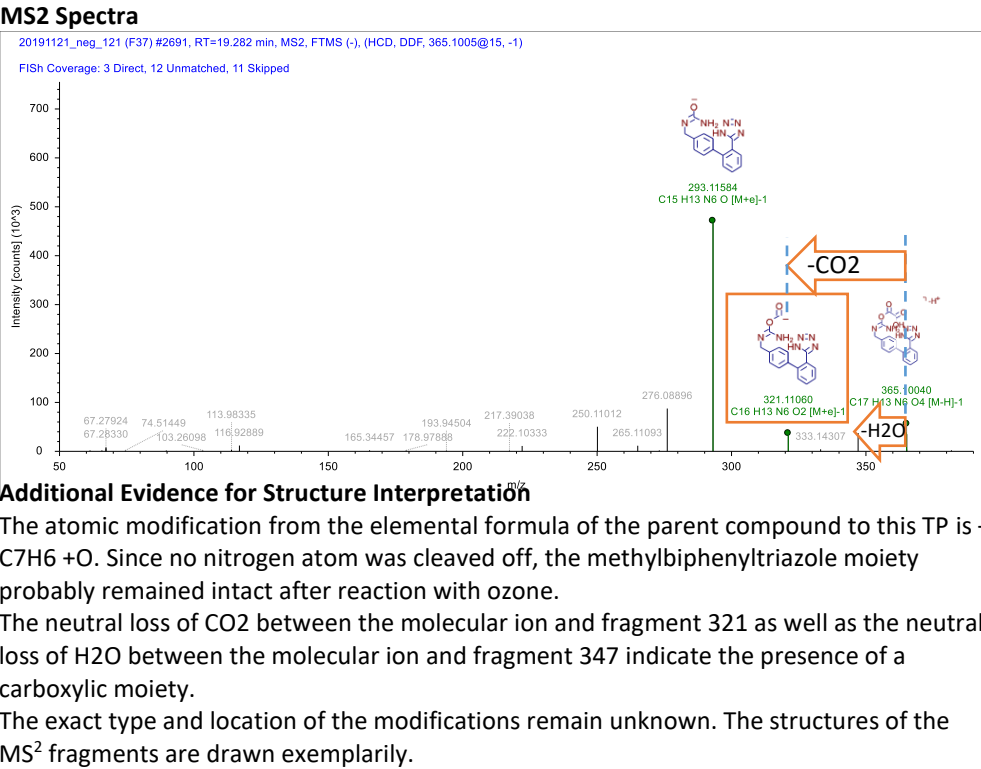
Formula
C17H14O4N6

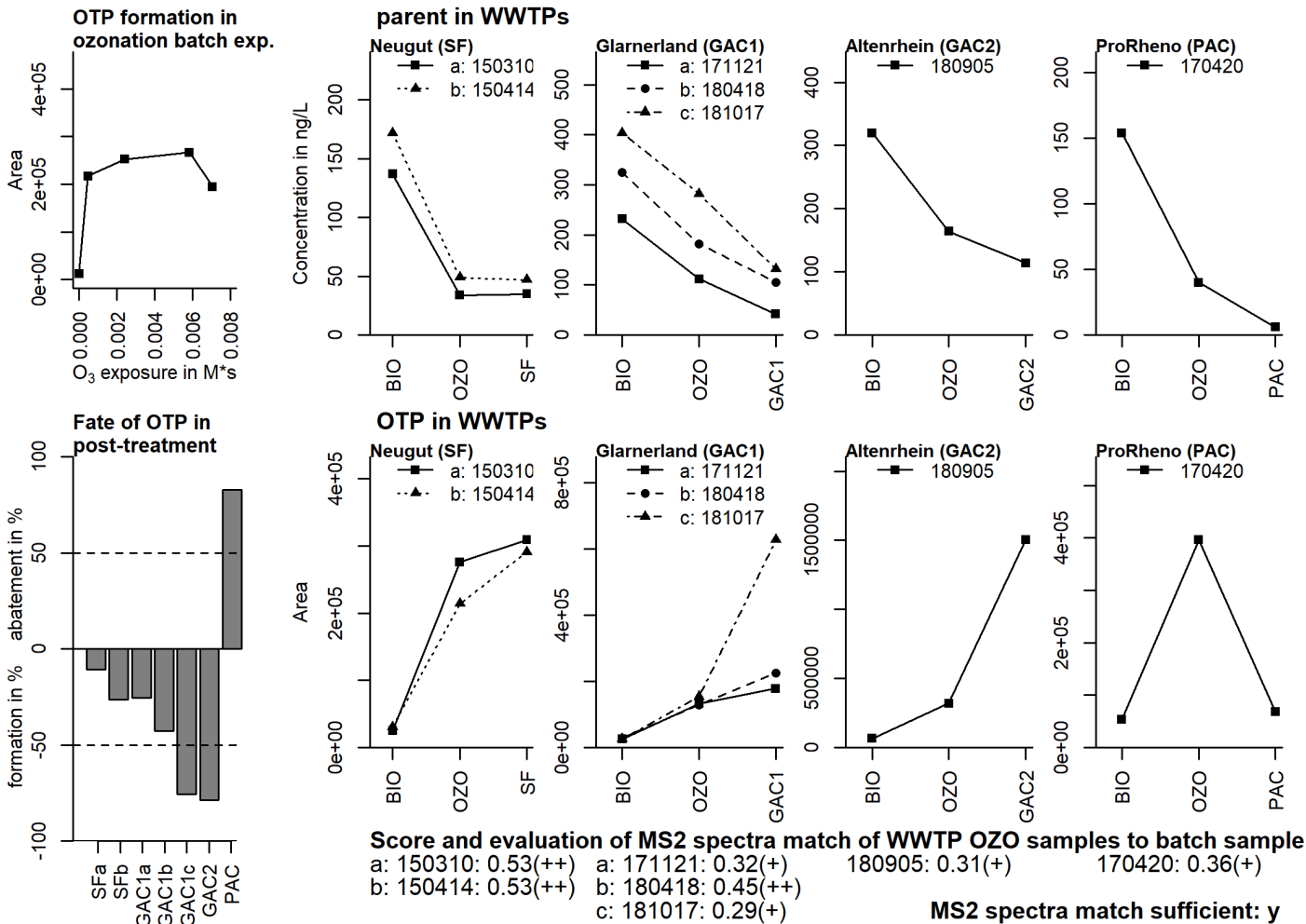
Atomic modification
-C7H6 +O



Confidence Level
Level 3

Massbank ID
ET404801





MS Spectra

Pos 207.0766 [m+H]⁺
Neg 205.0616 [m-H]⁻

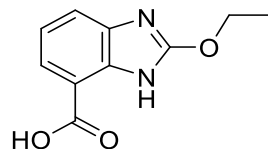
Formula

C₁₀H₁₀O₃N₂

Atomic modification

-C₁₄H₁₀N₄

Proposed Structure



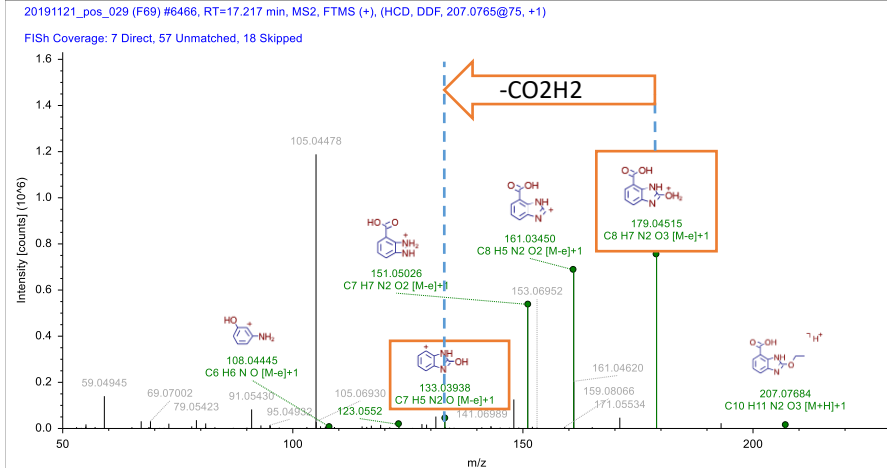
Confidence Level

Level 3

Massbank ID

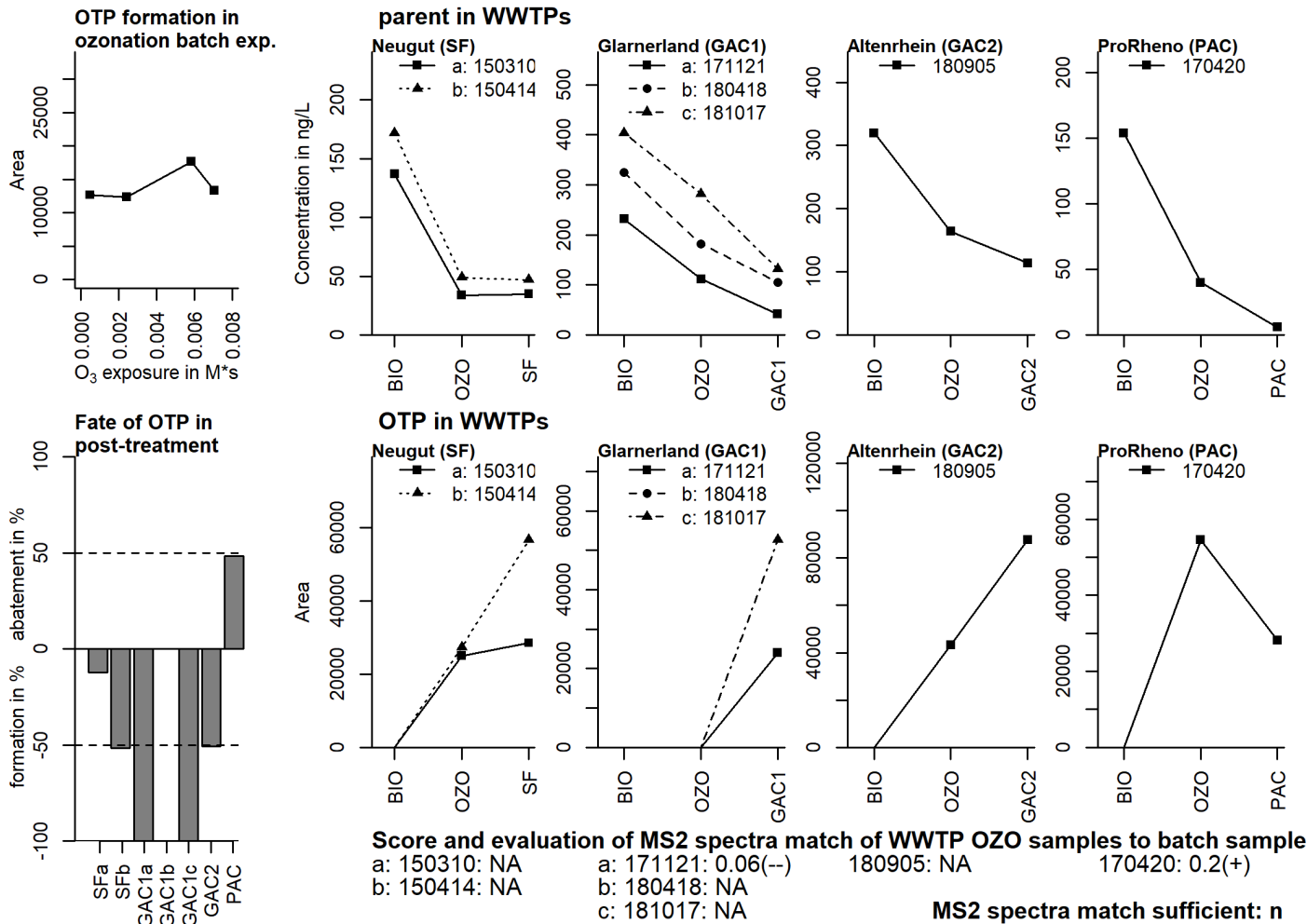
ET400601

MS2 Spectra



Additional Evidence for Structure Interpretation

The atomic modification of C₁₄H₁₀N₄ from the parent structure to this TP fits to a cleavage of the moiety, which is common for all sartans (tetrazolebiphenylmethyl). The remaining part of the molecule fits to the formula for this TP. Furthermore, this TP was also detected in the negative MS spectrum and a neutral loss between fragments 179 and 133 indicates the presence of a carboxylic acid moiety, meaning that the carboxylic acid moiety is conserved. The proposed structure of the TP Sartan_p_251.0924_17.9 fits to the cleaved tetrazolebiphenylmethyl moiety, to which an oxygen atom was added.



MS Spectra

Pos 341.1130 [m+H]⁺
Neg 339.0983 [m-H]⁻

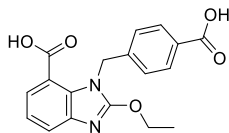
Formula

C₁₈H₁₆O₅N₂

Atomic modification

-C₆H₄N₄ + O₂

Proposed Structure



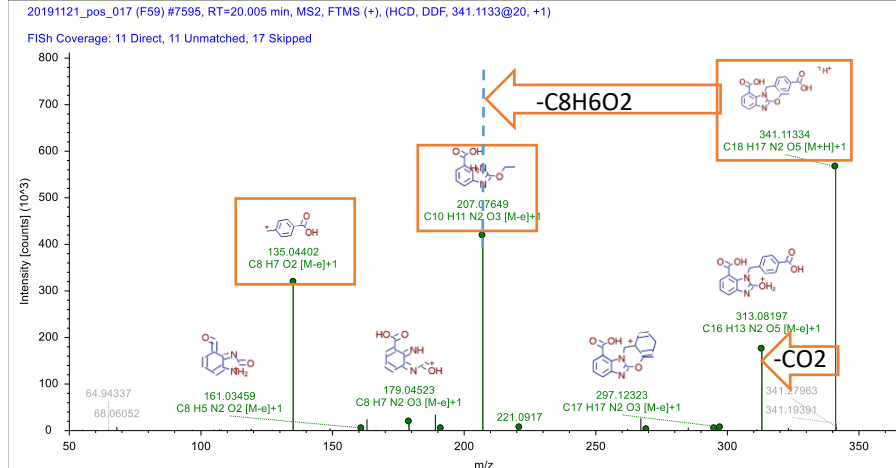
Confidence Level

Level 3

Massbank ID

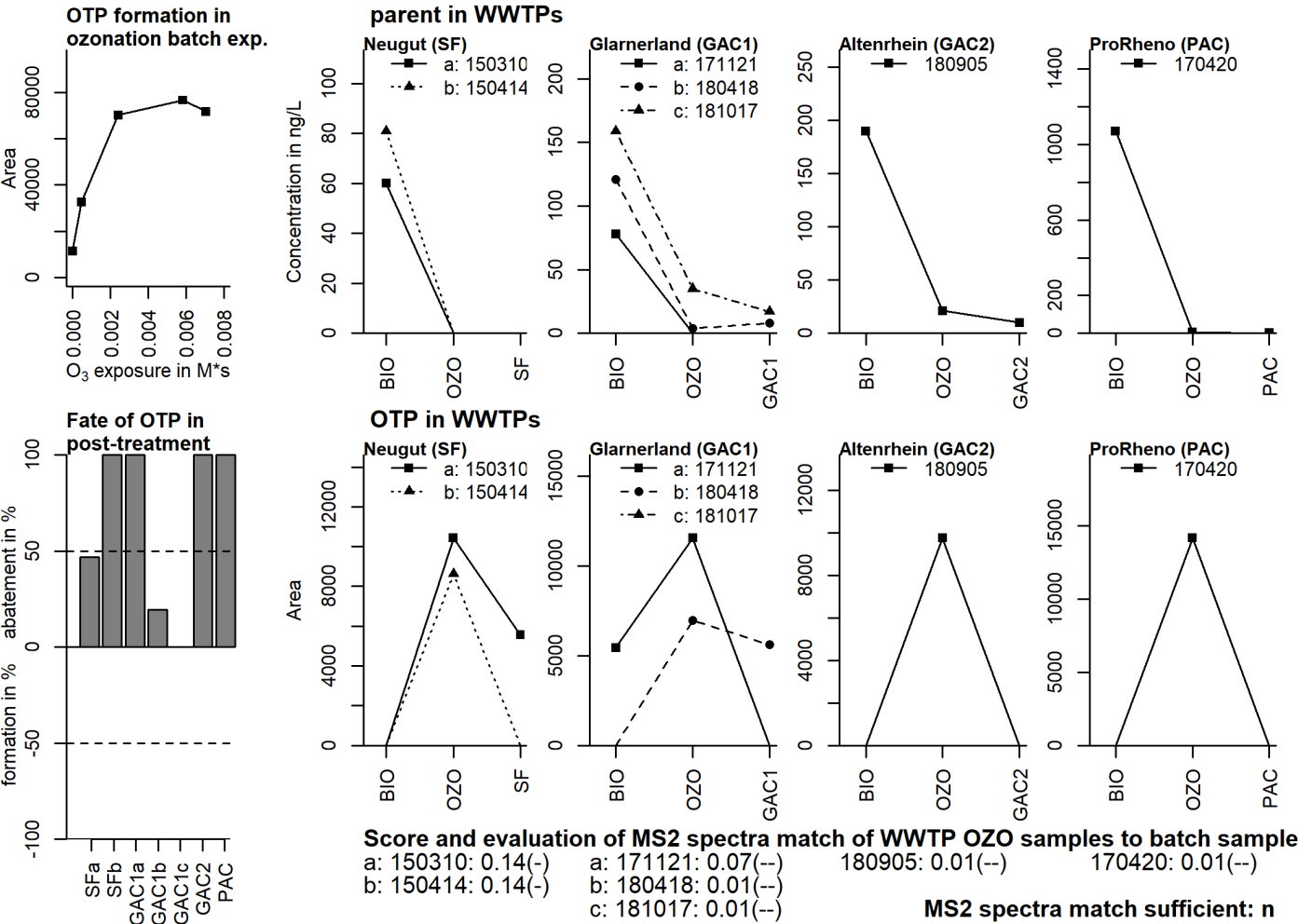
ET400701

MS2 Spectra



Additional Evidence for Structure Interpretation

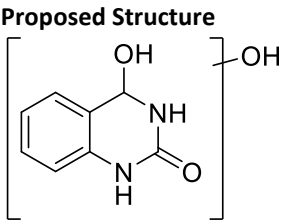
The atomic modification from the elemental formula of the parent compound to this TP is -C₆H₄N₄ + O₂. This fits to a cleavage within the benzol moiety of the tetrazolebiphenylmethyl part and the modification of -H₂ + O₂ atoms. The fragment 135 as well as the neutral loss of C₈H₆O₂ between the precursor and fragment 207 indicate that the two oxygens are connected to the p-methylbenzyl moiety. The exact type and position remain unknown. Aromatic compounds are known to form cis,cis-Muconic acid like structures (von Sonntag & von Gunten 2012). Therefore, the formation of this kind of structure, followed by a C-C bond cleavage, resulting in a p-benzoic acid moiety, is probable. There is however no further evidence. The structures of the MS² fragments are drawn exemplarily.



MS Spectra
Neg 179.0461 [m-H]-

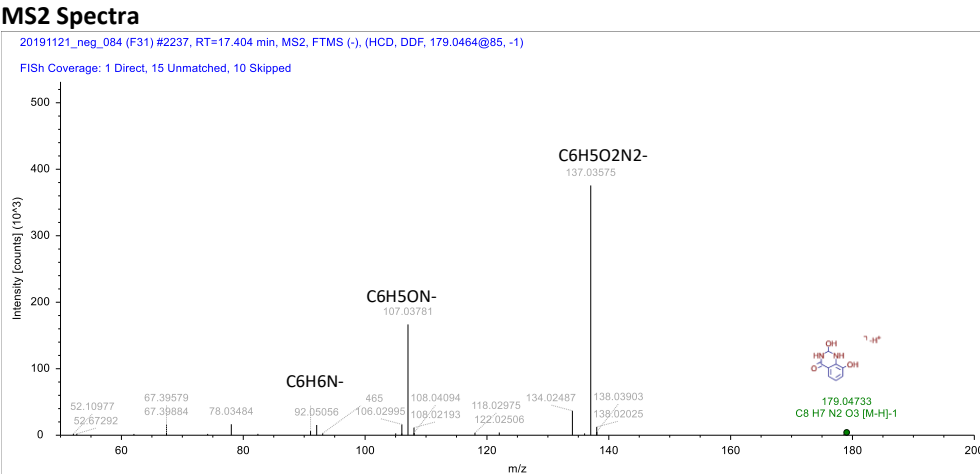
Formula
C8H8O3N2

Atomic modification
-C7H4 + O2



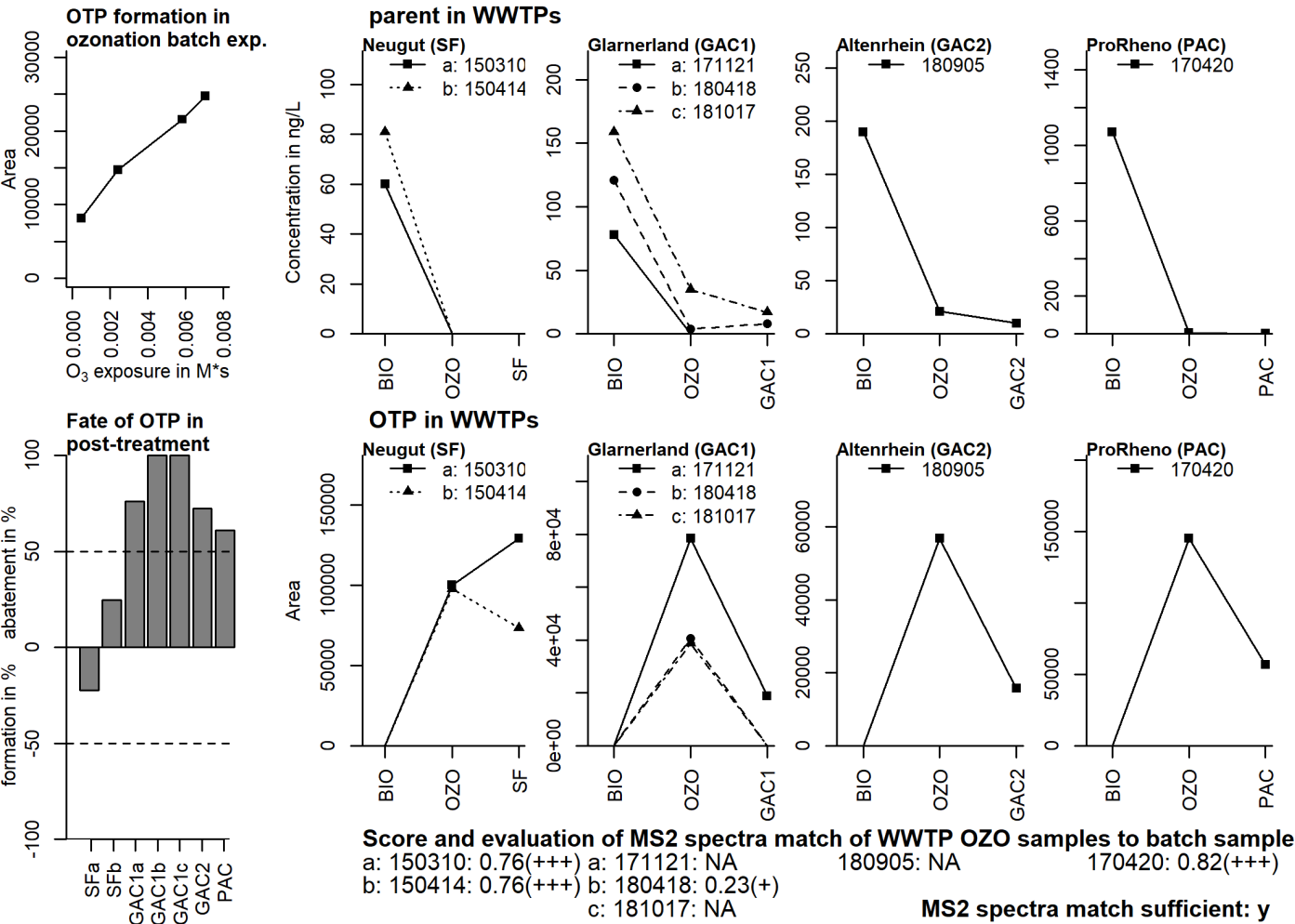
Confidence Level
Level 3

Massbank ID
ET405001



Additional Evidence for Structure Interpretation

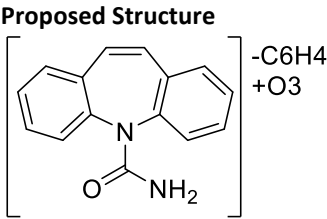
Azais et al. (2017) identified the drawn structure on the MS² spectra as ozonation transformation product of carbamazepine, more precisely as degradation product of BQD and/or BaQD. Since both BQD and BaQD were detected, it is possible that these structures further reacted to form this TP. Another possibility is the formation a hydroxylamine moiety. Both the hydroxylamine and the phenol moiety quickly react further during ozonation. Therefore, the exact structure of this TP remains unclear. The structures of the MS² fragments are drawn exemplarily.



MS Spectra
Neg 207.0409 [m-H]-

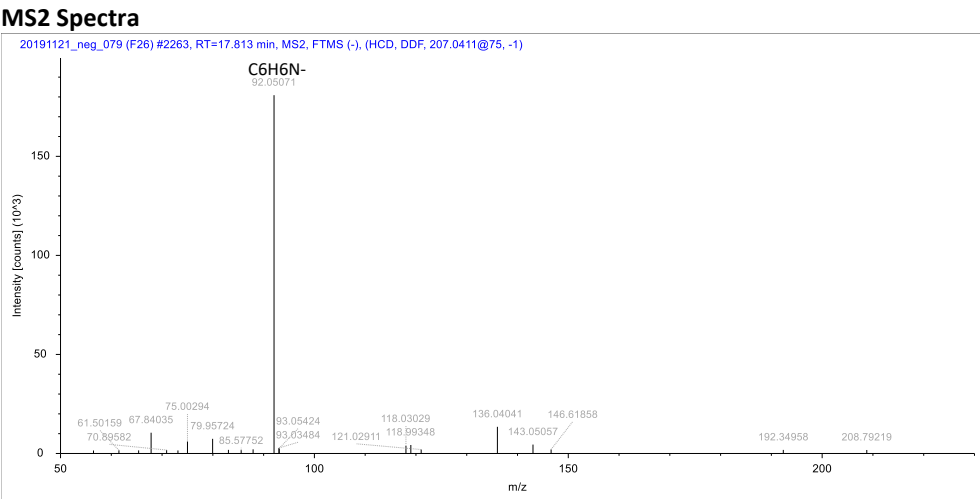
Formula
C9H8O4N2

Atomic modification
-C6H4 +O3



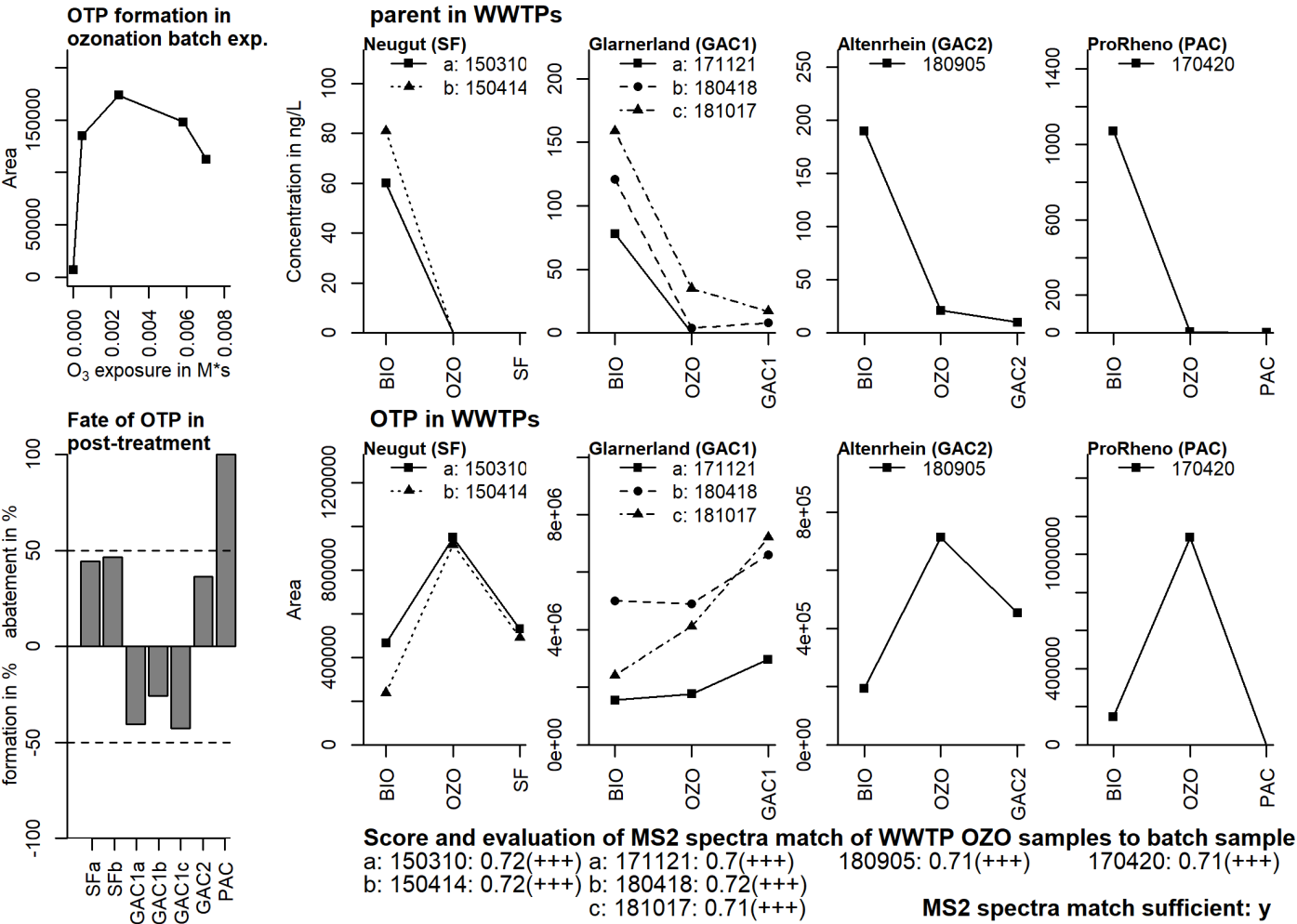
Confidence Level
Level 3

Massbank ID
ET405101



Additional Evidence for Structure Interpretation

The atomic modification from the elemental formula of the parent compound to this TP is – C6H4 +O3. This modification fits to a cleavage of a phenyl moiety and the addition of 3 oxygen atoms. There is however no further evidence. The exact type and location of the modifications remain unknown.

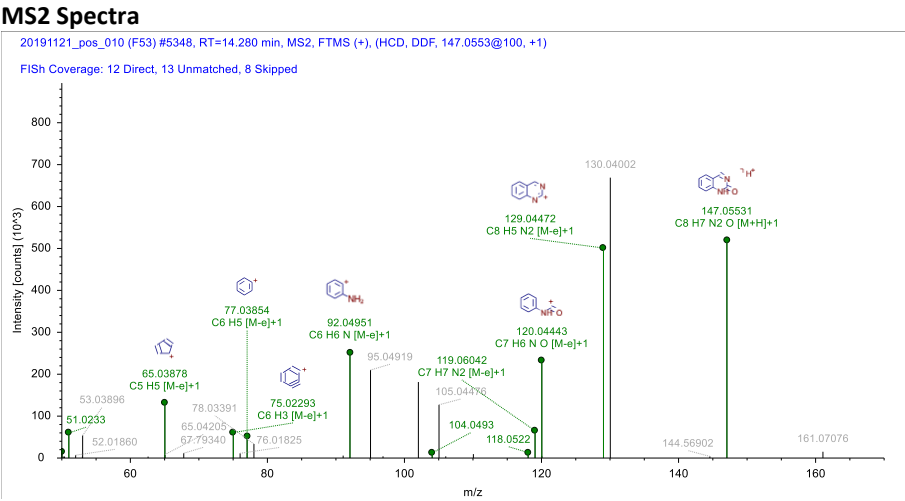
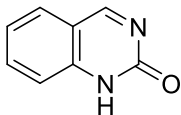


MS Spectra
Pos 147.0552 [m+H]⁺

Formula
C₈H₆ON₂

Atomic modification
-C₇H₄

Proposed Structure

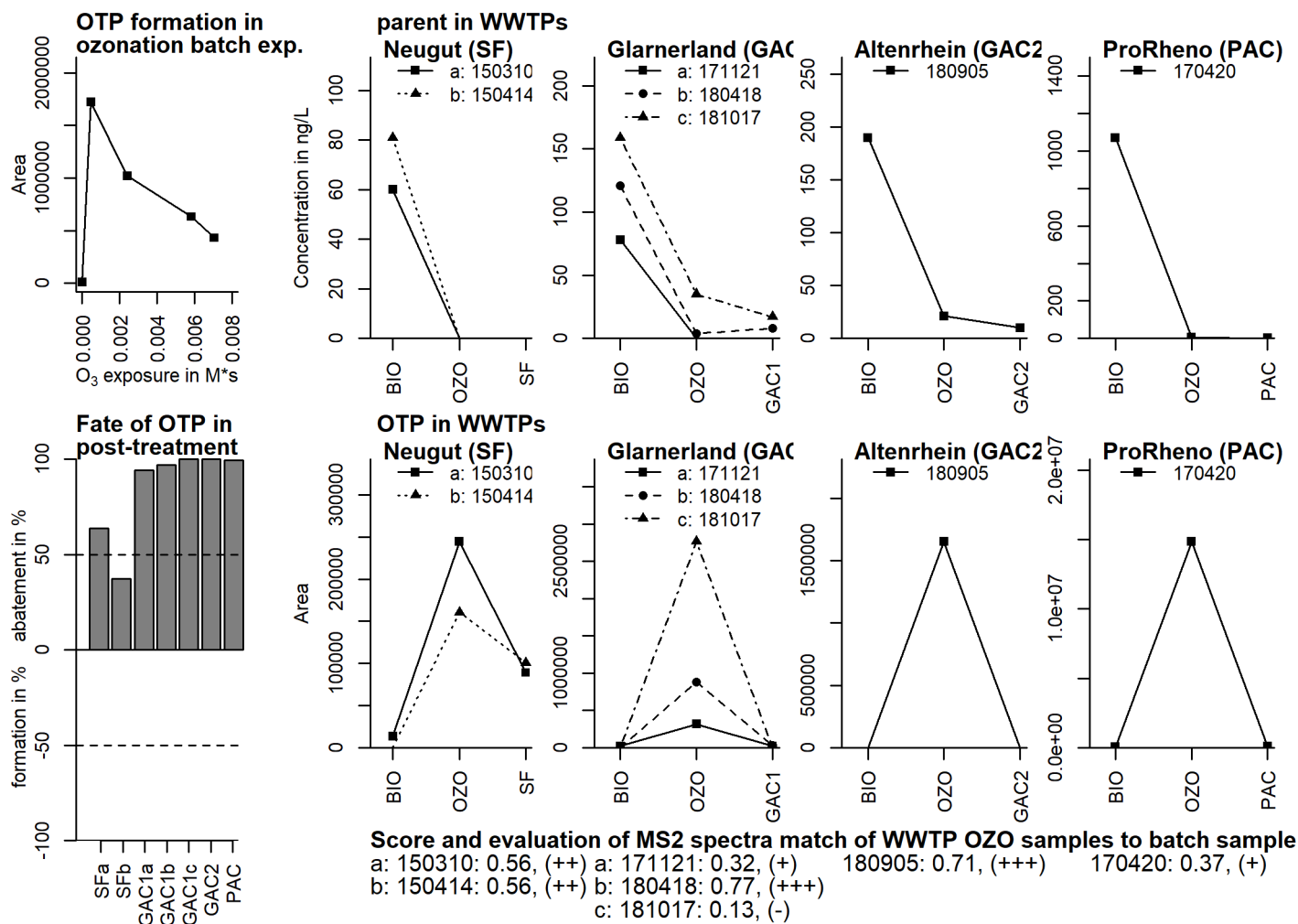


Additional Evidence for Structure Interpretation

A fragment with the mass 147 and the elemental formula C₈H₆ON₂ was observed in the MS² spectrum of CBZ_p_251.0812_17.9, which was identified as BQM. This indicates that this OTP structure is a substructure of BQM. This fits a cleavage of the benzaldehyde moiety, which was suggested by Azaïs et al. (2017) as a degradation product of BQM.

Confidence Level
Level 3

Massbank ID
ET400801

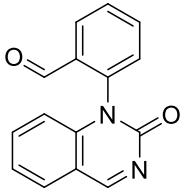


MS Spectra
Pos 251.0812 [m+H]⁺

Formula
C₁₅H₁₀O₂N₂

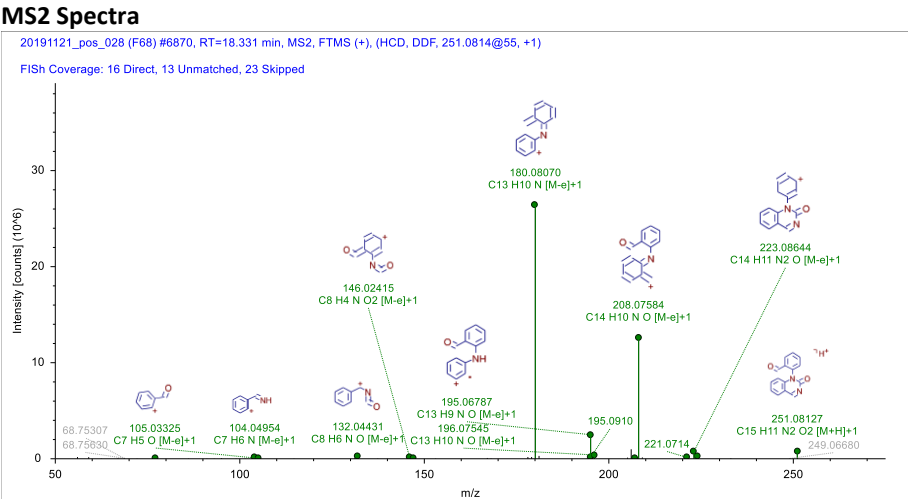
Atomic modification
-H₂ + O

Proposed Structure



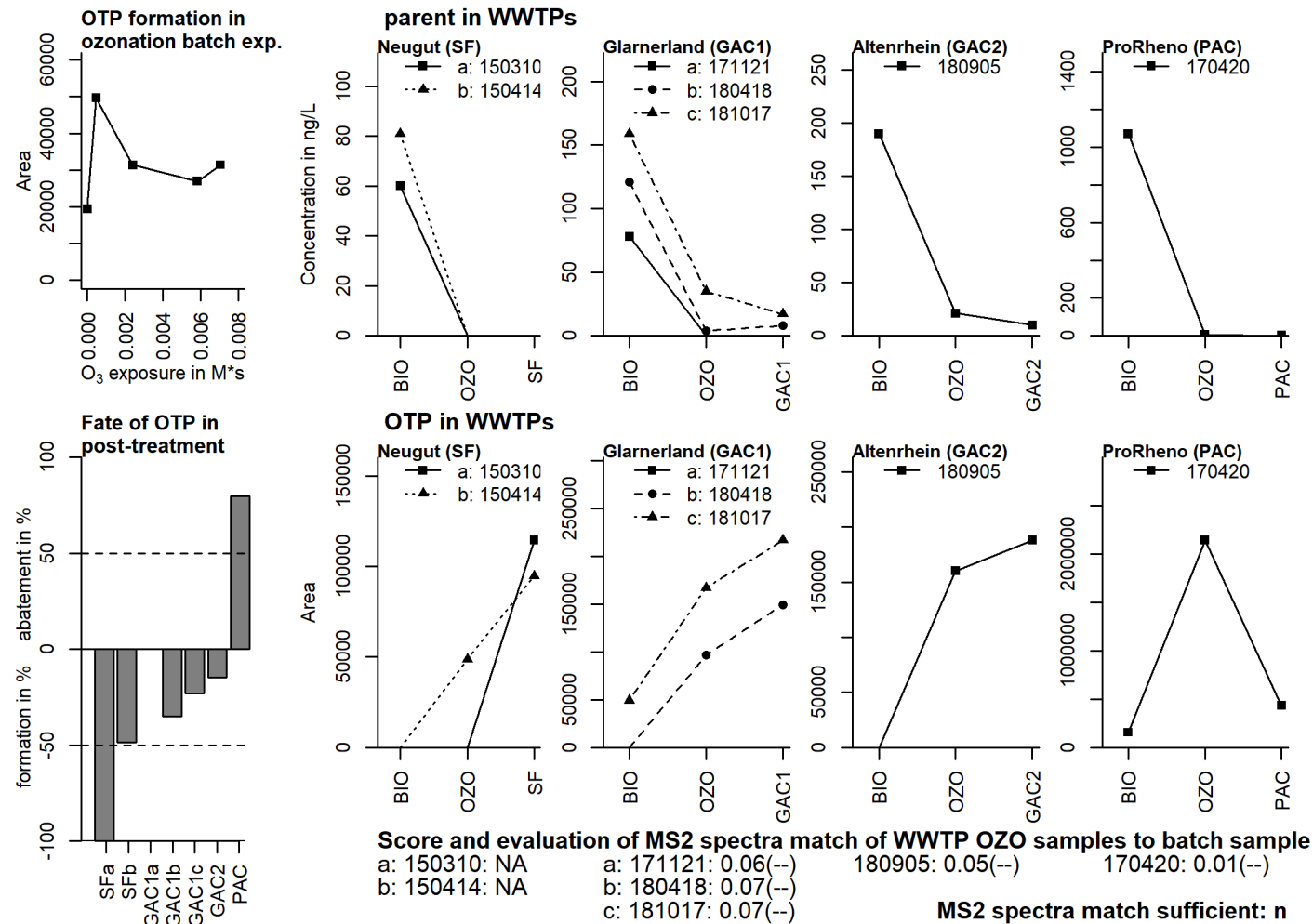
Confidence Level
Level 2a

Massbank ID
ET400901



Additional Evidence for Structure Interpretation

McDowell et al. (2005) identified 1-(2-benzaldehyde)-4-hydro-(1H,3H)-quinazoline-2-one (BQM) as an OTP for carbamazepine. The observed evidence from our MS and MS² spectrum match the one from McDowell et al. (2005). This OTP was later identified in two further studies (Hübner et al. 2014, Azais et al. 2017).



MS Spectra

Pos 267.0762 [m+H]⁺
Neg 265.0617 [m-H]⁻

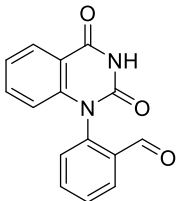
Formula

C₁₅H₁₀O₃N₂

Atomic modification

-H₂ +O₂

Proposed Structure



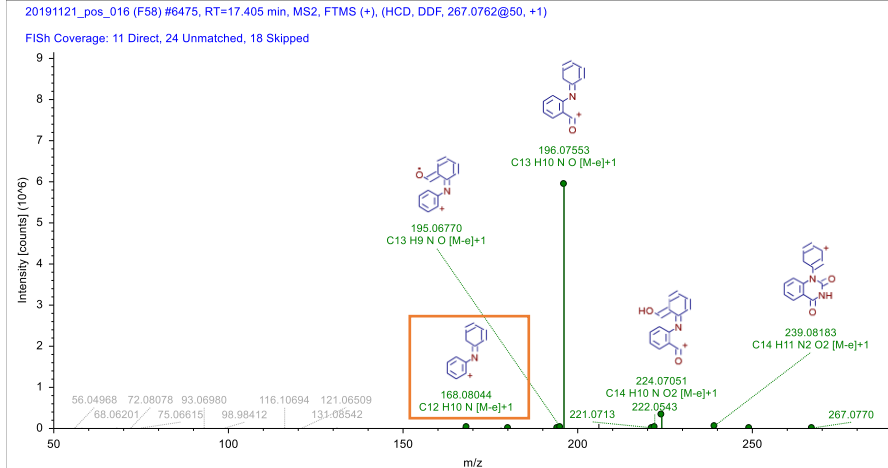
Confidence Level

Level 2a

Massbank ID

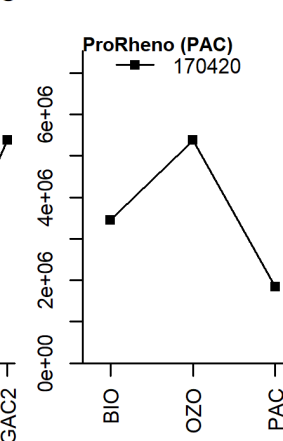
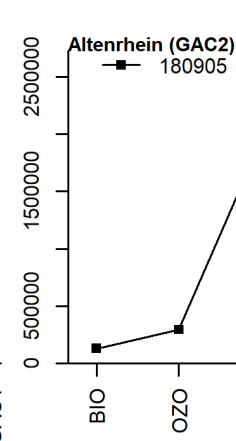
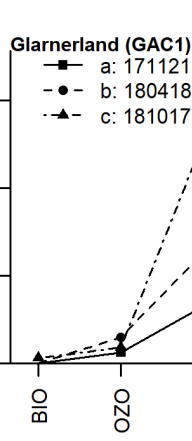
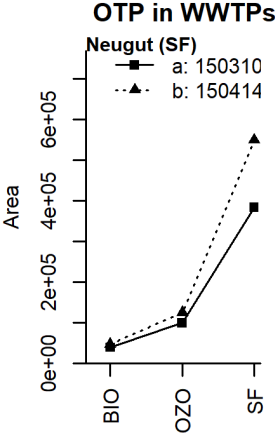
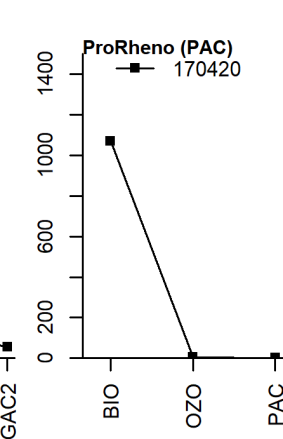
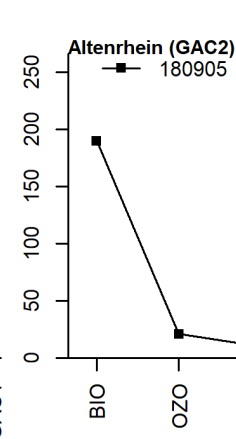
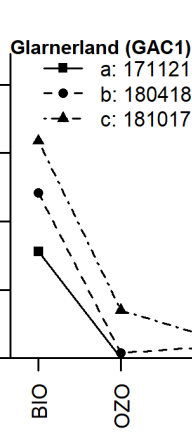
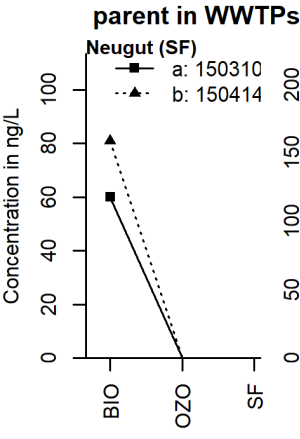
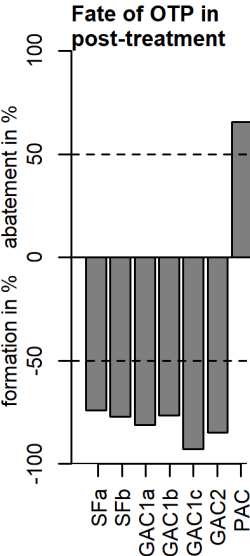
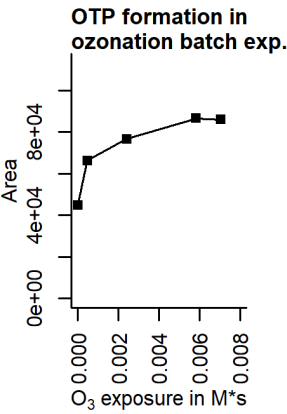
ET401001

MS2 Spectra



Additional Evidence for Structure Interpretation

The MS² fragment at the nominal mass 168 was also observed for the parent compound. It indicates that the structure drawn for fragment 168 is a substructure of this TP. McDowell et al. (2005) identified 1-(2-benzaldehyde)-(1H,3H)-quinazoline-2,4-dione (BQD) as an OTP for carbamazepine. The evidence from our MS and MS² spectrum match the one from McDowell et al. (2005). This OTP was later identified in two further studies (Hübner et al. 2014, Azais et al. 2017).



Score and evaluation of MS2 spectra match of WWTP OZO samples to batch sample

a: 150310: 0.78(+++) a: 171121: 0.86(+++) 180905: 0.84(+++) 170420: 1(+++)

b: 150414: 0.78(+++) b: 180418: 0.64(+++) c: 181017: 0.65(+++)

MS2 spectra match sufficient: y

MS Spectra

Pos 283.0711 [m+H]⁺
Neg 281.0565 [m-H]⁻

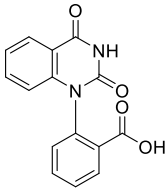
Formula

C₁₅H₁₀O₄N₂

Atomic modification

-H₂ +O₃

Proposed Structure



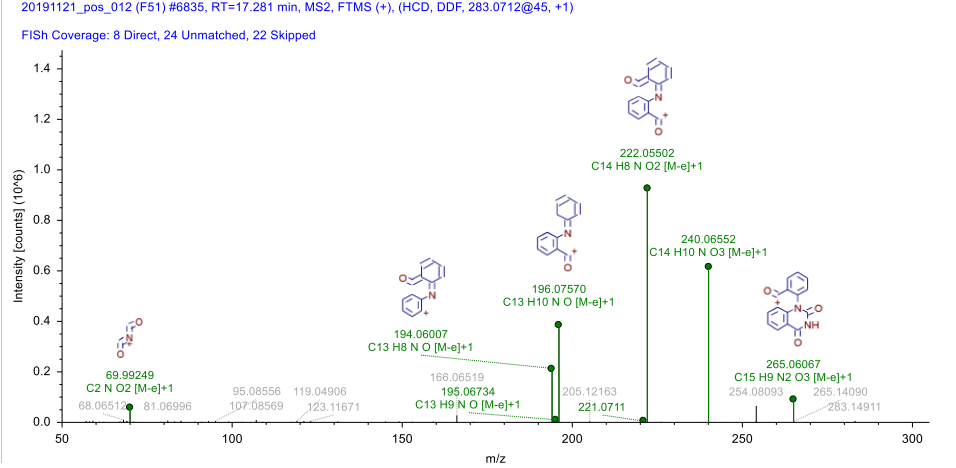
Confidence Level

Level 2a

Massbank ID

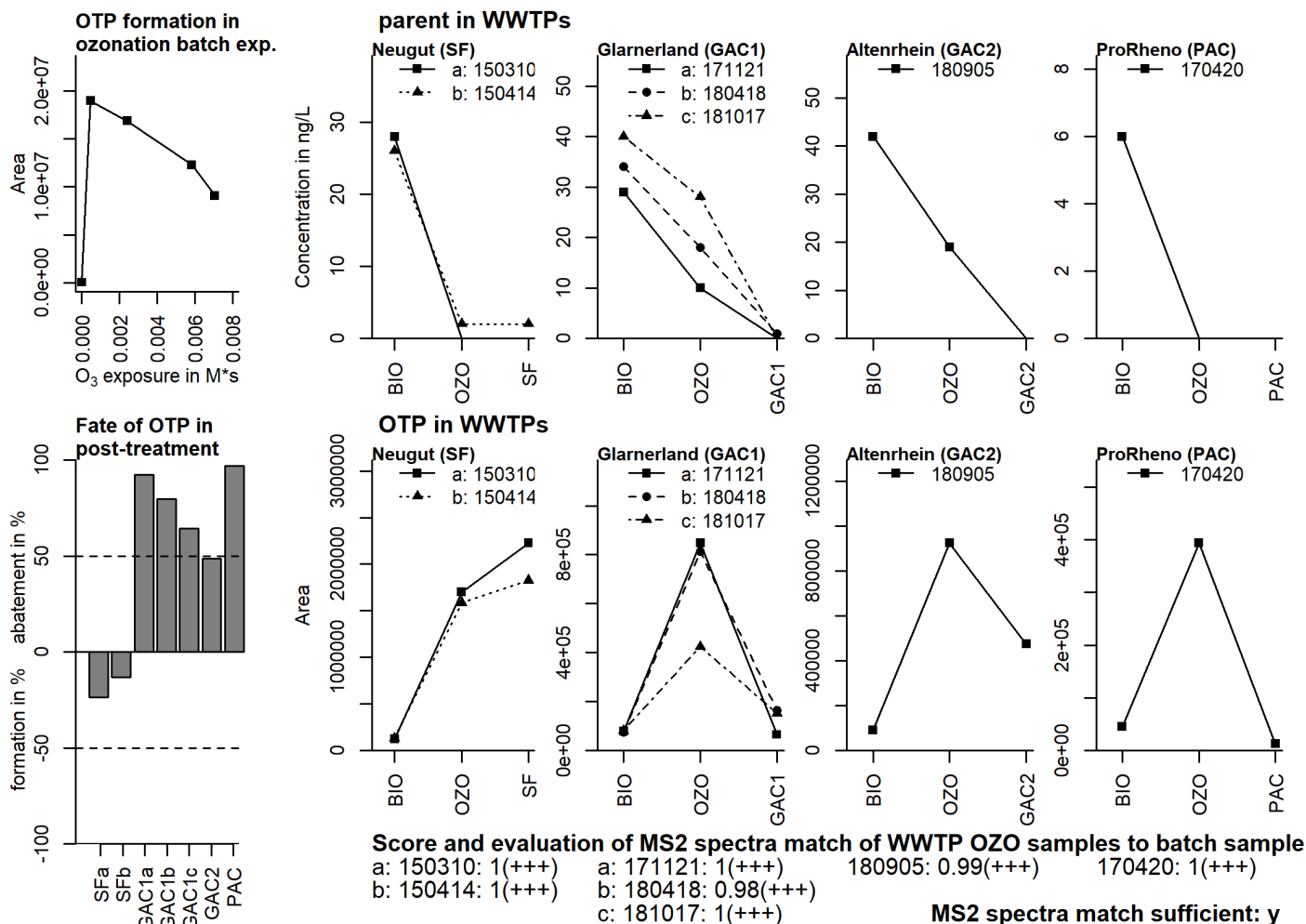
ET401101

MS2 Spectra



Additional Evidence for Structure Interpretation

1-(2-benzoic acid)-(1H,3H)-quinazoline-2,4-dione (BaQD) was identified by McDowell et al. (2005) as an oxidation product of carbamazepine after ozonation. The observed evidence from our MS and MS² spectrum match the one from by McDowell et al. (2005).



MS Spectra

Pos 341.1661 [m+H]⁺

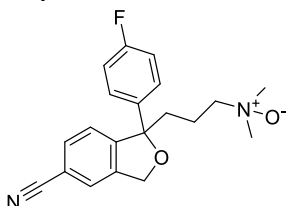
Formula

C₂₀H₂₁O₂N₂F

Atomic modification

+O

Proposed Structure



Confidence Level

Level 1

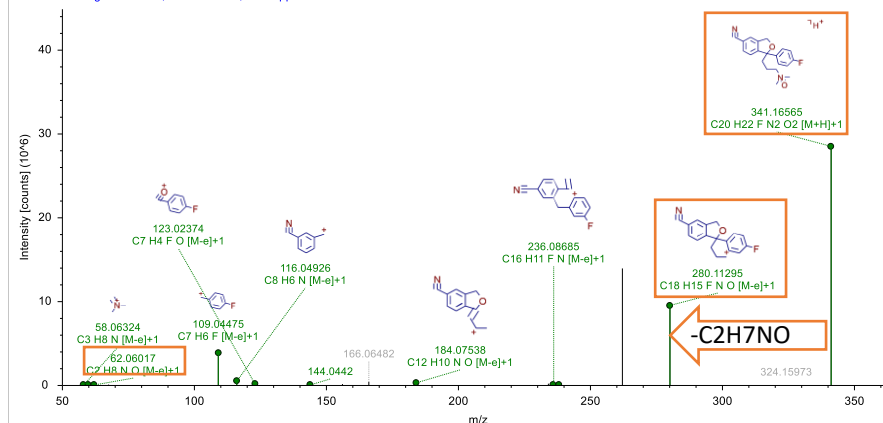
Massbank ID

ET401201

MS2 Spectra

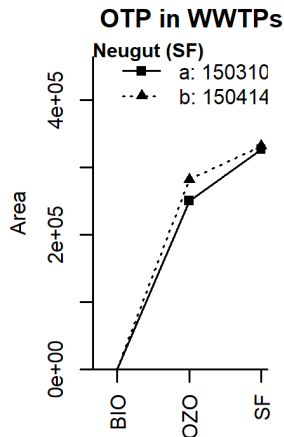
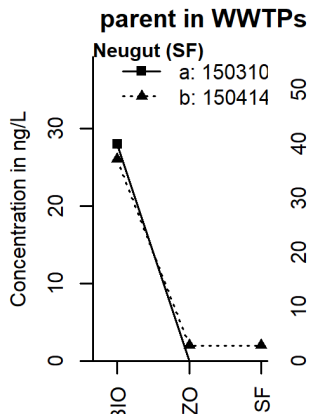
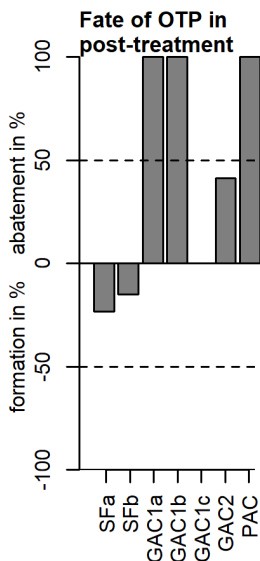
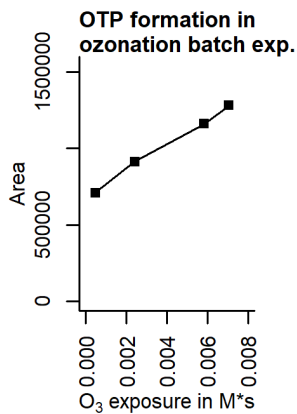
20191121_pos_017 (F59) #6384, RT=16.973 min, MS₂, FTMS (+), (HCD, DDF, 341.1660@20, +1)

FISH Coverage: 14 Direct, 8 Unmatched, 14 Skipped



Additional Evidence for Structure Interpretation

The MS² fragments at the nominal masses 280, 236, 184, 144, 123, 116, 109 and 58 were also observed for the parent compound. This indicates that the structure drawn for the fragment 280 is a substructure of this OTP. The atomic modification from the parent compound to this OTP is +O. This would fit to an N-oxidation or a hydroxylation. Tertiary amines are very reactive to form N-oxides during ozonation (von Sonntag & von Gunten 2012). Therefore, the formation of a N-oxide at the dimethylpropylamine is very likely. Hörsing et al. (2012) identified Citalopram N-oxide as major OTP. The neutral loss of C₂H₇NO between the precursor and fragment 280 as well as the fragment 62 indicate that the modification of +O took place at the dimethylamine moiety.



Score and evaluation of MS2 spectra match of WWTP OZO samples to batch sample

Sample	Score	Evaluation
a: 150310	1(+++)	
b: 150414	1(+++)	
a: 171121	0.83(+++)	
b: 180418	0.77(+++)	
c: 181017	0.71(+++)	
180905	0.86(+++)	
170420	0.98(+++)	

MS2 spectra match sufficient: y

MS Spectra

Pos 355.1451
Neg 399.1359

[m+H]⁺
[m+FA-H]⁻

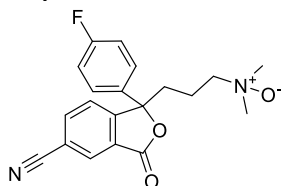
Formula

C₂₀H₁₉O₃N₂F

Atomic modification

-H₂ +O₂

Proposed Structure



Confidence Level

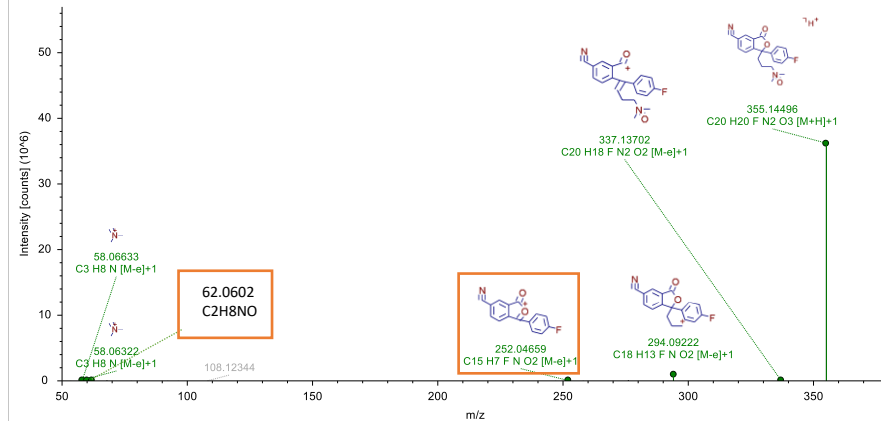
Level 3

Massbank ID

ET401301

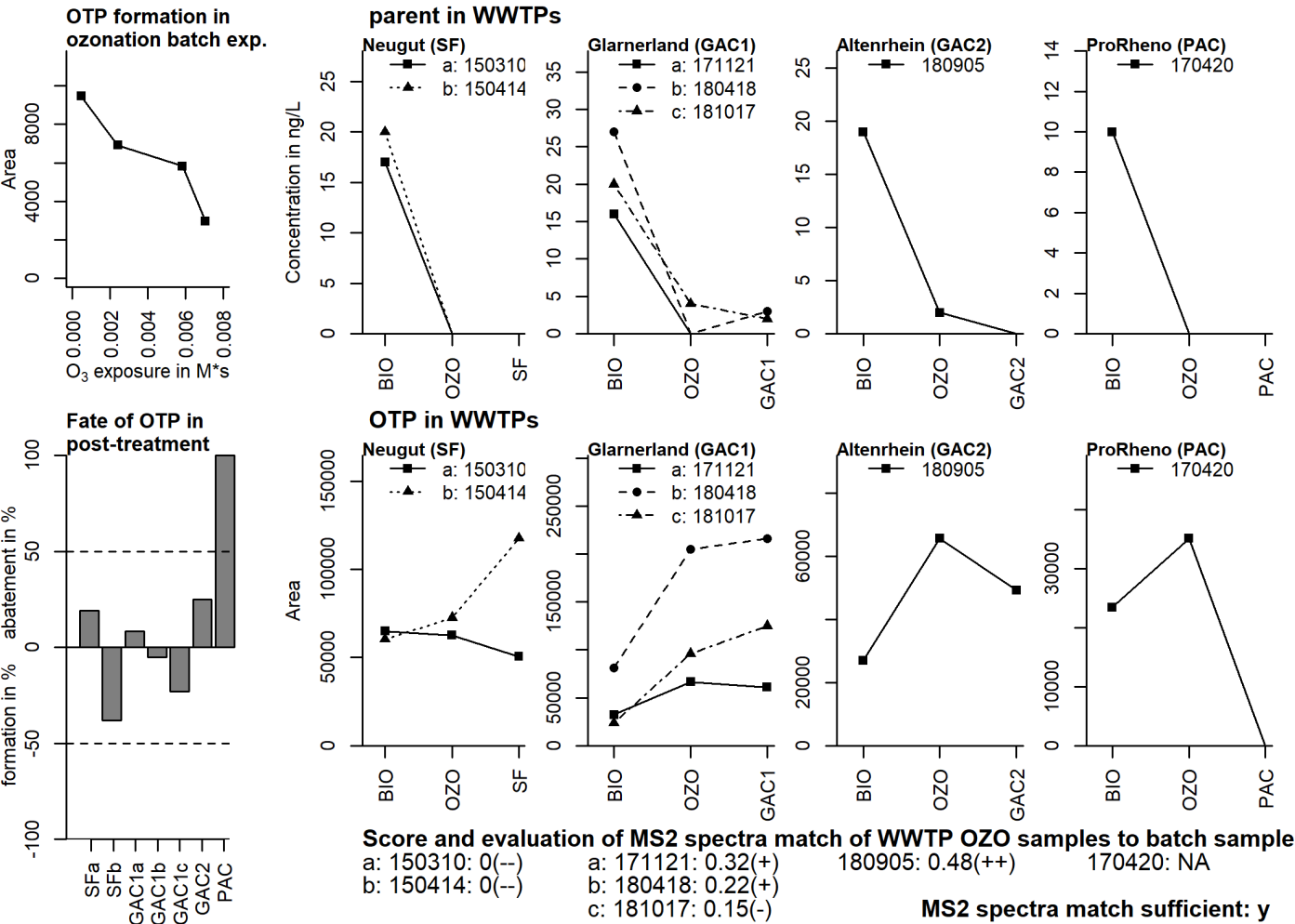
MS2 Spectra

20191121_pos_006 (F49) #6124, RT=16.497 min, MS2, FTMS (+), (HCD, DDF, 355.1452@15, +1)
FISH Coverage: 8 Direct, 3 Unmatched, 7 Skipped



Additional Evidence for Structure Interpretation

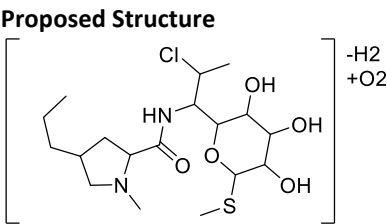
The atomic modification from the elemental formula of the parent compound to this TP is -H₂ +O₂. A fragment at the exact mass of 62.0602 corresponding to an elemental formula of C₂H₈NO was observed in this MS² spectrum as well as in the MS² spectrum of CIT_p_341.1661_16.2, which was identified as citalopram N-oxide. This indicates that an N-oxide moiety is also likely for this TP. Fragment 252 with a formula of C₁₅H₇FN O₂ indicates that a second oxygen atom was added at the citalopram carbon backbone without the propyldimethylamine moiety. The modification of -H₂ +O fits to the formation of a carbonyl group. Hörsing et al. (2012) suggested that the methylene group of the benzofuran moiety is a likely location for the formation of a carbonyl group. However, this mechanism is not entirely understood.



MS Spectra
Neg 453.1466 [m-H]-

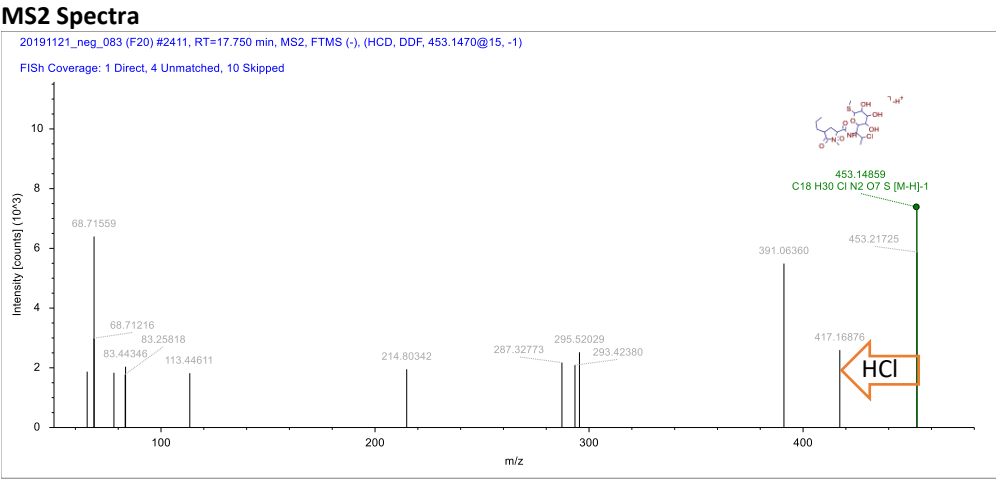
Formula
C18H31O7N2ClS

Atomic modification
-H2 +O2



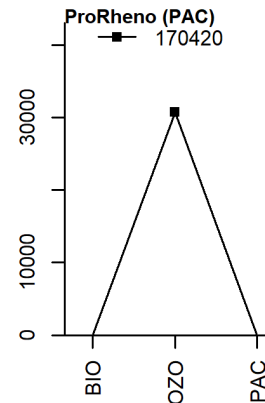
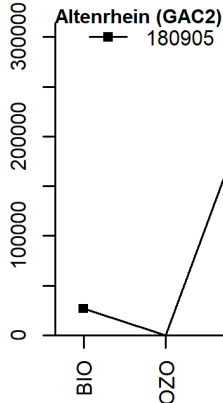
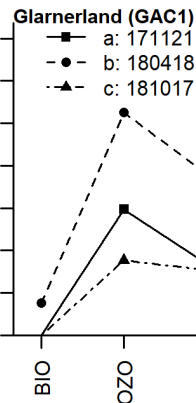
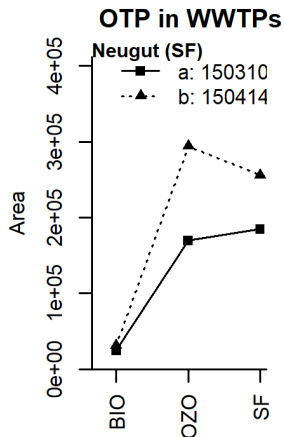
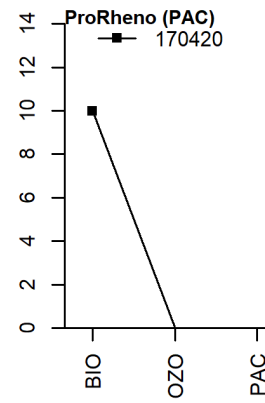
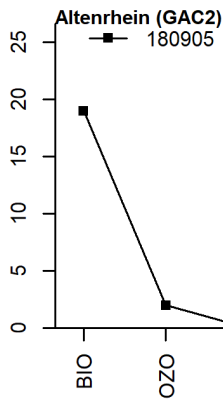
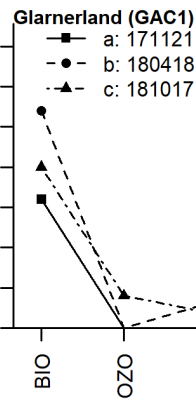
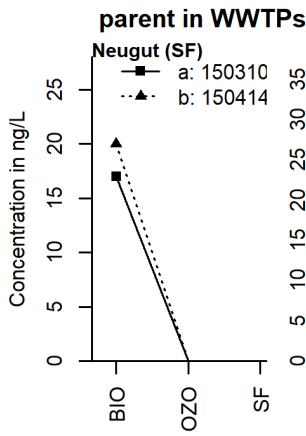
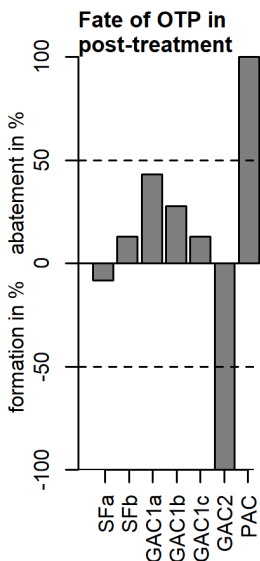
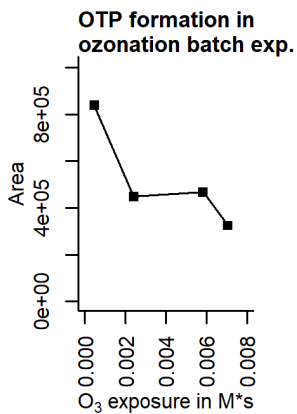
Confidence Level
Level 3

Massbank ID
ET405201



Additional Evidence for Structure Interpretation

The atomic modification from the elemental formula of the parent compound to this TP is -H2 +O2. This modification fits to multiple known reactions with ozone such as the formation of an N-oxide (+O), a sulfoxide (+O) or the addition of a hydroxyl group after attack of a hydroxyl radical on a CH2 group (+O) (von Sonntag & von Gunten 2012). A possibility would be one of the mentioned reactions and the formation of a carbonyl moiety (-H2 +O) at a CH2 group of the mother substance. There is however no further evidence for the exact type and location of the modifications. The structures of the MS² fragments are drawn exemplarily.



Score and evaluation of MS2 spectra match of WWTP OZO samples to batch sample

a: 150310: 1(+++) a: 171121: 1(+++) 180905: 0.96(+++) 170420: 0.64(+++)

b: 150414: 1(+++) b: 180418: 1(+++) c: 181017: 0.91(+++)

MS2 spectra match sufficient: y

MS Spectra

Pos 457.1769 [m+H]⁺
Neg 455.1623 [m-H]⁻

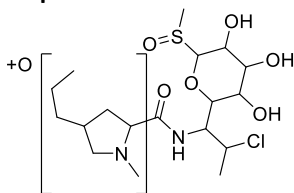
Formula

C₁₈H₃₃O₇N₂ClS

Atomic modification

+O₂

Proposed Structure



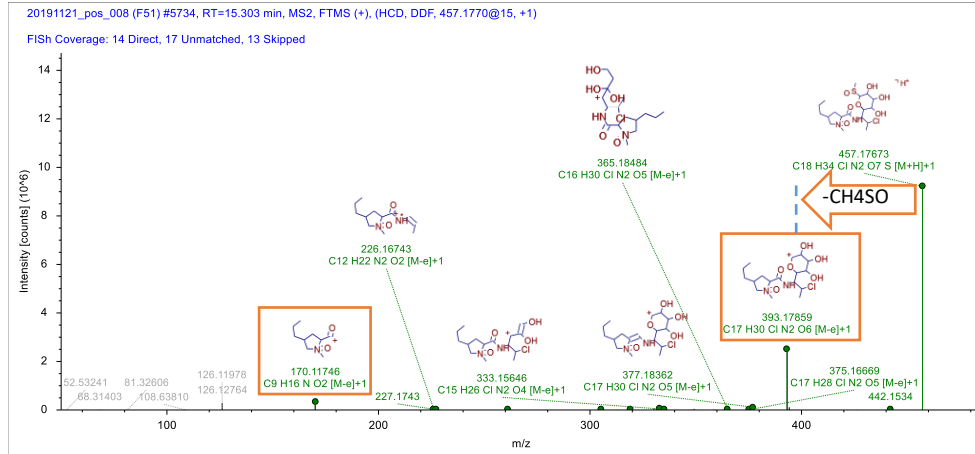
Confidence Level

Level 3

Massbank ID

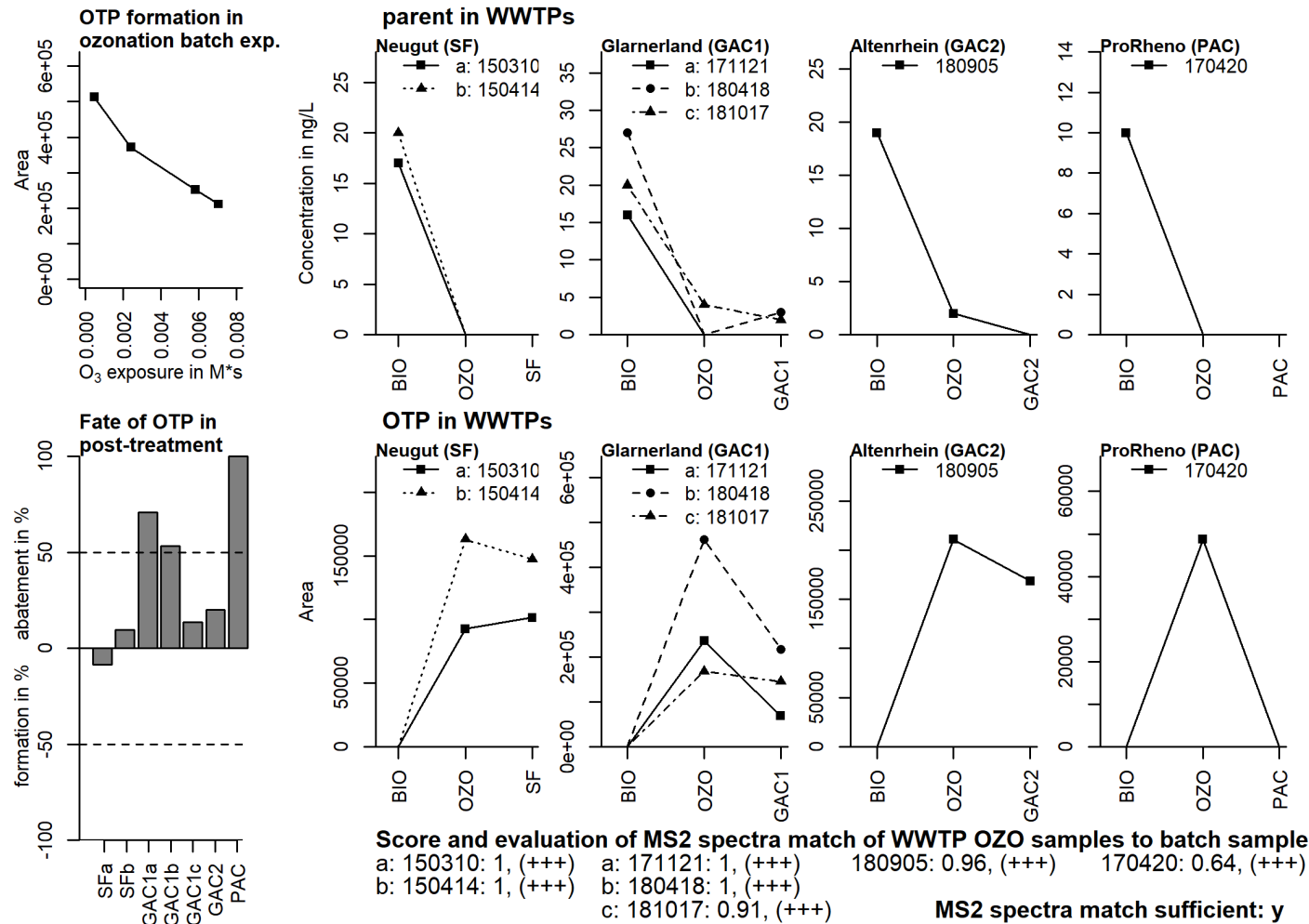
ET401401

MS2 Spectra



Additional Evidence for Structure Interpretation

For the parent compound clindamycin two OTPs (CLI_p_457.1769_14.7 and CLI_p_457.1769_15.1) with the exact mass of 457.1769 were observed at different retention times (14.7 and 15.1 min). The MS² spectra of these OTPs are very similar. The atomic modification from the elemental formula of the parent compound to these OTPs is +O₂. A modification of +O can be realized by the formation of sulfoxides from thioethers, by the formation of N-oxides from tertiary amines, or by hydroxylation. All three reactions are known to occur during ozonation (von Sonntag & von Gunten 2012). The neutral loss of CH₄SO between the precursor and fragment 393 fits to the cleavage of a methylsulfoxide moiety and indicates the formation of this substructure. Fragment 170 indicates that the second oxygen was added at the methylpropylpyrrolidine moiety. One OTP could for example have been formed by N-oxidation and the other by hydroxylation.

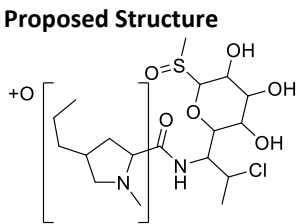


MS Spectra

Pos 457.1769 [m+H]⁺
Neg 455.1623 [m-H]⁻

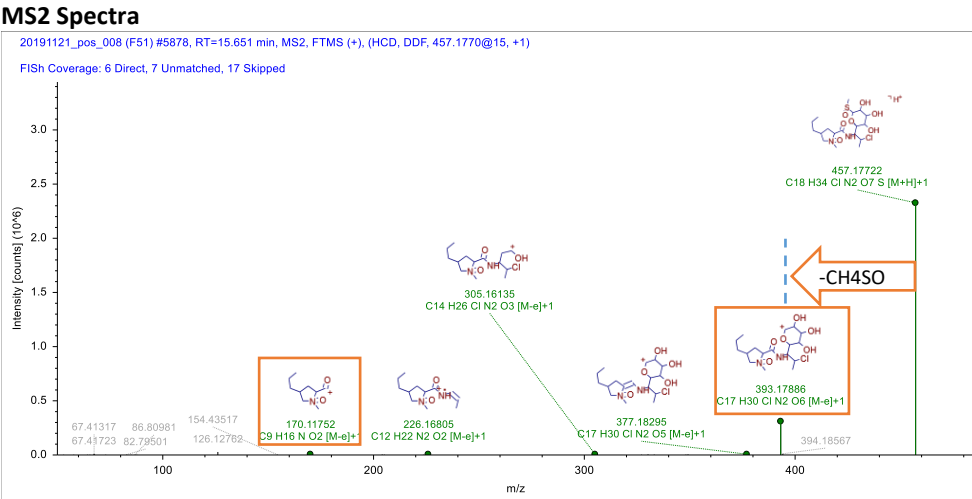
Formula
C18H33O7N2ClS

Atomic modification
+O2



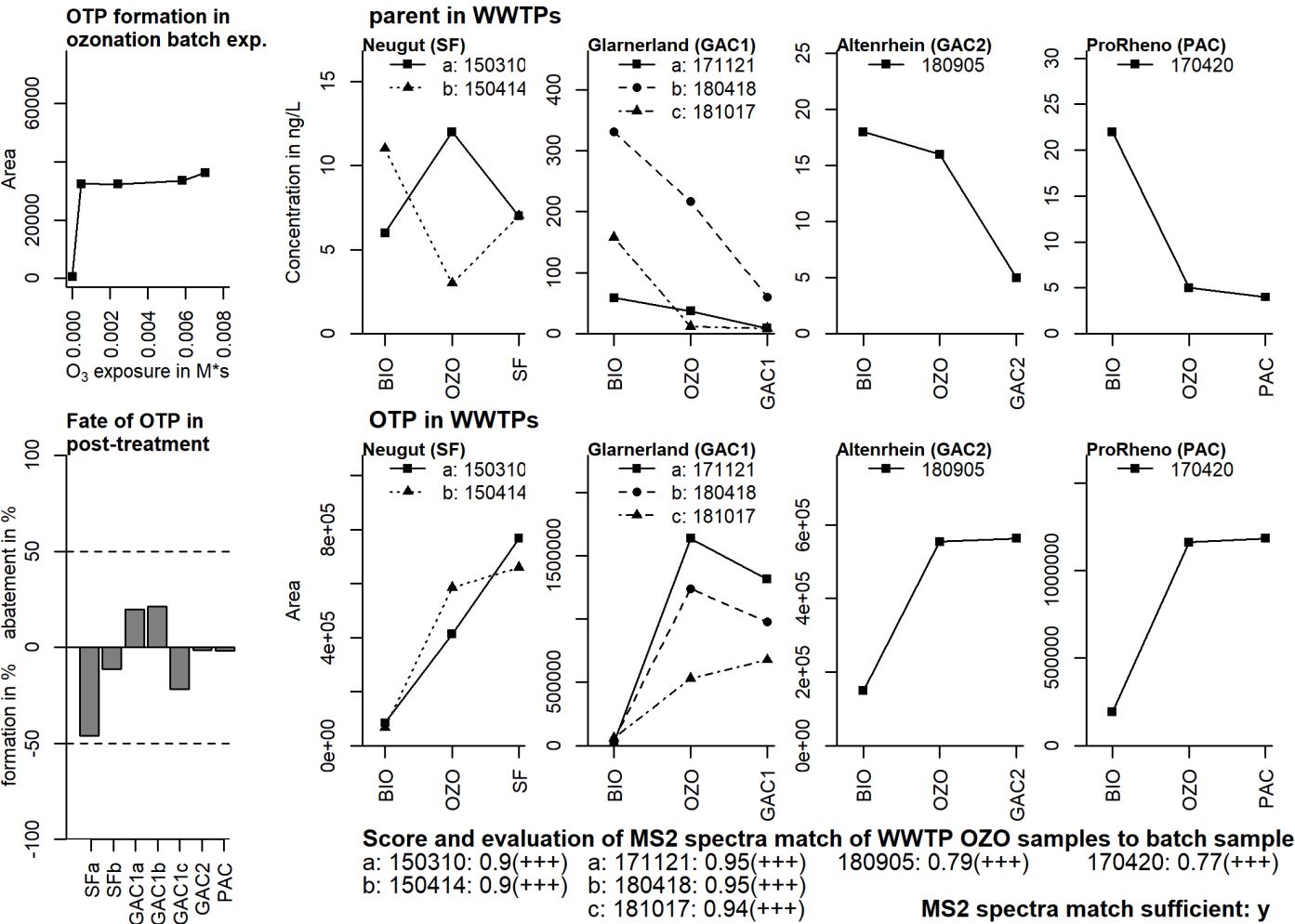
Confidence Level
Level 3

Massbank ID
ET401501



Additional Evidence for Structure Interpretation

For the parent compound clindamycin two OTPs (CLI_p_457.1769_14.7 and CLI_p_457.1769_15.1) with the exact mass of 457.1769 were observed at different retention times (14.7 and 15.1 min). The MS² spectra of these OTPs are very similar. The atomic modification from the elemental formula of the parent compound to these OTPs is +O₂. A modification of +O can be realized by the formation of sulfoxides from thioethers, by the formation of N-oxides from tertiary amines, or by hydroxylation. All three reactions are known to occur during ozonation (von Sonntag & von Gunten 2012). The neutral loss of CH₄SO between the precursor and fragment 393 fits to the cleavage of a methylsulfoxide moiety and indicates the formation of this substructure. Fragment 170 indicates that the second oxygen was added at the methylpropylpirolidine moiety. One OTP could for example have been formed by N-oxidation and the other by hydroxylation.

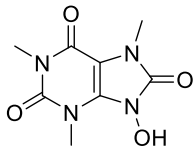


MS Spectra
Neg 225.0627 [m-H]-

Formula
C₈H₁₀O₄N₄

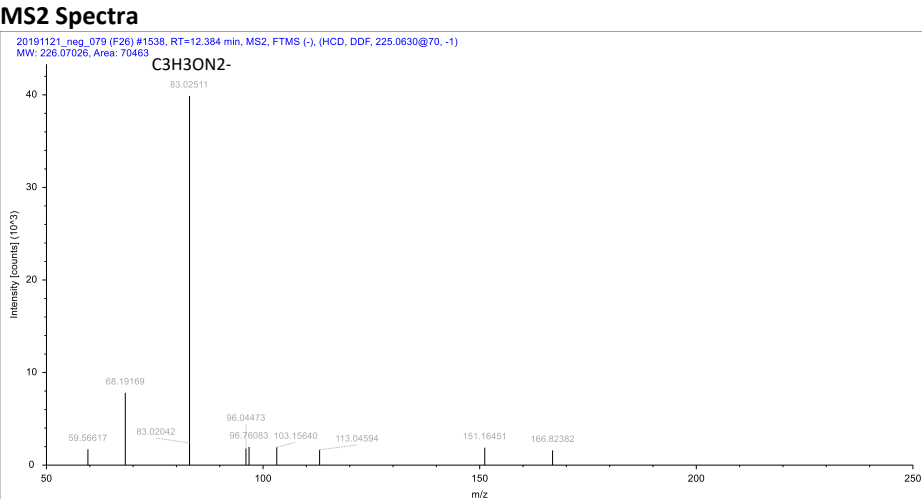
Atomic modification
+O₂

Proposed Structure



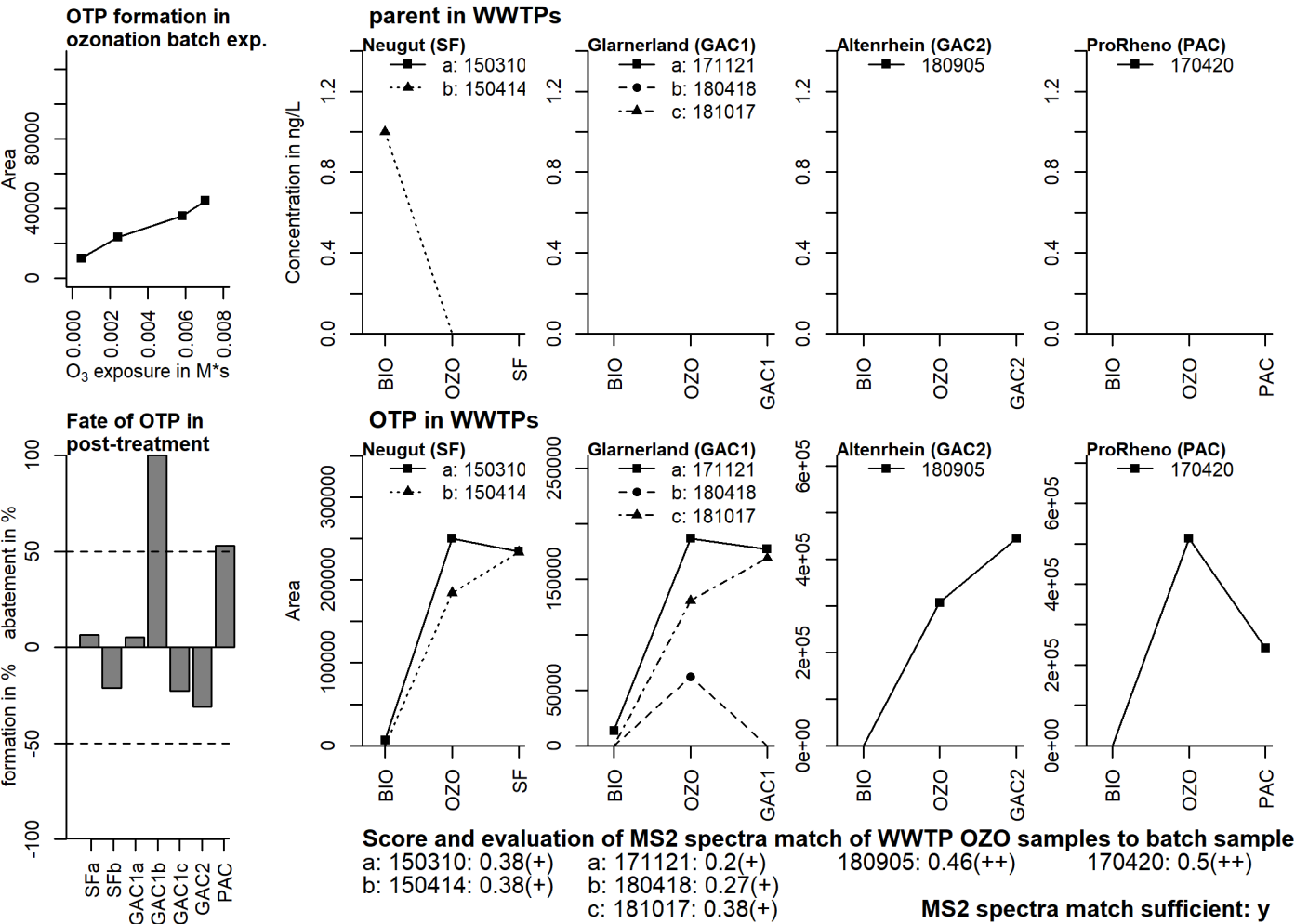
Confidence Level
Level 3

Massbank ID
ET405301



Additional Evidence for Structure Interpretation

The atomic modification from the elemental formula of the parent compound to this TP is +O₂. This modification fits to multiple reactions with ozone and/or hydroxyl radicals. Rosal et al. (2009) detected an OTP of caffeine with this mass and suggested the formation of a carbonyl group at the C-2 and the addition of an hydroxyl group at the N-2 of the methylimidazole moiety. There is however no further evidence for the exact type and location of the modifications.

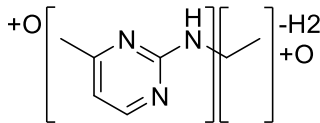


MS Spectra
Pos 168.0766 [m+H]⁺

Formula
C7H9O2N3

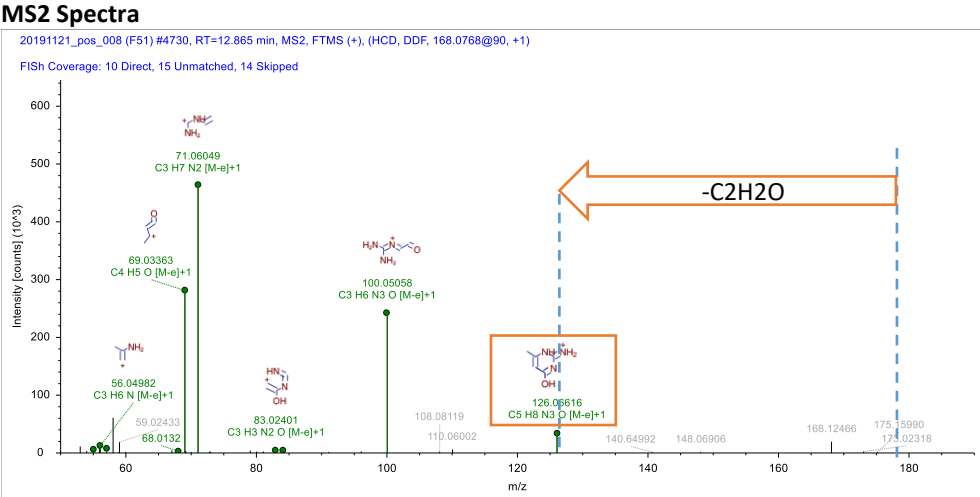
Atomic modification
-C7H6+O2

Proposed Structure



Confidence Level
Level 3

Massbank ID
ET401601



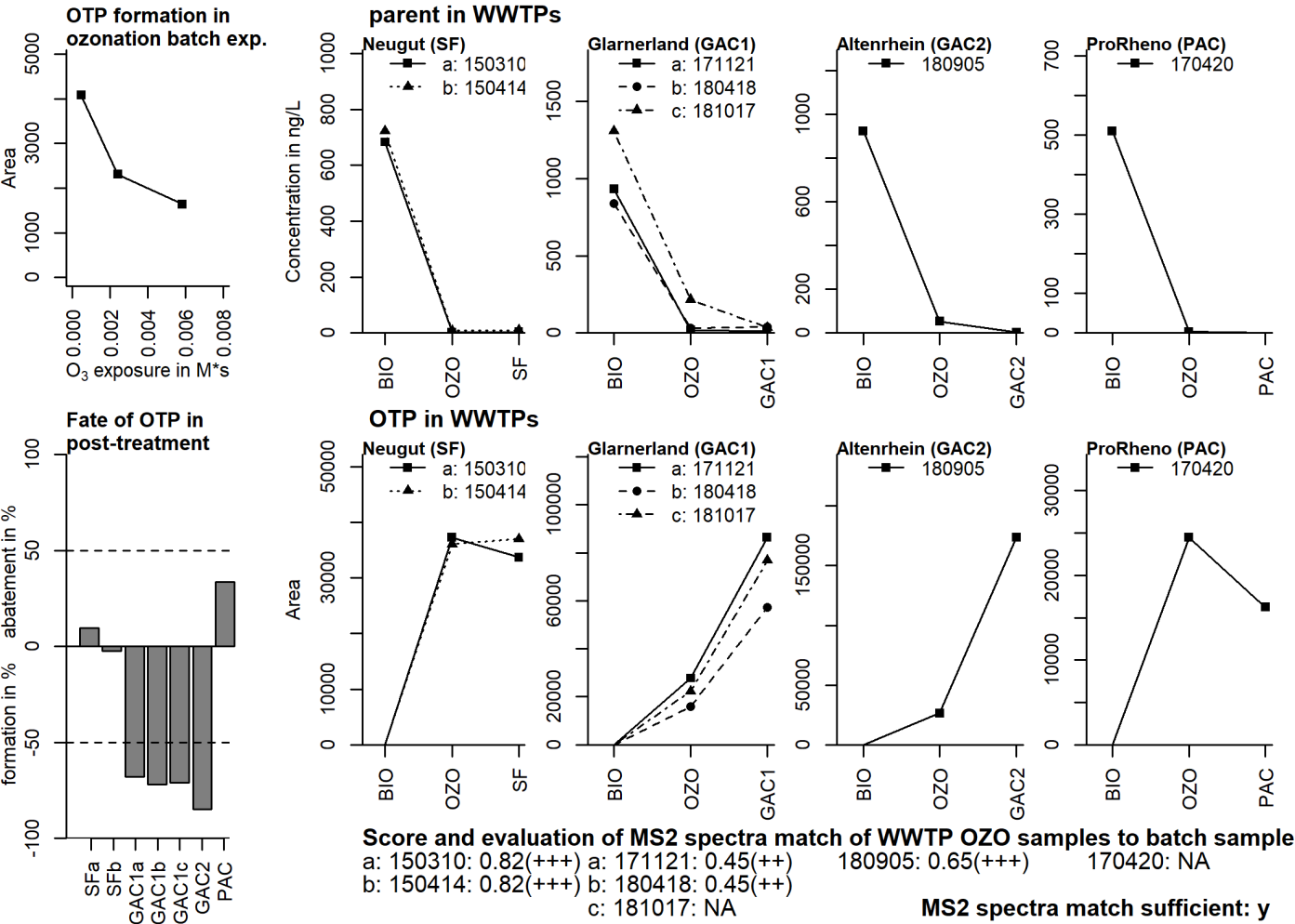
Additional Evidence for Structure Interpretation

The atomic modification from the elemental formula of the parent compound to this OTP is -C7H6+O2. This fits to a cleavage within the phenyl moiety and the cleavage of the cyclopropyl moiety.

The neutral loss of C2H2O between the precursor and fragment 126 indicates that an oxygen atom was added at a C2H2 moiety. This fits to a Criegee addition, which is known to occur within phenyl moieties (von Sonntag & von Gunten 2012), followed by a C-C bond cleavage.

Fragment 126 indicates that the second oxygen atom was added at the methylpyrimidin-amine like moiety of the TP.

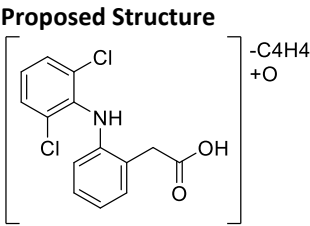
The exact type and position of the modifications remain unknown. The structures of the MS² fragments are drawn exemplarily.



MS Spectra
Neg 257.9729 [m-H]-

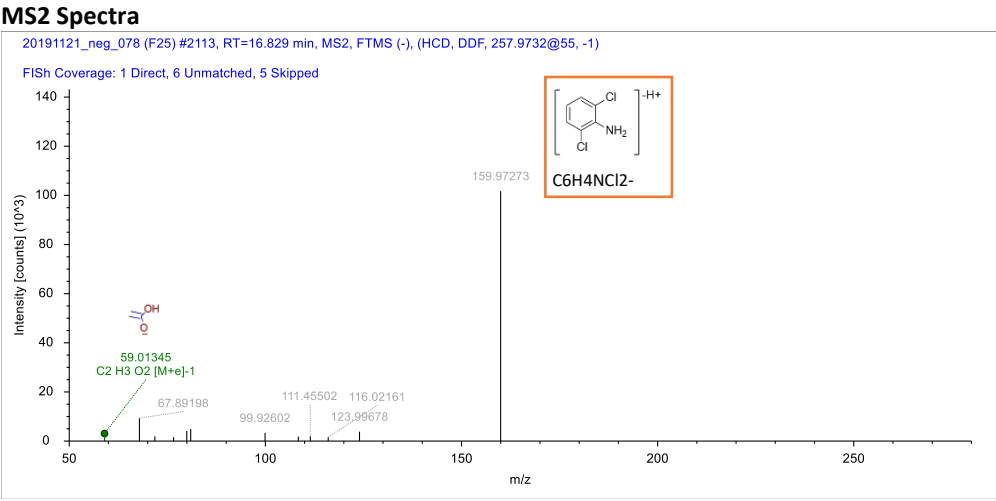
Formula
C10H7O3NCI2

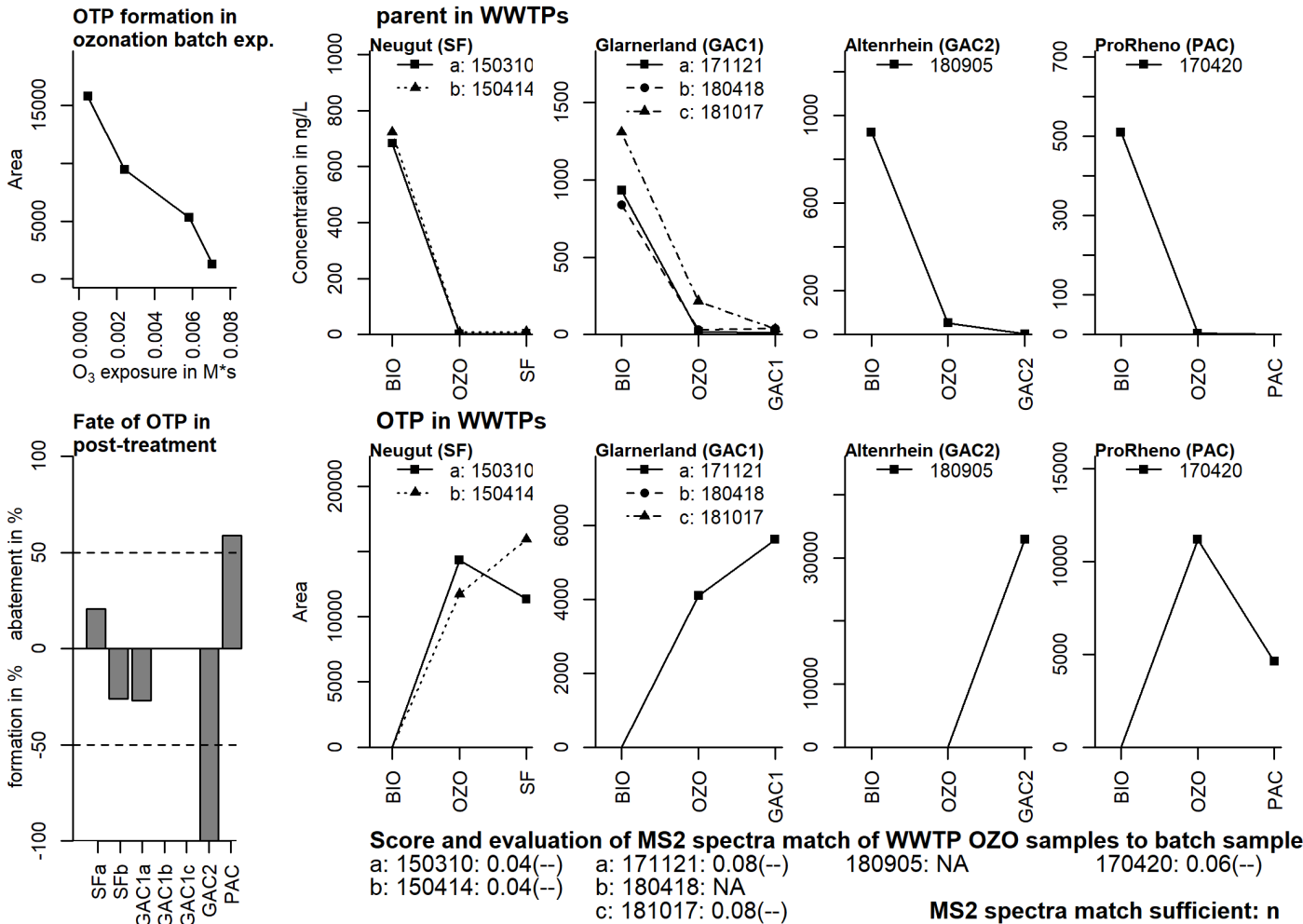
Atomic modification
-C4H4+O



Confidence Level
Level 3

Massbank ID
ET405401

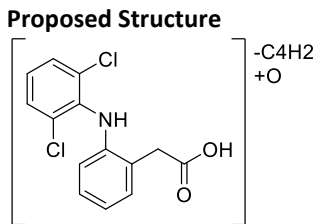




MS Spectra
Neg 259.9873 [m-H]-

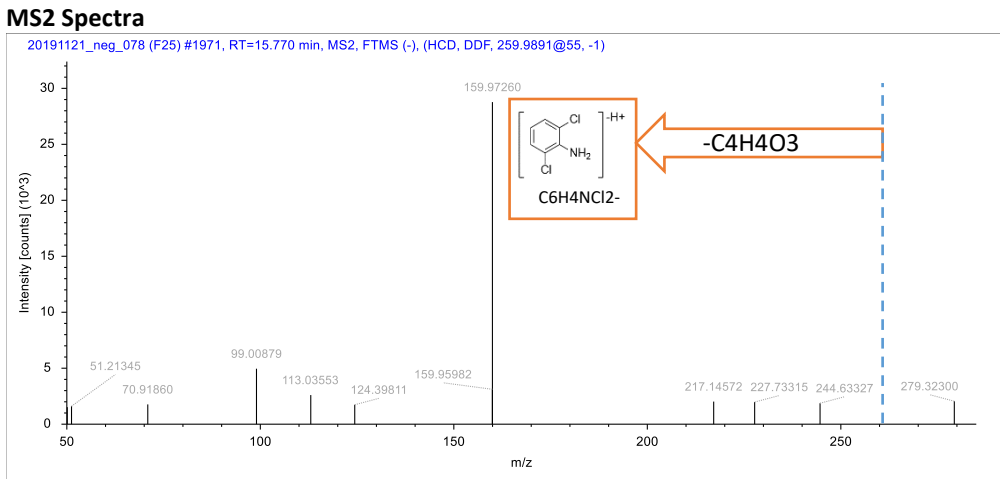
Formula
C10H9O3NCI2

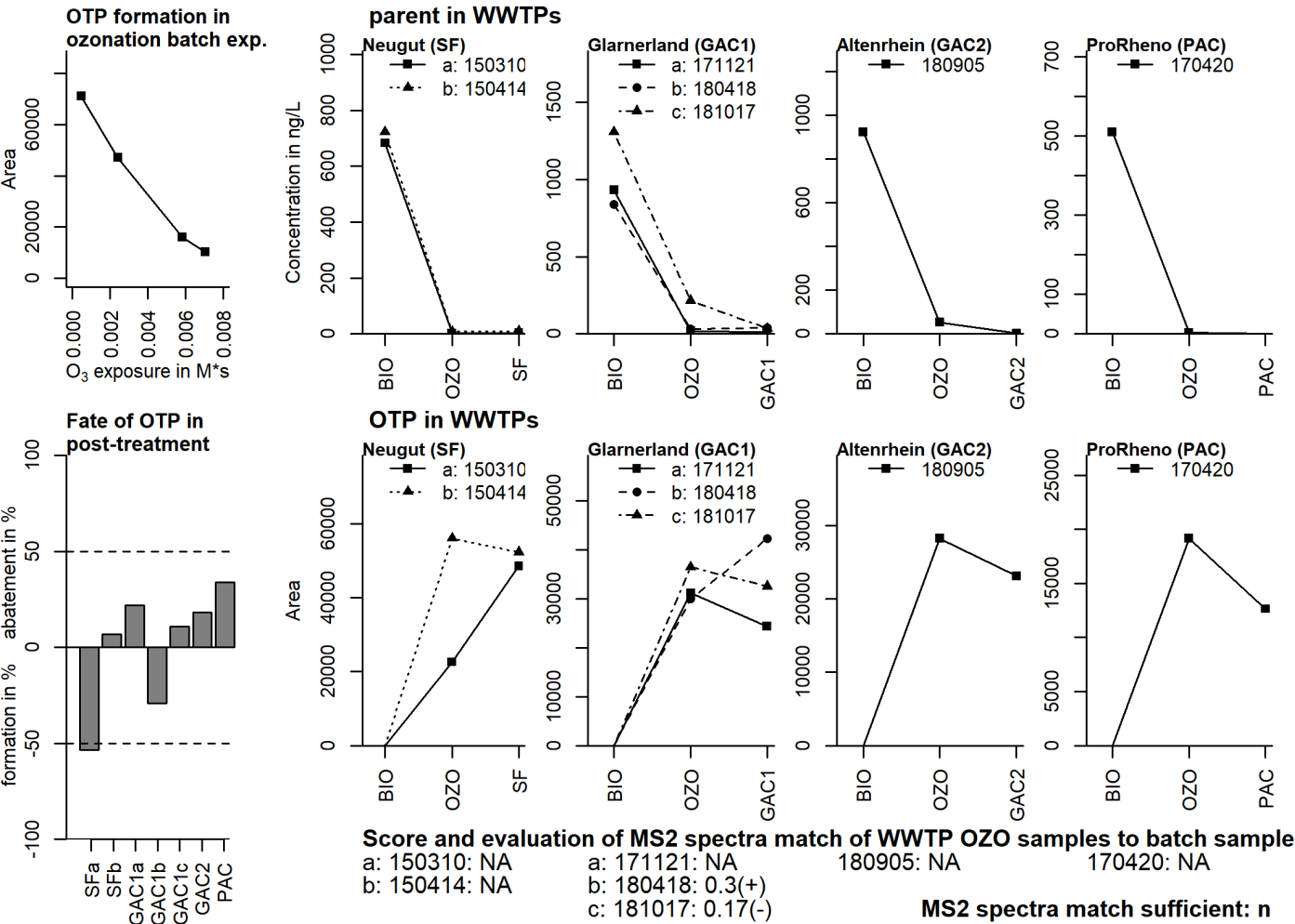
Atomic modification
-C4H2+O



Confidence Level
Level 3

Massbank ID
ET405501



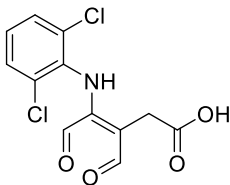


MS Spectra
Neg 299.9834 [m-H]-

Formula
C12H9O4NCI2

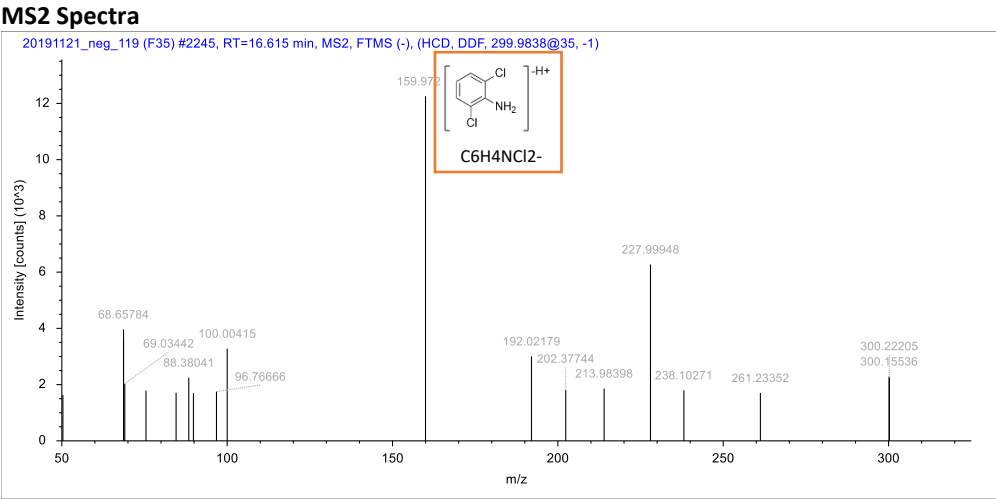
Atomic modification
-C2H2 +O2

Proposed Structure



Confidence Level
Level 3

Massbank ID
ET405601

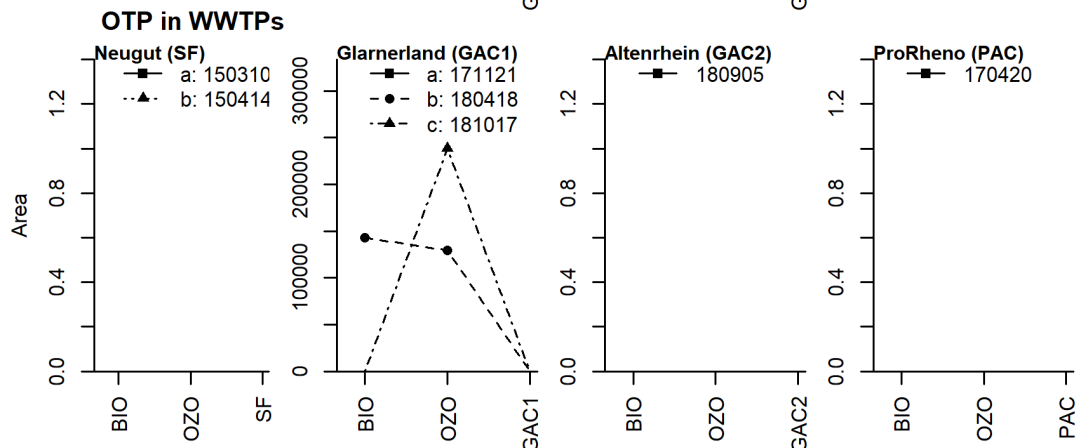
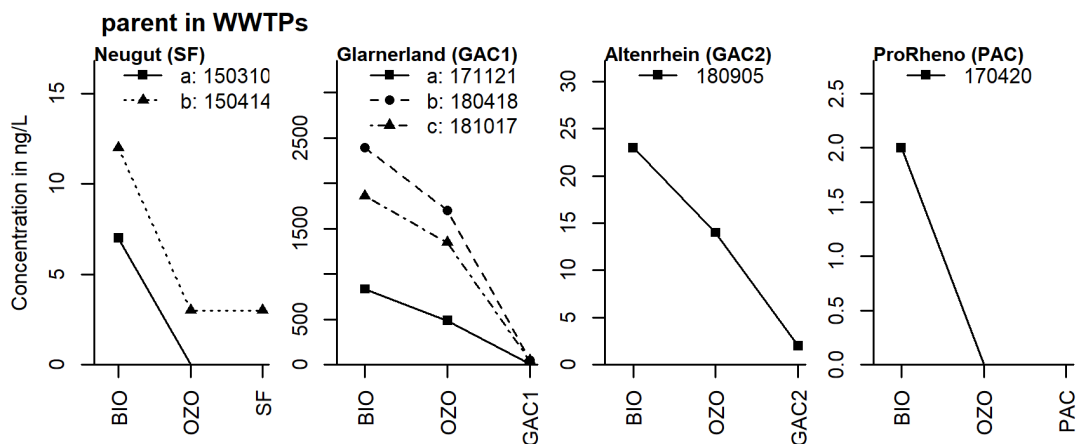
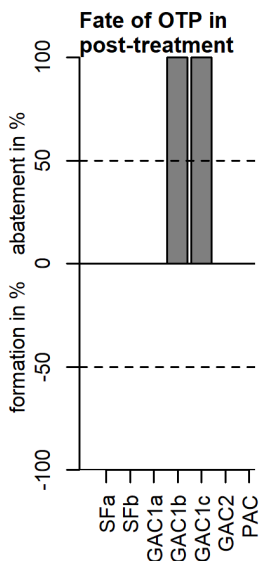


Additional Evidence for Structure Interpretation

Fragment 159 corresponds to the dichloroaniline moiety and indicates the modification didn't take place at this part of the molecule.

The atomic modification fits to a cleavage within the other aromatic moiety and the addition of 2 oxygen atoms. Ozone is known to react with phenyl moieties to form malealdehyde type moieties (Tay et al. 2013, Müller et al. 2012). Therefore, the formation of this TP is likely. There is however no further evidence.

The MS² fragments at the nominal masses 183, 167, 166, 165, 152, 151 were also observed for the parent compound. It indicates that the structure drawn for the fragment 183 is a substructure of this OTP. The neutral loss of C₄H₁₁NO as well as the fragment 88 with corresponding formula of C₄H₁₀NO indicate that the modification took place on the dimethylethyl moiety and not at the diphenyl moiety of the parent structure. Tertiary amines are very reactive to form *N*-oxides during ozonation (von Sonntag & von Gunten 2012). Therefore, the formation of a *N*-oxide at the dimethylethylamine is very likely. Merel et al. (2017) identified diphenhydramine *N*-oxide as major OTP.



Score and evaluation of MS2 spectra match of WWTP OZO samples to batch sample
a: 150310: NA a: 171121: 0.12(-) 180905: NA 170420: NA
b: 150414: NA b: 180418: 0.33(+) **MS2 spectra match sufficient: y**
c: 181017: NA

MS2 spectra match sufficient: y

Pos	215.0582	[m+H] ⁺
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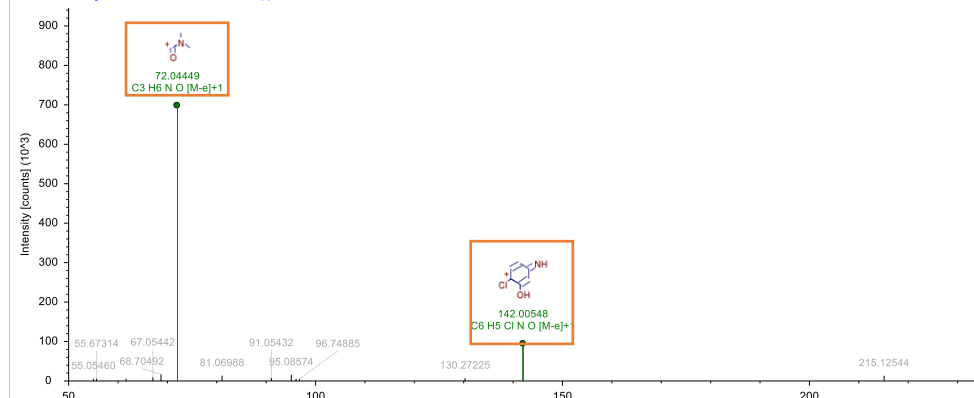
C9H11O2N2Cl
$$-\text{Cl} + \text{OH}$$
CN(C)C(=O)Nc1cc(ClO)c(ClO)cc1

Level 3

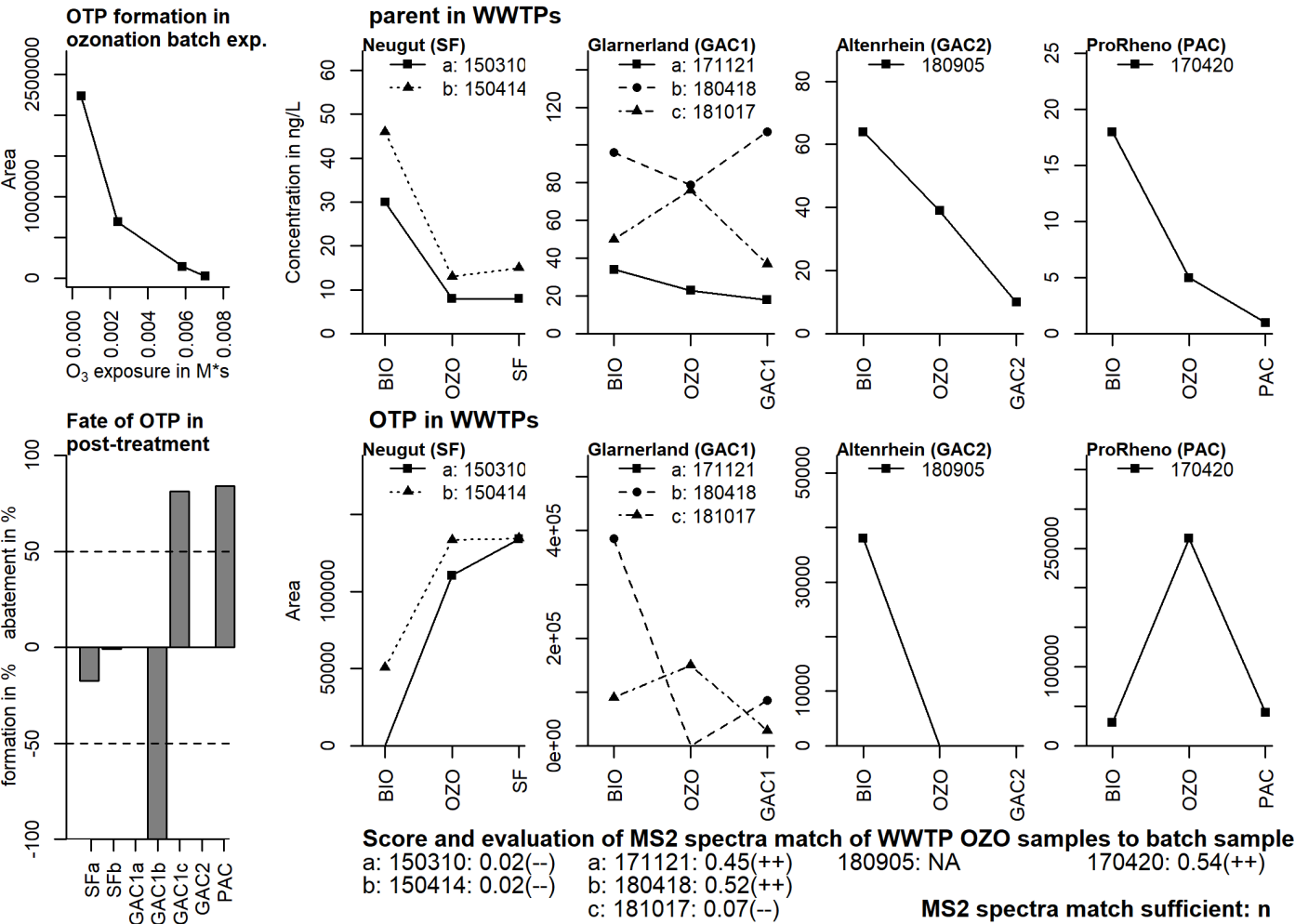
ET401801

20191121_pos_018 (F60) #5690, RT=15.205 min, MS2, FTMS (+), (HCD, DDF, 215.0582@70, +1)

FISh Coverage: 2 Direct, 5 Unmatched, 7 Skipped



The atomic modification from the elemental formula of the parent compound to this OTP is $-Cl + OH$. This fits to an abstraction of the chlorine atom and the addition of a hydroxy group. The MS^2 fragment at the nominal mass 72 was also observed for the parent compound. It indicates that the structure drawn for the fragment 72 is a substructure of this TP. The MS^2 fragment at the nominal mass 142 corresponds to the fragment 159 of the parent with an addition of an oxygen and hydrogen atom and abstraction of a chlorine. This indicates that the hydroxy group was added on the aromatic moiety. Feng et al. (2008) suggested a structure where a chlorine atom is replaced by a hydroxyl group as OTP of diuron (likely by hydroxyl reaction). It is likely that one chlorine was replaced by a hydroxyl group. However, it is unclear at which position. The structures of the MS^2 fragments are drawn exemplarily.

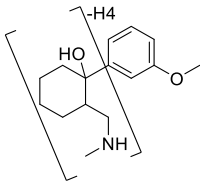


MS Spectra
Pos 246.1489 [m+H]⁺

Formula
C₁₅H₁₉NO₂

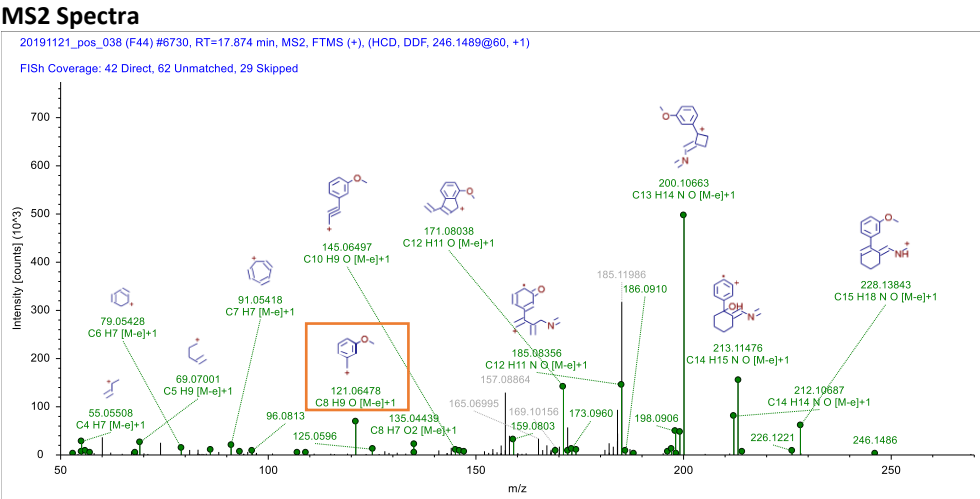
Atomic modification
-H₄

Proposed Structure



Confidence Level
Level 3

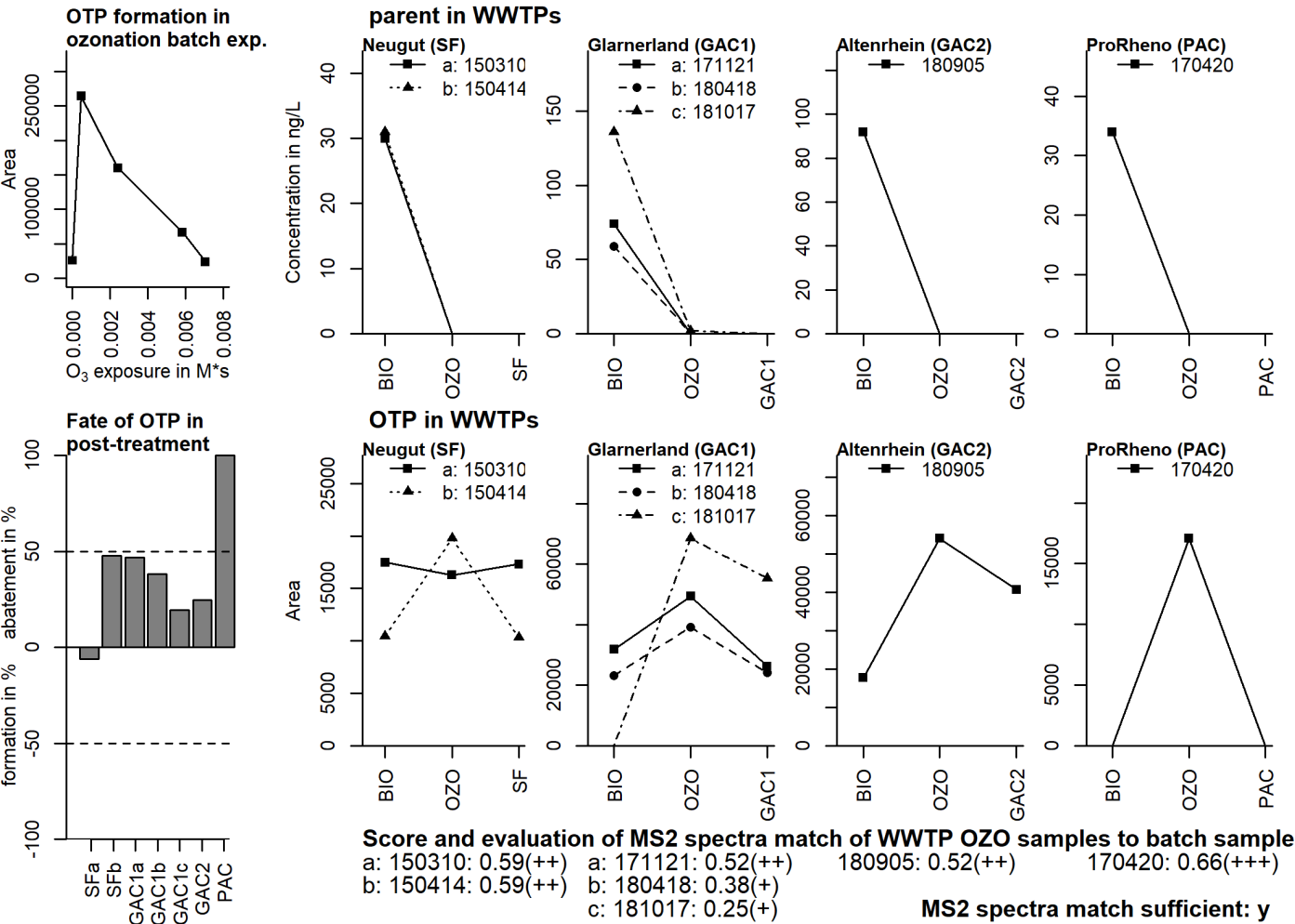
Massbank ID
ET401901



Additional Evidence for Structure Interpretation

Fragment 121 was also observed for the parent compound. It indicates that the methoxybenzyl moiety is a substructure of this TP.

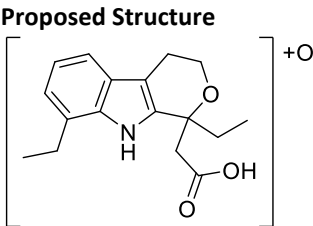
A possible abstraction of H₂ is the formation of an imine moiety, which then undergo hydrolysis. It is unclear where the two remaining hydrogen atoms were abstracted. The exact type and location remain unknown. The structures of the MS² fragments are drawn exemplarily.



MS Spectra
Neg 302.1395 [m-H]-

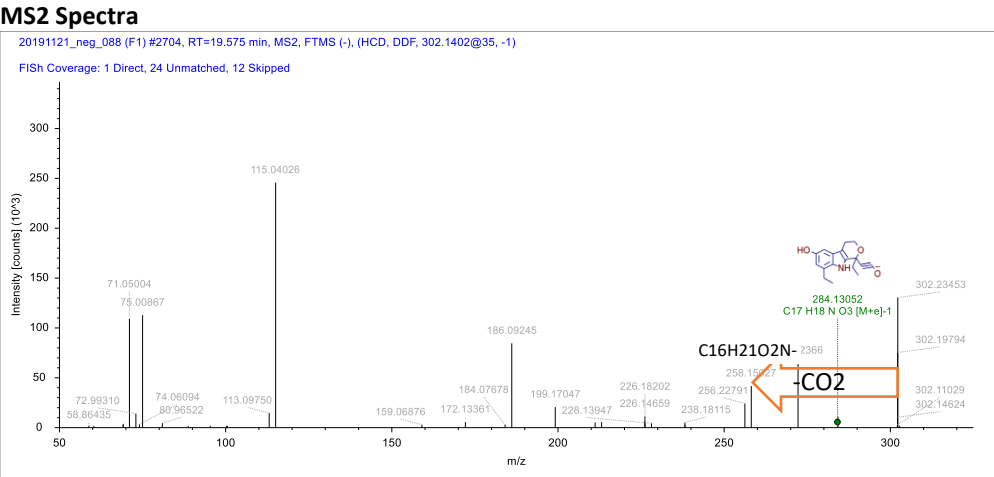
Formula
C17H21O4N

Atomic modification
+O



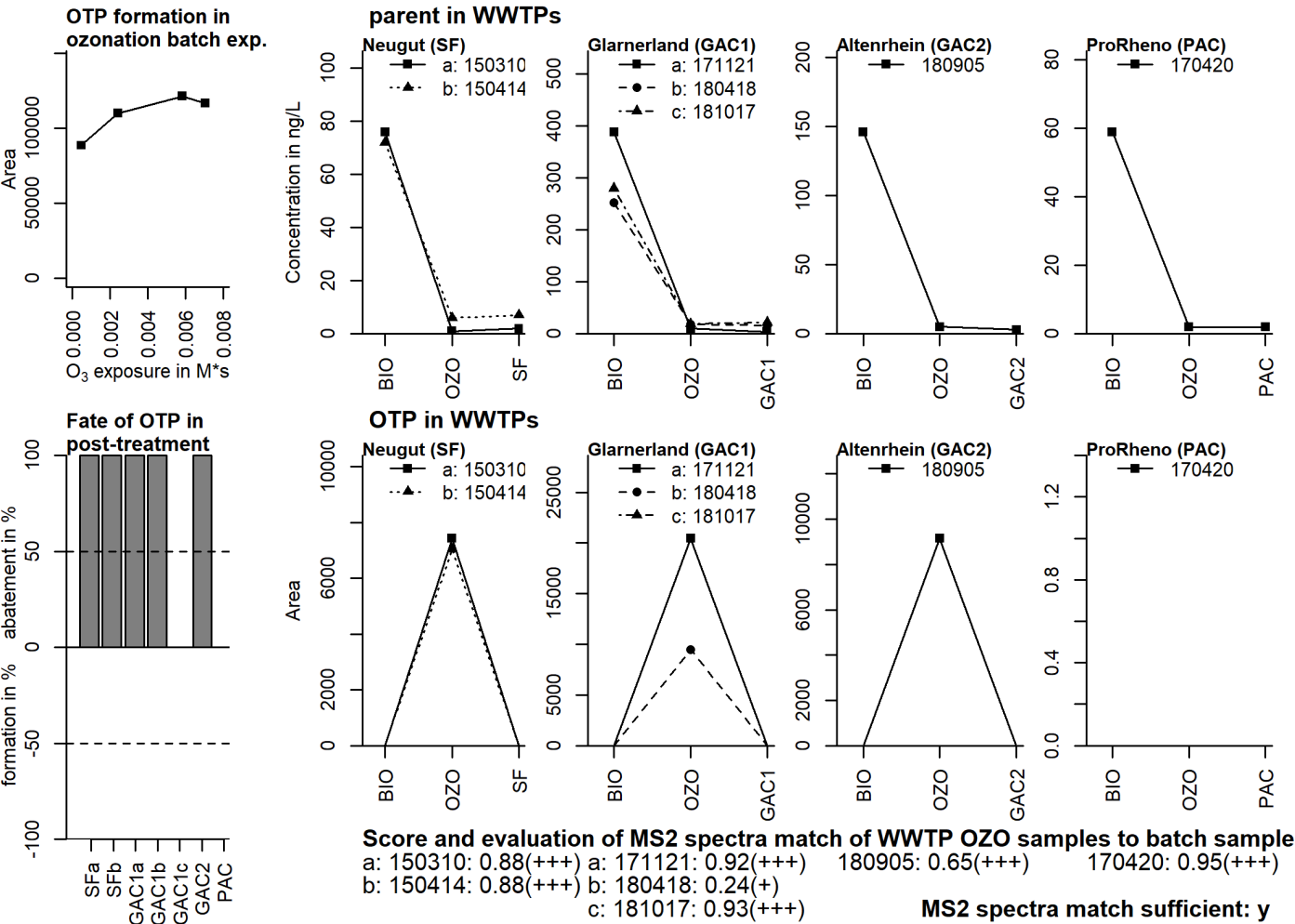
Confidence Level
Level 3

Massbank ID
ET405701



Additional Evidence for Structure Interpretation

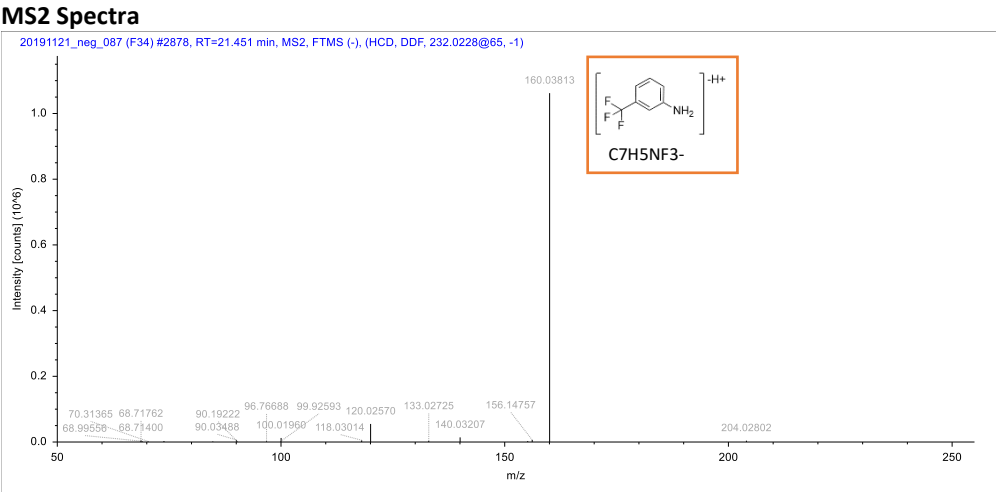
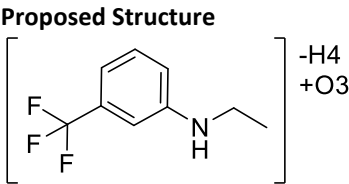
The atomic modification from the elemental formula of the parent compound to this TP is +O. This fits to the addition of a hydroxy group via an OH radical. This likely did not occur at the benzene moiety, because once formed, phenol moieties react further quickly during ozonation. There is no further evidence for the exact type and location of the modification. The structures of the MS² fragments are drawn exemplarily.



MS Spectra
Neg 232.0224 [m-H]-

Formula
C₉H₆O₃NF₃

Atomic modification
-C₅H₄ + O

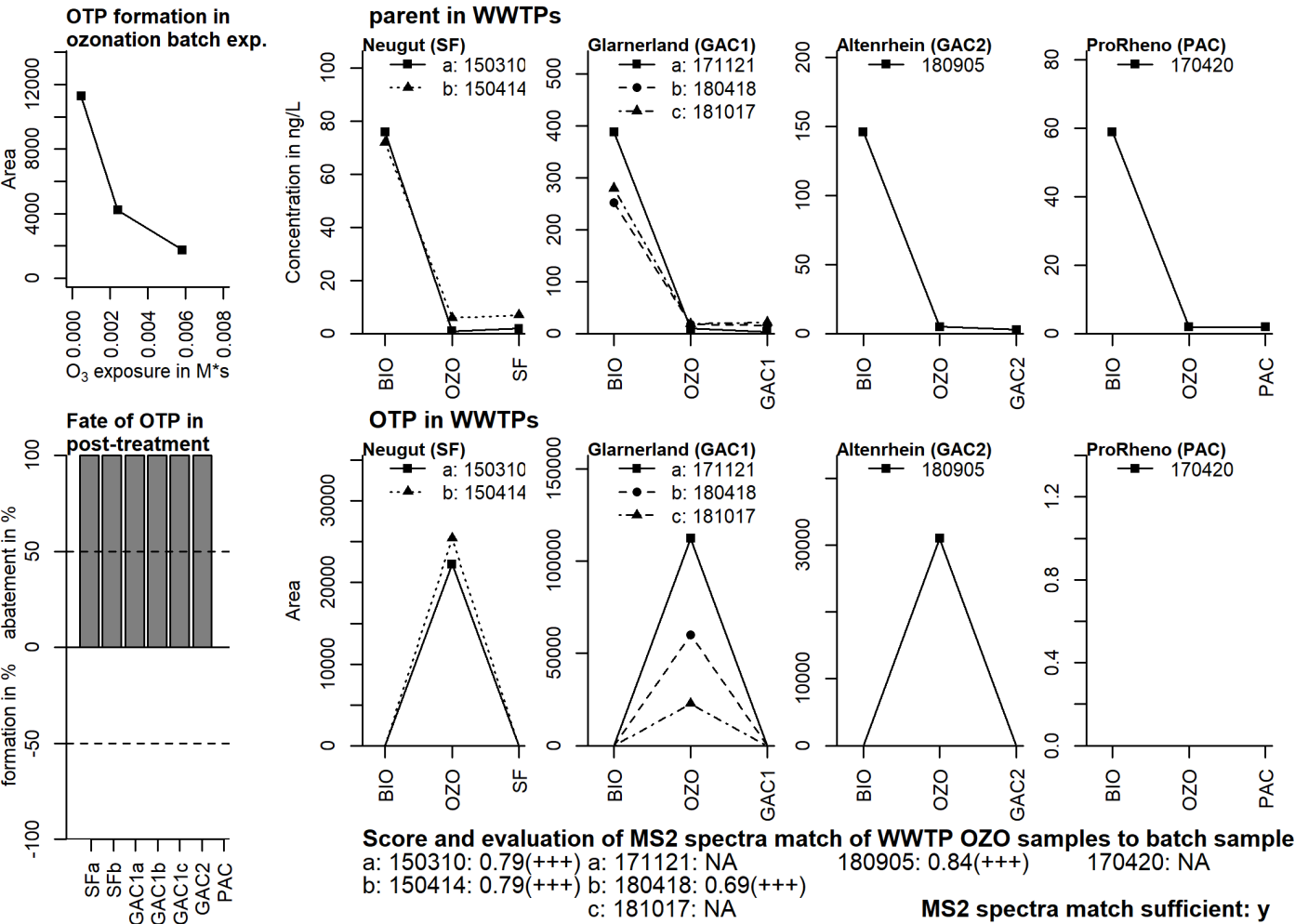


Additional Evidence for Structure Interpretation

The atomic modification from the elemental formula of the parent compound to this TP is -C₅H₄ + O. Fragment 160 indicates that the 5 C atoms were cleaved off from the benzoic acid moiety. There is no further evidence for the location of the oxygen atoms.

Confidence Level
Level 3

Massbank ID
ET405901

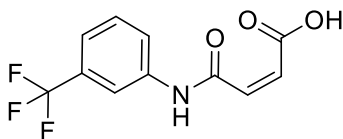


MS Spectra
Neg 258.0384 [m-H]-

Formula
C11H8O3NF3

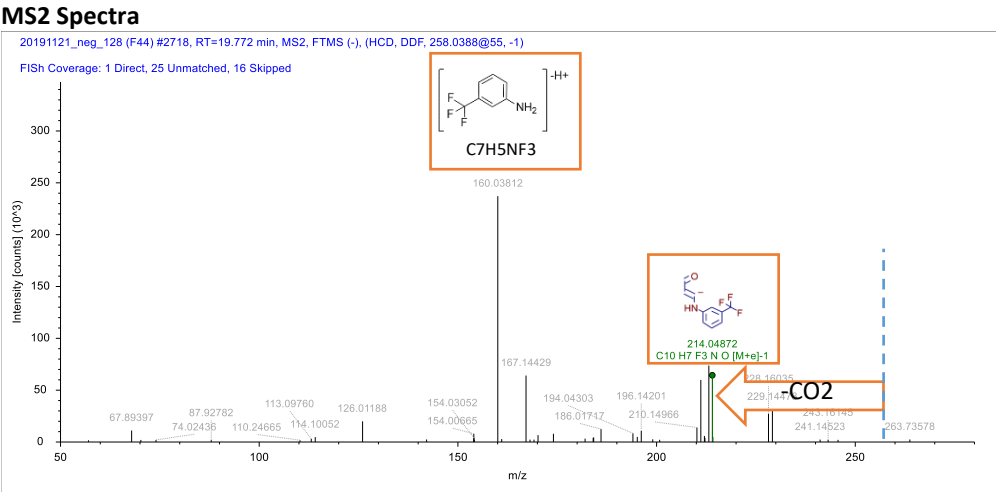
Atomic modification
-C3H2 +O

Proposed Structure



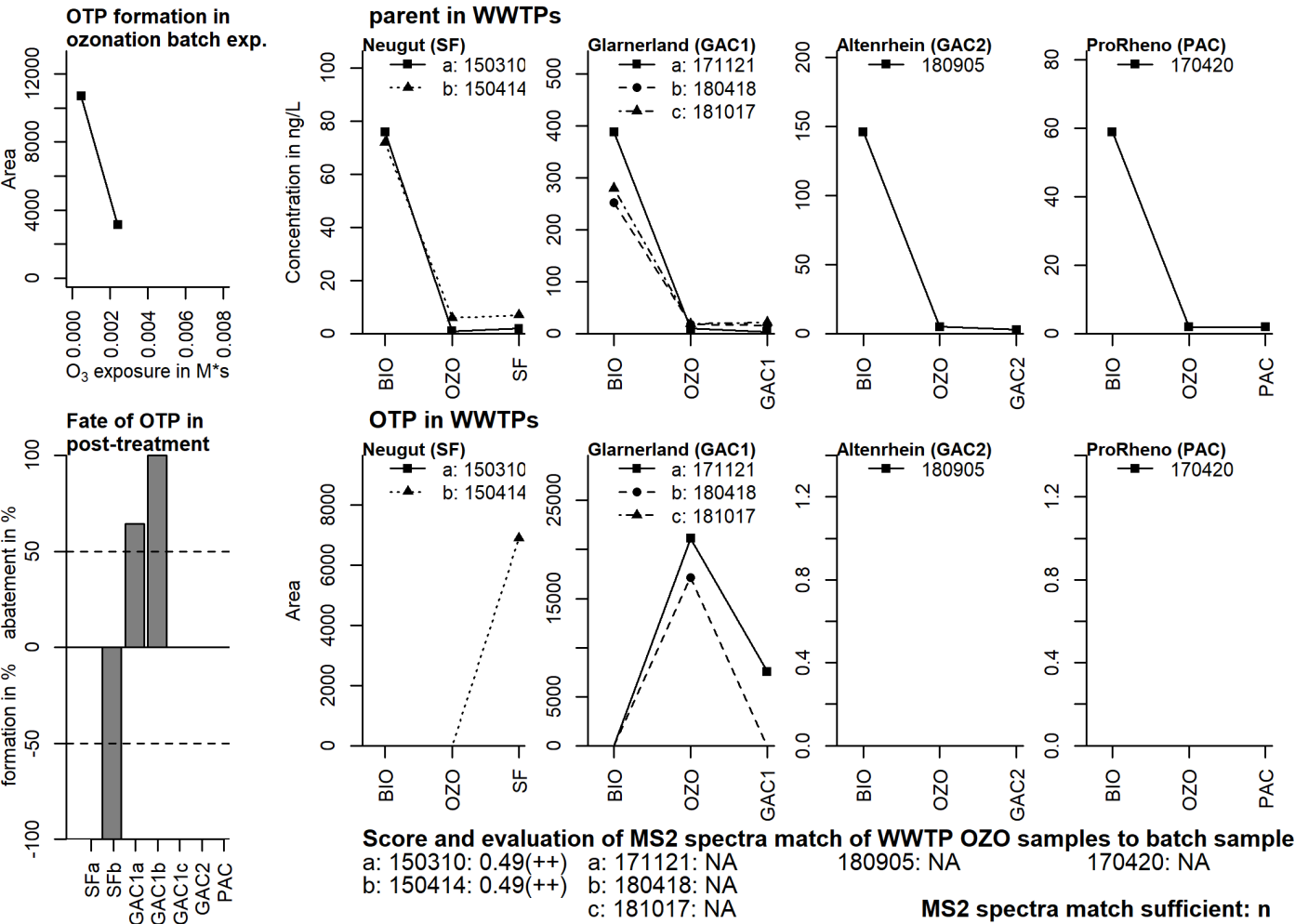
Confidence Level
Level 3

Massbank ID
ET406001



Additional Evidence for Structure Interpretation

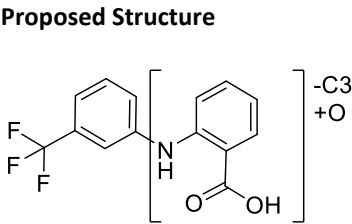
The atomic modification from the elemental formula of the parent compound to this TP is -C3H2 +O. Fragment 160 indicates that the 3 C atoms were cleaved off from the benzoic acid moiety. The neutral loss of CO2 between the molecular ion and fragment 214 indicates the presence of a carboxylic acid moiety. The drawn suggested structure fits the molecular formula of this OTP and could have been formed after two Criegee additions on the benzoic acid moiety, leading to the cleaving off of three carbon atoms, including the carboxylic acid moiety. The exact type and location of the modification remain unknown. The structures of the MS2 fragments are drawn exemplarily.



MS Spectra
Neg 260.0537 [m-H]-

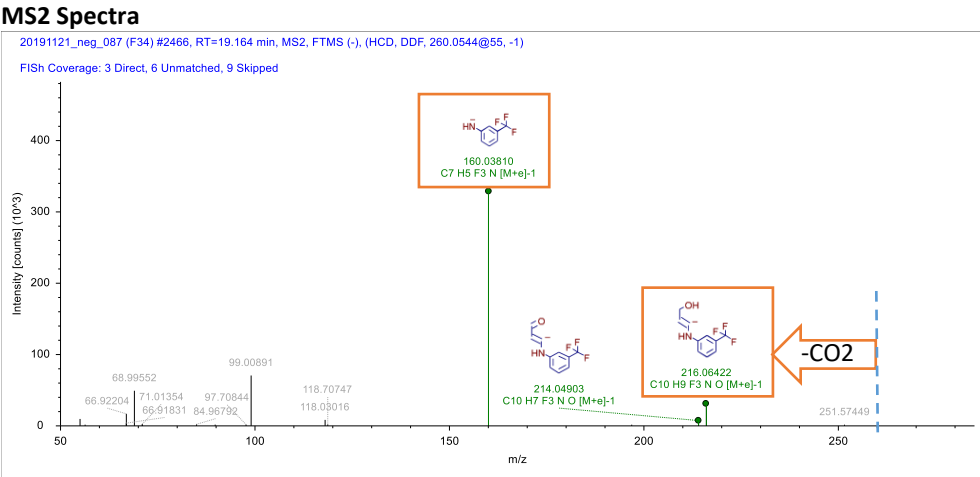
Formula
C11H10O3NF3

Atomic modification
-C3 +O



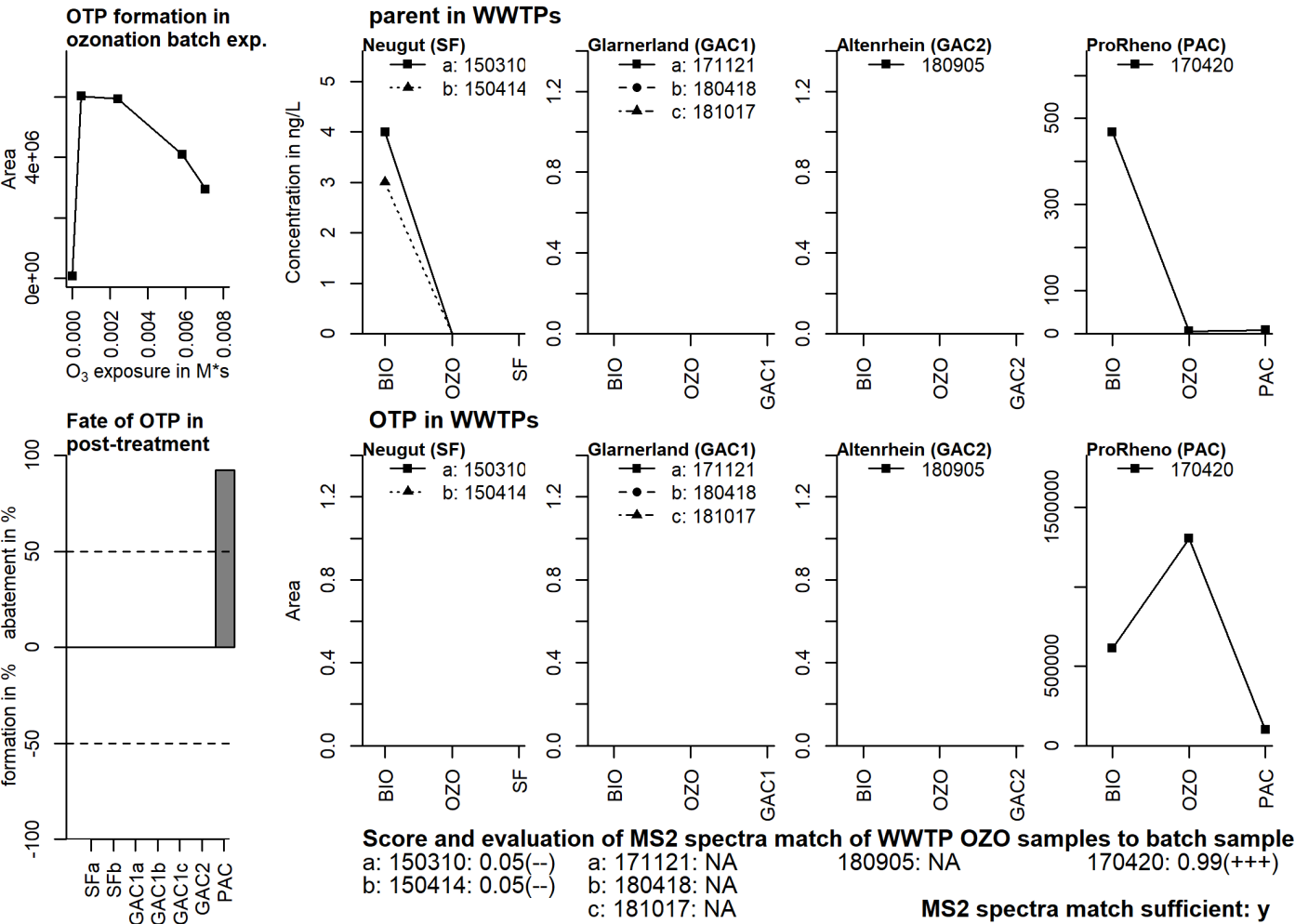
Confidence Level
Level 3

Massbank ID
ET406101



Additional Evidence for Structure Interpretation

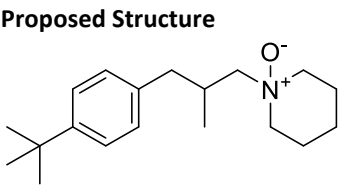
The atomic modification from the elemental formula of the parent compound to this TP is -C3 +O. Fragment 160 indicates that the 3 C atoms were cleaved off from the benzoic acid moiety. The neutral loss of CO2 between the molecular ion and fragment 216 as well as the fact that this OTP was detected in the negative MS spectrum suggest the presence of a carboxylic acid moiety. The exact type and location of the cleavage/modification remain unknown. The structures of the MS² fragments are drawn exemplarily.



MS Spectra
Pos 290.2478 [m+H]⁺

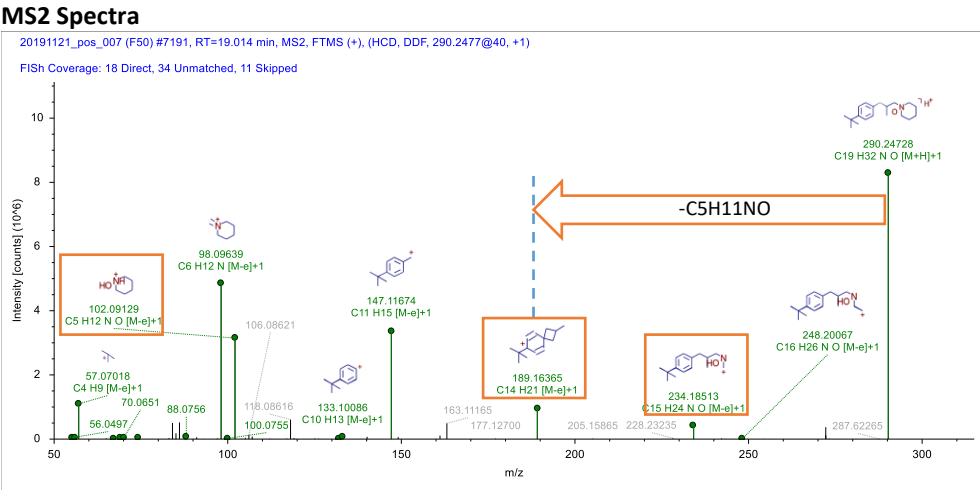
Formula
C₁₉H₃₁ON

Atomic modification
+O



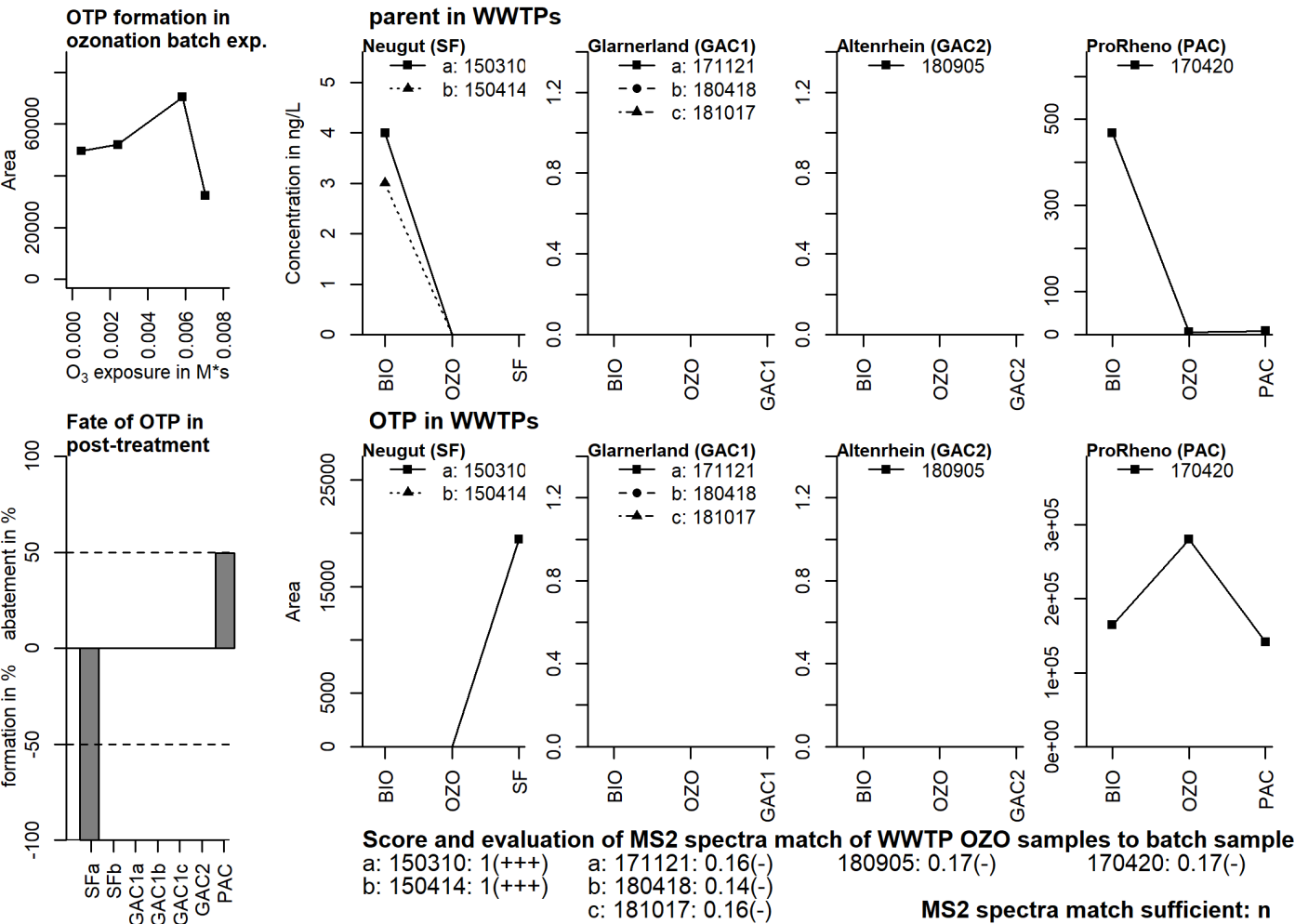
Confidence Level
Level 3

Massbank ID
ET402001



Additional Evidence for Structure Interpretation

The MS² fragments at the nominal masses 189, 147, 133, 69 and 57 were also observed for the parent compound. The atomic modification from the elemental formula of the parent compound to this TP is +O. The neutral loss of C₅H₁₁NO between the precursor and fragment 189 as well as fragment 102 indicate that the modification took place at the piperidine moiety. Tertiary amines are very reactive to form *N*-oxides during ozonation (von Sonntag & von Gunten 2012). Therefore, the formation of a *N*-oxide at the piperidine moiety is very likely (Borowska et al. 2016).



MS Spectra

Pos 320.2220 [m+H]⁺

Neg 318.2072 [m-H]⁻

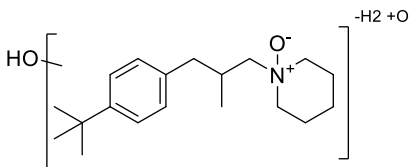
Formula

C₁₉H₂₉O₃N

Atomic modification

-H₂ +O₃

Proposed Structure

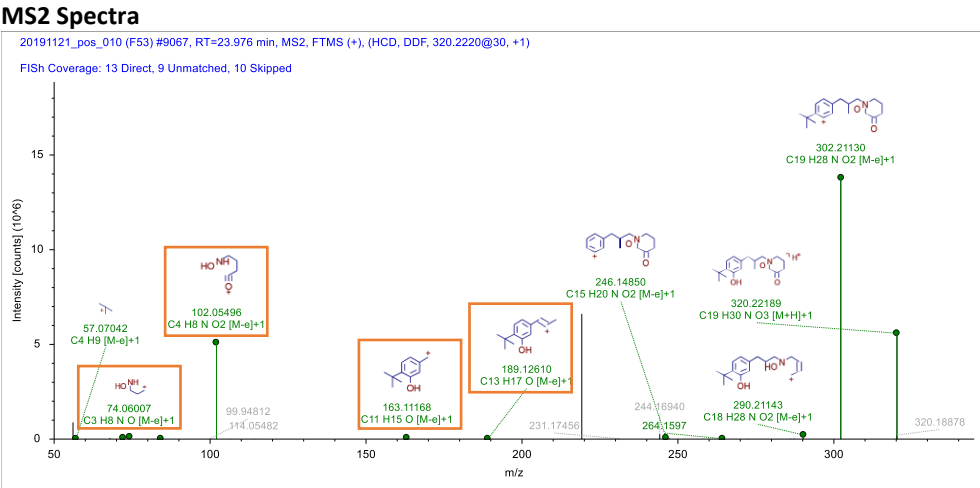


Confidence Level

Level 3

Massbank ID

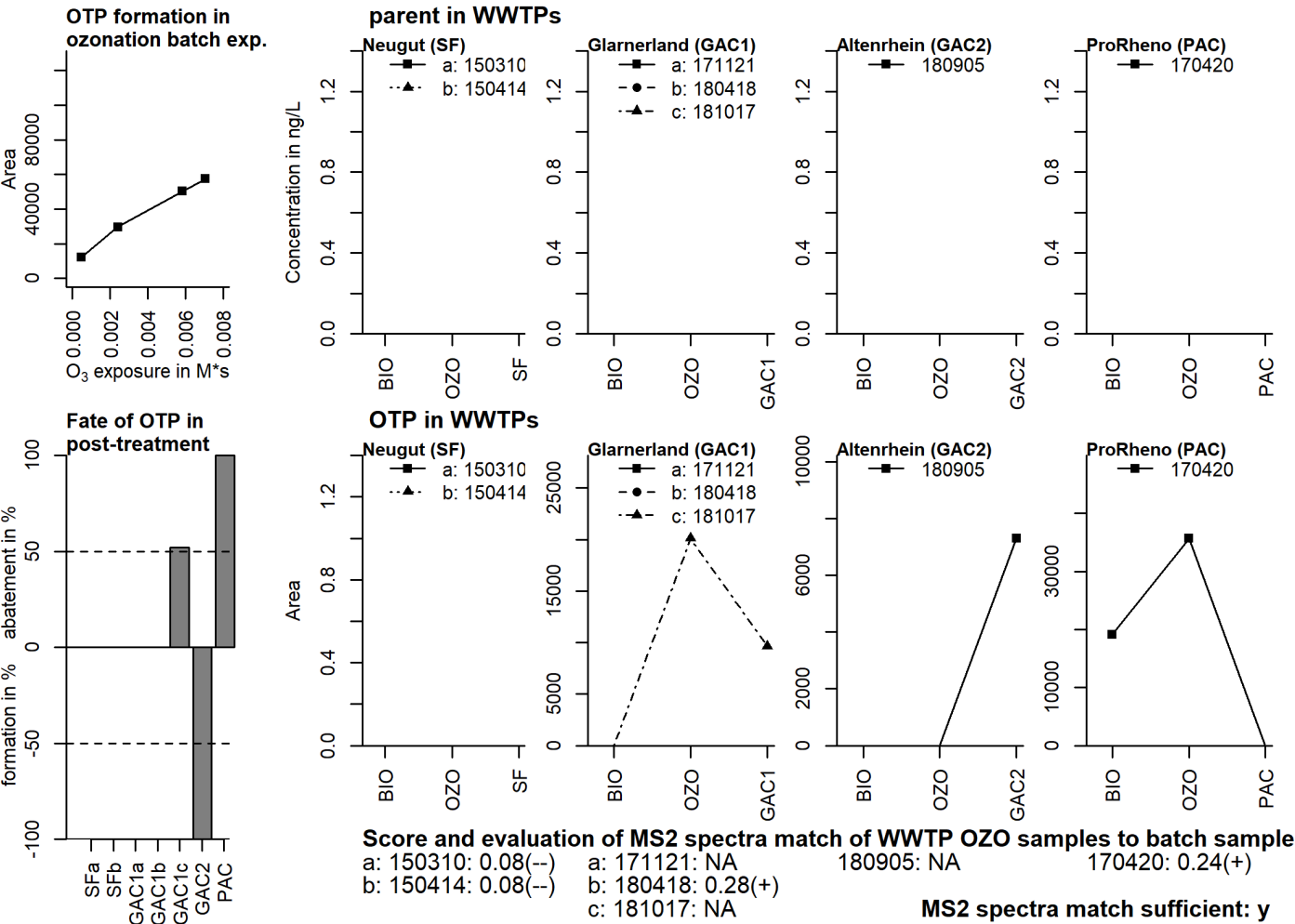
ET402101



Additional Evidence for Structure Interpretation

The atomic modification from the elemental formula of the parent compound to this TP is -H₂ +O₃. Fragment 74 was also found in the MS² spectrum of FEN_p_290.2478_18.3, which was identified as fenpropidin *N*-oxide. This indicates that a *N*-oxide formation is also likely for this TP.

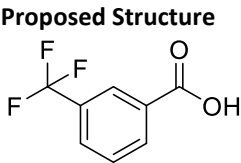
The second oxygen atom was likely added by hydroxylation of an aliphatic C- atom with an OH radical. Hydroxylation of the benzene moiety is unlikely because once formed, phenol moieties react further quickly during ozonation. Fragments 163 and 189 suggest that an oxygen atom was added at the 1-(tert-Butyl)-4-methylbenzyl moiety. Fragment 102 indicates that the third oxygen atom was added at the piperidine moiety. The exact position and type of modification remain unknown. The structures of the MS² fragments are drawn exemplarily.



MS Spectra
Neg 189.0168 [m-H]-

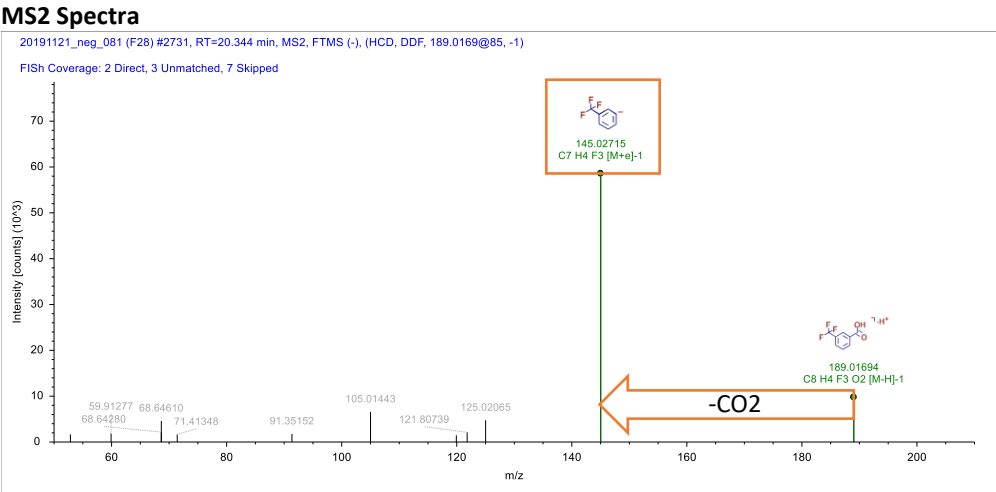
Formula
C8H5O2F3

Atomic modification
-C4H9N +O2



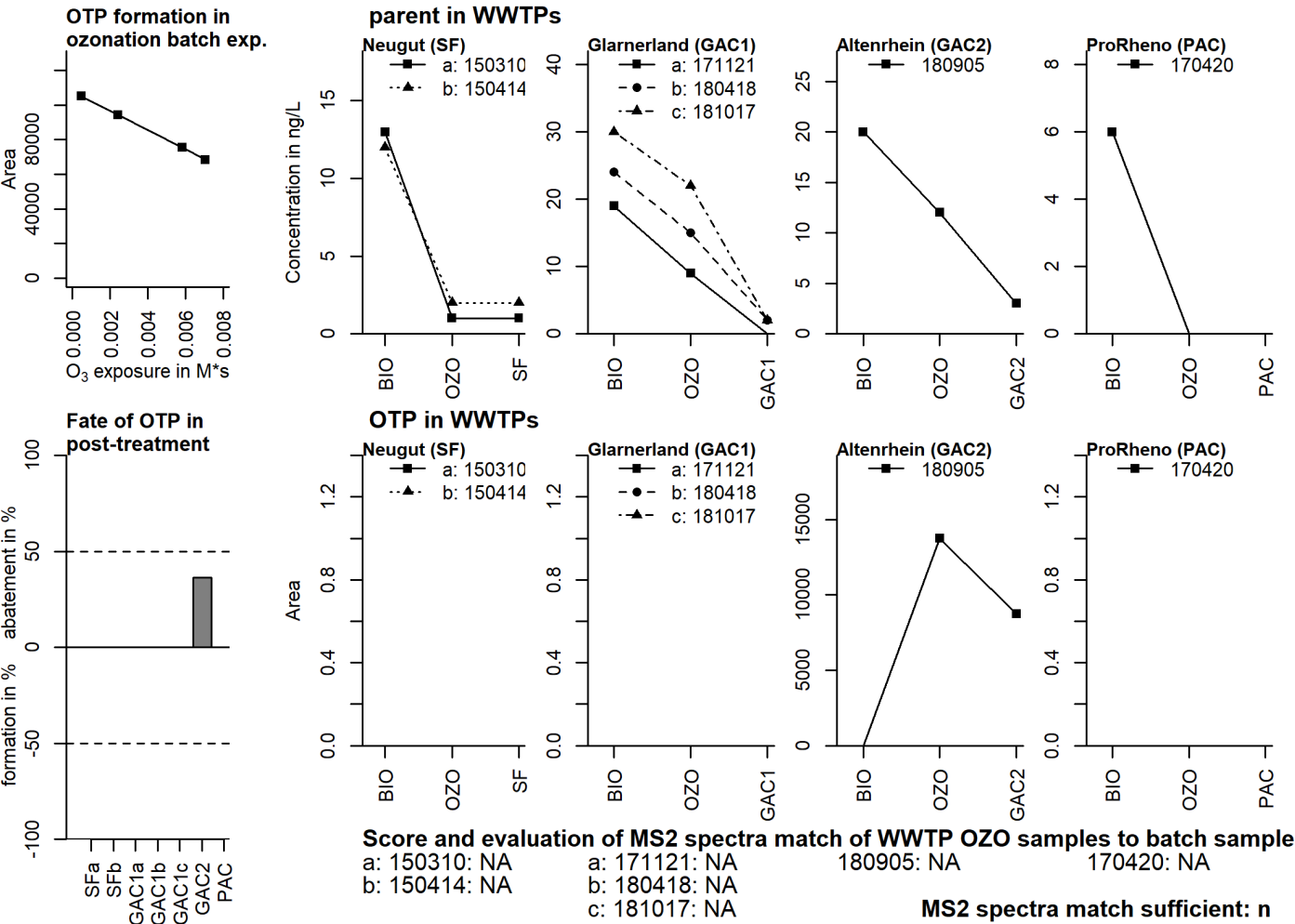
Confidence Level
Level 3

Massbank ID
ET405801



Additional Evidence for Structure Interpretation

The atomic modification fits for fenfluramine to a cleavage of the diethylamine moiety. The addition of 2 oxygen atoms fits to the formation of carboxylic acid moiety on the remaining C atom of the side chain. The neutral loss of 44 between the proposed structure and fragment 145 also indicates the formation of a carboxylic acid moiety.



MS Spectra

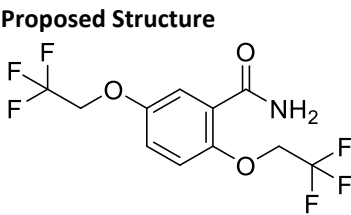
Neg	362.0466	[m+FA-H]-
Neg	316.0415	[m-H]-

Formula

C11H9O3NF6

Atomic modification

-C6H11N

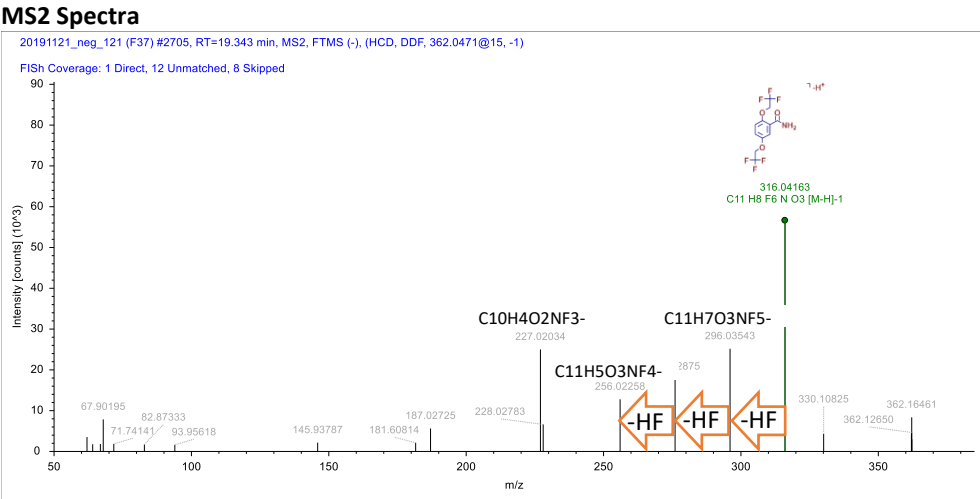


Confidence Level

Level 3

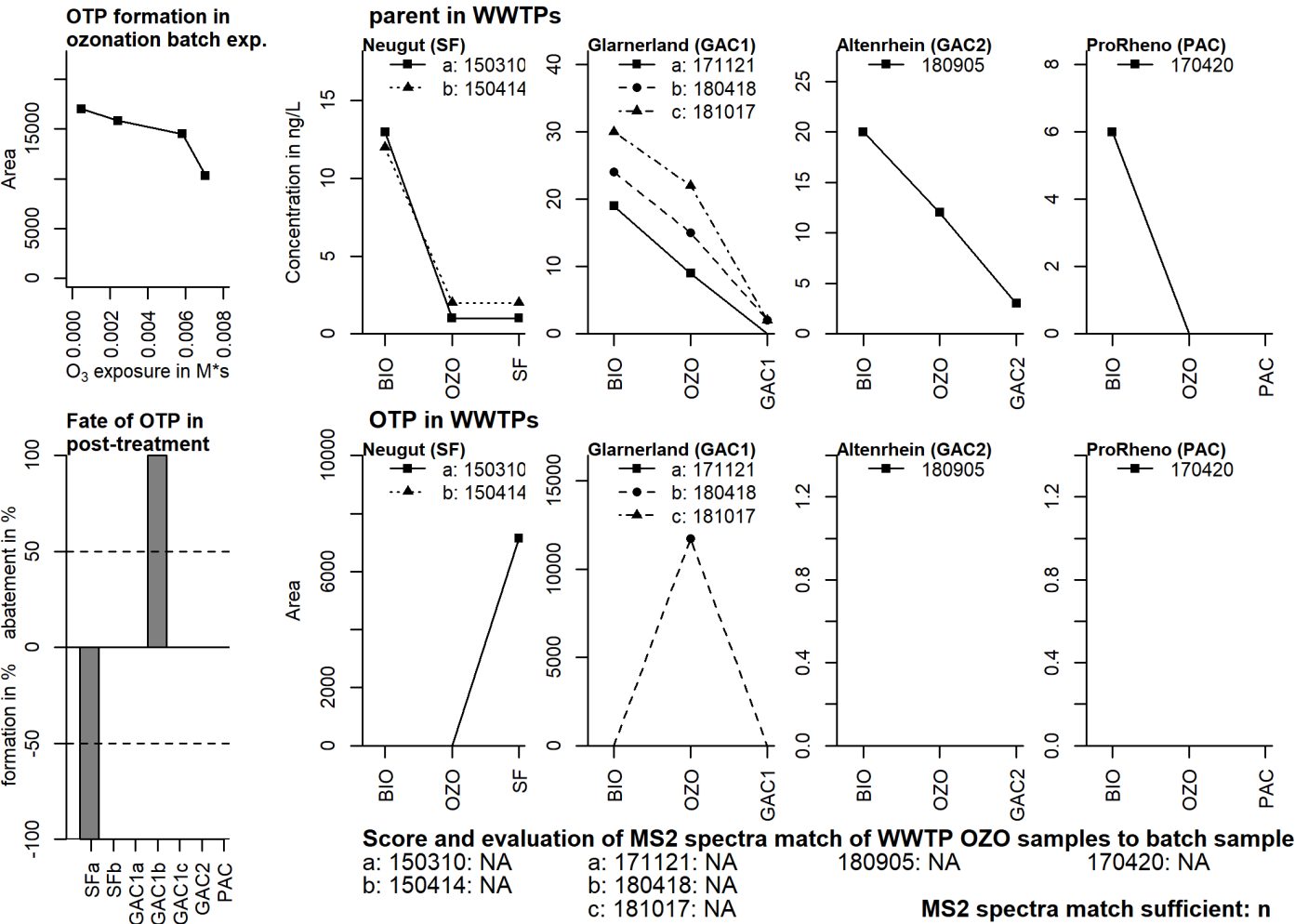
Massbank ID

ET406267



Additional Evidence for Structure Interpretation

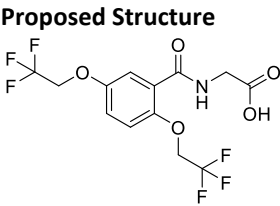
The atomic modification from the elemental formula of the parent compound to this OTP is -C6H11N. This modification fits to a splitting-off of the methylpiperidine moiety of the parent compound. A N-dealkylation reaction is known to occur during ozonation (von Sonntag & von Gunten 2012). Therefore, for this reaction the proposed structure is likely.



MS Spectra
Neg 374.0465 [m-H]-

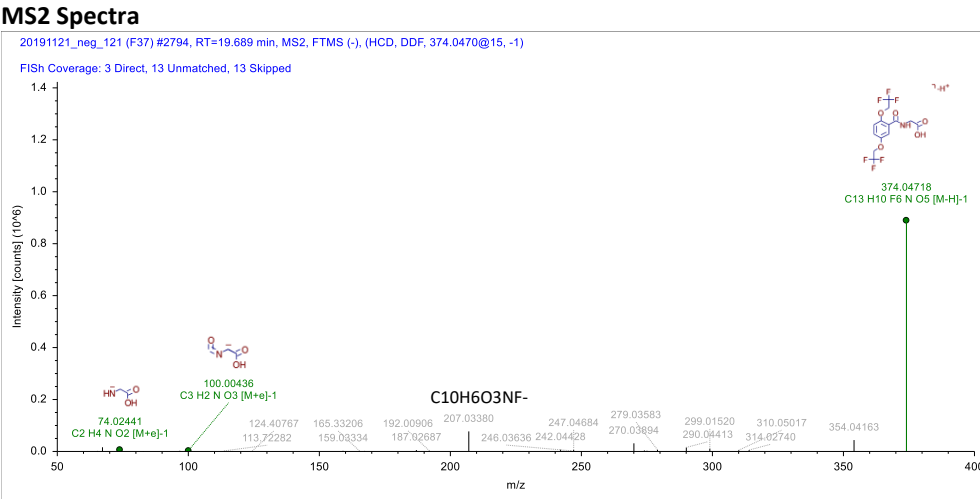
Formula
C₁₃H₁₁O₅NF₆

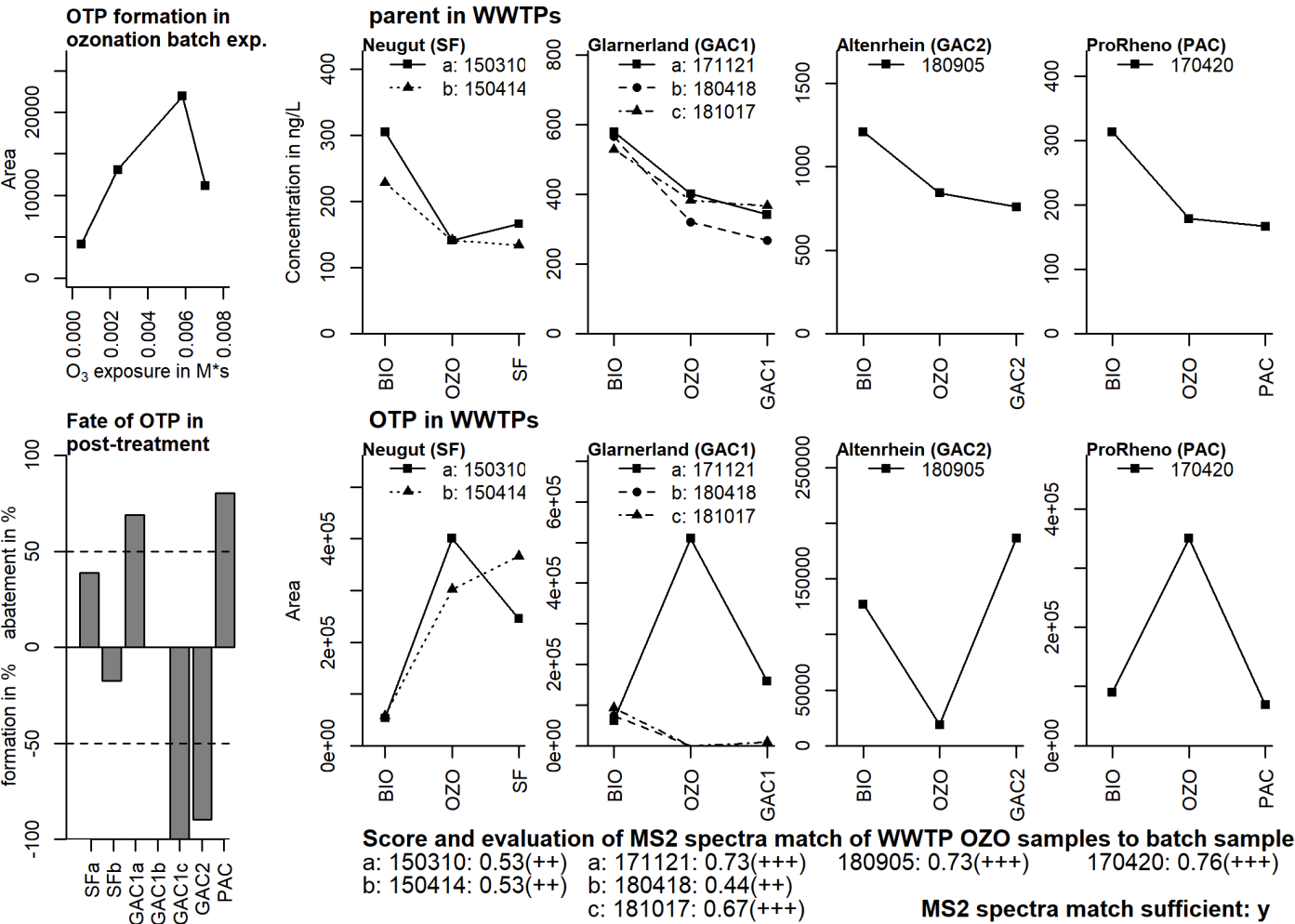
Atomic modification
-C₄H₉N +O₂



Confidence Level
Level 3

Massbank ID
ET406301



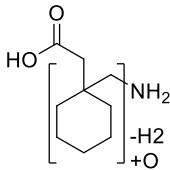


MS Spectra
Pos 186.1124 [m+H]⁺

Formula
C₉H₁₅O₃N

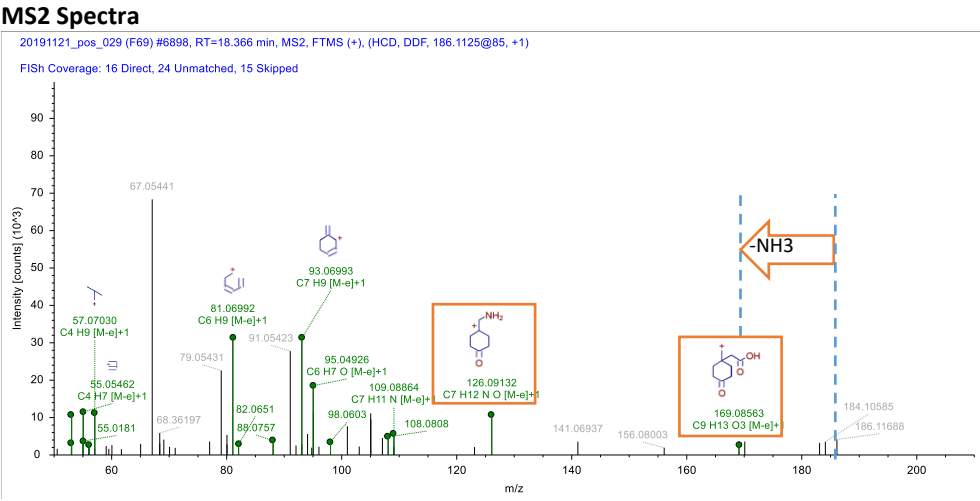
Atomic modification
-H₂ + O

Proposed Structure



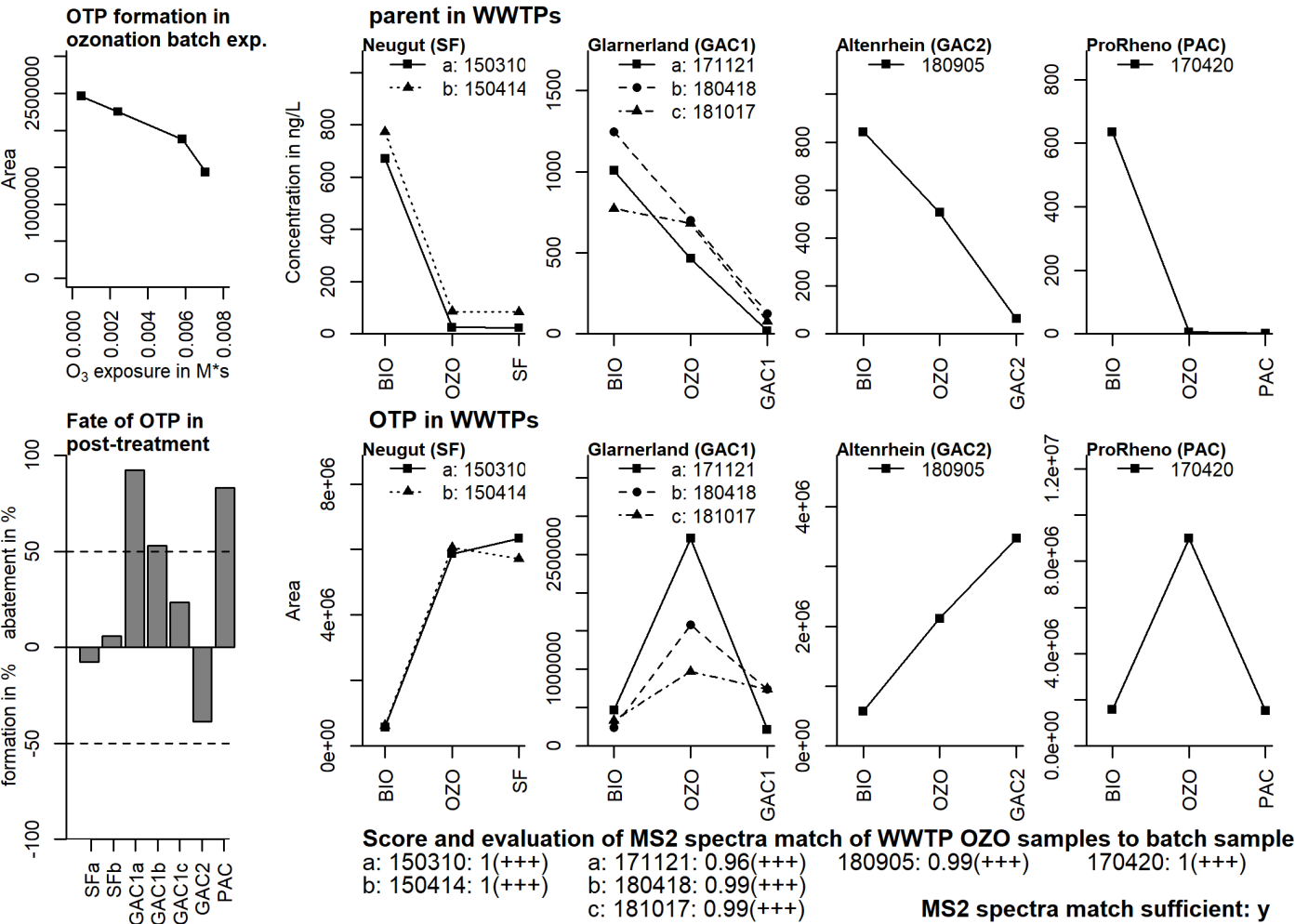
Confidence Level
Level 3

Massbank ID
Et402201



Additional Evidence for Structure Interpretation

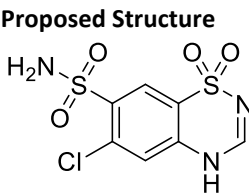
The atomic modification from the elemental formula of the parent compound to this OTP is -H₂ + O. The neutral loss of NH₃ between the precursor and fragment 169 indicates that the oxidation did not take place at the primary amine. Fragment 126 indicates that the oxidation took place at the methylcyclohexyl moiety. Different reactions could have led to this molecular formula, most likely the formation of a carbonyl moiety at the α-C atom of the primary amine. However, the exact position and type of modification remain unknown. The structures of the MS² fragments are drawn exemplarily.



MS Spectra
Neg 293.9415 [m-H]-
Pos 295.9561 [m+H+]

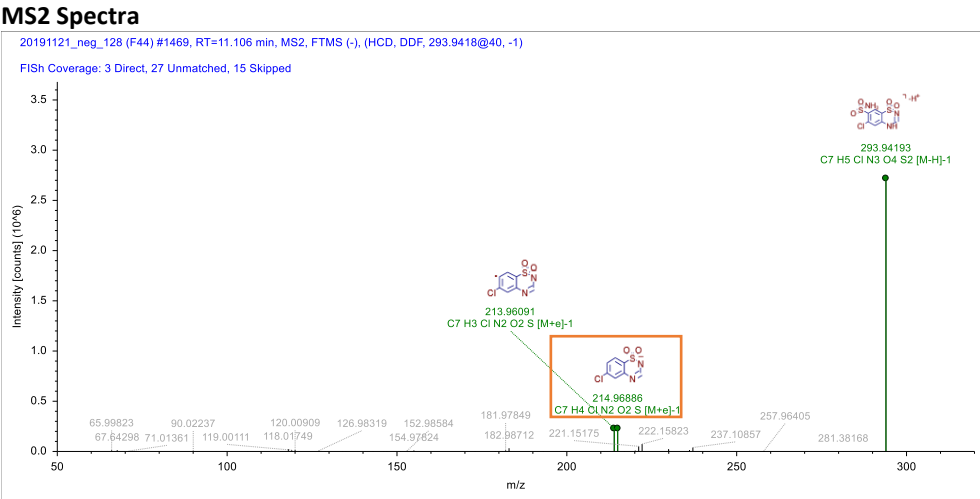
Formula
C7H6O4N3ClS2

Atomic modification
-H2



Confidence Level
Level 1

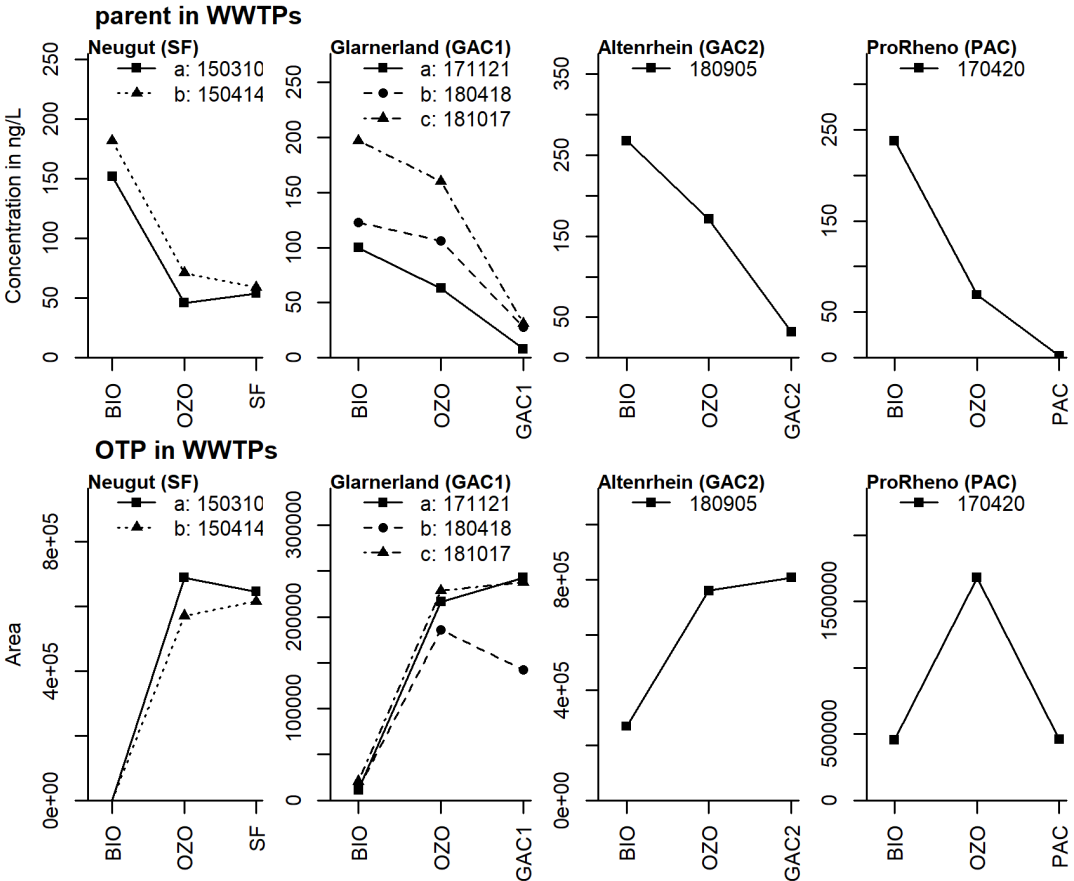
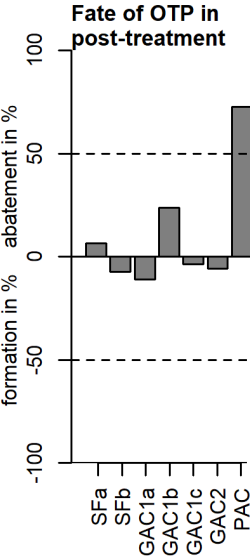
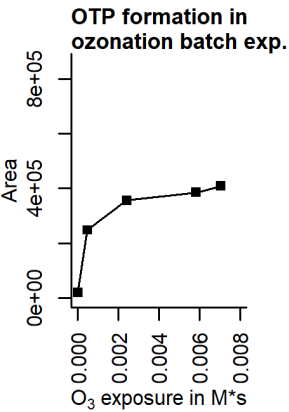
Massbank ID
ET406401



Additional Evidence for Structure Interpretation

The atomic modification from the elemental formula of the parent compound to this OTP is -H2. This fits to the oxidation of a C-N bond. A fragment with mass 216 was observed for the parent compound and corresponds to fragment 214 here with abstraction of 2 hydrogen atoms.

The drawn structure, chlorthiazide, was identified as OTP of hydrochlorthiazide by Borowska et al. (2016). The structure was confirmed with a reference standard.

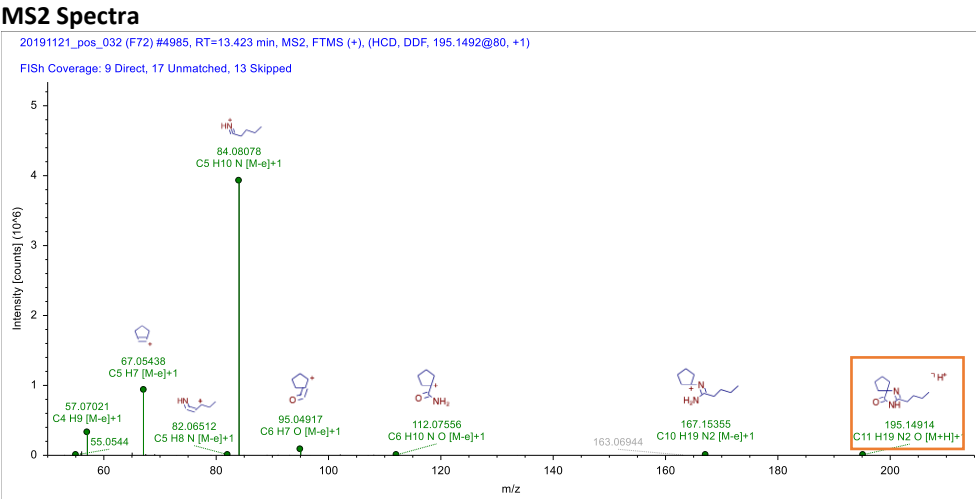
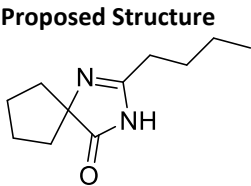


MS2 spectra match sufficient: y

MS Spectra
Pos 195.1491 [m+H]⁺

Formula
C₁₁H₁₈ON₂

Atomic modification
-C₁₄H₁₀N₄

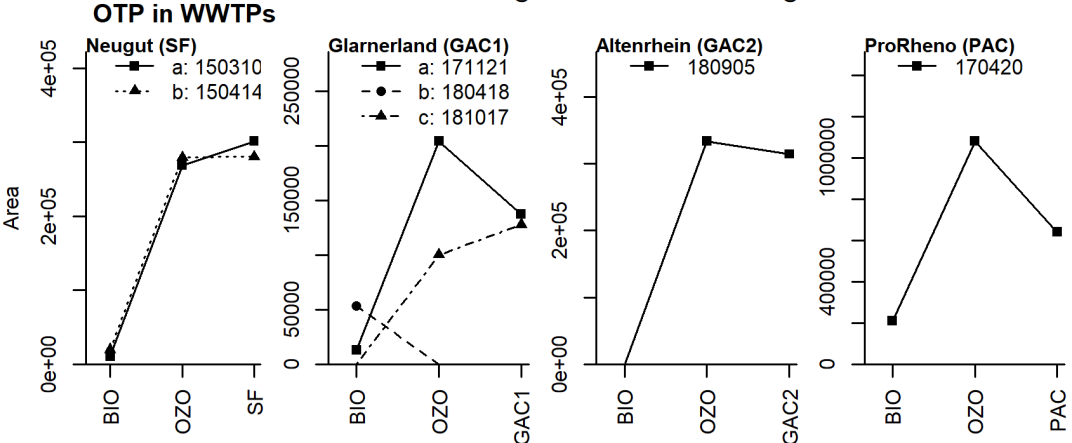
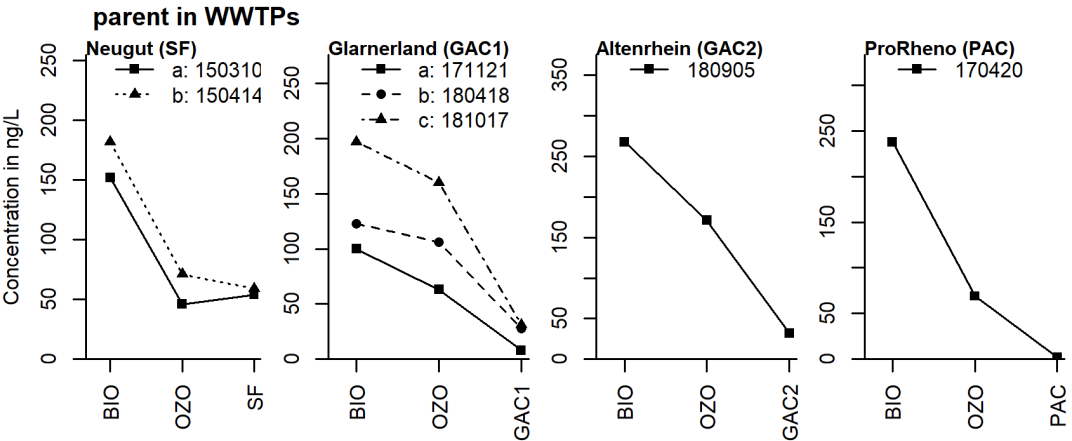
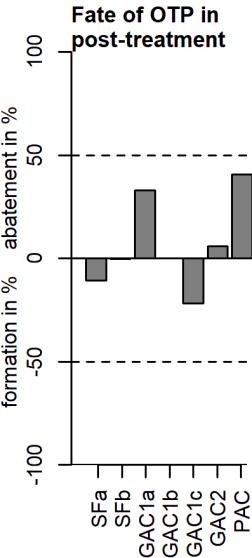
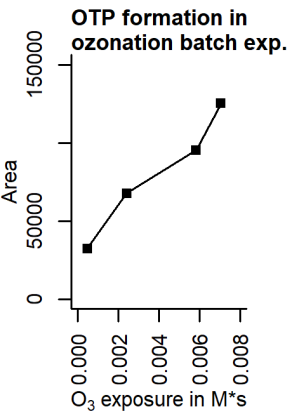


Additional Evidence for Structure Interpretation

A fragment with the nominal mass 195 was also observed for the parent compound and corresponds to this precursor. It indicates that the proposed structure was conserved. The atomic modification from the elemental formula of the parent compound to this OTP fits to a cleavage of the tetrazolebiphenylmethyl moiety, which is common for all sartans.

Confidence Level
Level 3

Massbank ID
ET402301



Score and evaluation of MS2 spectra match of WWTP OZO samples to batch sample

Sample	Score	Evaluation
a: 150310	0.78	(+++)
a: 171121	0.08	(--)
a: 180905	0.04	(--)
a: 170420	0.75	(+++)
b: 150414	0.78	(+++)
b: 180418	0.07	(--)
c: 181017	0.06	(--)

MS2 spectra match sufficient: y

MS Spectra

Pos 253.1445 [m+H]⁺
Neg 251.1399 [m-H]⁻

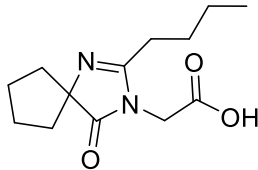
Formula

C₁₃H₂₀O₃N₂

Atomic modification

-C₁₂H₈N₄ +O₂

Proposed Structure



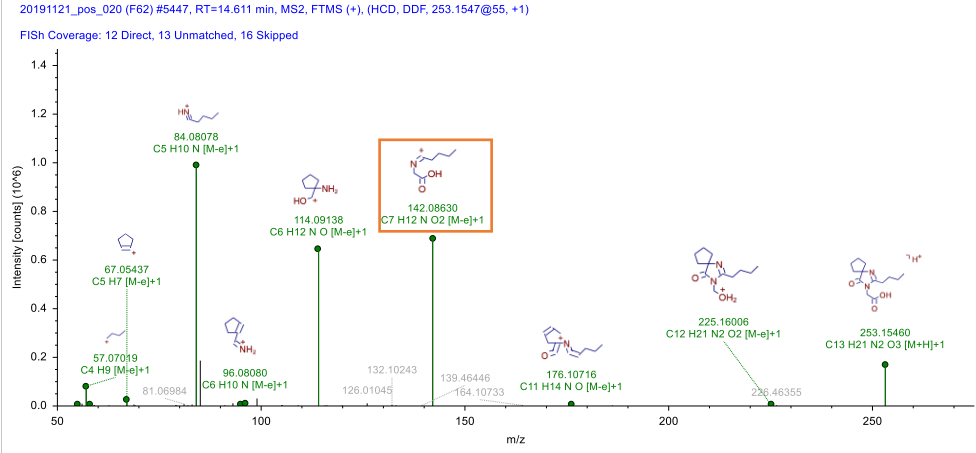
Confidence Level

Level 3

Massbank ID

ET402401

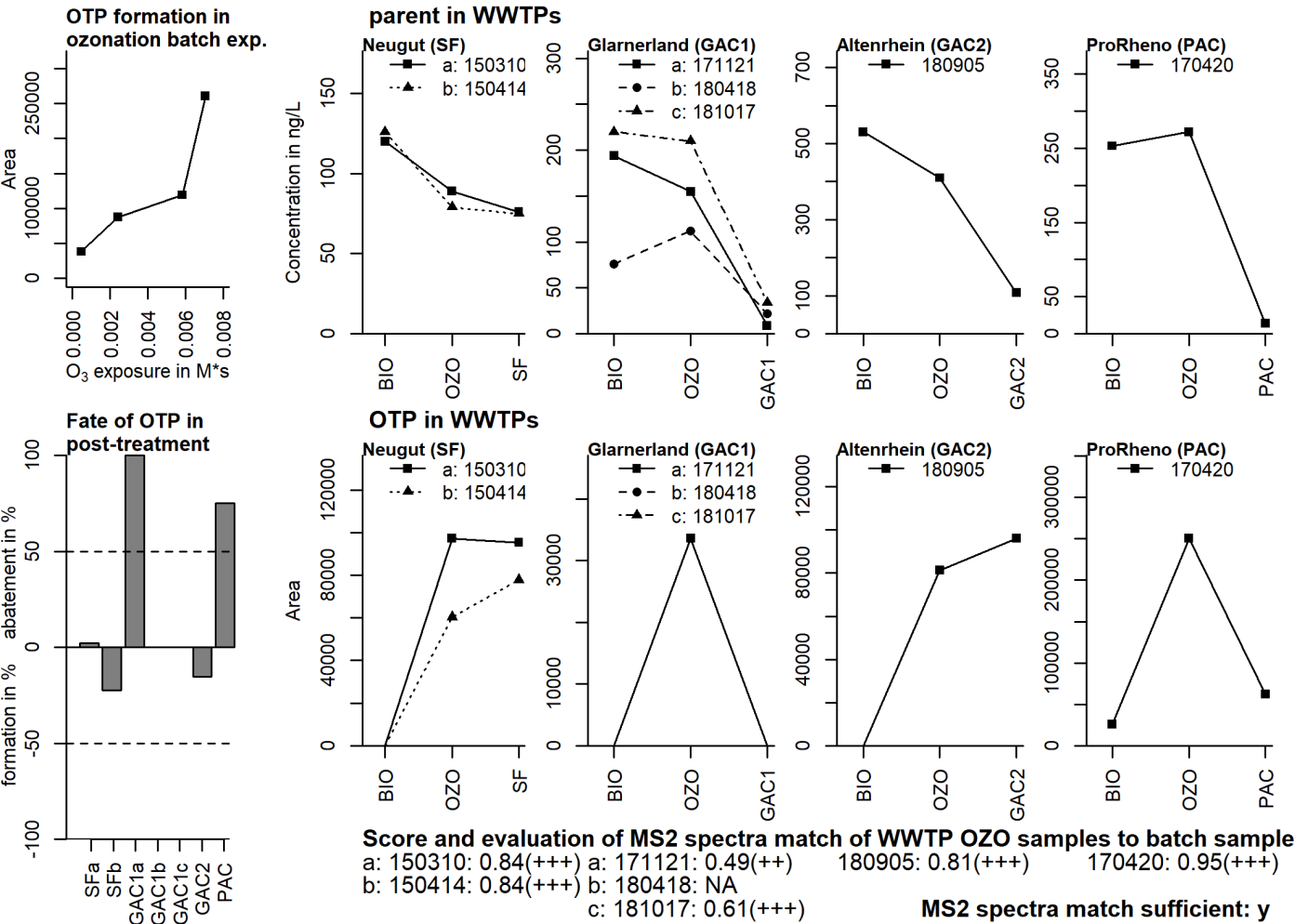
MS2 Spectra



Additional Evidence for Structure Interpretation

The atomic modification from the elemental formula of the parent compound to this OTP is -C₁₂H₈N₄ +O₂. This fits to a cleavage within the benzene moiety of the tetrazolebiphenylmethyl part, which is common for all sartans, and the addition of two oxygen atoms. Fragment 142 indicates that the two oxygens are connected to the ethylpentylamine moiety. This TP was detected in negative MS spectrum, which indicates the presence of a carboxylic acid moiety. There is no further evidence in the MS² spectrum about the exact position. However, aromatic compounds are known to form cis,cis-muconic acid like structures (von Sonntag & von Gunten 2012). Therefore, the formation of this kind of structure, followed by a C-C bond cleavage, resulting in an aminoacetic acid structure, is likely.

The atomic modification from the elemental formula of the parent compound to this OTP is $-C_6H_4N_4 + O_2$. This fits to a cleavage within the benzene moiety of the tetrazolebiphenylmethyl part, which is common for all sartans, and the addition of two oxygen atoms. Fragment 135 indicates that the two oxygens are connected to the dimethylphenyl moiety. This OTP was also detected in negative MS spectrum, which indicates the presence of a carboxylic acid moiety. Furthermore, the precursor showed a neutral loss of CO_2 in the negative MS spectrum. There is no further evidence in the MS^2 spectrum about the exact position. However, aromatic compounds are known to form *cis,cis*-muconic acid like structures (von Sonntag & von Gunten 2012). Therefore, the formation of this kind of structure, followed by a C-C bond cleavage, resulting in a benzoic acid structure, is likely.



MS Spectra

Pos 259.9988 [m+H]⁺

Neg 257.9842 [m-H]⁻

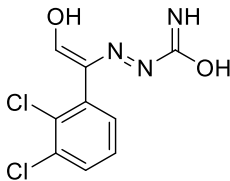
Formula

C₉H₇O₂N₃Cl₂

Atomic modification

-N₂ +O₂

Proposed Structure

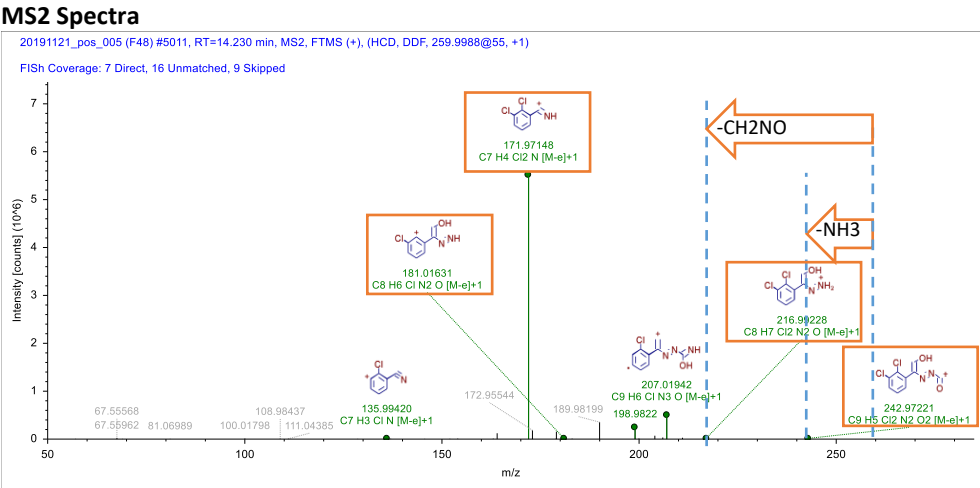


Confidence Level

Level 3

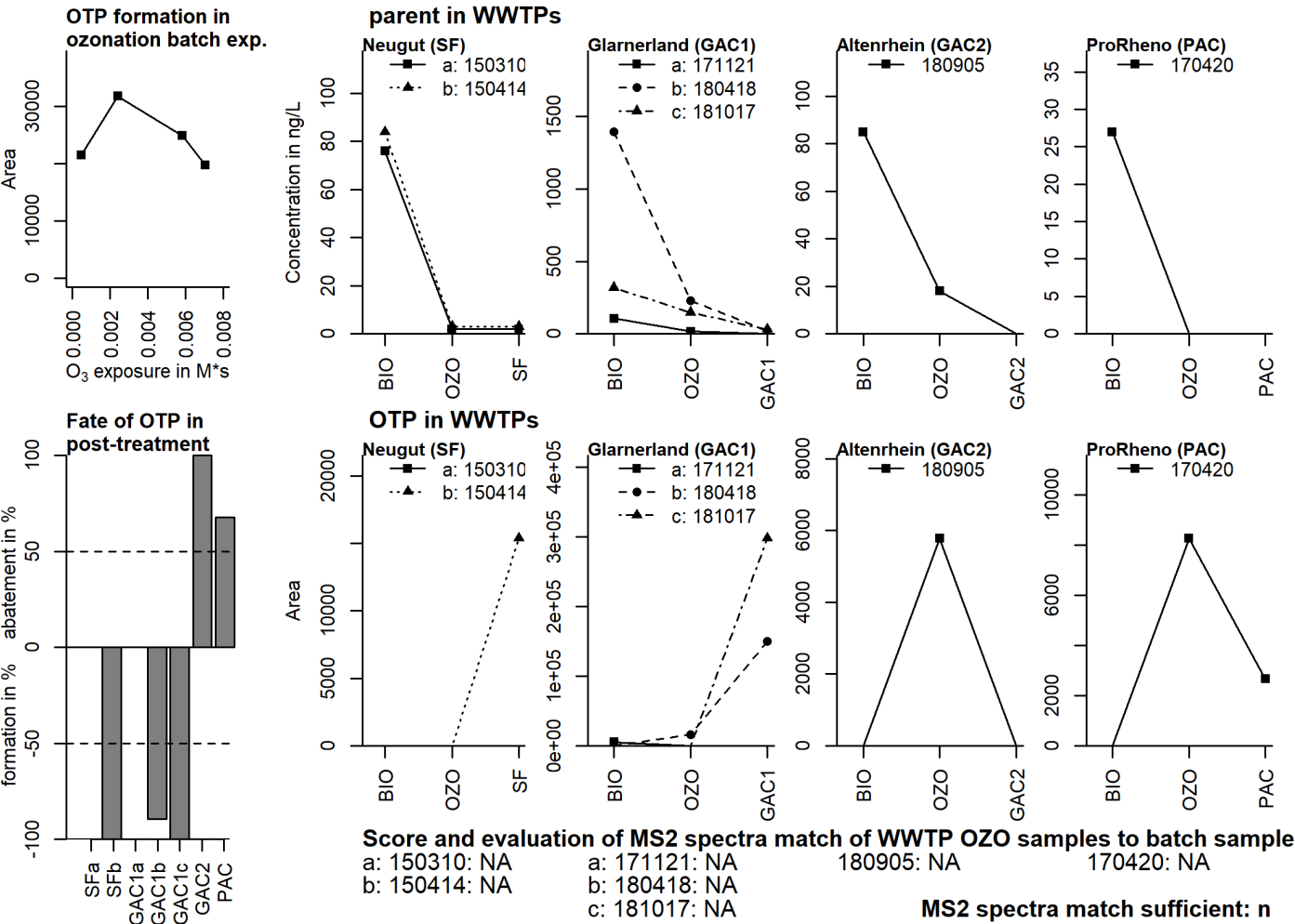
Massbank ID

ET402601



Additional Evidence for Structure Interpretation

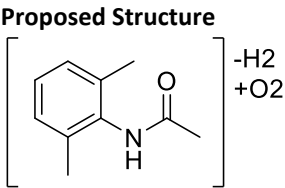
The MS² fragment at the nominal mass 171 was also observed for the parent compound. It indicates that the structure drawn for the fragment 171 is a substructure of this OTP. The atomic modification from the elemental formula of the parent compound to this OTP is -N₂ +O₂. This fits to a cleavage of two nitrogen atoms within the triazin-5-ylmethyl moiety and the addition of two oxygen atoms. Keen et al. (2014) suggested this structure as OTP of lamotrigine. The neutral loss of NH₃ between the precursor and fragment 242 indicates that this OTP contains a terminal nitrogen atom. The neutral loss of CH₂NO between the precursor and fragment 216 fits to a cleavage of a formimidic acid like moiety. Bollmann et al. (2016) however suggested the formation of the N₂-oxide. What is an N₂-oxide?



MS Spectra
Neg 192.0665 [m-H]-

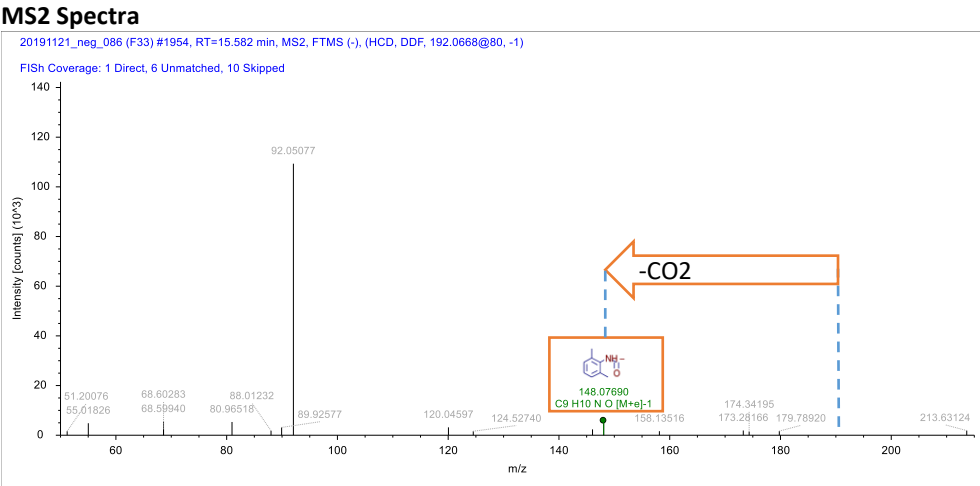
Formula
C10H11O3N

Atomic modification
-C4H11N +O2



Confidence Level
Level 3

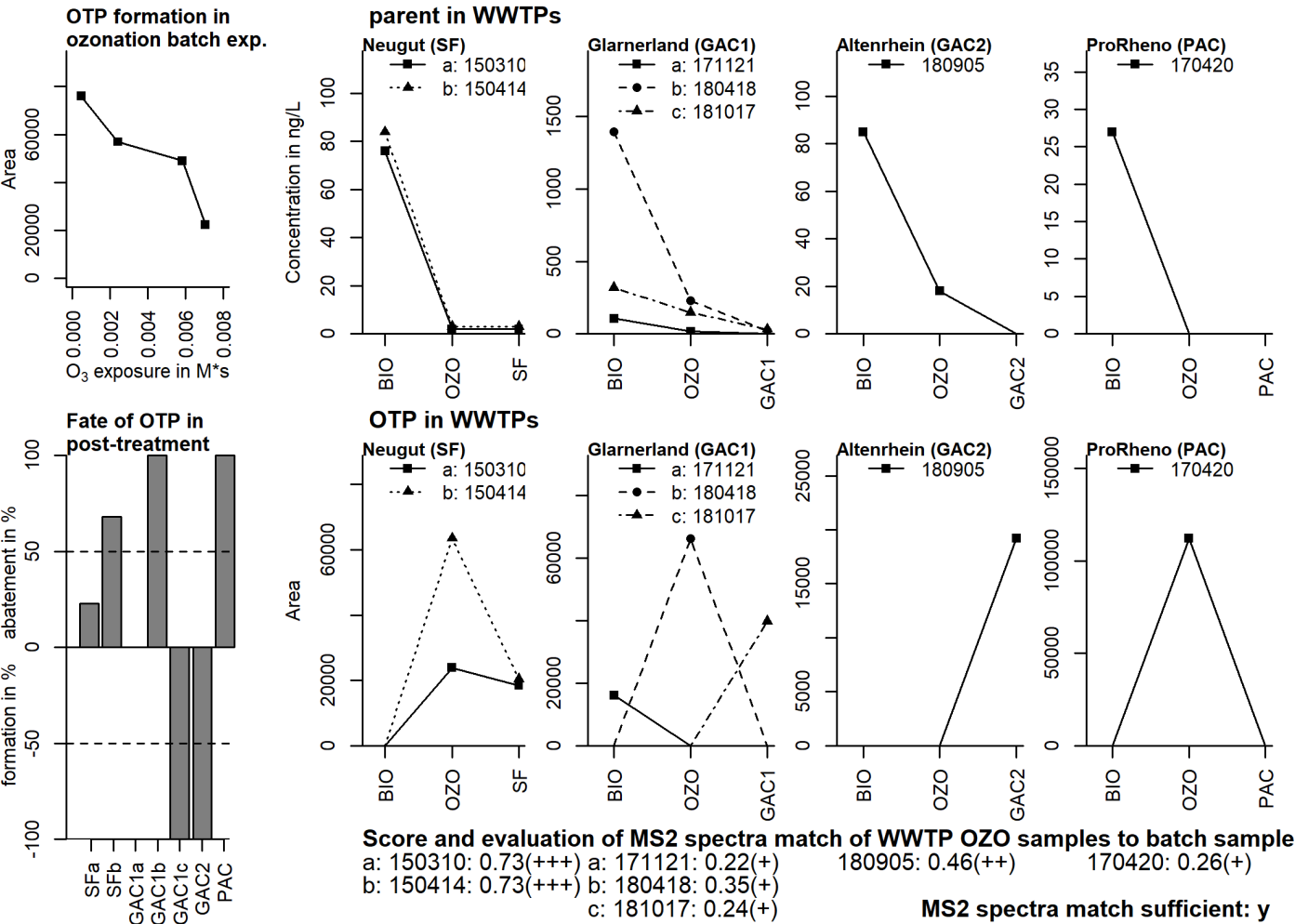
Massbank ID
ET407001



Additional Evidence for Structure Interpretation

The loss of C4H11N fits to the cleavage of diethylamine moiety. Two oxygen atoms were added on the remaining dimethylacetanilide moiety.

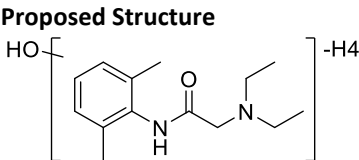
The neutral loss of CO2 (-44) between the proposed structure and fragment 148 indicates the presence of a carboxylic acid. Its exact location remains unknown.



MS Spectra
Pos 247.1438 [m+H]⁺

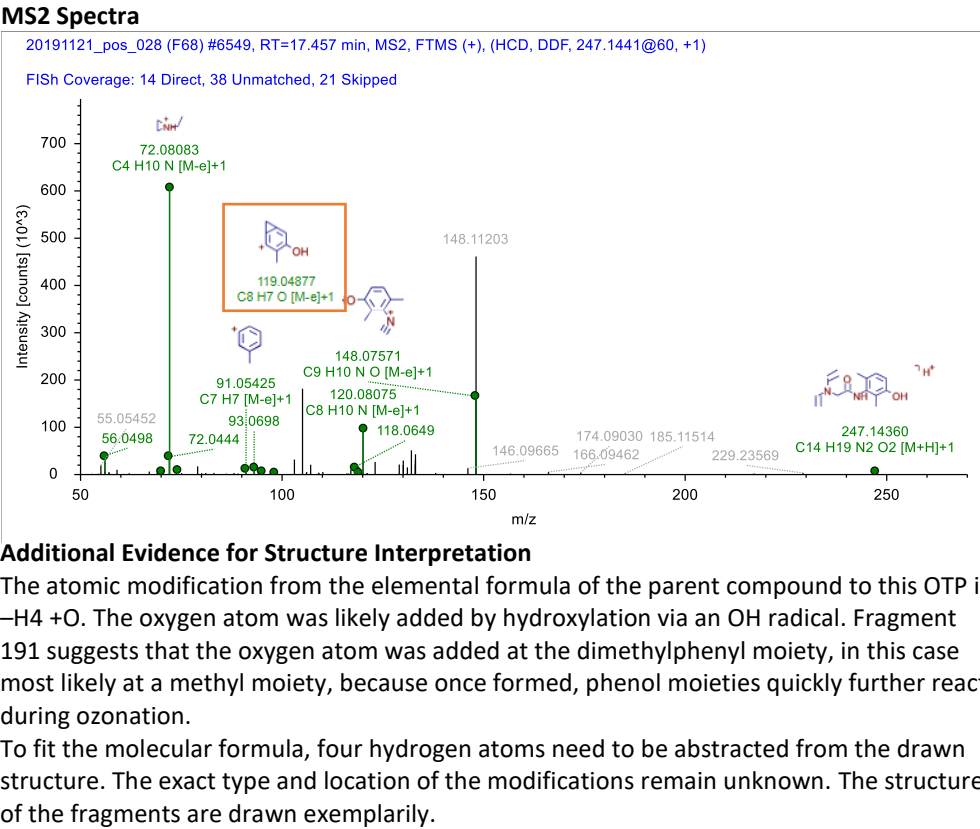
Formula
C₁₄H₁₈O₂N₂

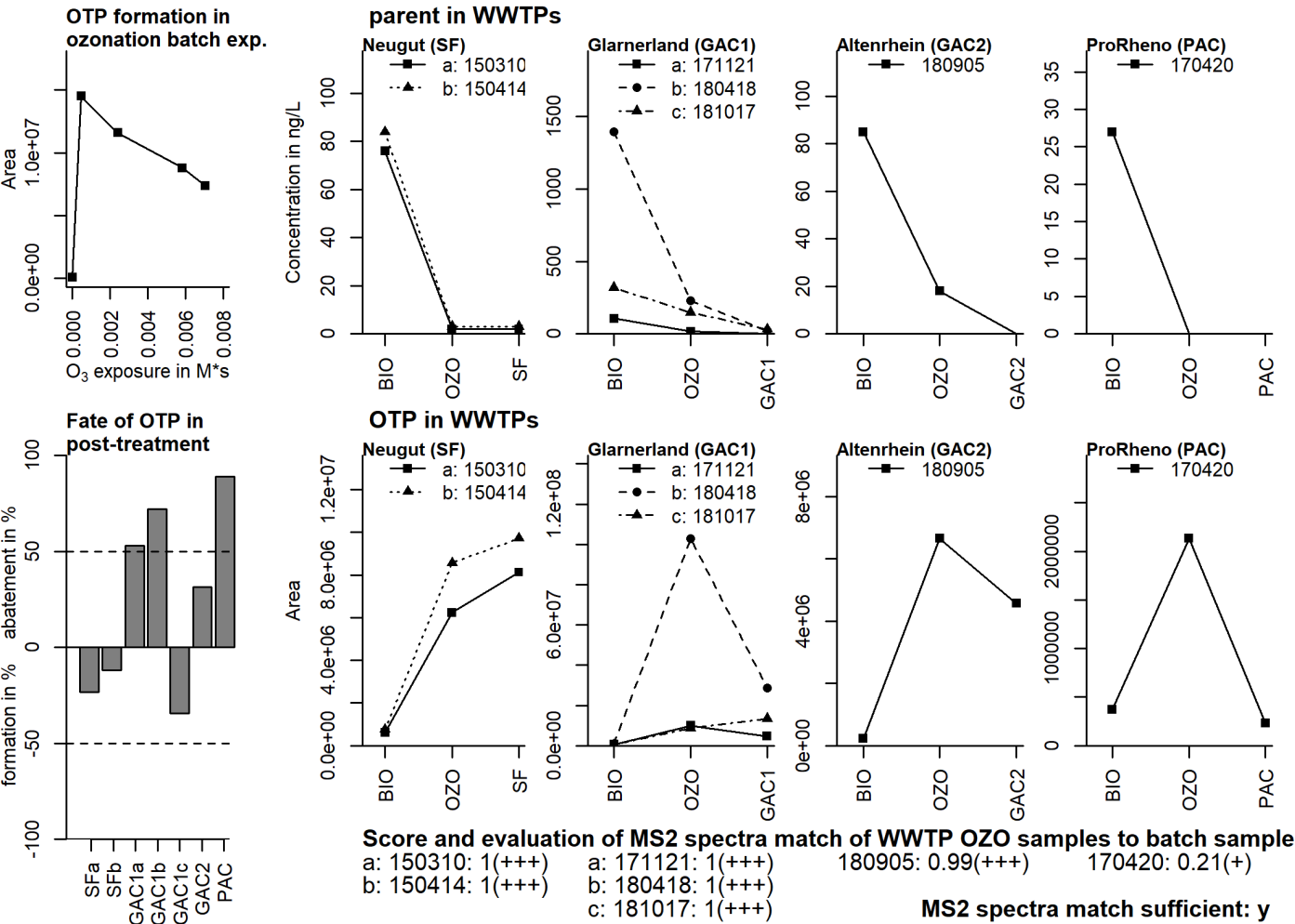
Atomic modification
-H₄ + O



Confidence Level
Level 3

Massbank ID
ET402701



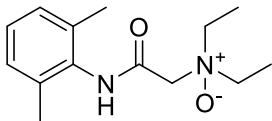


MS Spectra
Pos 251.1752 [m+H]⁺

Formula
C₁₄H₂₂O₂N₂

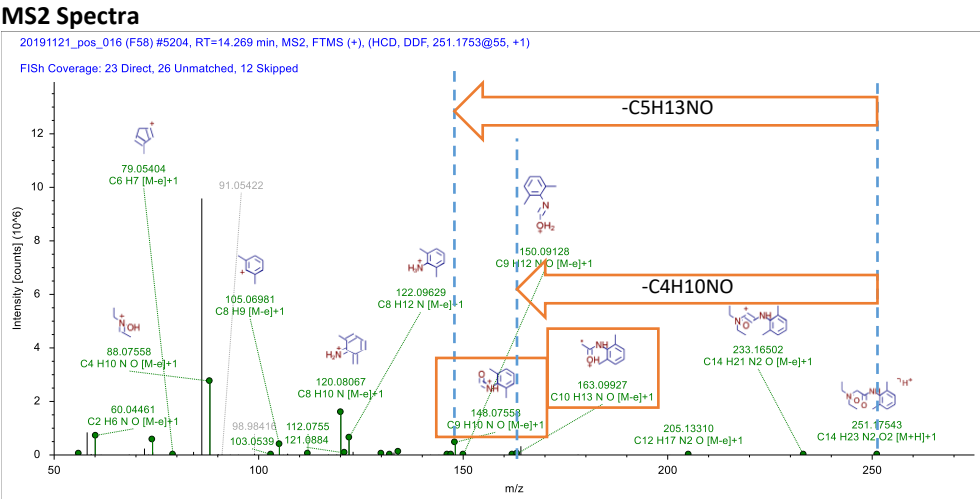
Atomic modification
+O

Proposed Structure



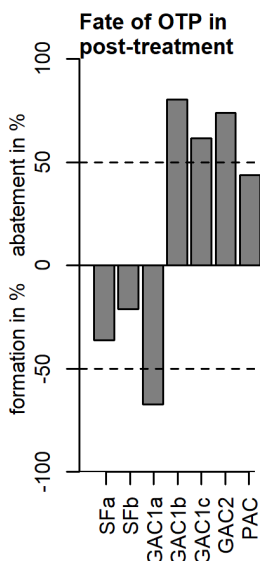
Confidence Level
Level 1

Massbank ID
ET402801



Additional Evidence for Structure Interpretation

Tertiary amines are very reactive to form *N*-oxides during ozonation (von Sonntag & von Gunten 2012). Therefore, the formation of a *N*-oxide at the diethylmethylamine moiety is very likely. The neutral loss of C₅H₁₃NO between the precursor and fragment 148 fits to a cleavage of the diethylmethylamine *N*-oxide moiety. The neutral loss of C₅H₁₀NO between the precursor and fragment 148 fits to a cleavage of the diethylamine *N*-oxide moiety. Merel et al. (2017) identified lidocaine *N*-oxide as major ozone transformation product. The MS² spectrum here matches the reference spectrum we recorded for lidocaine *N*-oxide.

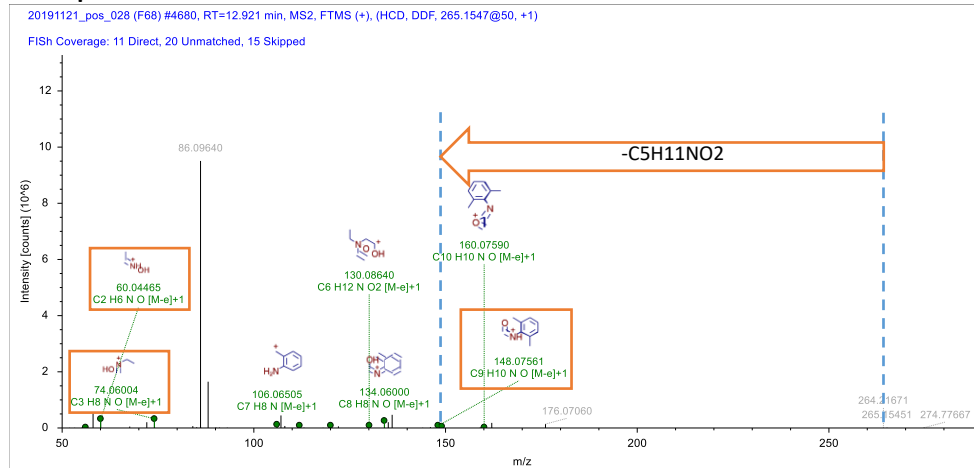


Pos	265.1545	[m+H] ⁺
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Atomic modification
-H₂ +O₂

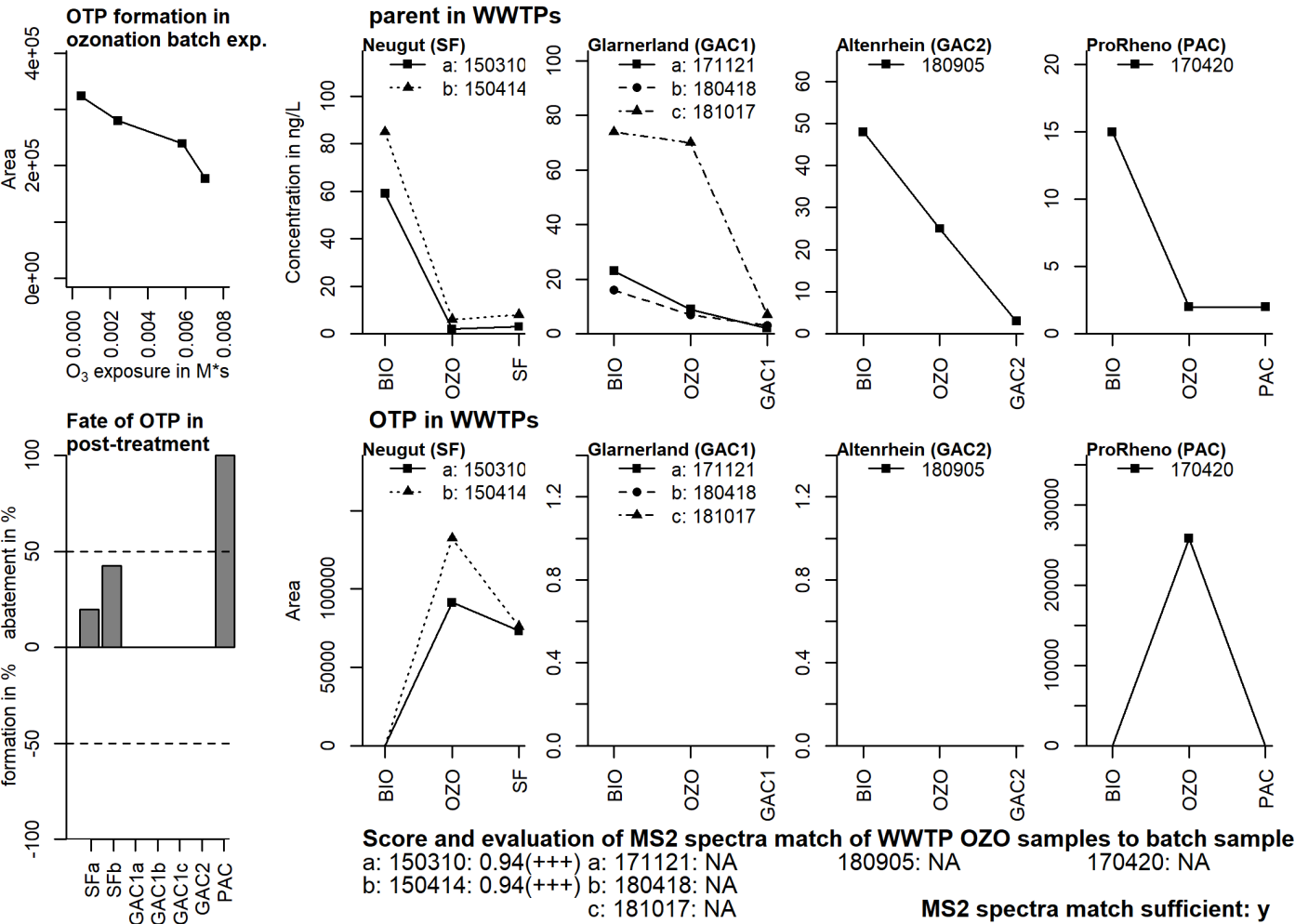
CC1=CC=C(C=C1)NC(=O)CC[N+](C)(C)[O-]

Massbank ID
ET402901



The atomic modification from the elemental formula of the parent compound to this OTP is $-H_2 + O_2$. Since tertiary amines are very reactive to form *N*-oxides during ozonation (von Sonntag & von Gunten 2012), it is probable that the *N*-oxide was formed in this OTP as well, which is supported by fragments 74 and 60. The OTP LID_p_251.1752_13.6 was identified as lidocaine *N*-oxide.

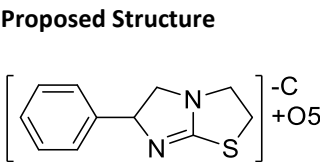
The neutral loss of $C_5H_{11}NO_2$ between precursor and fragment 148 indicates also that the modification $+O_2 -H_2$ occurred at the diethylmethylamine. This modification fits to an *N*-oxidation in combination with a formation of a carbonyl group, likely at a CH_2 of an ethyl moiety. The exact type and position of the modification remain unknown. The structures of the MS^2 fragments are drawn exemplarily.



MS Spectra
Neg 272.0392 [m-H]-

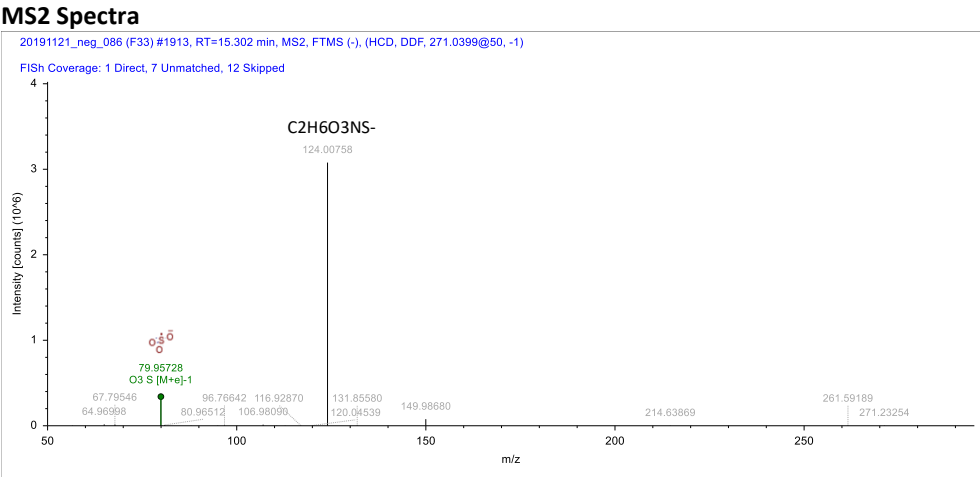
Formula
C10H12O5N2S

Atomic modification
-C +O5



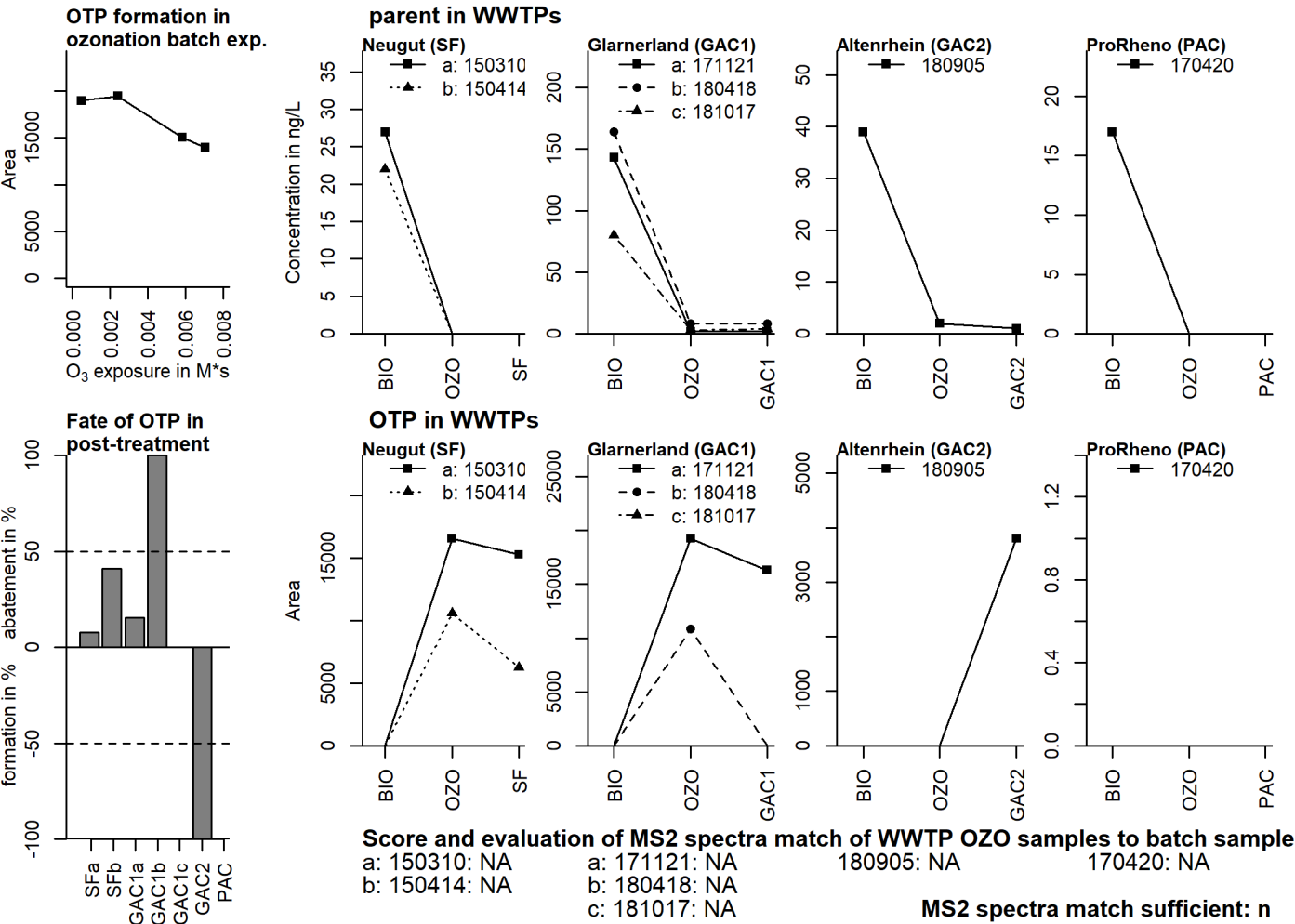
Confidence Level
Level 3

Massbank ID
ET406501



Additional Evidence for Structure Interpretation

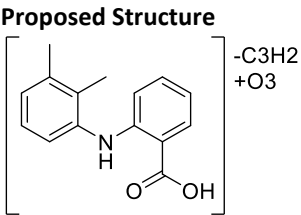
Fragment 79 indicates the formation of a sulfonic acid moiety. The exact type and location of the remaining modification(s) remain unknown.



MS Spectra
Neg 250.0719 [m-H]-

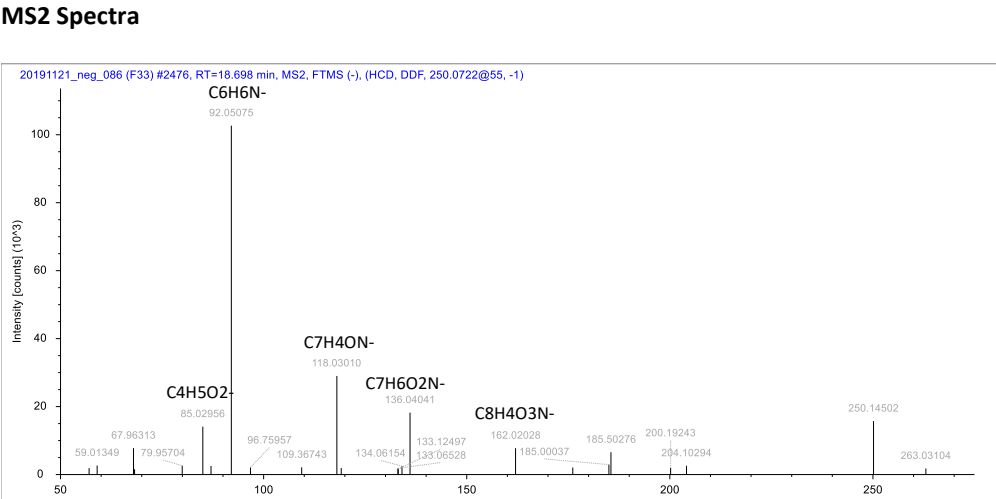
Formula
C₁₂H₁₃O₅N

Atomic modification
-C₃H₂ +O₃



Confidence Level
Level 3

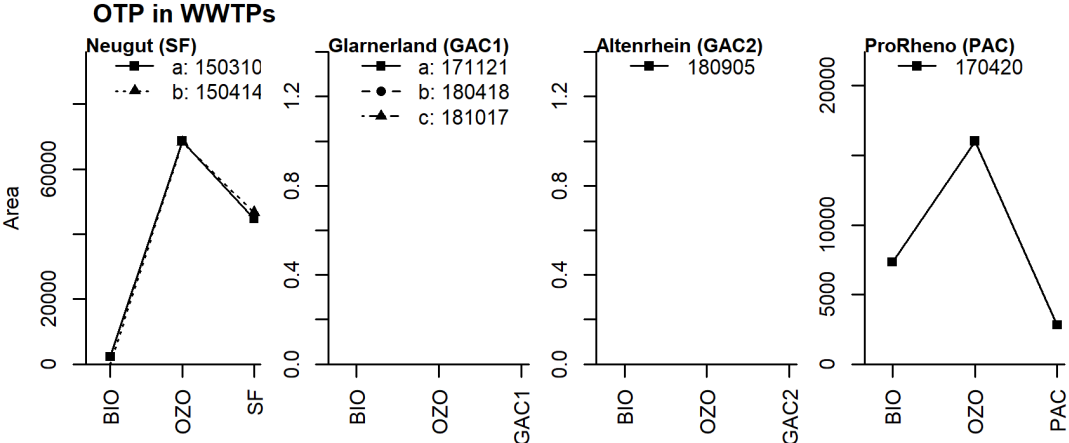
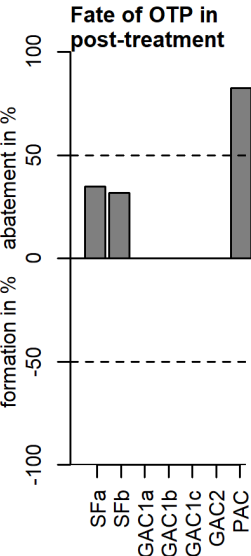
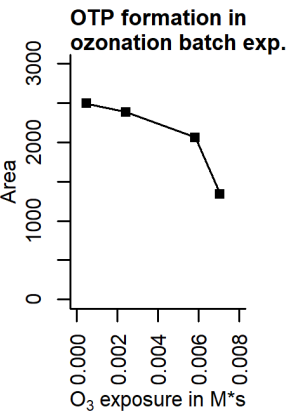
Massbank ID
ET406601



Additional Evidence for Structure Interpretation

The atomic modification from the elemental formula of the parent compound to this OTP is -C₃H₂ +O₃. This OTP was likely formed by opening of an aromatic ring, which is known to happen during ozonation (von Sonntag & von Gunten 2012). Fragment 92 with formula C₆H₆N indicates that one of the two rings was not cleaved. The exact type and location of the modifications remain unknown.

Name nk_n_293.1148_16.2



Score and evaluation of MS2 spectra match of WWTP OZO samples to batch sample

Sample	Score	Evaluation
a: 150310	0.94	+++
a: 171121	NA	
a: 180905	NA	
a: 170420	0.96	+++
b: 150414	0.94	+++
b: 180418	0.92	+++
c: 181017	NA	

MS2 spectra match sufficient: y

MS Spectra

Neg 293.1148 ?

Formula

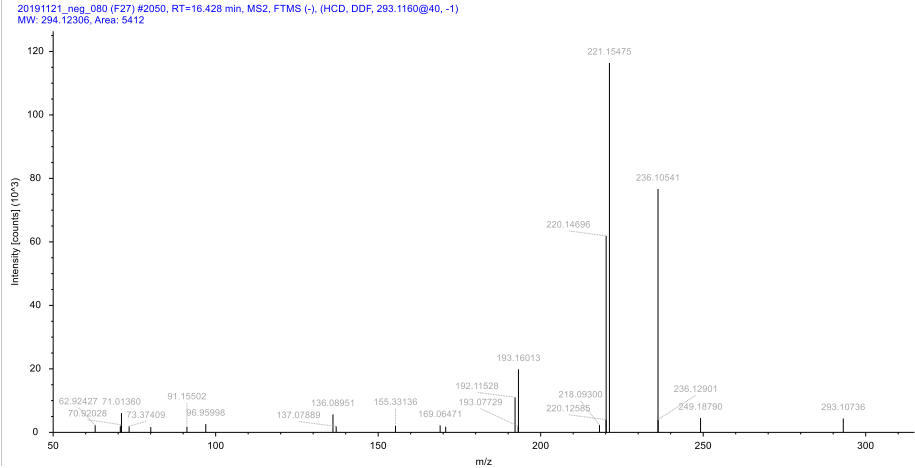
?

Atomic modification

?

Proposed Structure

MS2 Spectra



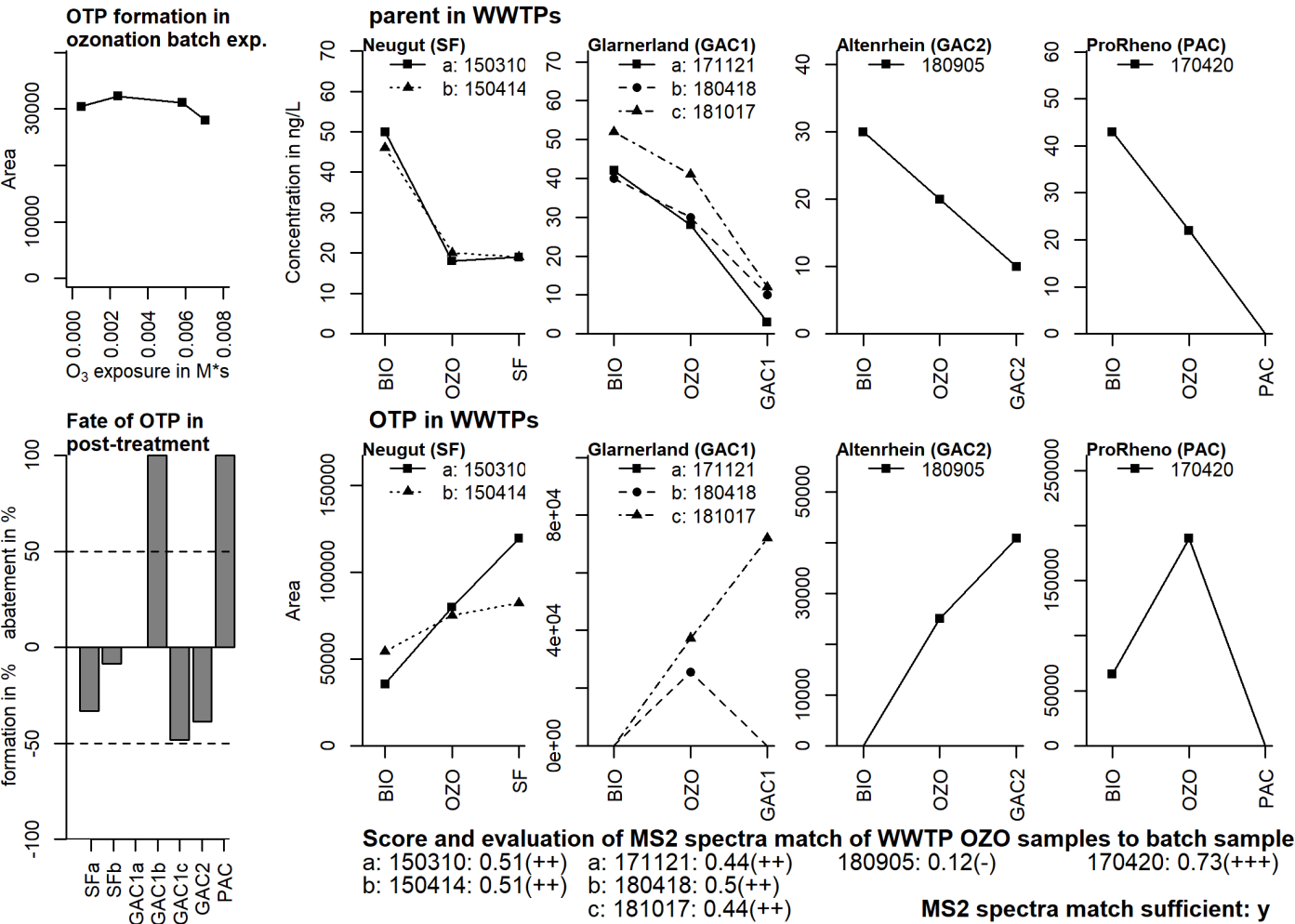
Additional Evidence for Structure Interpretation

Confidence Level

Level 5

Massbank ID

ET406901

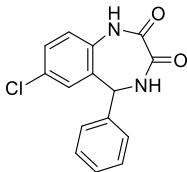


MS Spectra
Pos 285.0423 [m+H]⁺

Formula
C₁₅H₉O₂N₂Cl

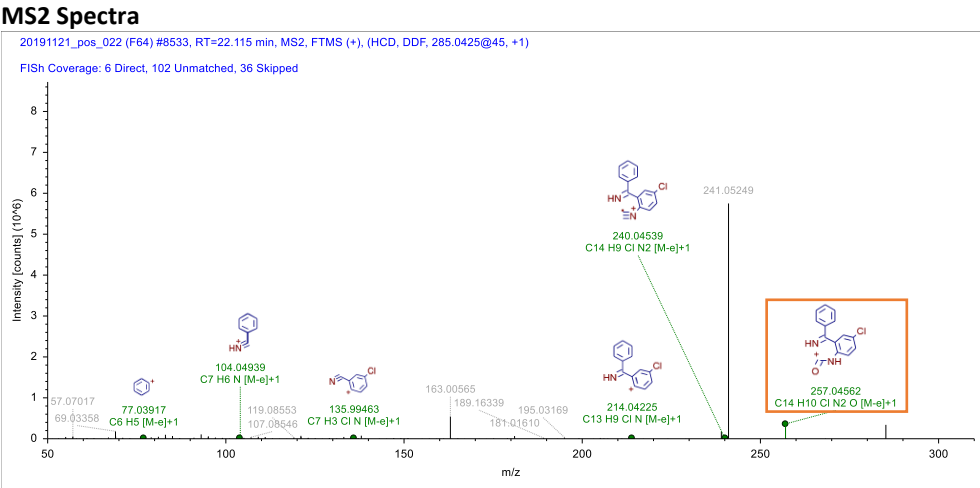
Atomic modification
-H₂

Proposed Structure



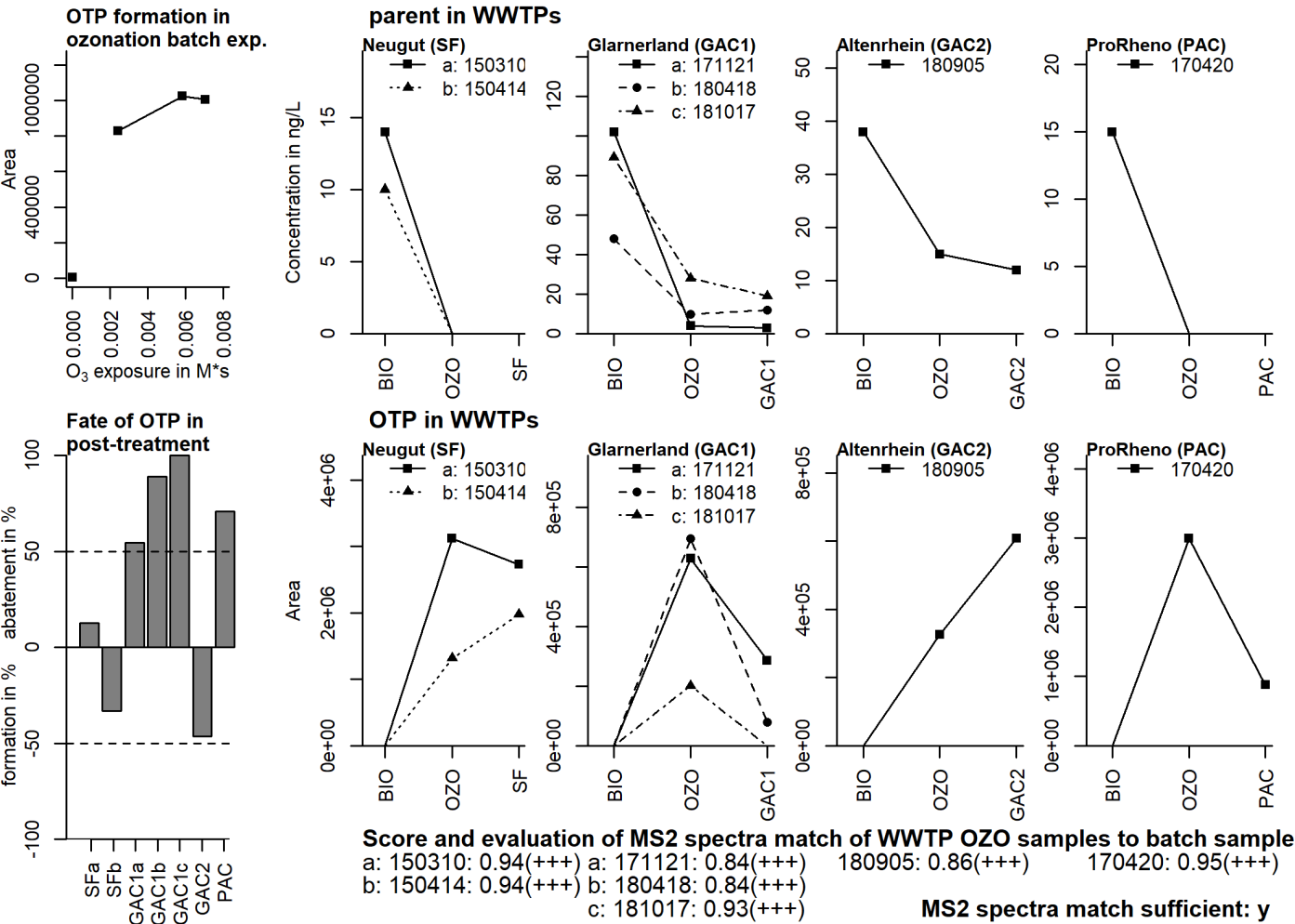
Confidence Level
Level 3

Massbank ID
ET403001



Additional Evidence for Structure Interpretation

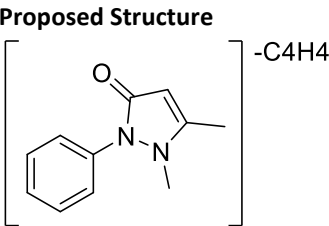
The MS² fragments at the nominal masses 257, 214 and 104 were also observed for the parent compound. They indicate that the structure drawn for the fragment 257 is a substructure of this OTP. This OTP probably was formed by oxidation of the C-OH bond to a carbonyl group.



MS Spectra
Pos 137.0708 [m+H]⁺

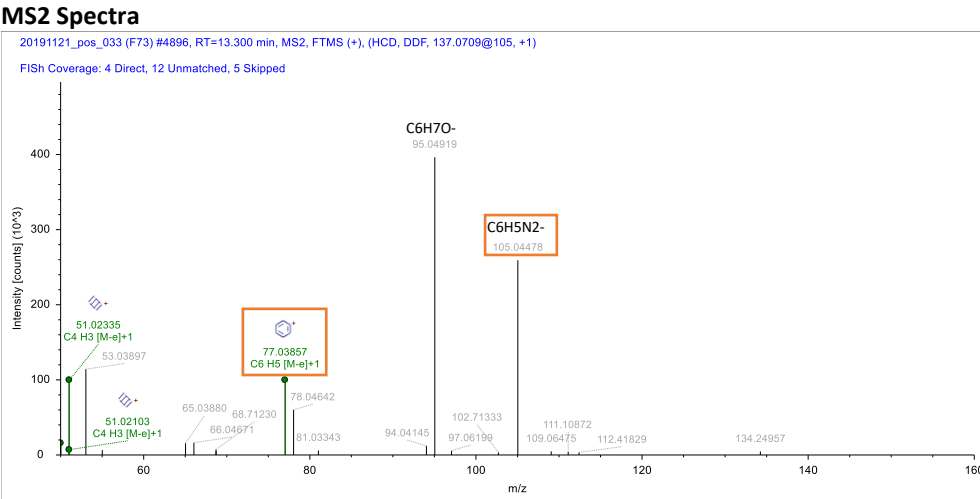
Formula
C₇H₈ON₂

Atomic modification
-C₄H₄



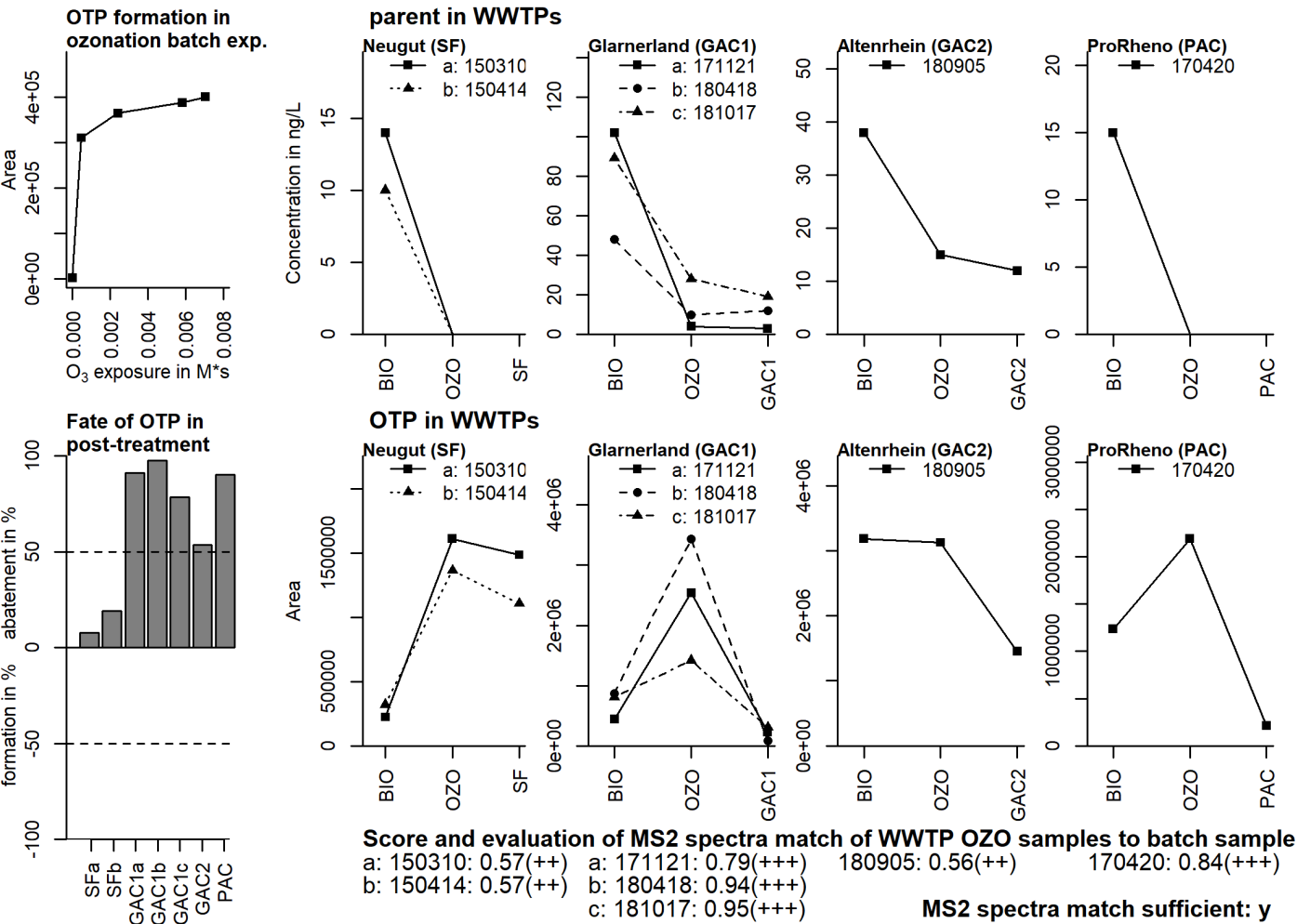
Confidence Level
Level 3

Massbank ID
ET403101



Additional Evidence for Structure Interpretation

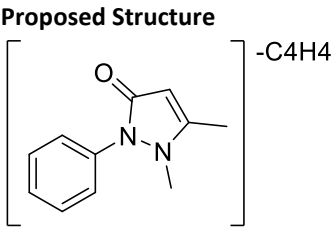
For the parent compound phenazone two OTPs (PHE_p_137.0709_14.1 and PHE_p_137.0709_12.9) with the exact mass of 137.0708 were observed at different retention times (14.1 and 12.9 min). The MS² spectra of these OTPs are very similar. The MS² fragment at the nominal masses 77 was also observed for the parent compound. It indicates that a phenyl moiety is a substructure of this TP. Fragment 92 and 105 fit to the formula C₆H₆N and C₆H₅N respectively. Both TPs were probably formed after a rearrangement, resulting in the formation of two isomers. The exact type of the modification remains unknown.



MS Spectra
Pos 137.0708 [m+H]⁺

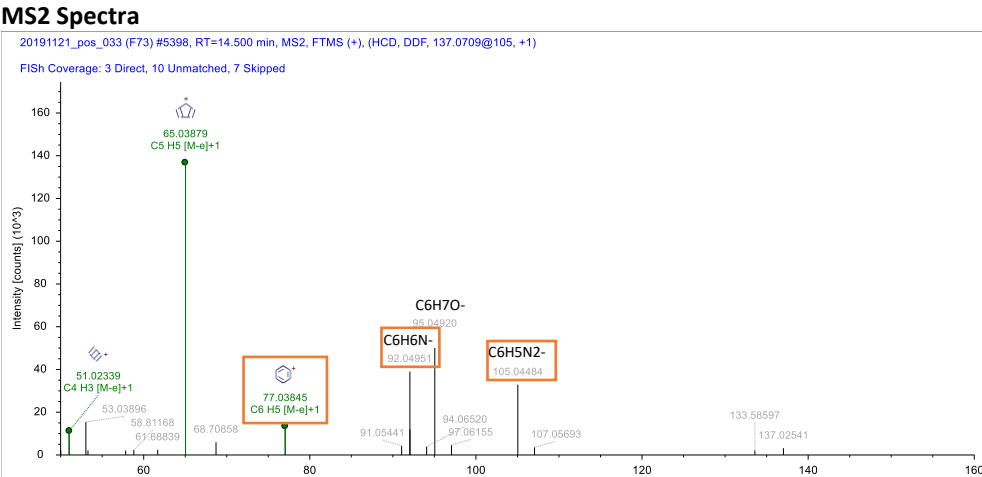
Formula
C₇H₈ON₂

Atomic modification
-C₄H₄



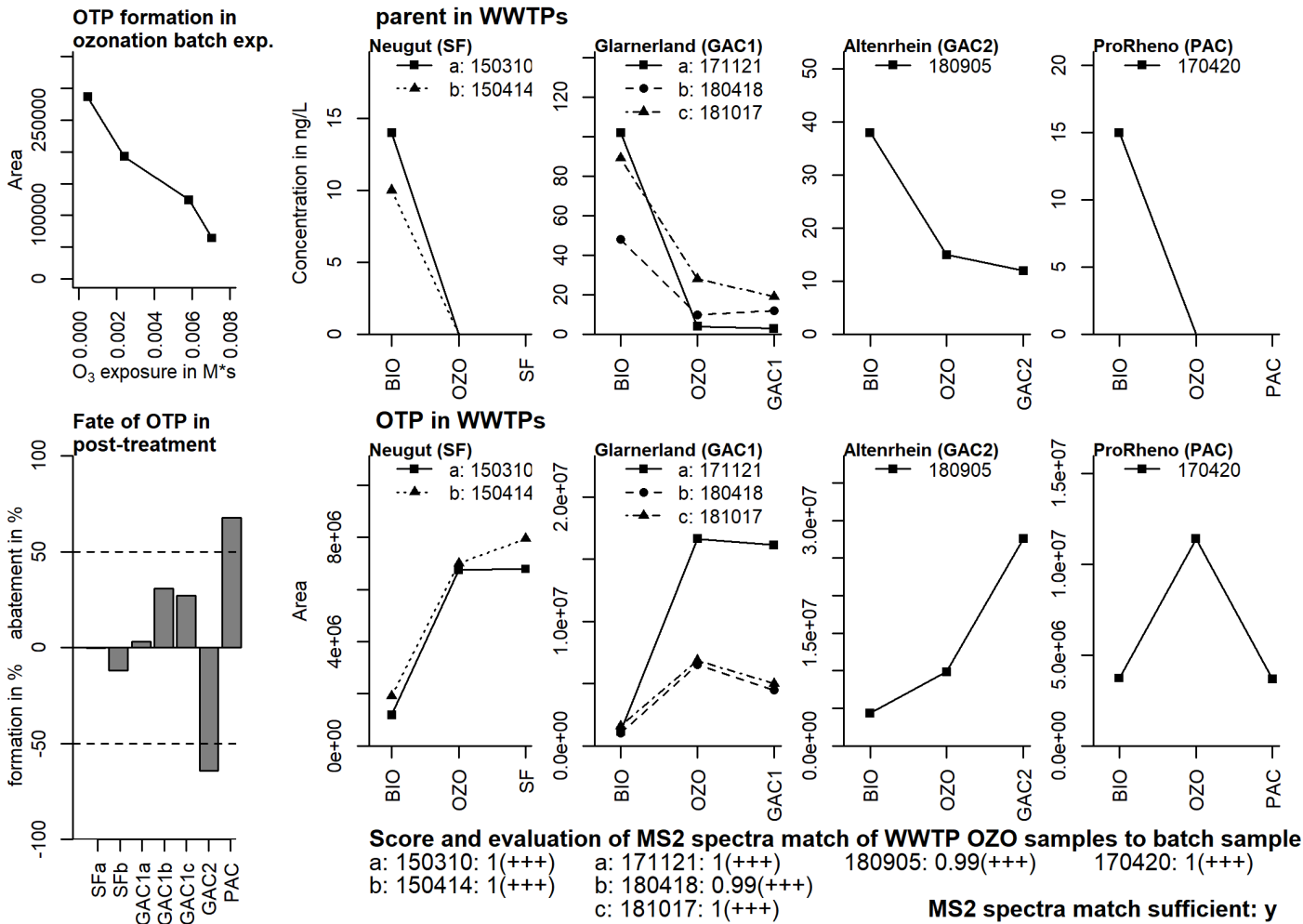
Confidence Level
Level 3

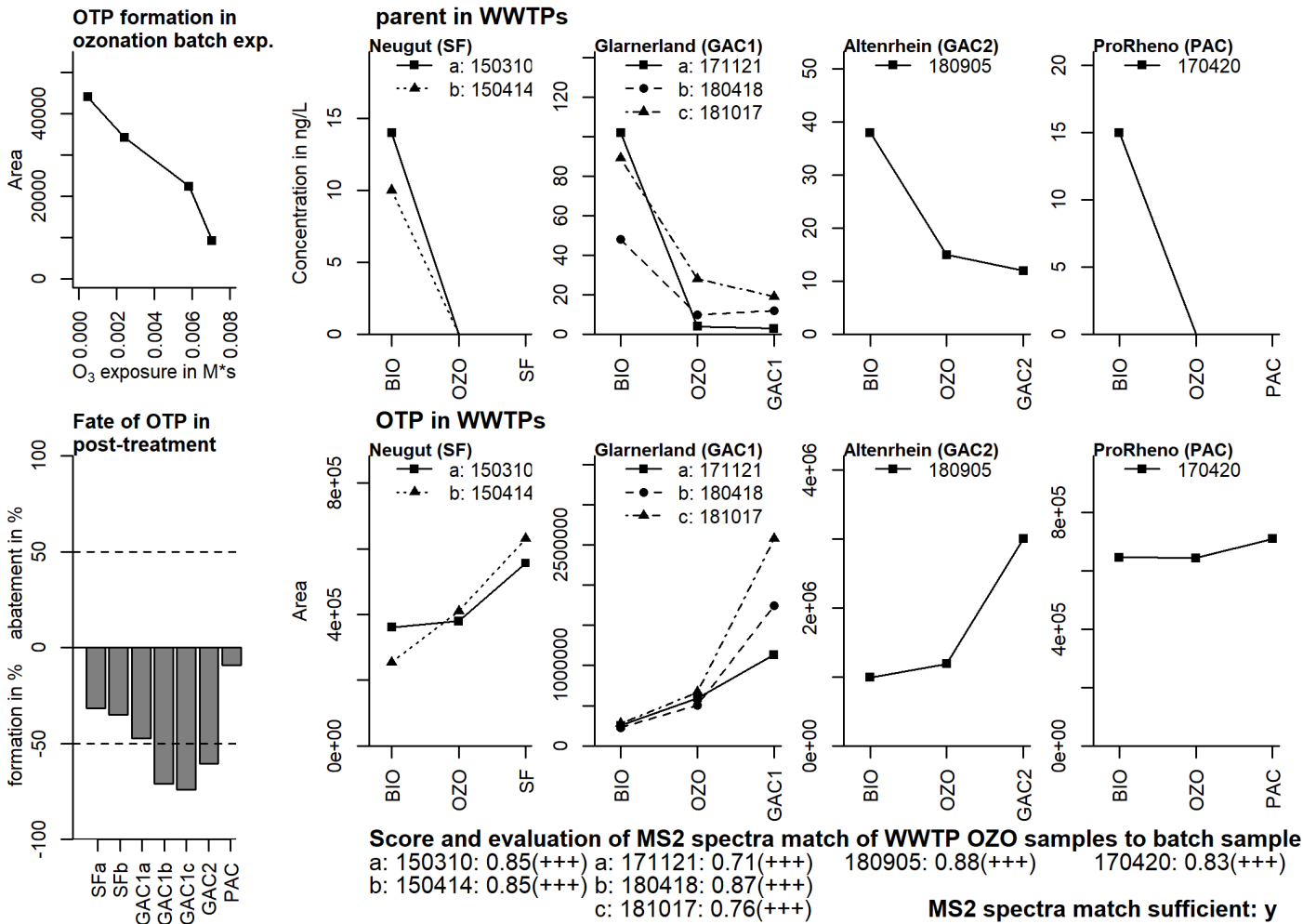
Massbank ID
ET403201



Additional Evidence for Structure Interpretation

For the parent compound phenazone two OTPs (PHE_p_137.0709_14.1 and PHE_p_137.0709_12.9) with the exact mass of 137.0708 were observed at different retention times (14.1 and 12.9 min). The MS² spectra of these OTPs are very similar. The MS² fragment at the nominal masses 77 was also observed for the parent compound. It indicates that a phenyl moiety is a substructure of this TP. Fragment 92 and 105 fit to the formula C₆H₆N and C₆H₅N respectively. Both TPs were probably formed after a rearrangement, resulting in the formation of two isomers. The exact type of the modification remains unknown.





MS Spectra

Pos 237.0868 [m+H]⁺
Neg 235.0722 [m-H]⁻

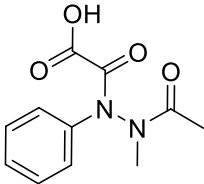
Formula

C₁₁H₁₂O₄N₂

Atomic modification

+O₃

Proposed Structure



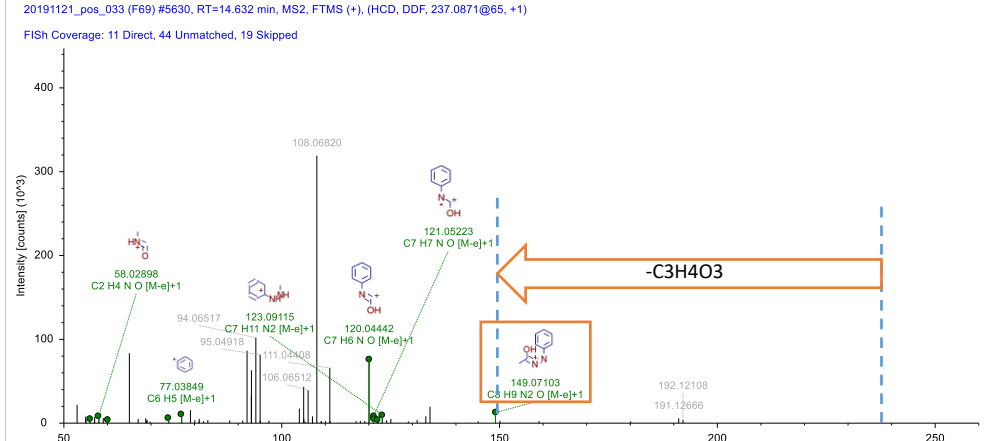
Confidence Level

Level 3

Massbank ID

ET403401

MS2 Spectra



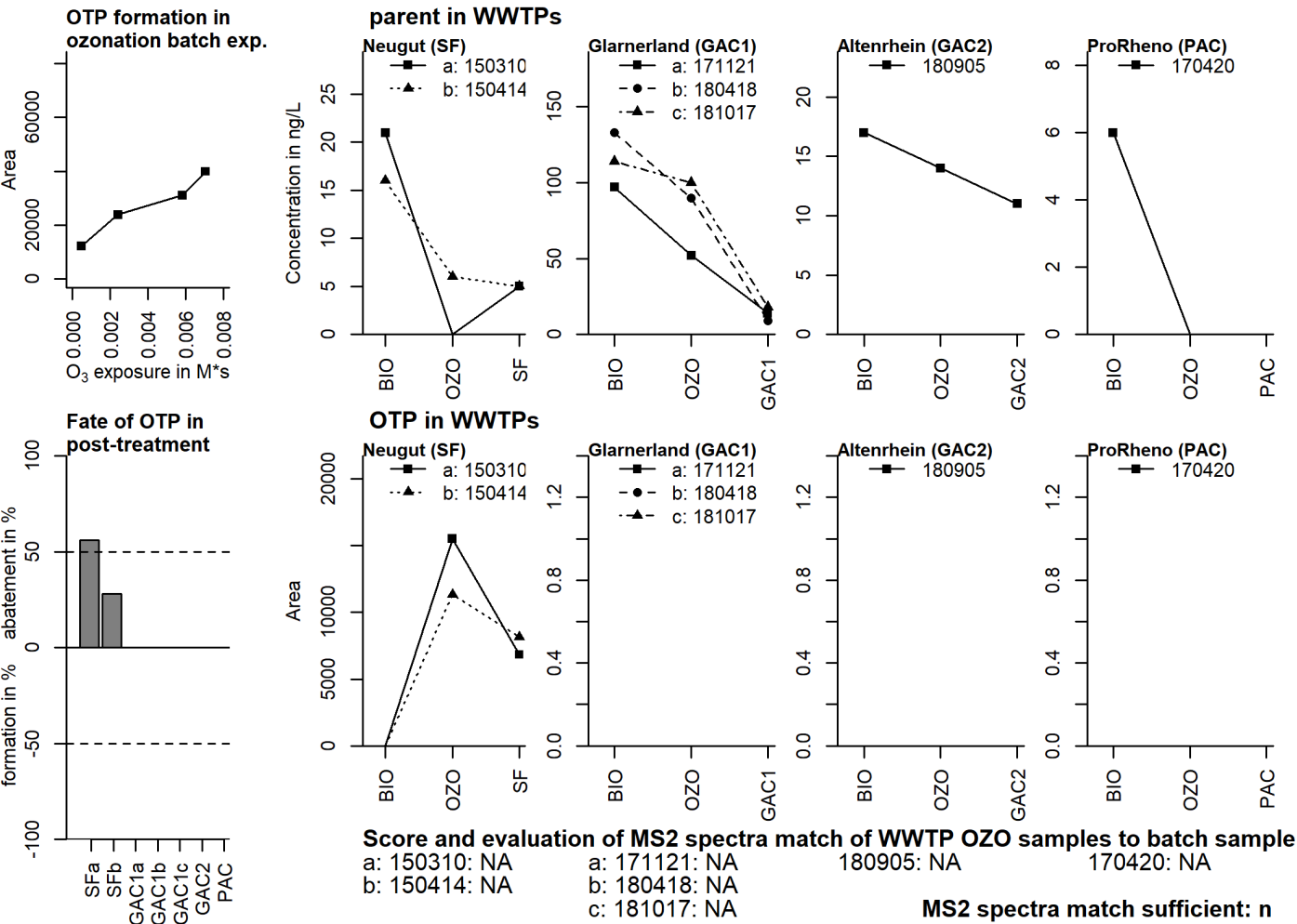
Additional Evidence for Structure Interpretation

The MS² fragments at the nominal masses 149 and 77 were also observed for the parent compound. They indicate that the structure drawn for the fragment 149 is a substructure of this OTP.

The neutral loss of C₃H₄O₃ between the precursor and fragment 149 fits to the cleavage of a methyl group and an oxoacetic like structure.

This structure has been suggested as OTP of phenazone by Miao et al. (2015).

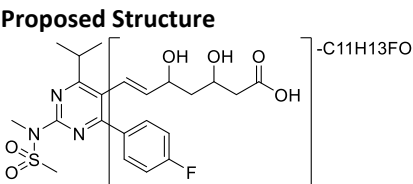
The MS² fragment at the nominal mass 144 was also observed for the parent compound. It indicates that the drawn structure of the MS² fragment 144 is a substructure of this OTP and that the modification likely did not take place at the aromatic moiety. Ozonolysis of alkynes are known to give diketone products (Bailey 1982). Furthermore, the neutral loss of CO₂ between the proposed structure and fragment 148 indicates the presence of a carboxylic acid moiety. Therefore, the formation of the drawn structure is likely.



MS Spectra
Neg 300.0656 [m-H]-

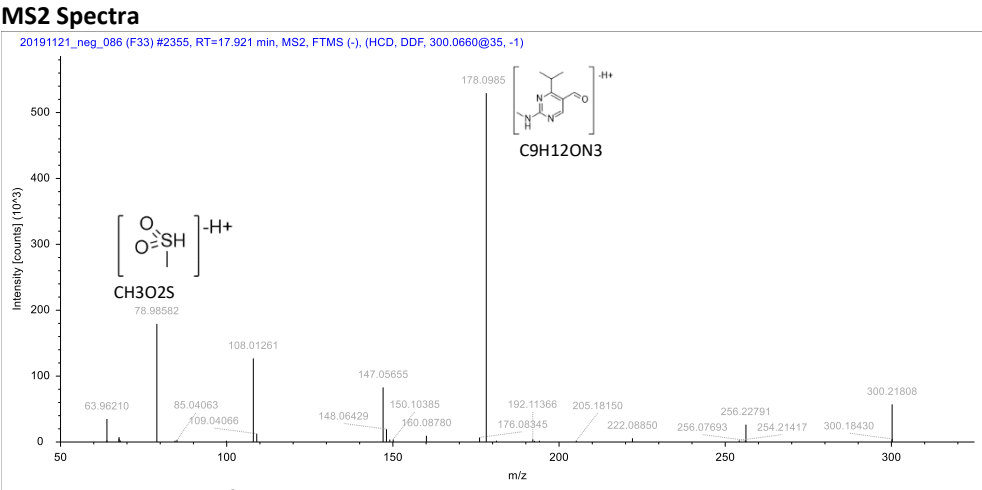
Formula
C11H15O5N3S

Atomic modification
-C11H13FO



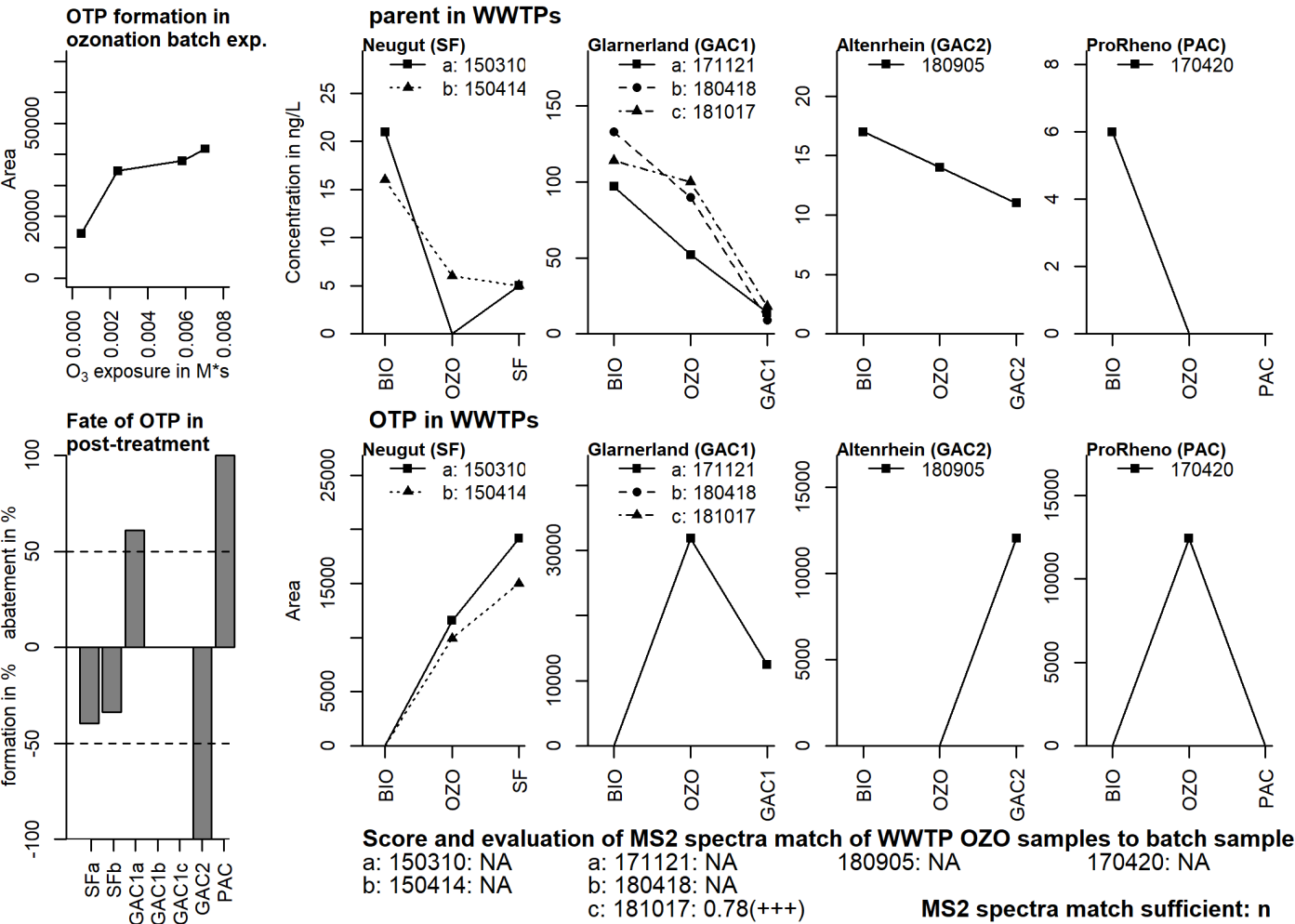
Confidence Level
Level 3

Massbank ID
ET407201



Additional Evidence for Structure Interpretation

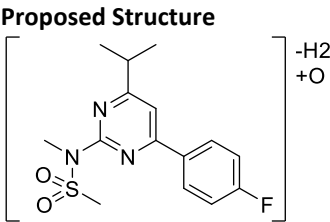
The atomic modification from the elemental formula of the parent compound to this OTP is -C11H13FO. This fits to cleavages and the addition of oxygen atoms within the dihydroxyheptenoic acid moiety and the fluorophenyl moiety, likely at the C-C double bond since ozone is known to react with olefins (von Sonntag & von Gunten 2012). The exact type and location of the modification(s) remain unknown. The structures of the MS² fragments are drawn exemplarily.



MS Spectra
Neg 336.0821 [m-H]-

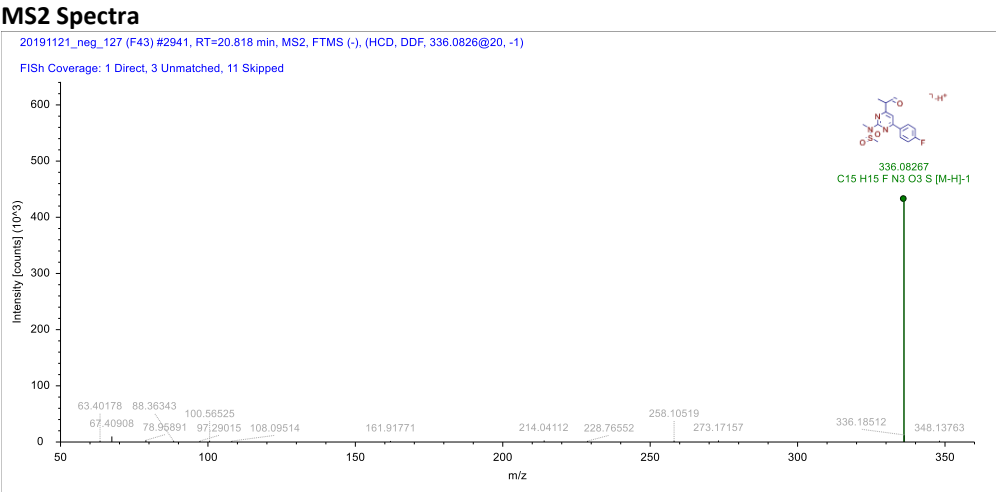
Formula
C15H16O3N3FS

Atomic modification
-C7H12O3



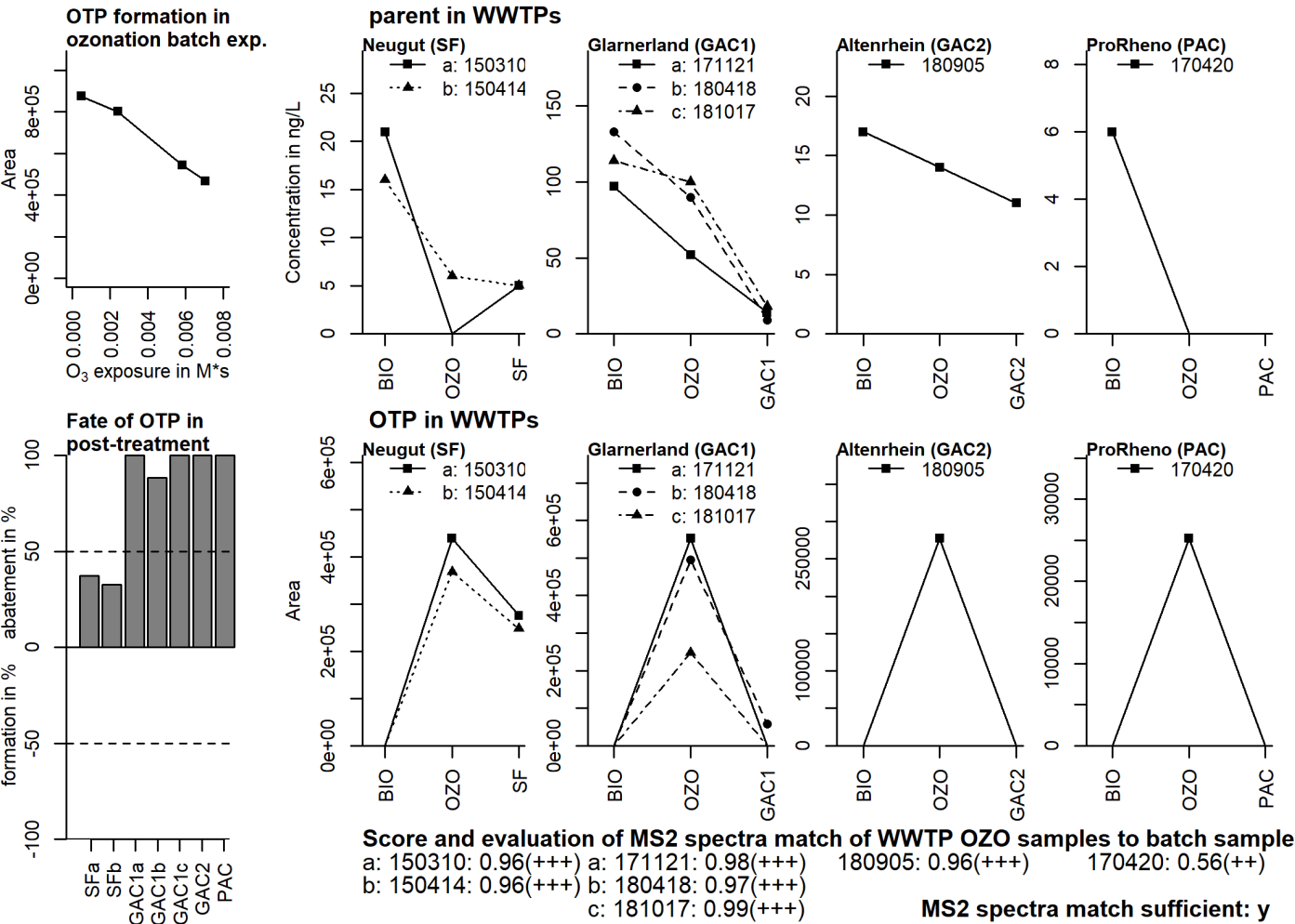
Confidence Level
Level 3

Massbank ID
ET407301



Additional Evidence for Structure Interpretation

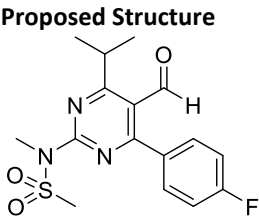
The atomic modification from the elemental formula of the parent compound to this OTP is —C7H12O3. This fits the cleavage of the dihydroxyheptenoic acid moiety and likely the formation of a carbonyl group. The exact type and location of the modification(s) remain unknown. The structures of the MS² fragments are drawn exemplarily.



MS Spectra
Pos 352.11236 [m+H]⁺

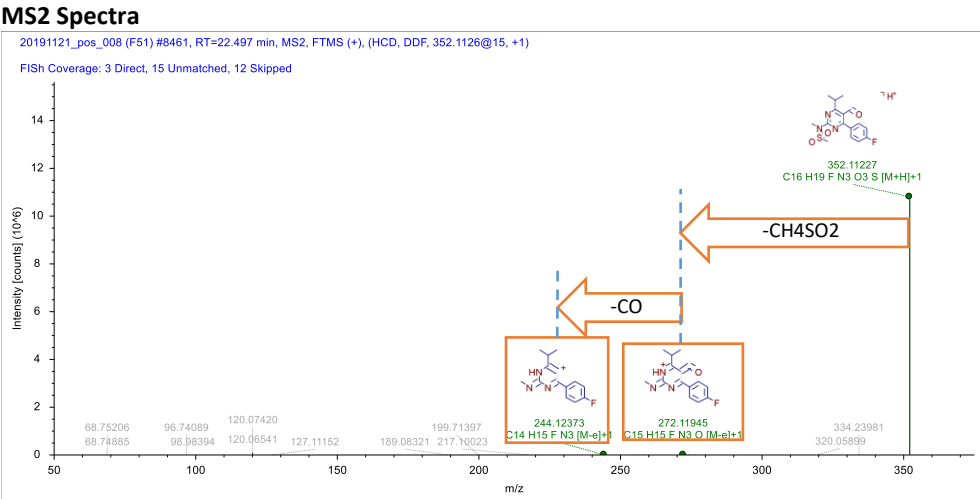
Formula
C₁₆H₁₈O₃N₃FS

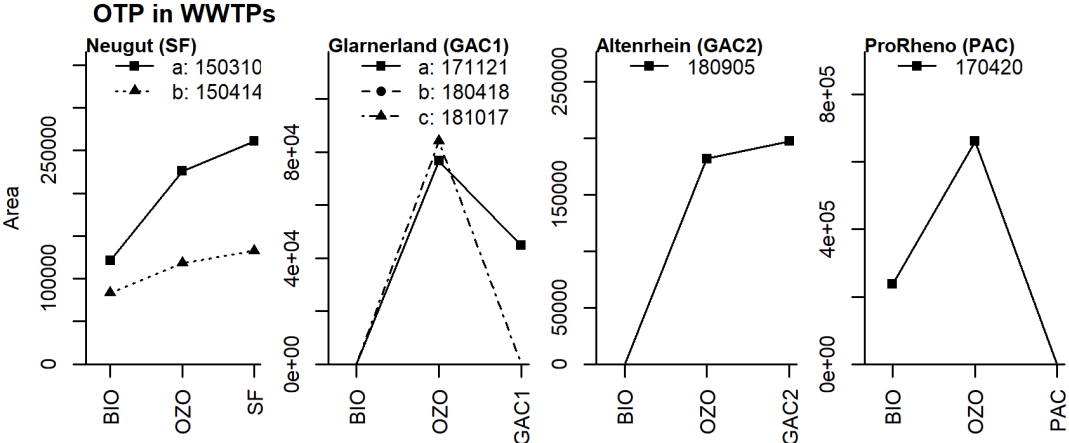
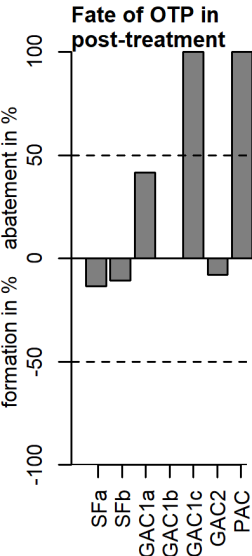
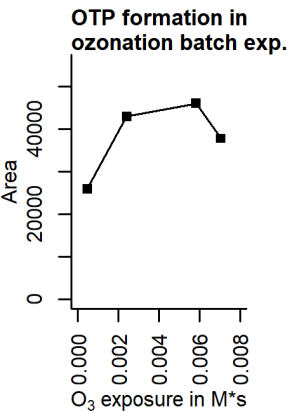
Atomic modification
-C₆H₁₀O₃



Confidence Level
Level 3

Massbank ID
ET403501





Score and evaluation of MS2 spectra match of WWTP OZO samples to batch sample

Sample	Score	Evaluation
a: 150310	0.56	(++)
b: 150414	0.56	(++)
a: 171121	0.46	(++)
b: 180418	0.74	(+++)
c: 181017	0.19	(-)
180905	0.72	(+++)
170420	0.32	(+)

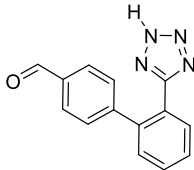
MS2 spectra match sufficient: y

MS Spectra
Pos 251.0924 [m+H]⁺

Formula
C₁₄H₁₀ON₄

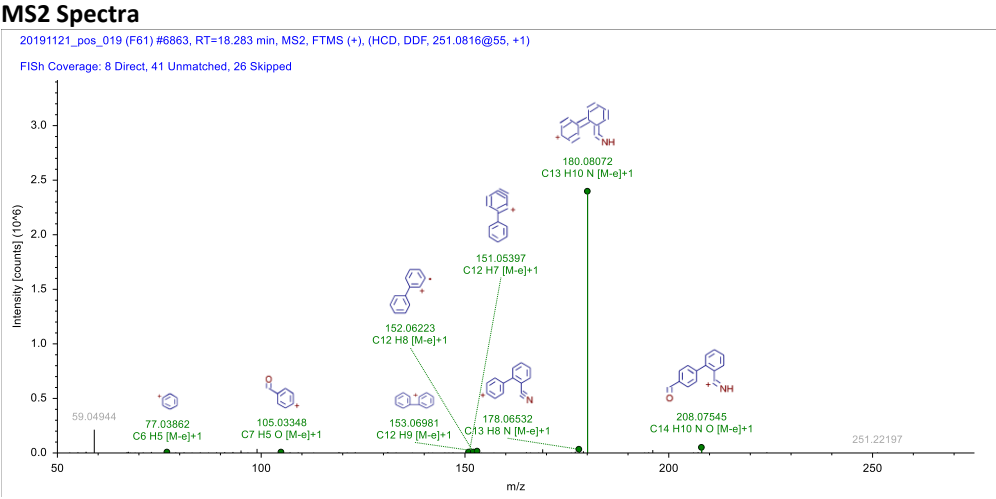
Atomic modification
Multiple parent substances

Proposed Structure



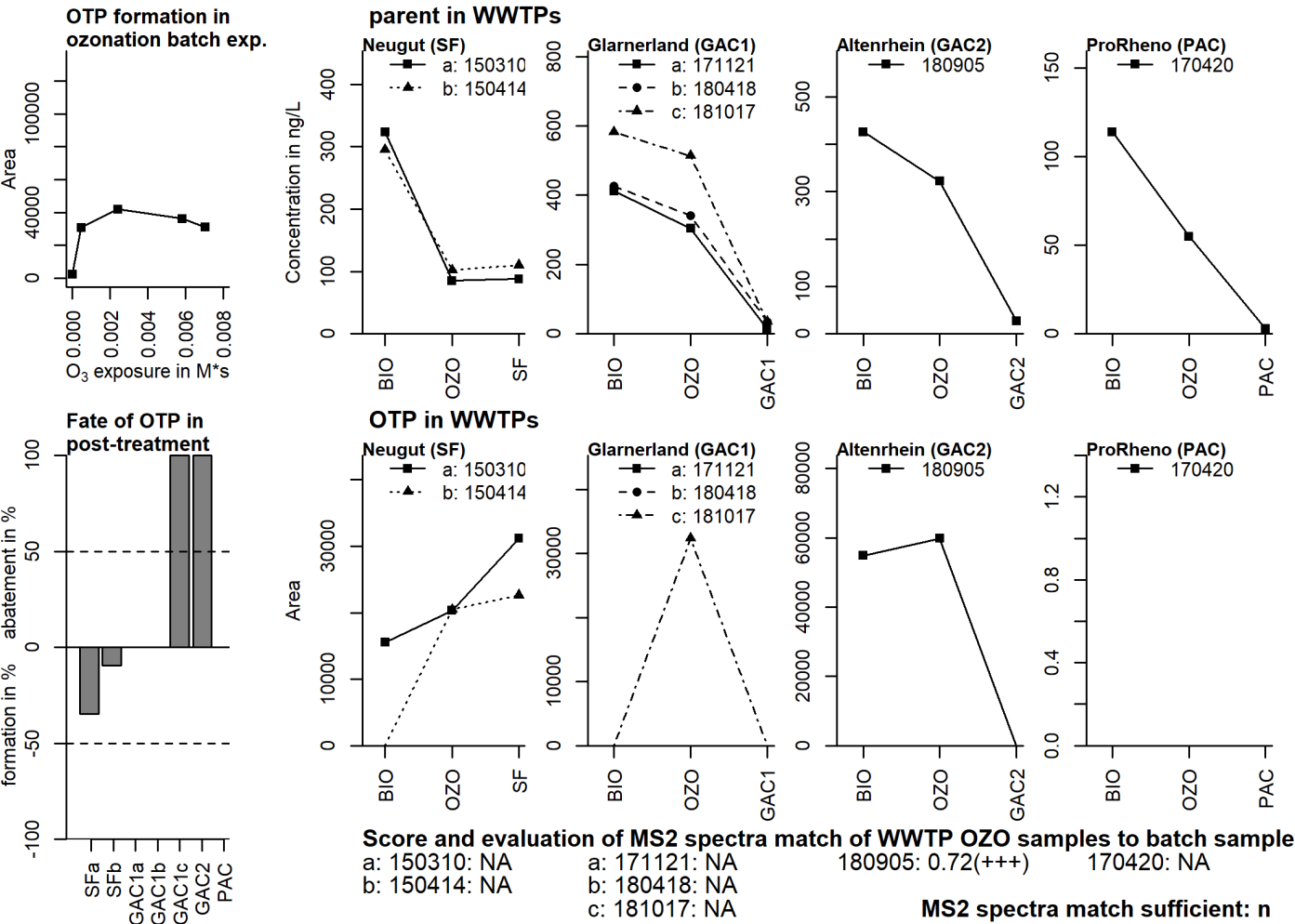
Confidence Level
Level 3

Massbank ID
ET403601



Additional Evidence for Structure Interpretation

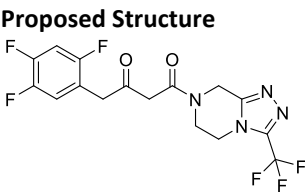
The chemical formula for this TP is C₁₄H₁₀ON₄ and fits to the tetrazolebiphenylmethyl moiety, which is common for all sartans and to which an oxygen atom was added. Diehle et al. (2019) suggested this OTP as a probable OTP for valsartan (Level 2b). The observed evidence from our MS spectrum and the MS² spectrum match the one from Diehle et al. (2019).



MS Spectra
Neg 405.0789 [m-H]-

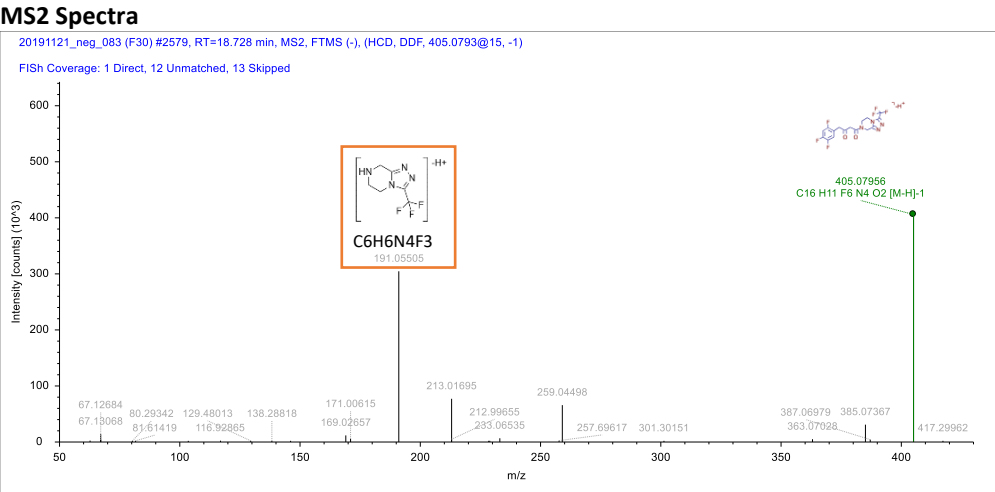
Formula
C16H12O2N4F6

Atomic modification
-NH3 +O



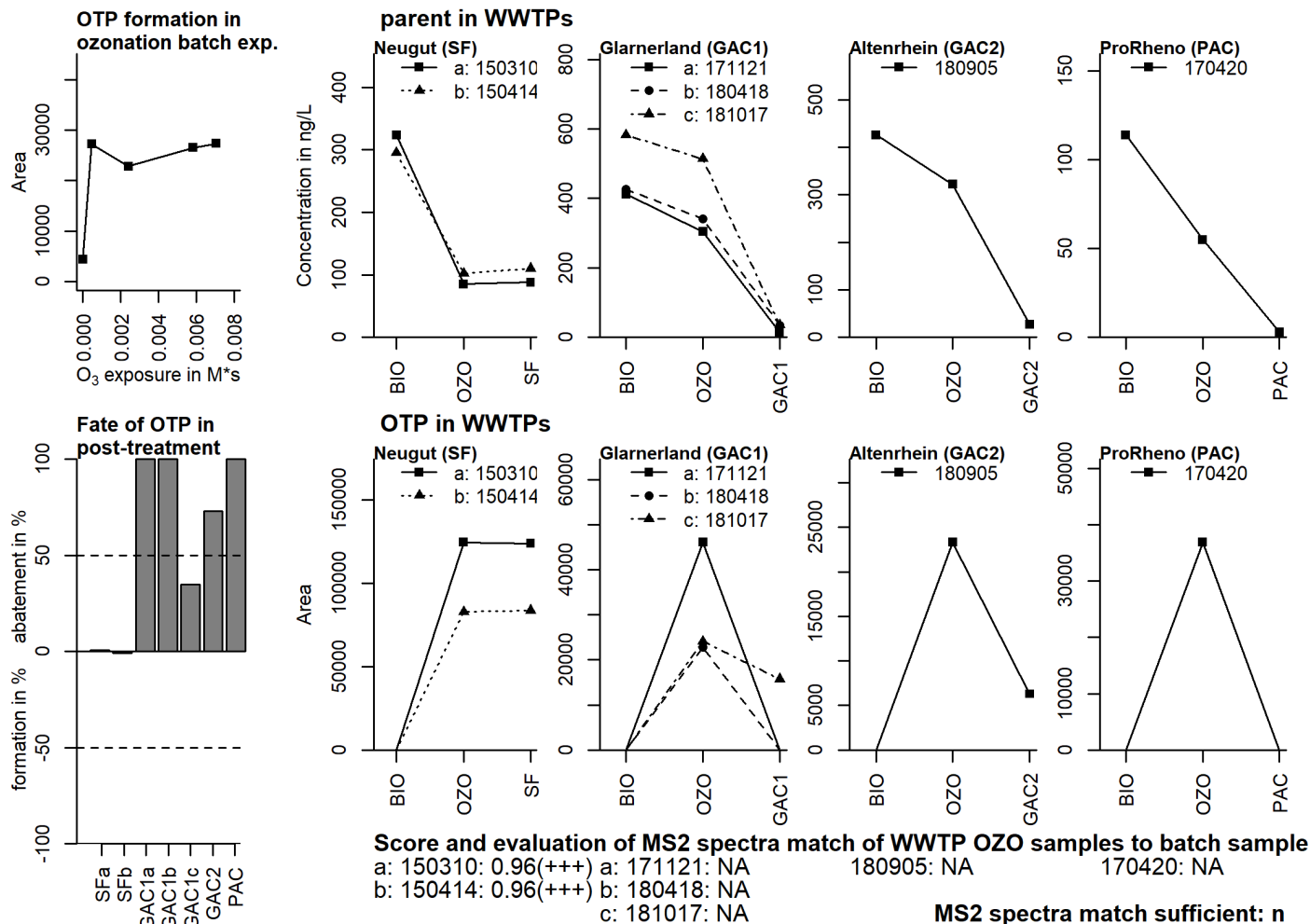
Confidence Level
Level 2a

Massbank ID
ET407401



Additional Evidence for Structure Interpretation

The MS² fragment at the nominal mass 191 was also observed for the parent compound. It indicates that the drawn structure of the MS² fragment 191 is a substructure of this OTP and that the modification likely did not take place at the aromatic moiety. The atomic modification from the elemental formula of the parent compound to this OTP fits to the cleavage of the primary amine moiety and the formation of a carbonyl moiety. The drawn structure was confirmed (level 1) by Hermes et al. (2020) as OTP of sitagliptin and the evidence from our MS and MS² spectrum match the one from Hermes et al. (2020).



MS Spectra

Neg 420.0889 [m-H]-
Neg 466.0952 m+FA-H
Pos 422.1045 [m+H]+

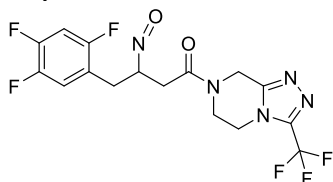
Formula

$C_{16}H_{13}O_2N_5F_6$

Atomic modification

-H2 +O

Proposed Structure



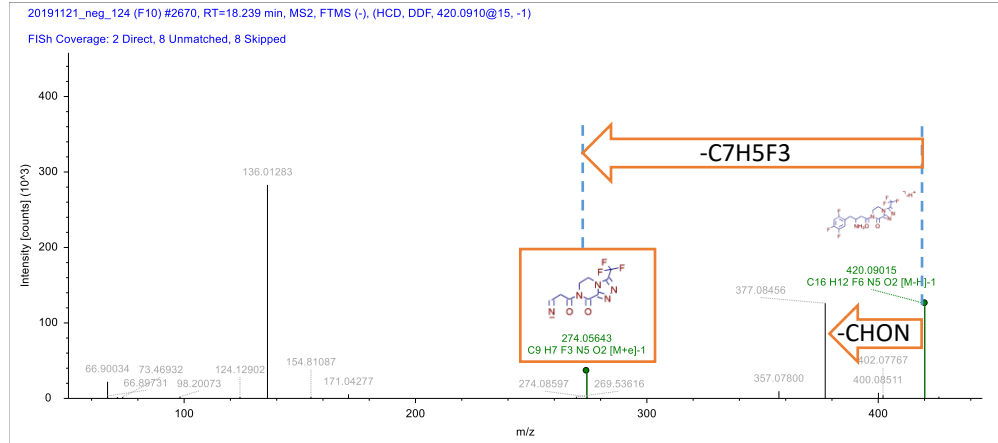
Confidence Level

Level 3

Massbank ID

ET407501

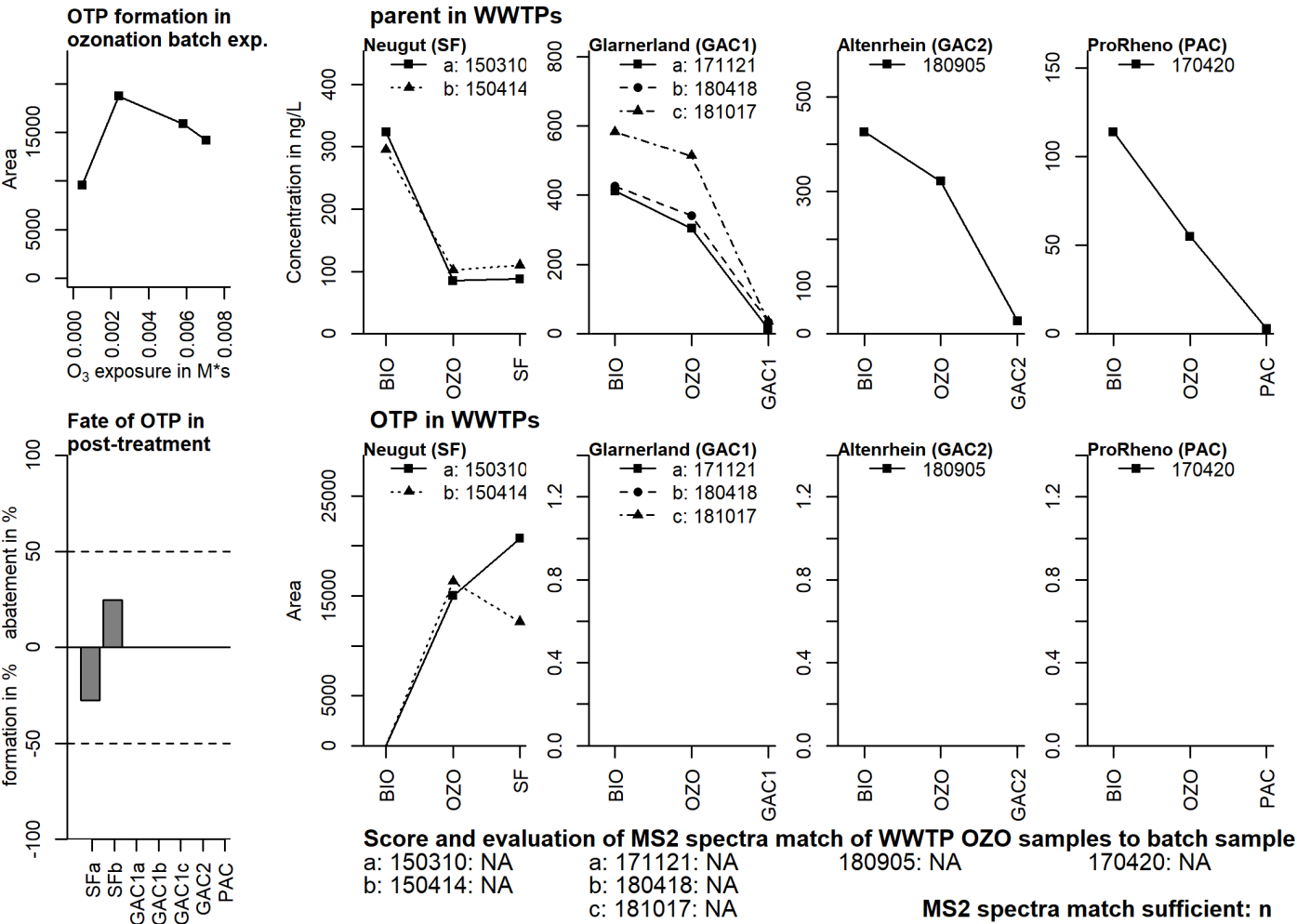
MS2 Spectra



Additional Evidence for Structure Interpretation

The atomic modification from the elemental formula of the parent compound to this OTP is -H2 +O. The neutral loss of C7H5F3 between the molecular ion and fragment 274 indicates that the modification did not take place at the methyltrifluorophenyl moiety.

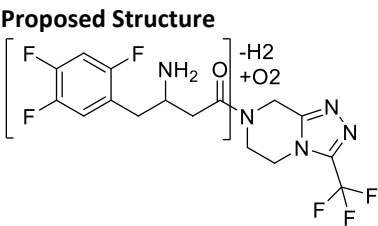
This atomic modification fits to the formation the nitroso group, a reaction that is known to happen during ozonation (Lim et al. 2019). Hermes et al. (2020) detected three OTPs of sitagliptin with this exact same mass and also suggested the formation of nitroso-sitagliptin, which tautomerizes with the oxime form.



MS Spectra		
Neg	436.0847	[m-H]-
Neg	482.0847	[m+FA-H]-
Pos	438.0995	[m+H+]

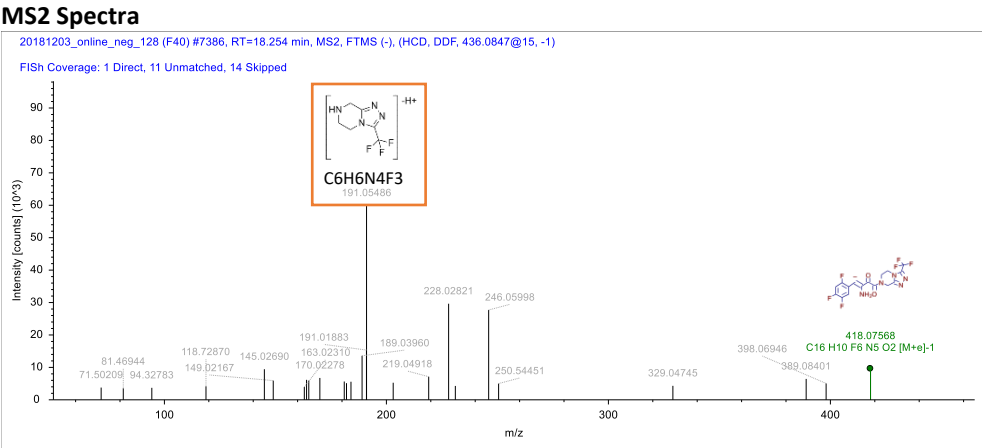
Formula
C16H13O3N5F6

Atomic modification
-H2 +O2



Confidence Level
Level 3

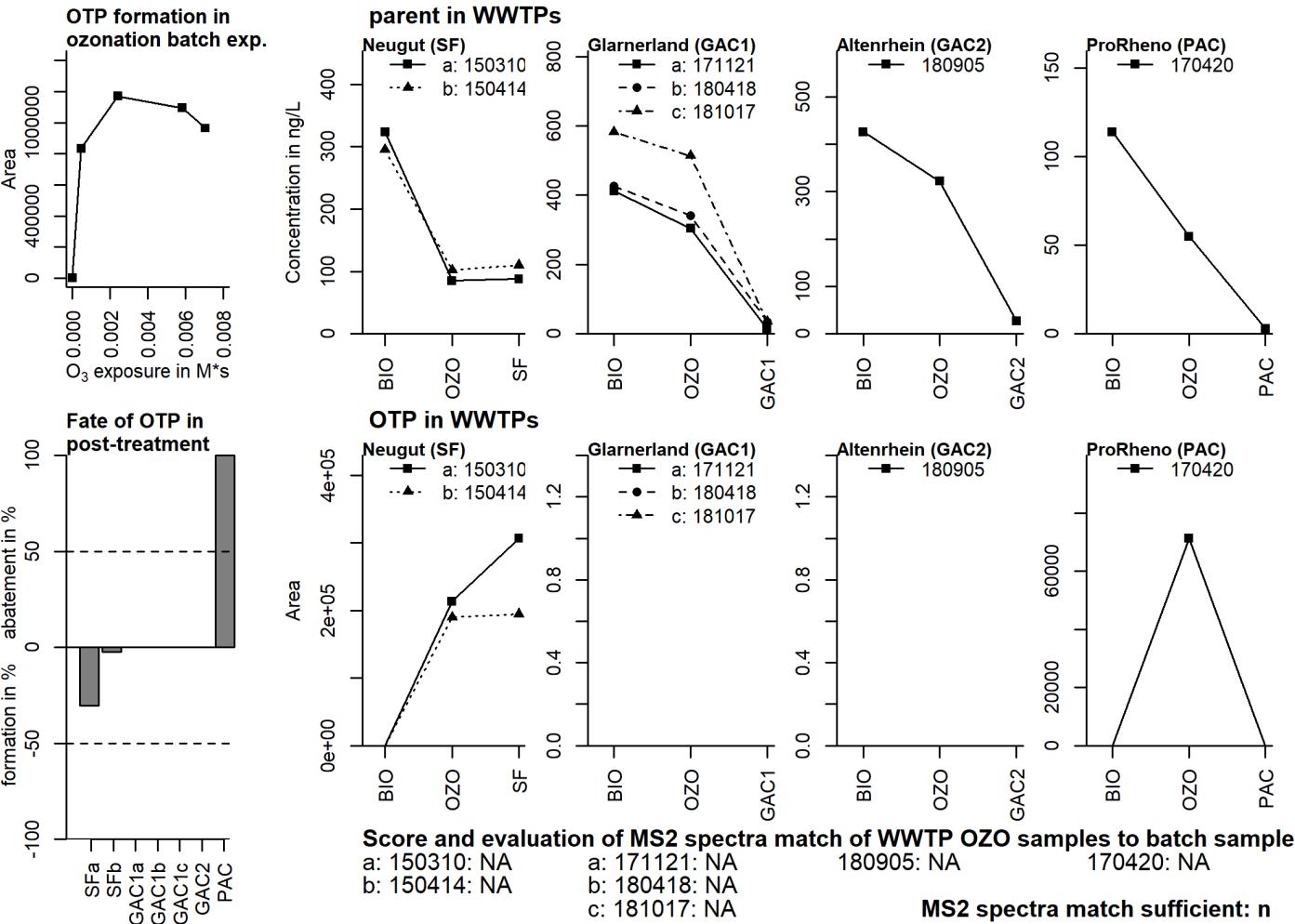
Massbank ID
ET407601



Additional Evidence for Structure Interpretation

For the parent compound sitagliptin two OTPs (SIT_n_436.0847_18.3 and SIT_n_436.0847_19.2) with the exact mass of 436.0847 were observed at different retention times (18.3 and 19.2 min). The MS² spectra of these OTPs are very similar. The atomic modification from the elemental formula of the parent compound to this OTP is -H₂ +O₂. The MS² fragment at the nominal mass 191 was also observed for the parent compound and corresponds to the trifluoromethyltriazolopyrazine moiety, which indicates that no modification occurred at this moiety.

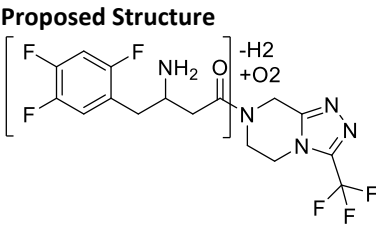
This atomic modification fits to the formation of e.g. a nitro group at the primary amine, a carbonyl group and the addition of an OH group or the formation of a hydroxylamine and a carbonyl moiety. The exact type and location of the modification(s) remain unknown. The structures of the MS² fragments are drawn exemplarily. Hermes et al. (2020) suggested the nitro structure as OTP of sitagliptin.



MS Spectra
Neg 436.0847 [m-H]-

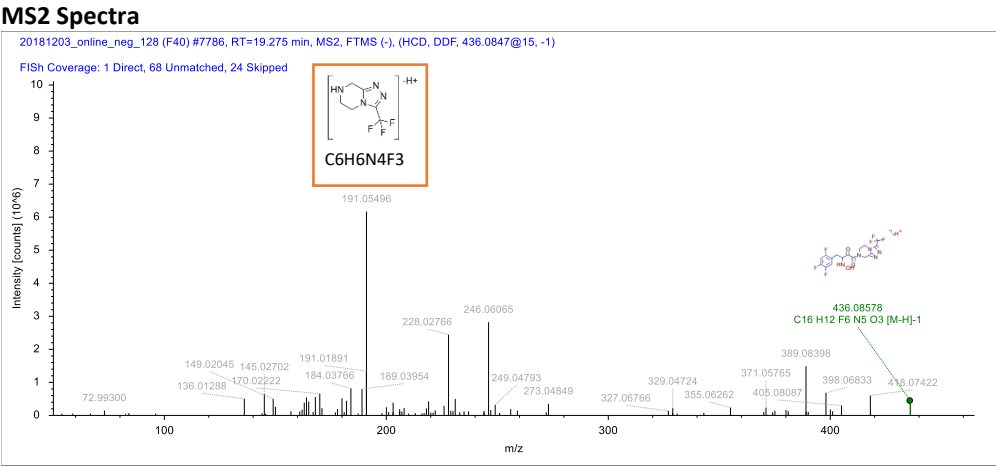
Formula
C16H13O3N5F6

Atomic modification
-H2 +O2



Confidence Level
Level 3

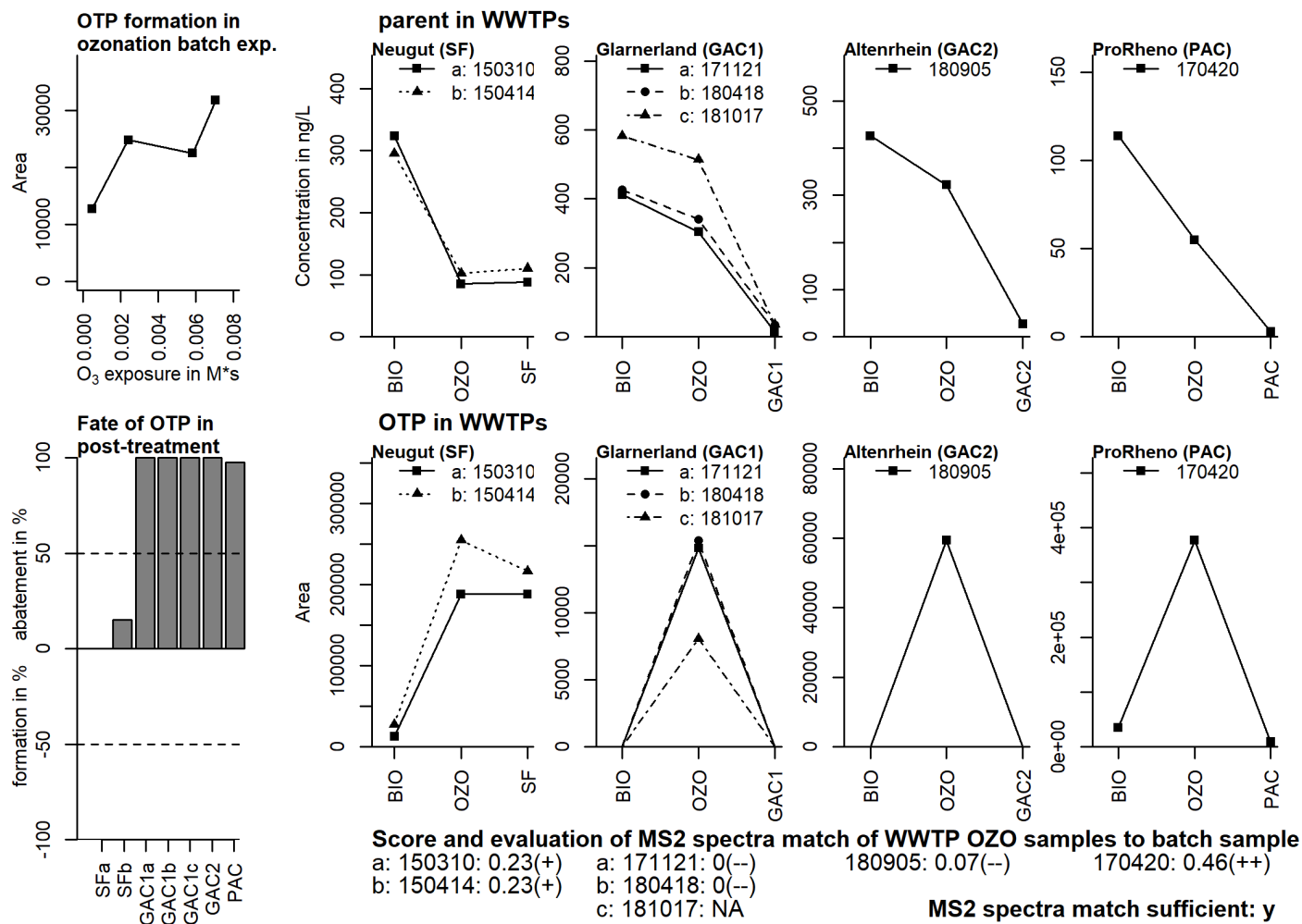
Massbank ID
ET407701



Additional Evidence for Structure Interpretation

For the parent compound sitagliptin two OTPs (SIT_n_436.0847_18.3 and SIT_n_436.0847_19.2) with the exact mass of 436.0847 were observed at different retention times (18.3 and 19.2 min). The MS² spectra of these OTPs are very similar. The atomic modification from the elemental formula of the parent compound to this OTP is -H2 +O2. The MS² fragment at the nominal mass 191 was also observed for the parent compound and corresponds to the trifluoromethyltriazolopyrazine moiety, which indicates that no modification occurred at this moiety.

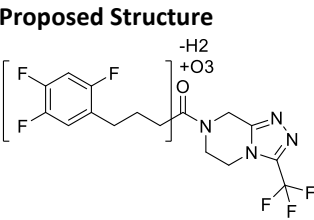
This atomic modification fits to the formation of e.g. a nitro group at the primary amine, a carbonyl group and the addition of an OH group or the formation of a hydroxylamine and a carbonyl moiety. The exact type and location of the modification(s) remain unknown. The structures of the MS² fragments are drawn exemplarily. Hermes et al. (2020) suggested the nitro structure as OTP of sitagliptin.



MS Spectra
Neg 437.0687 [m-H]-

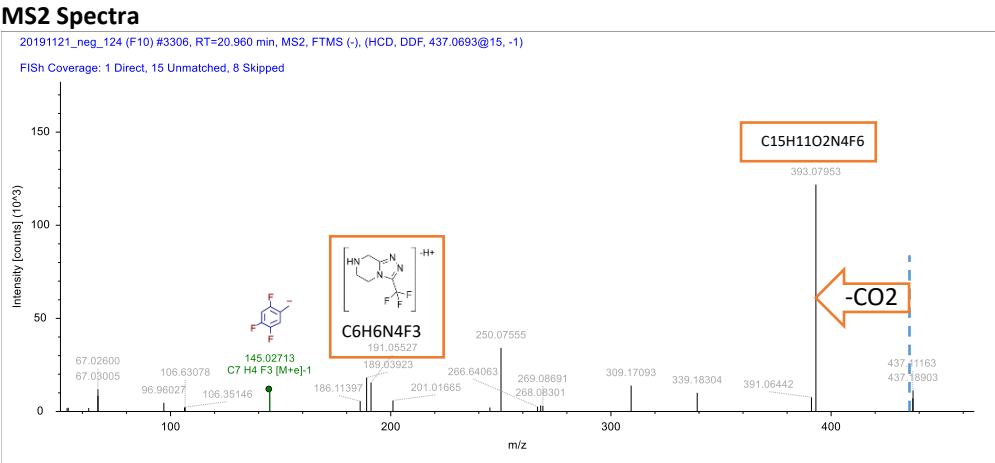
Formula
C16H12O4N4F6

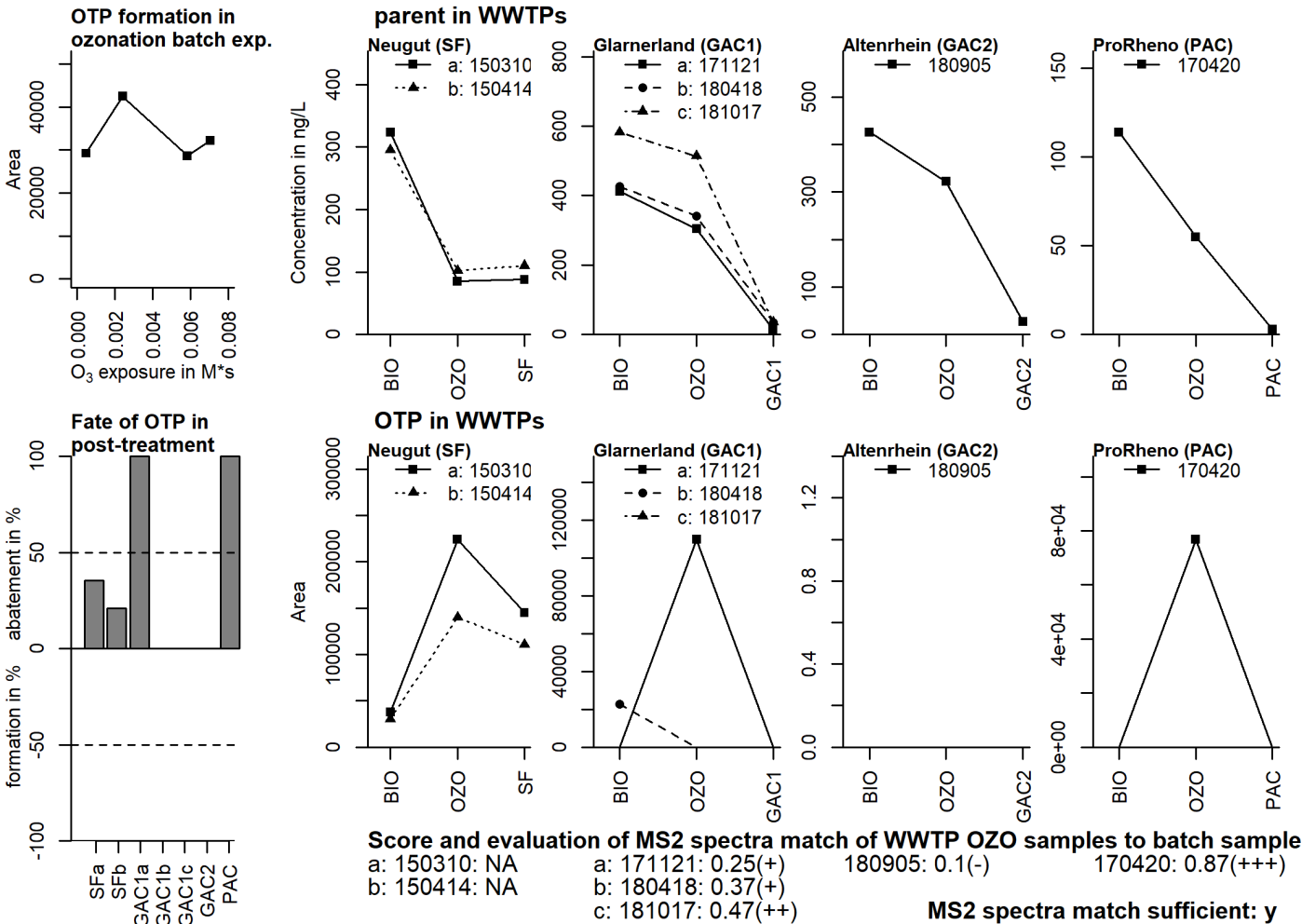
Atomic modification
-NH3 +O3



Confidence Level
Level 3

Massbank ID
ET407801





MS Spectra

Pos	391.0988	[m+H] ⁺
Neg	389.0841	[m-H] ⁻
Neg	435.0896	m+FA-H

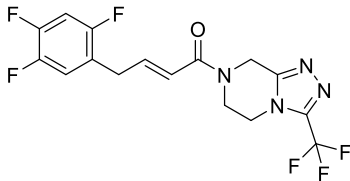
Formula

C16H12ON4F6

Atomic modification

-NH3

Proposed Structure



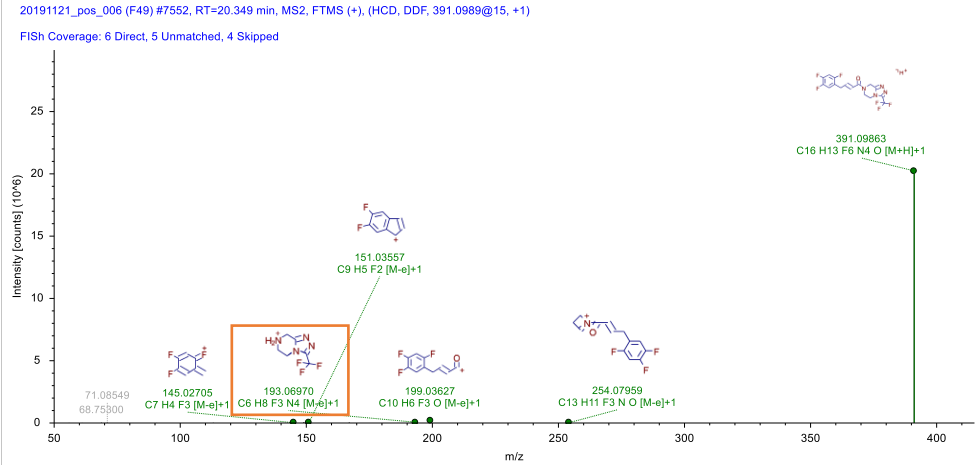
Confidence Level

Level 3

Massbank ID

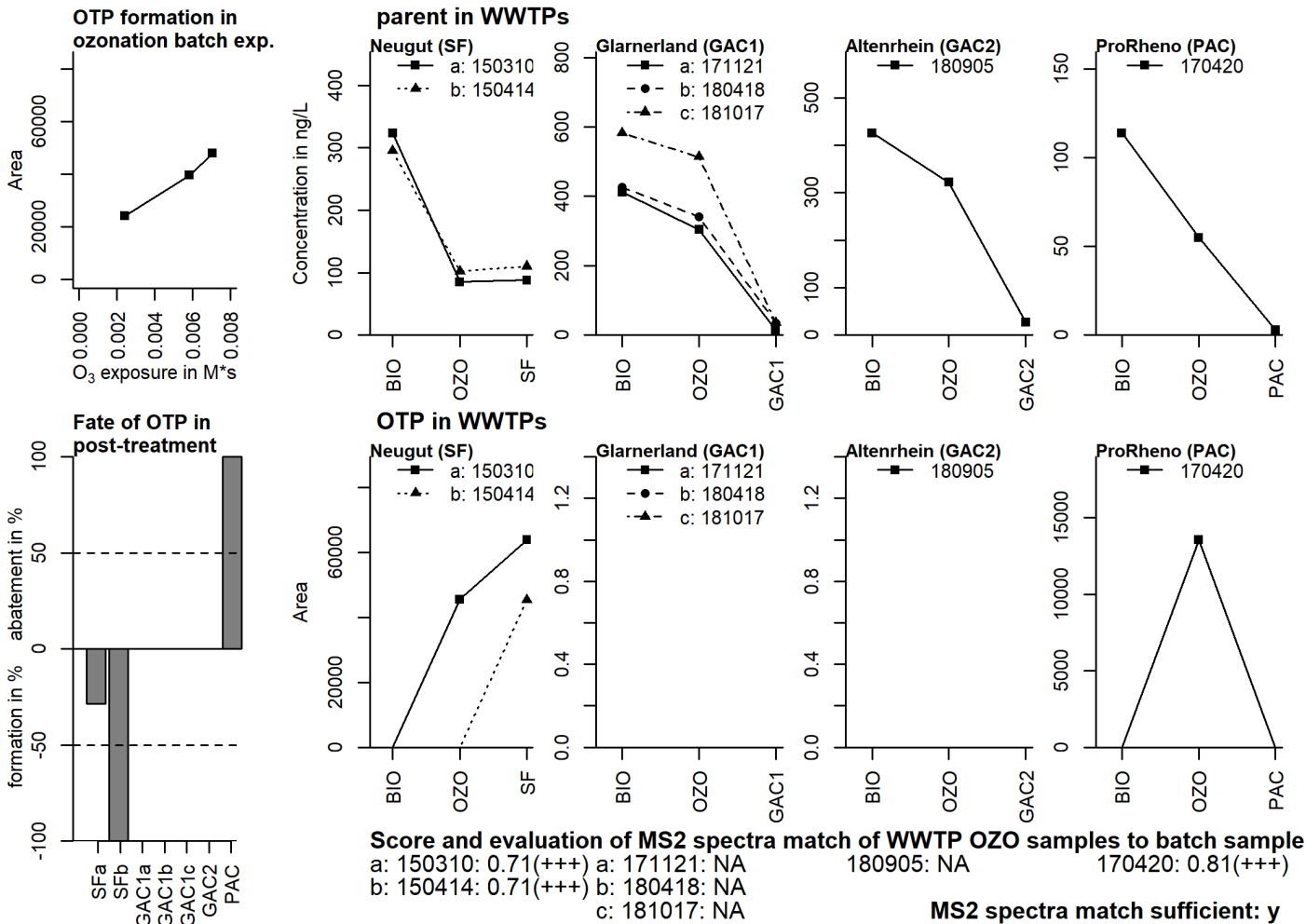
ET403701

MS2 Spectra



Additional Evidence for Structure Interpretation

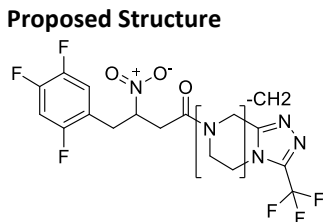
The atomic modification from the elemental formula of the parent compound to this OTP is -NH3. Fragment 193, which was also observed for the parent compound, indicates that the trifluoromethyltriazolopyrazine moiety is retained in the OTP. Therefore, the atomic modification fits to the cleavage of the primary amine moiety and the abstraction of two additional hydrogen atoms. The exact location of the modification remains unknown. The structures of the MS² fragments are drawn exemplarily. The drawn structure was suggested by Hermes et al. (2020) as OTP of sitagliptin and the evidence from our MS and MS² spectra match the one of Hermes et al. (2020). They suggested the formation of a C-C double bond at the location of the cleavage, although this is not reaction typically observed during ozonation. This remains unclear.



MS Spectra
Pos 426.0996 [m+H]⁺

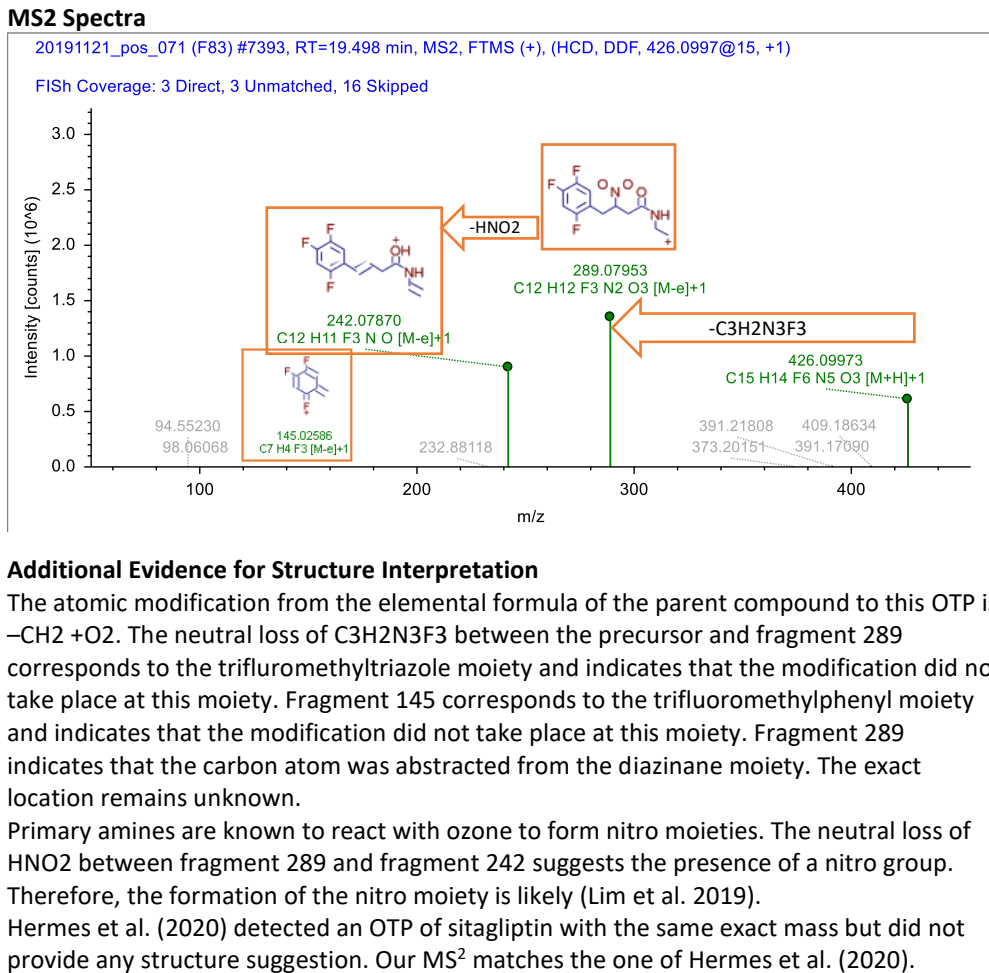
Formula
C₁₅H₁₃O₃N₅F₆

Atomic modification
-CH₂ +O₂

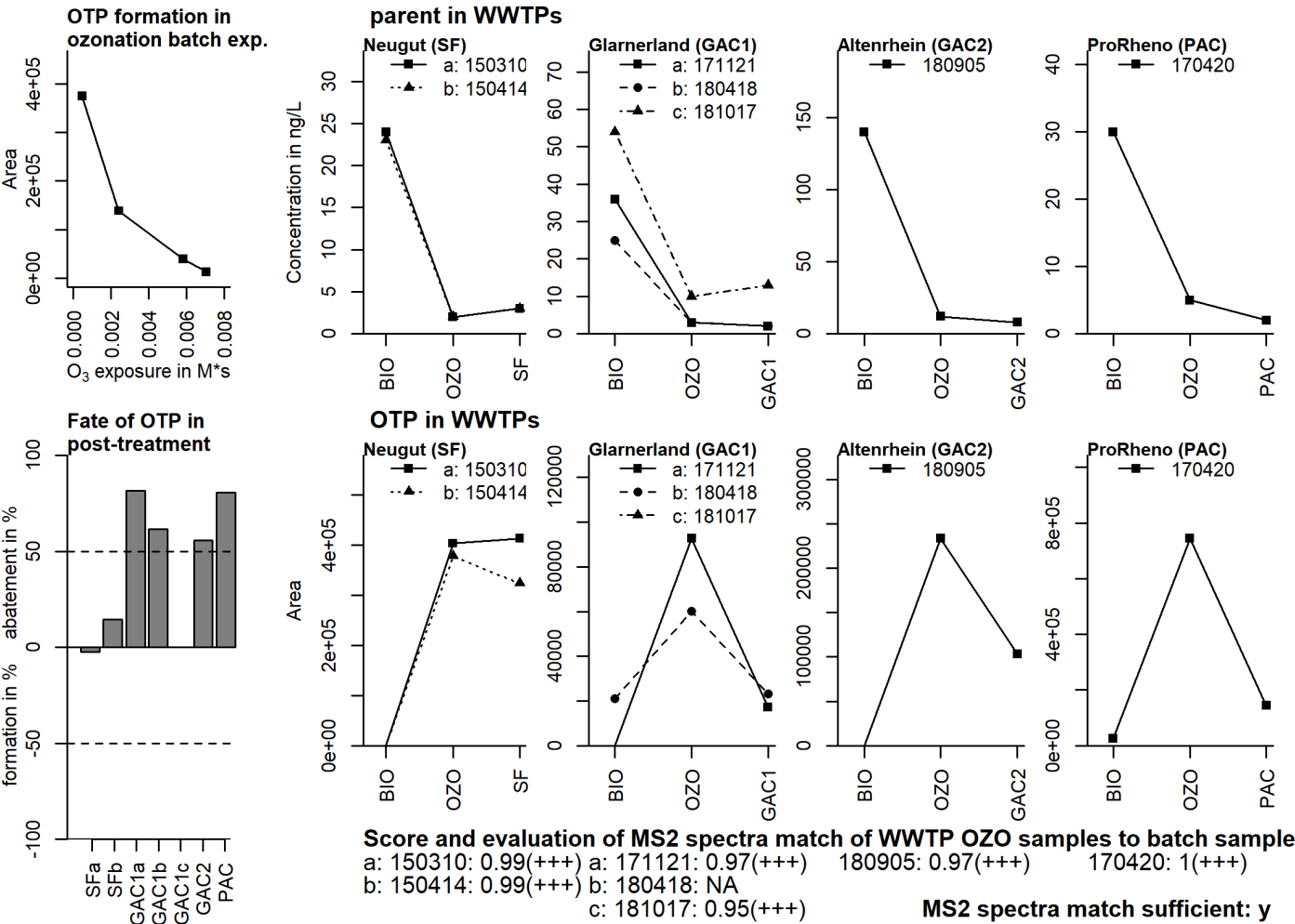


Confidence Level
Level 3

Massbank ID
ET403801



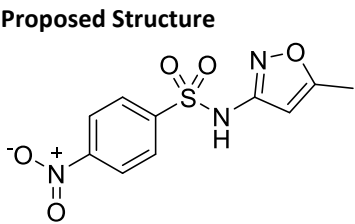
The atomic modification from the elemental formula of the parent compound to this OTP is $-C_3H_6 + O$, which fits to the formation of a nitro group and a cleavage within the methylisoxazol moiety. Aniline like moieties are known to react to nitro groups during ozonation (von Sonntag & von Gunten 2012). The neutral loss of NO_2 between the molecular ion and fragment 179 indicates the presence of a nitro group. Willach et al. (2017) detected an OTP of sulfomethoxazole with a similar mass but did not provide a structure suggestion. However, Martin De Vidales et al. (2012) detected an electrochemical oxidation product of sulfamethoxazole with exact an m/z of 225.9932 $[m-H]^-$ and suggested the proposed structure. The evidence from our MS and MS^2 spectra match the one from Martin De Vidales et al. (2012).



MS Spectra
Neg 282.0188 [m-H]-

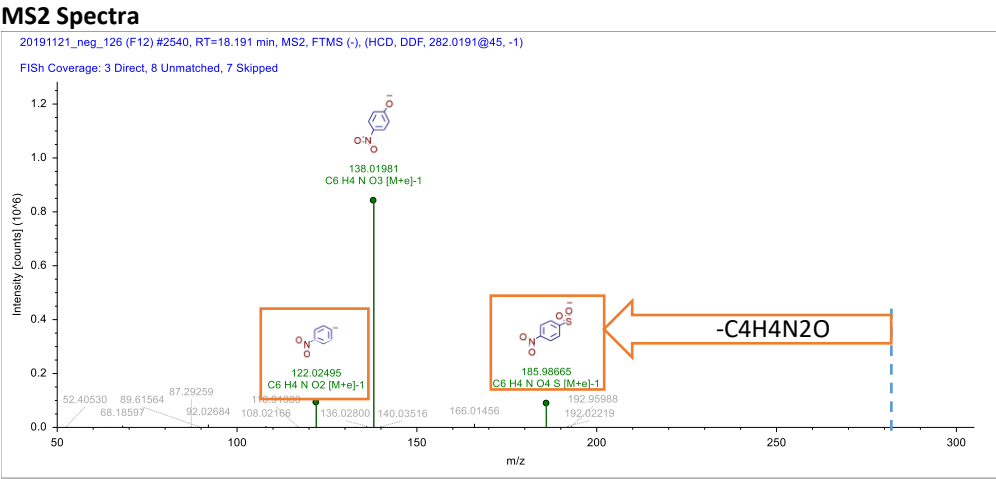
Formula
C10H9O5N3S

Atomic modification
-H2 +O2



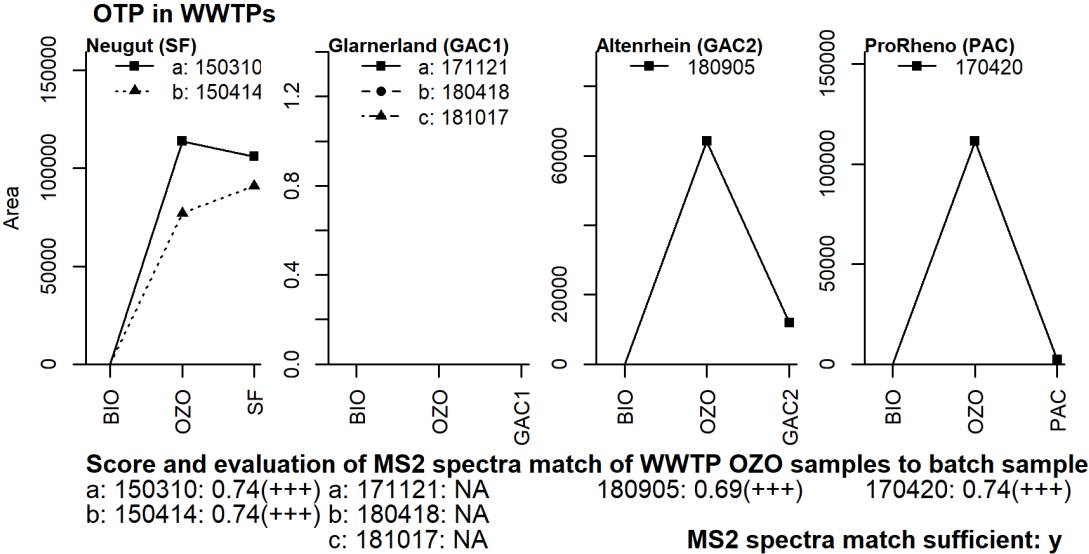
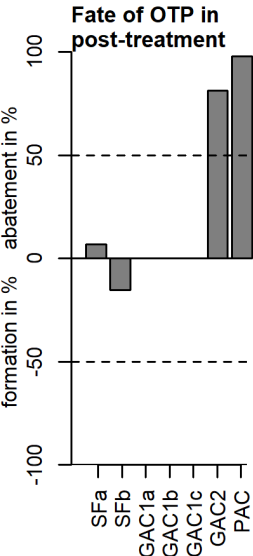
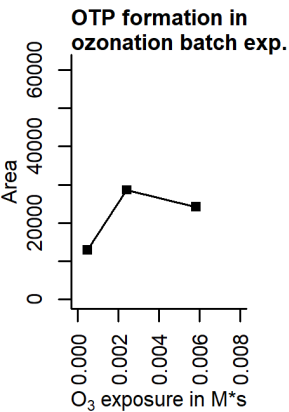
Confidence Level
Level 3

Massbank ID
ET408201



Additional Evidence for Structure Interpretation

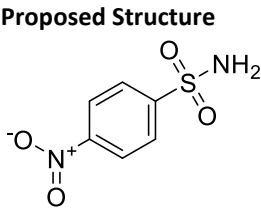
The neutral loss of C4H4N2O fits to the cleavage of the *N*-methyloxazol moiety. Fragment 122 indicates that the modification occurred at the aniline moiety. Aniline-like moieties are known to form nitro groups during ozonation (von Sonntag & von Gunten 2012). It is therefore likely that a nitro group was formed here. Nitro-sulfamethoxazole was proposed as ozonation transformation product in several studies (Abellan et al. 2008, Rodayan et al. 2010, Gomez-Ramos et al. 2011, Gao et al. 2014, Willach et al. 2017).



MS Spectra
Neg 200.9972 [m-H]-

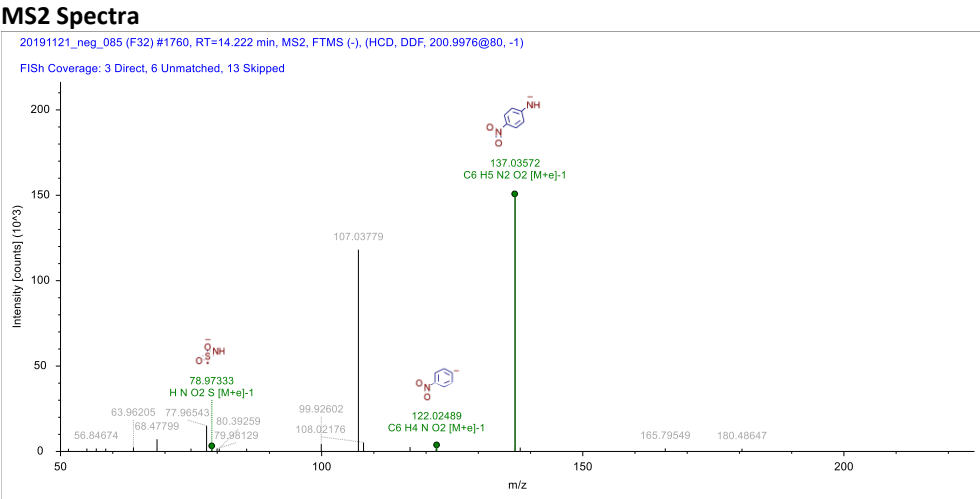
Formula
C₆H₆O₄N₂S

Atomic modification
SMX: -C₄H₅N + O
SMZ: -C₆H₈N₂ + O₂



Confidence Level
Level 3

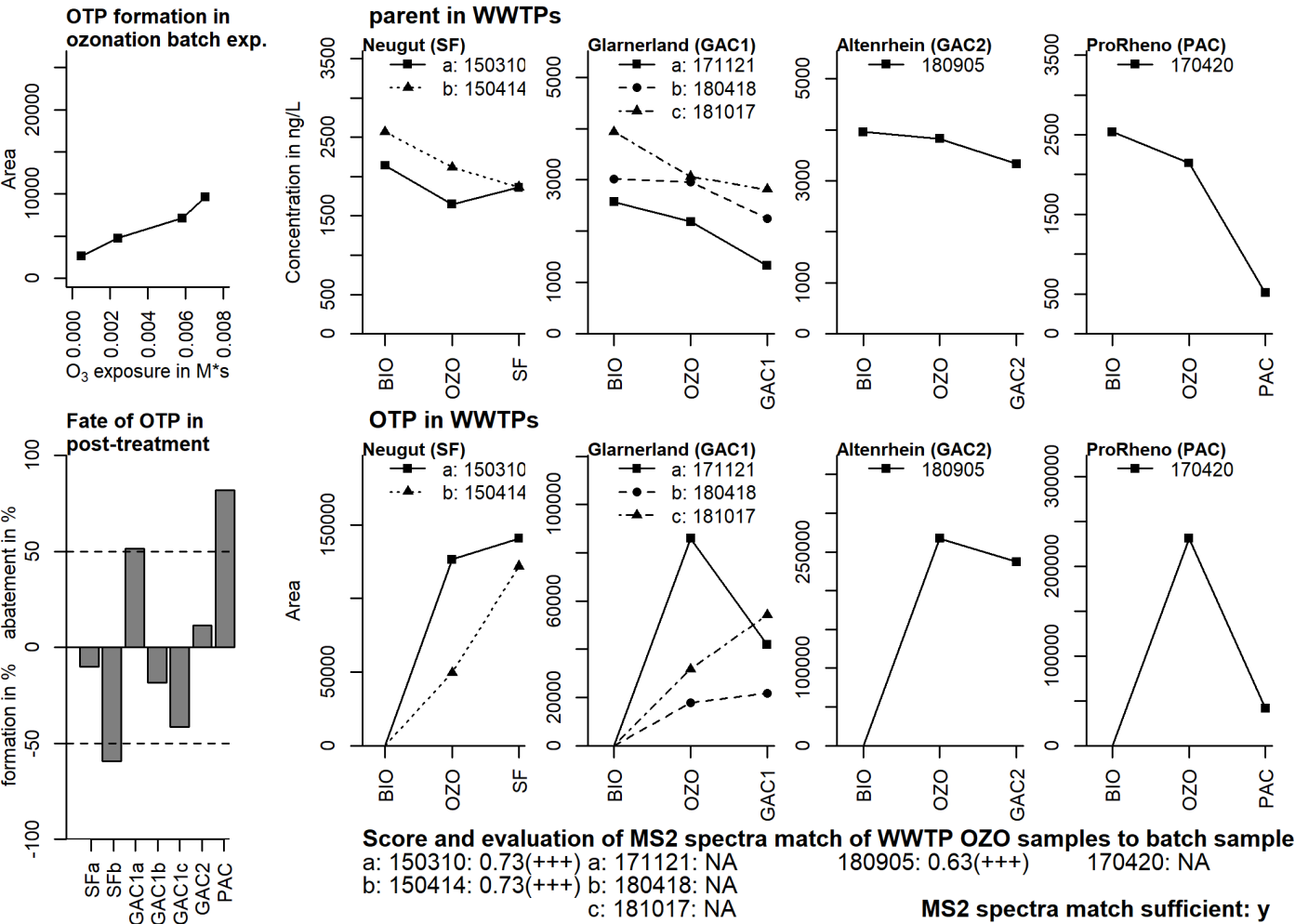
Massbank ID
ET408001



Additional Evidence for Structure Interpretation

For sulfamethazine the loss of C₆H₈N₂ fits to the cleavage of the dimethylpyrimidine moiety and for sulfamethoxazole the loss of C₄H₅N fits to the cleavage of the methyloxazole moiety. 2 oxygen atoms were added on the remaining benzenesulfonamide moiety.

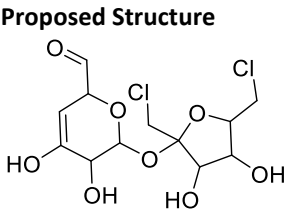
Aniline-like moiety are known to form nitro groups during ozonation (von Sonntag & von Gunten 2012). It is therefore likely that this was the case here. Nitro-sulfamethoxazole was proposed as ozonation transformation product in several studies (Abellan et al. 2008, Rodayan et al. 2010, Gomez-Ramos et al. 2011, Gao et al. 2014, Willach et al. 2017), as well as methyloxazole (Gomez-Ramos et al. 2011, Gao et al. 2014). Therefore, the proposed structure is likely to be formed.



MS Spectra
Neg 403.0202 [m+FA-H]-

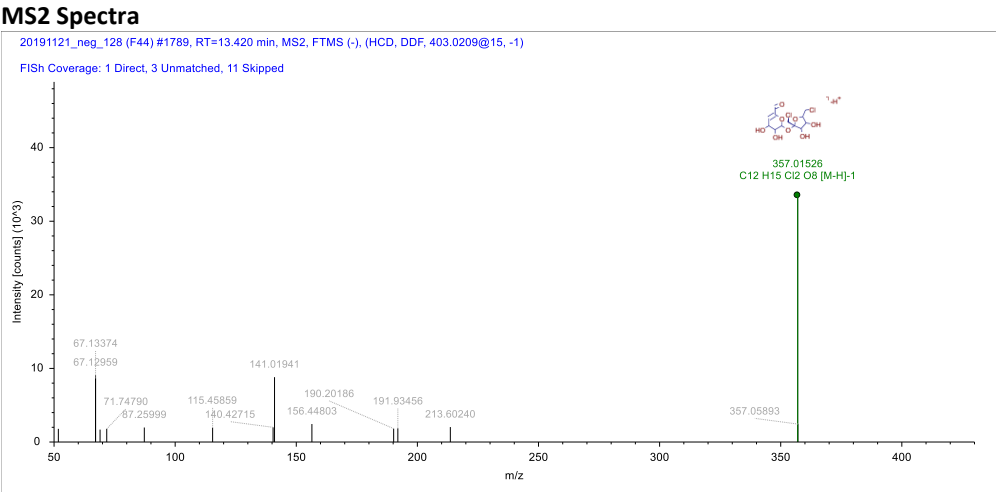
Formula
C₁₂H₁₆O₈Cl₂

Atomic modification
-H₃Cl



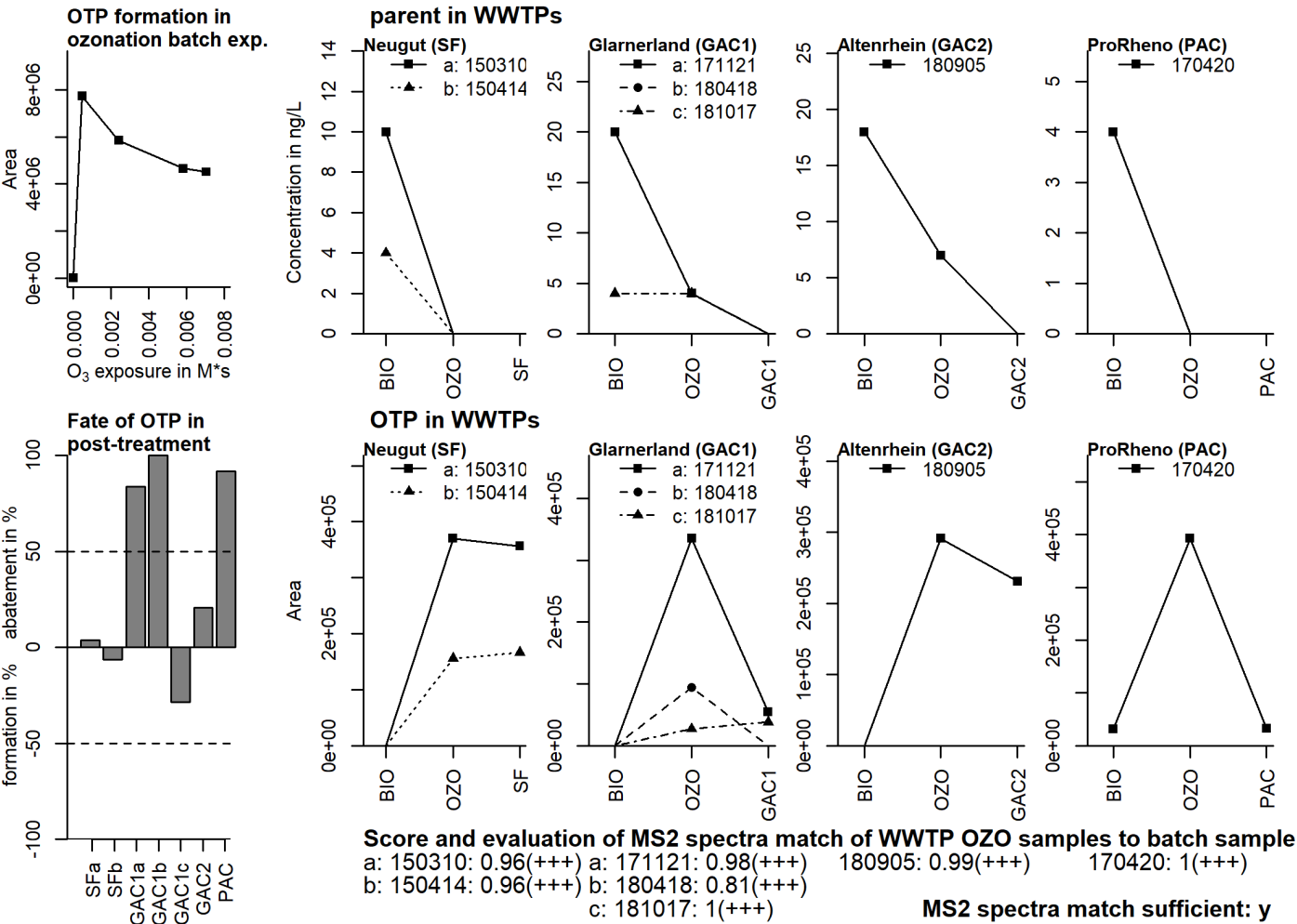
Confidence Level
Level 3

Massbank ID
ET408367



Additional Evidence for Structure Interpretation

The atomic modification from the elemental formula of the parent compound to this OTP is -H₃Cl. This modification fits to a cleavage of a chlorine atom and the oxidation of a C-O bond. Hu et al. (2017) also detected an OTP of sucralose with a m/z of 403 and suggested the structure drawn for fragment 357 based on the MS² spectra.

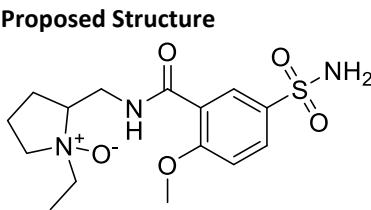


MS Spectra

Pos 358.1432 [m+H]⁺
Neg 356.1283 [m-H]⁻
Neg 402.1337 m+FA-H

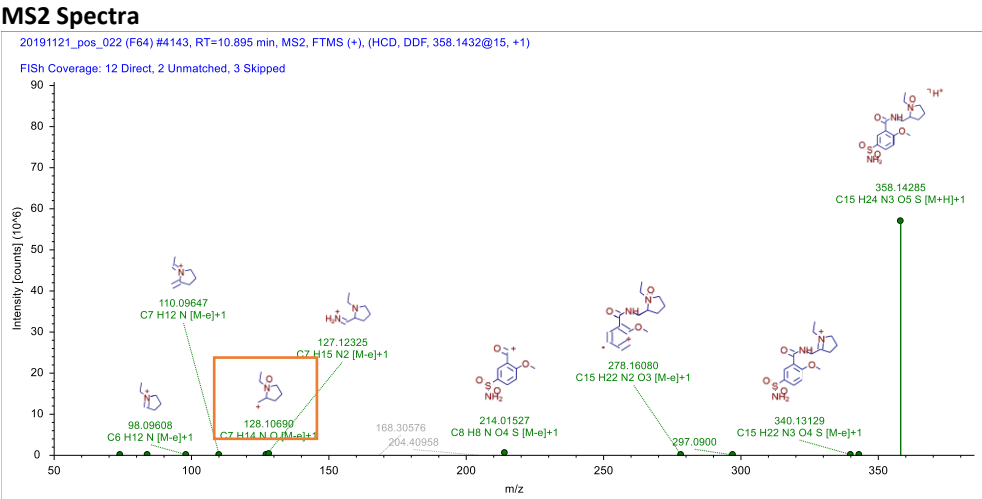
Formula
C₁₅H₂₃O₅N₃S

Atomic modification
+O



Confidence Level
Level 1

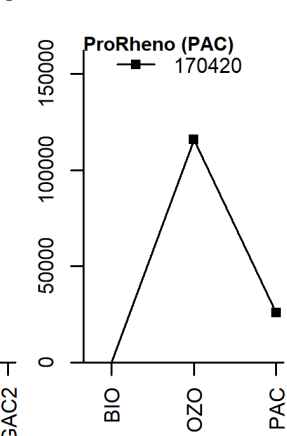
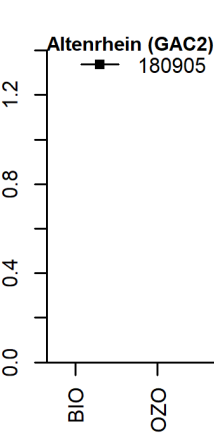
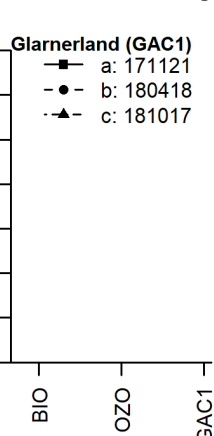
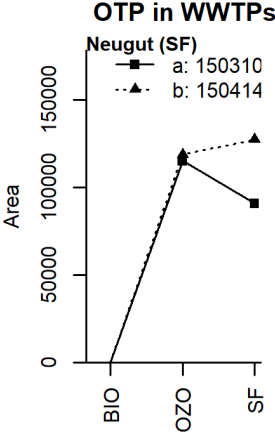
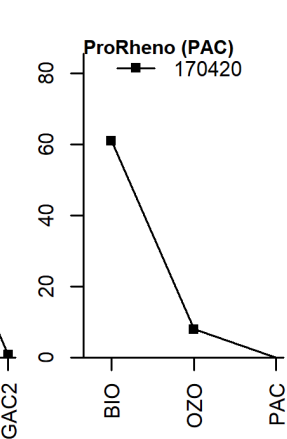
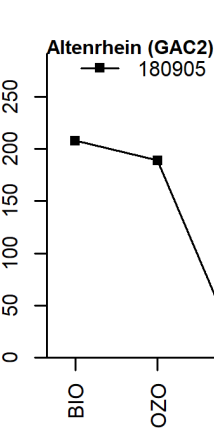
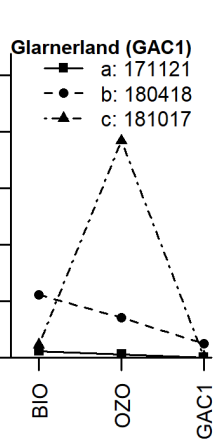
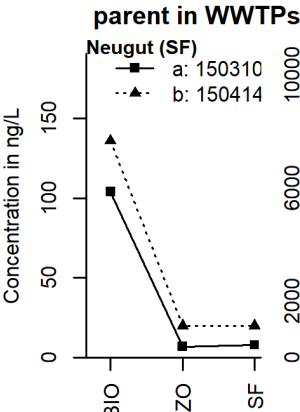
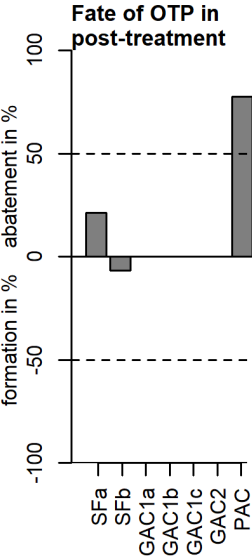
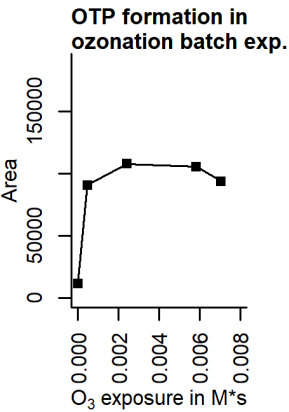
Massbank ID
ET403901



Additional Evidence for Structure Interpretation

A MS² fragment at the nominal mass of 112 was observed for the parent compound and corresponds to the ethylpyrrolidine moiety. The MS² fragment 128 here corresponds to the fragment 112 of the parent with an addition of an oxygen atom. Tertiary amines are very reactive to form *N*-oxides during ozonation (von Sonntag & von Gunten 2012). Therefore, the formation of a *N*-oxide at the ethylpyrrolidine is very likely.

Sulpiride *N*-oxide has been suggested as major OTP of Sulpiride by Bollmann et al. (2016) and Merel et al. (2017).



Score and evaluation of MS2 spectra match of WWTP OZO samples to batch sample

Sample	Score	Evaluation
a: 150310	0(--)	
b: 150414	0(--)	
a: 171121	0.46(++)	
b: 180418	0.51(++)	
c: 181017	0.46(++)	
180905	0.04(--)	
170420	0.12(-)	

MS2 spectra match sufficient: n

MS Spectra

Pos 266.1749 [m+H]⁺

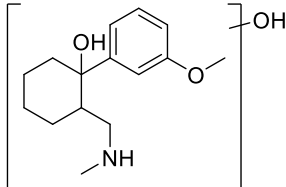
Formula

C₁₅H₂₃O₃N

Atomic modification

-CH₂ +O

Proposed Structure



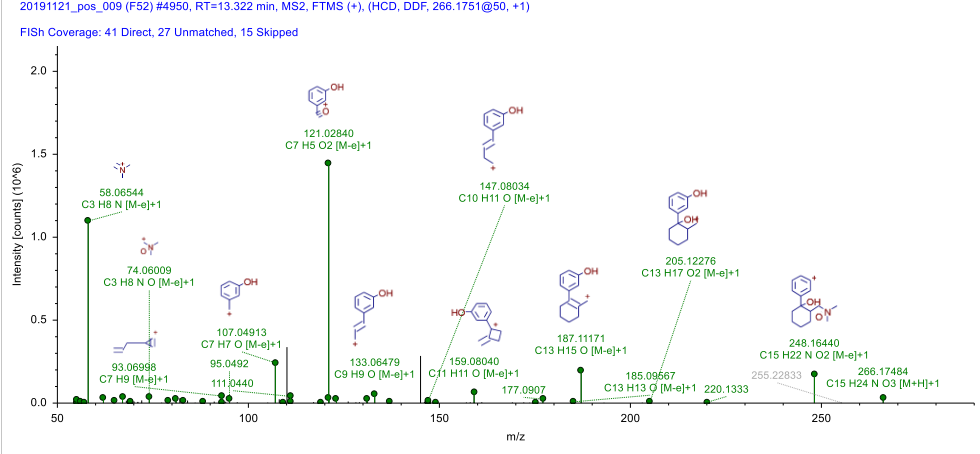
Confidence Level

Level 3

Massbank ID

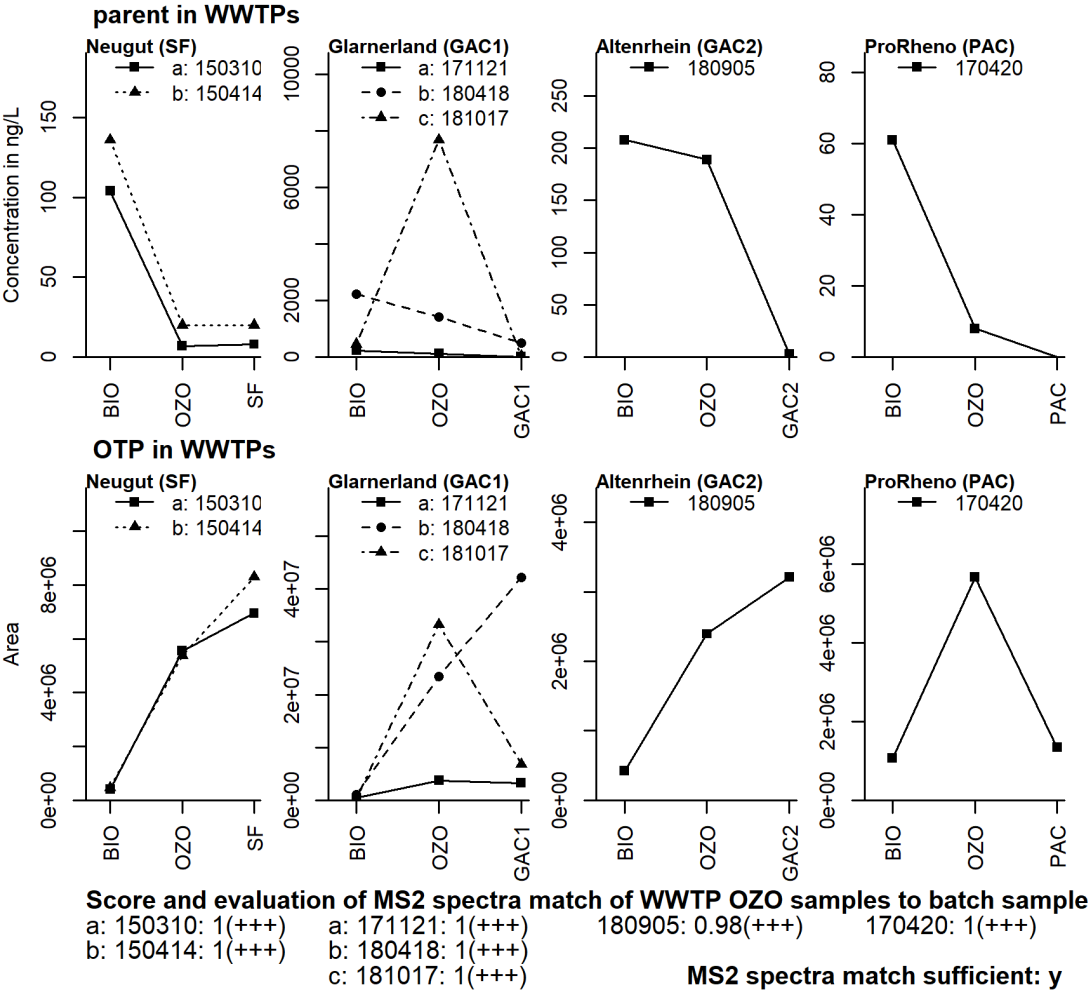
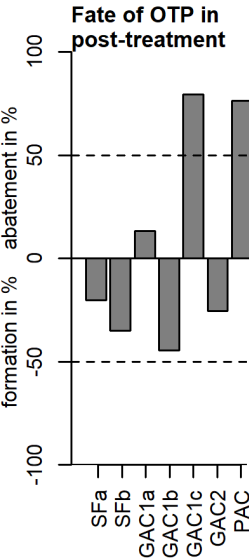
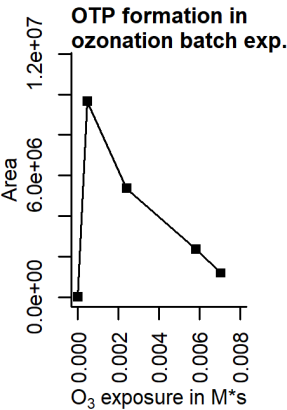
ET404001

MS2 Spectra



Additional Evidence for Structure Interpretation

Fragment 121 was also observed for TRA_p_250.1801_13.5, which was identified as *N*-desmethyltramadol. The atomic modification from the elemental formula of the parent compound to this OTP is -CH₂ +O. It is likely that *N*-demethylation occurred for this OTP as well at the dimethylamine moiety. The oxygen atom was likely added by hydroxylation via a hydroxyl radical, most likely at an aliphatic C-atom, because once formed, phenol moieties quickly react further during ozonation.

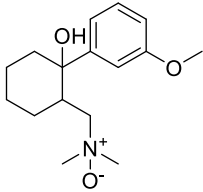


MS Spectra
Pos 280.1905 [m+H]⁺

Formula
C₁₆H₂₅O₃N

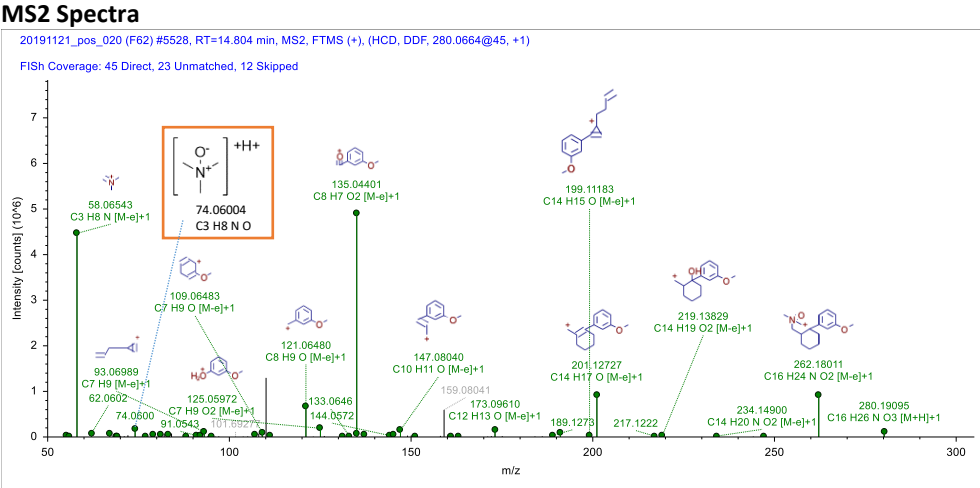
Atomic modification
+O

Proposed Structure



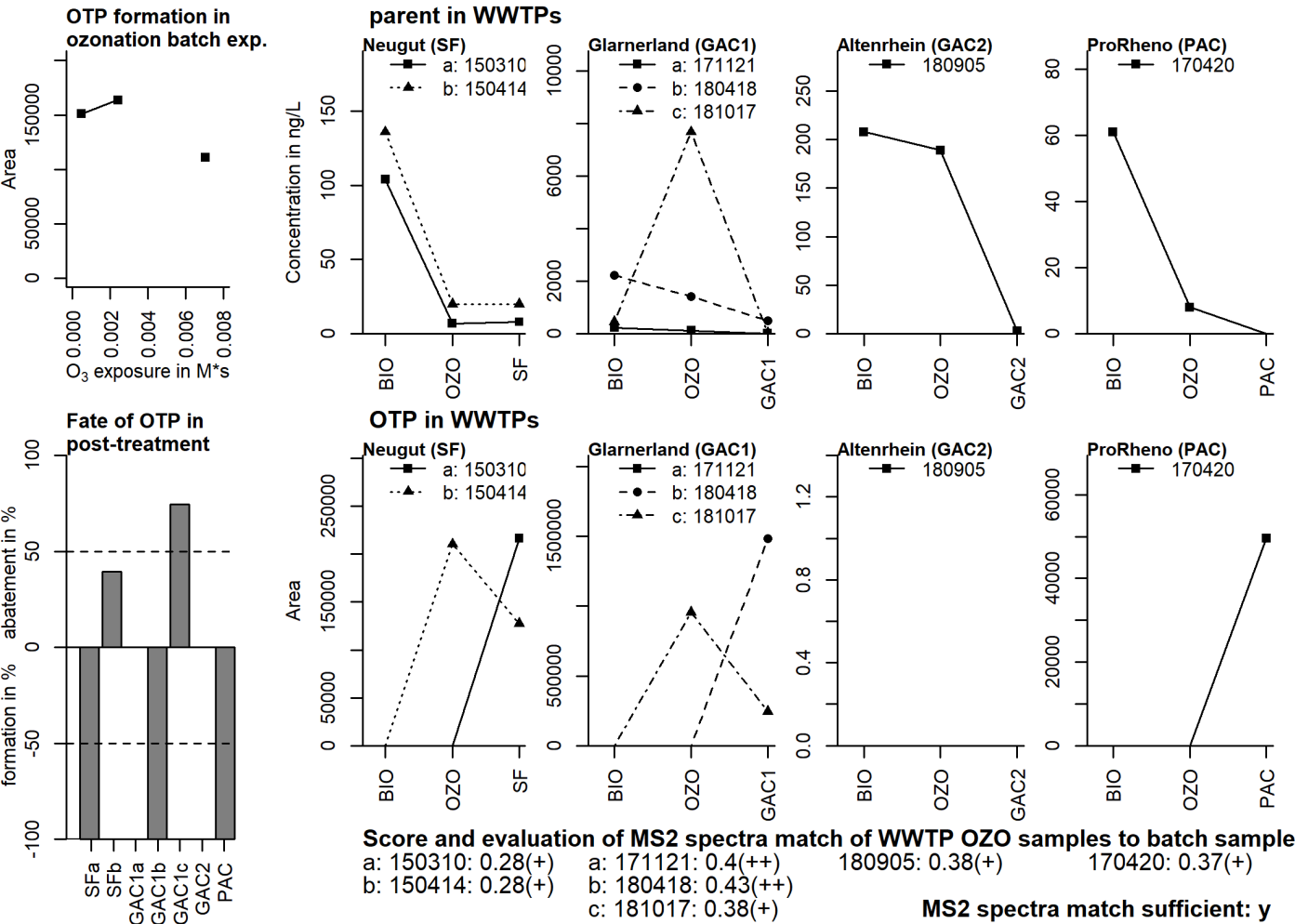
Confidence Level
Level 1

Massbank ID
ET404101



Additional Evidence for Structure Interpretation

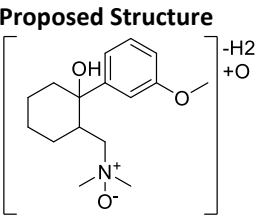
A MS² fragment at the nominal mass of 58 was observed for the parent compound and corresponds to the trimethylamine moiety. The MS² fragment 74 corresponds to the fragment 58 of the parent with an addition of an oxygen atom. Tertiary amines are very reactive to form *N*-oxides during ozonation (von Sonntag & von Gunten 2012). Therefore, the formation of a *N*-oxide at the trimethylamine moiety is very likely. Fragment 74 indicates the formation of the *N*-oxide. Tramadol *N*-oxide has been confirmed as major OTP of tramadol by Zimmermann et al. (2012) and Merel et al. (2017). The observed evidence from our MS spectrum and the MS² spectrum match with the previous studies. The MS² spectrum also matches the reference spectrum we recorded for tramadol *N*-oxide.

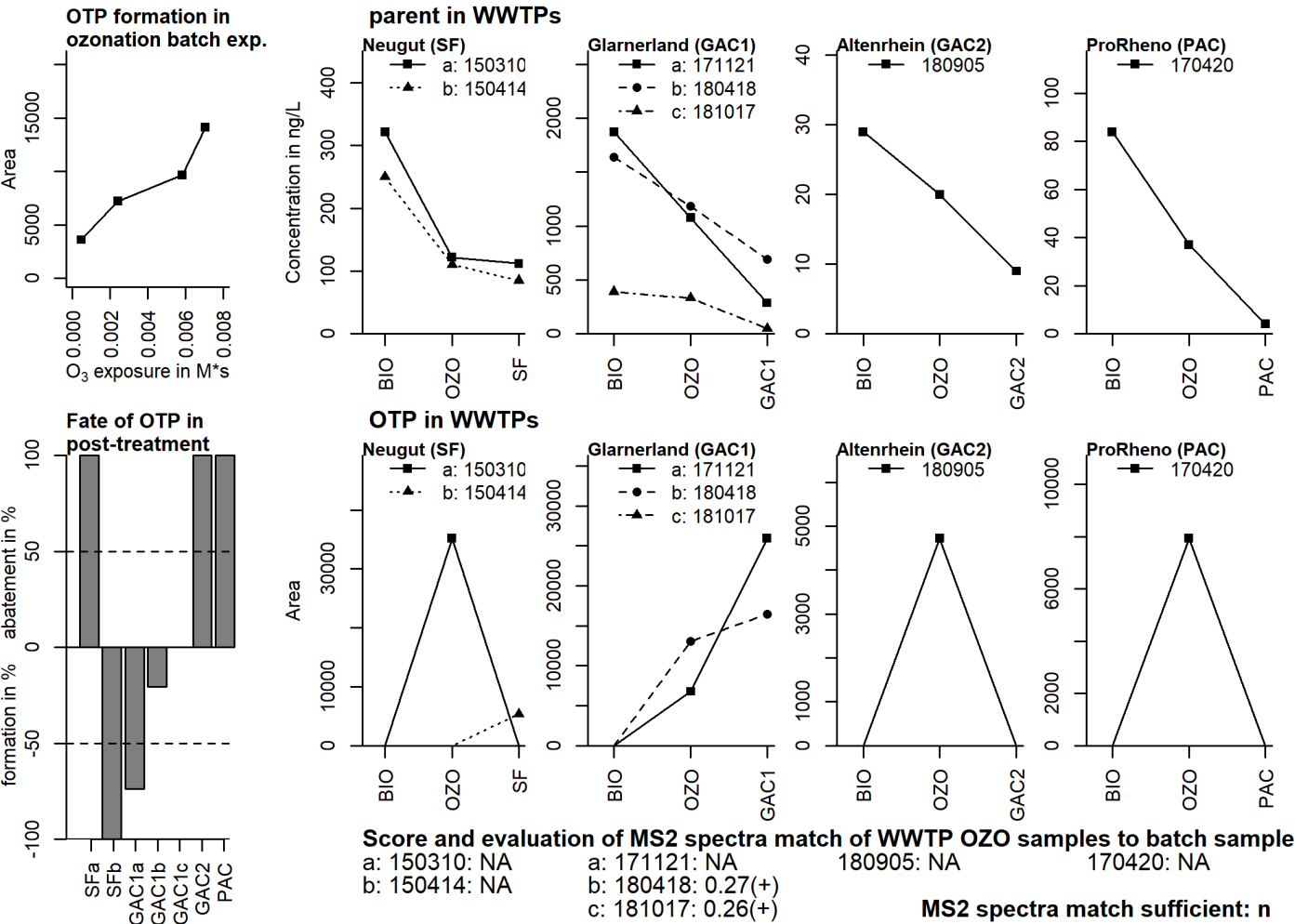


MS Spectra
Pos 294.1699 [m+H]⁺

Formula
C₁₆H₂₃N₂O₄

Atomic modification
-H₂ +O₂

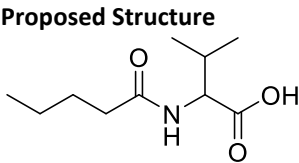




MS Spectra
Neg 200.1289 [m-H]-

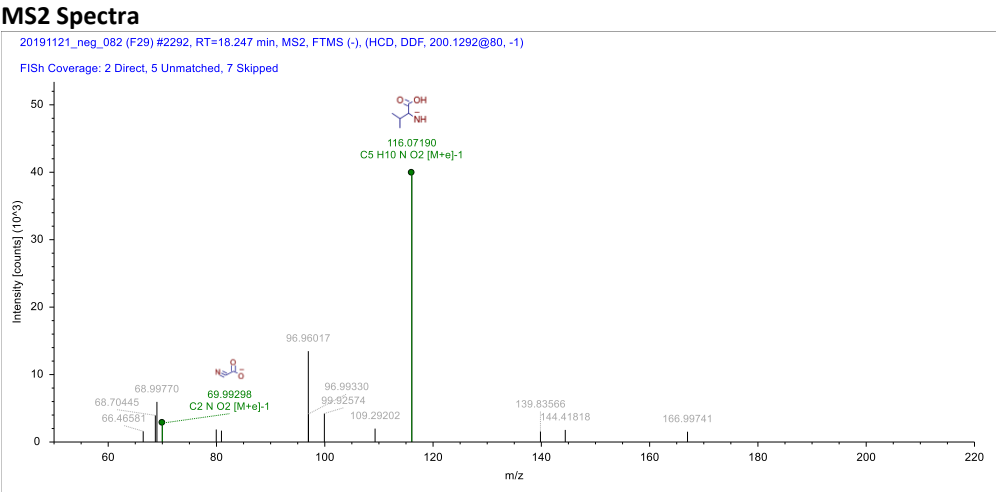
Formula
C10H19O3N

Atomic modification
-C14H10N4



Confidence Level
Level 3

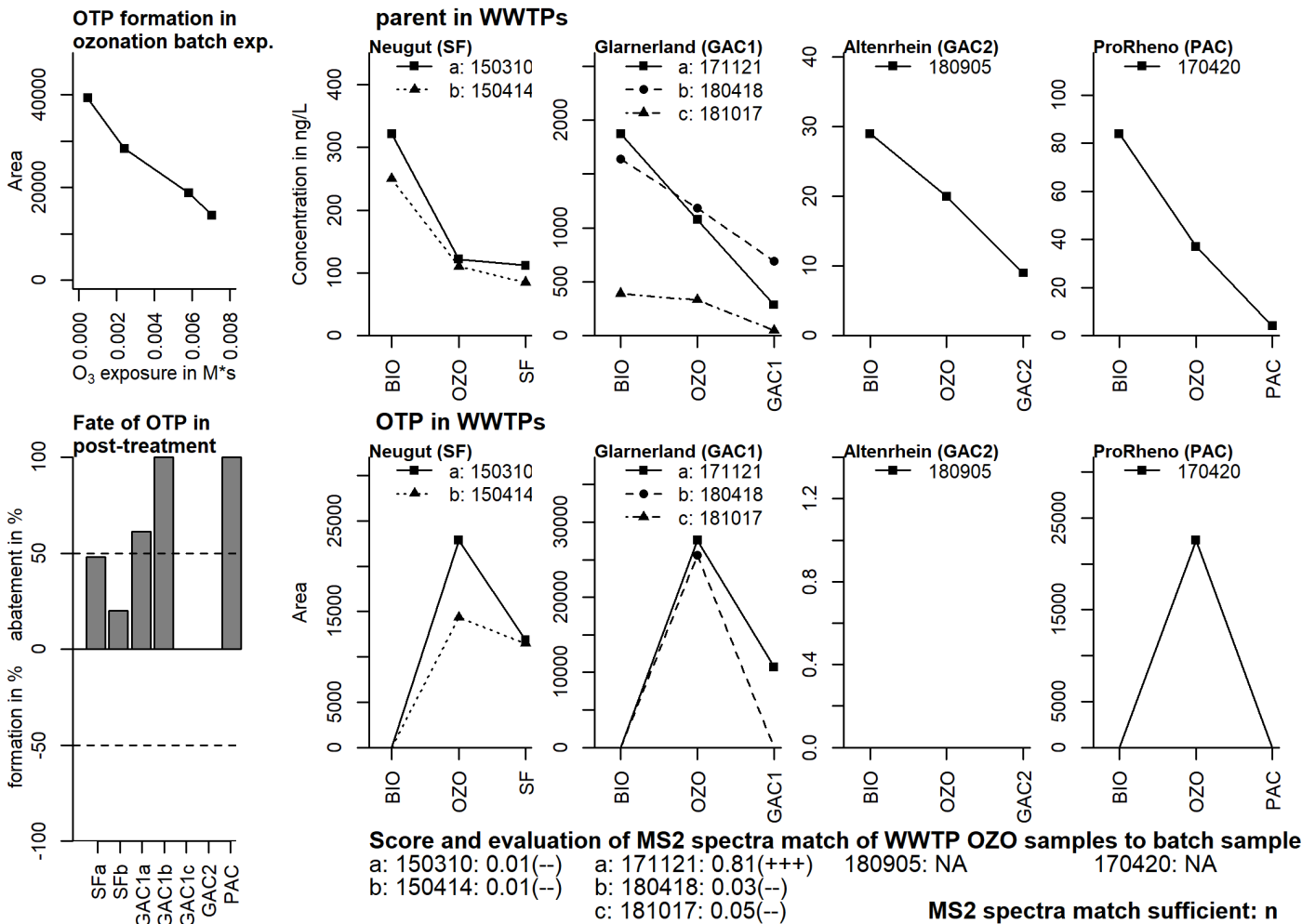
Massbank ID
ET408401



Additional Evidence for Structure Interpretation

A fragment with mass 200 was observed for the parent compound and corresponds to the molecular ion of this OTP. The atomic modification from the parent structure to this OTP fits to a cleavage of the moiety, which is common for all sartans (tetrazolebiphenylmethyl moiety).

Diehle et al. (2019) suggested the formation of this structure and our MS² spectra matches with the previous observations.



MS Spectra

Neg 234.1133 [m-H]-

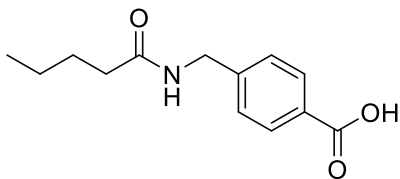
Formula

C13H17O3N

Atomic modification

-C11H12N4

Proposed Structure



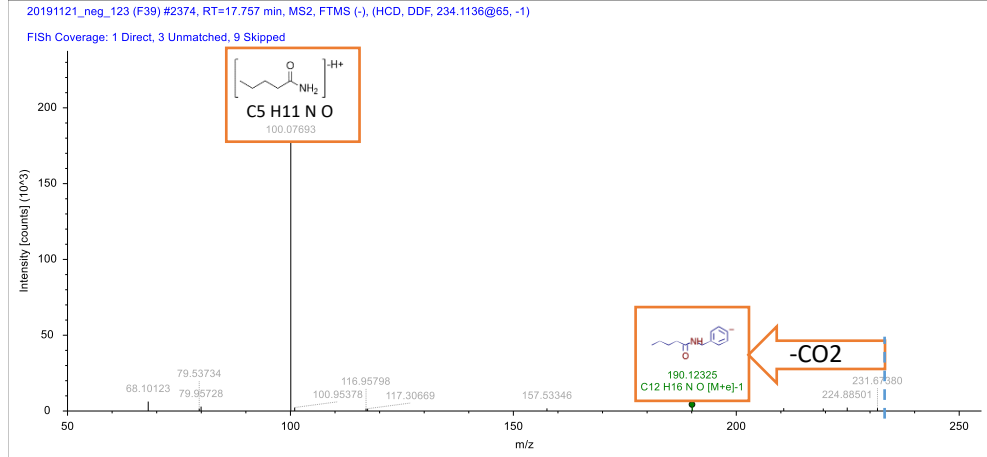
Confidence Level

Level 3

Massbank ID

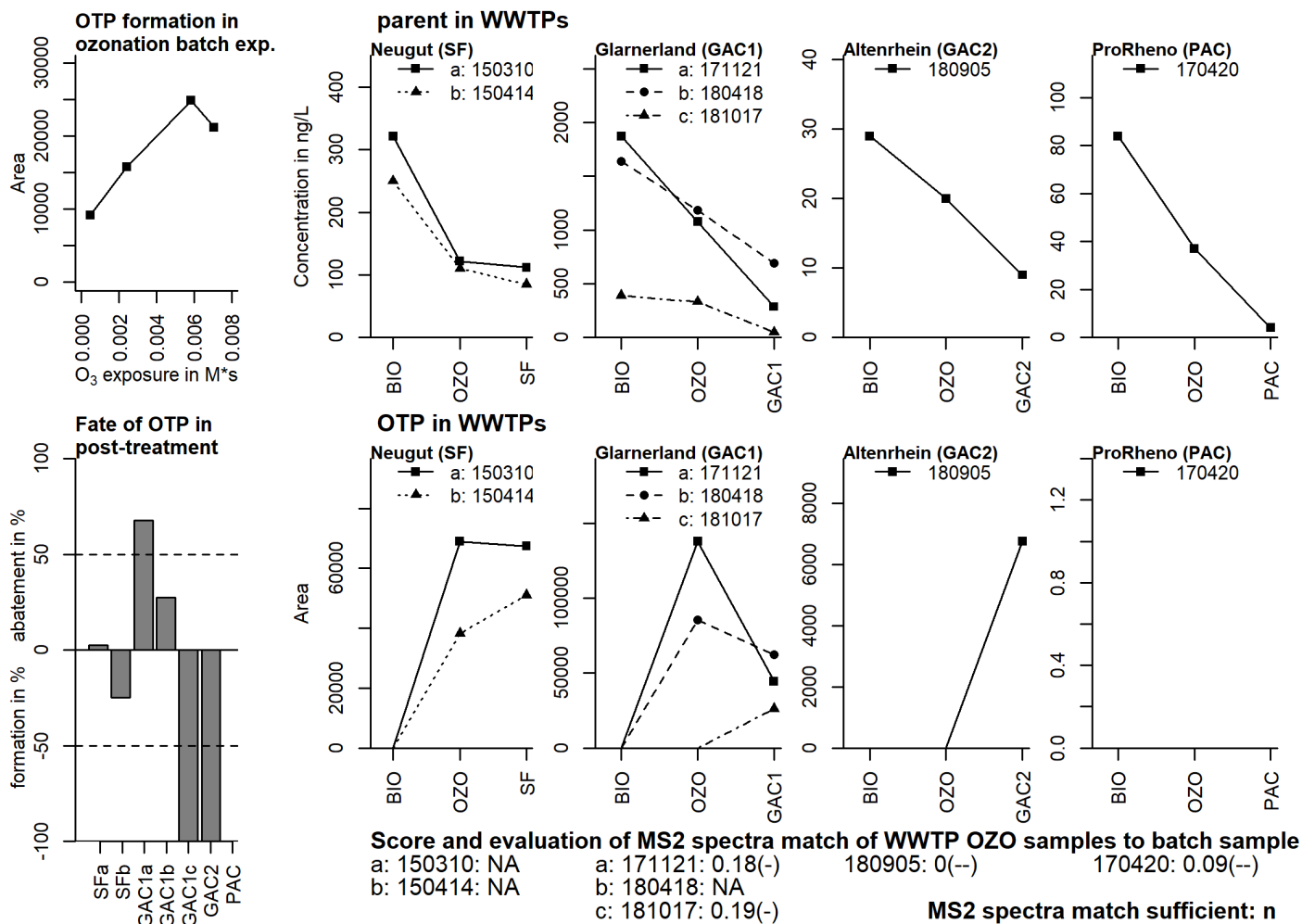
ET408501

MS2 Spectra



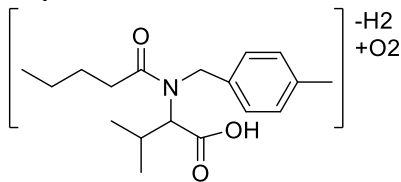
Additional Evidence for Structure Interpretation

The atomic modification for the elemental formula of the parent compound to this OTP is --C11H12N4. This fits to a cleavage within the benzene moiety of the tetrazolebiphenylmethyl part, which is common for all sartans, the cleavage of the methylbutanoic acid moiety and the addition of 2 oxygen atoms. Fragment 100 fits to the pentanamide moiety of the parent compound. The neutral loss of CO₂ between the molecular ion and fragment 190 indicates formation of a carboxylic acid moiety. Aromatic compounds are known to form cis,cis-Muconic acid like structures (von Sonntag & von Gunten 2012). Therefore, the formation of this kind of structure, followed by a C-C bond cleavage, resulting in a p-benzoic acid moiety, is probable. Diehle et al. (2019) suggested the formation of this structure and our MS² spectra matches with these previous observations.

**MS Spectra**

Neg 334.1657

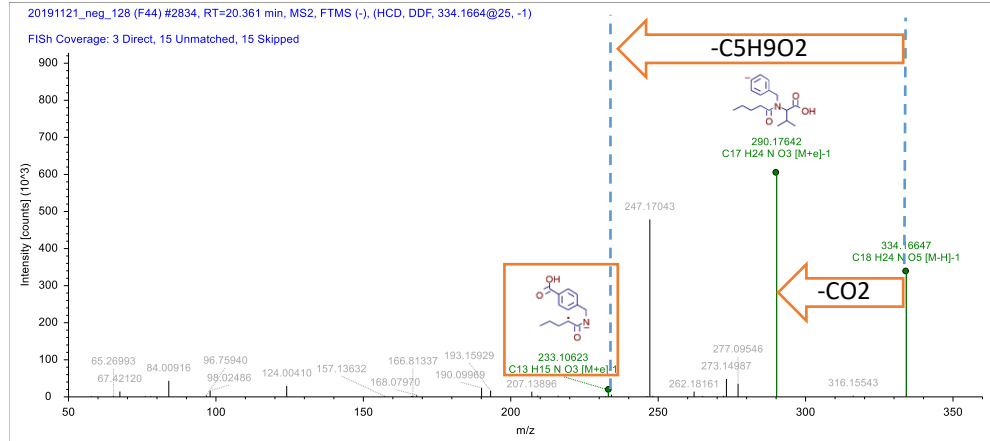
[m-H]-

FormulaC₁₈H₂₅O₅N**Atomic modification**-C₆H₄N₄ + O₂**Proposed Structure****Confidence Level**

Level 3

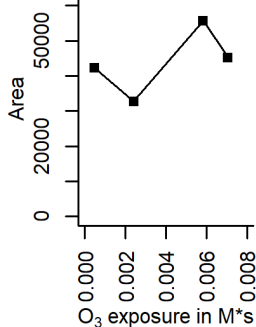
Massbank ID

ET408601

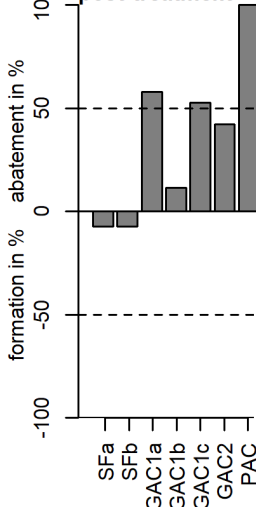
MS2 Spectra**Additional Evidence for Structure Interpretation**

The atomic modification for the elemental formula of the parent compound to this OTP is -C₆H₄N₄ + O₂. This fits to a cleavage within the benzene moiety of the tetrazolebiphenylmethyl part, which is common for all sartans, and the addition of 2 oxygen atoms. The neutral loss of C₅H₉O₂ between the precursor and fragment 207 indicates that the 2 oxygens are connected to the p-methylbenzyl or the pentanamide moiety. The exact type and position remain unknown. Aromatic compounds are known to form cis,cis-Muconic acid like structures (von Sonntag & von Gunten 2012). Therefore, the formation of this kind of structure, followed by a C-C bond cleavage, resulting in a p-benzoic acid moiety, is probable. There is however no further evidence. The structures of the MS² fragments are drawn exemplarily. Diehle et al. (2019) detected an OTP of valsartan with the exact same mass and suggested the same structure.

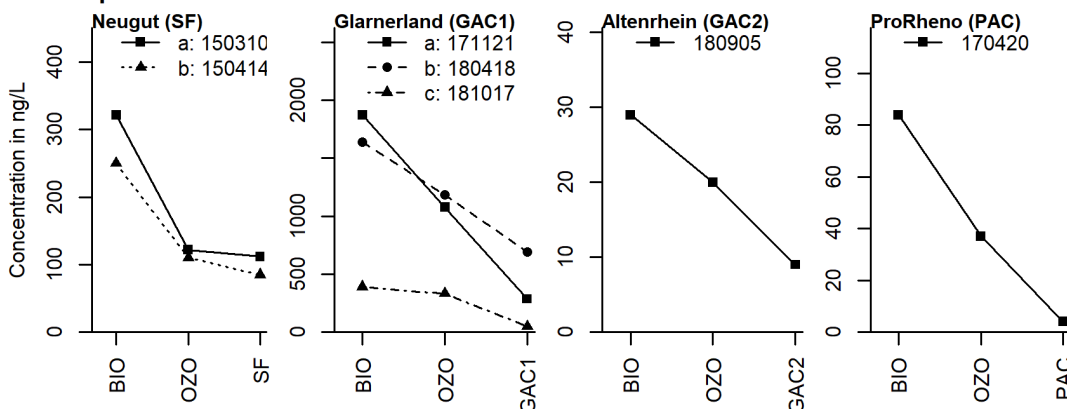
OTP formation in ozonation batch exp.



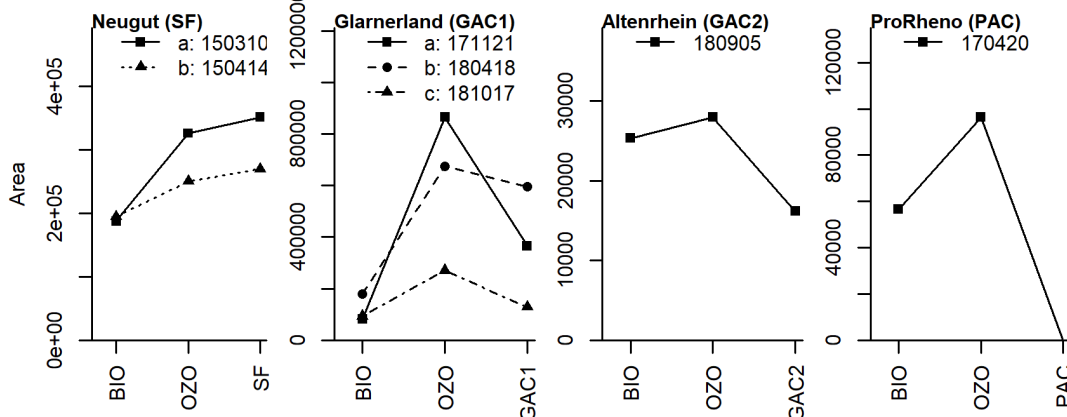
Fate of OTP in post-treatment



parent in WWTPs



OTP in WWTPs



Score and evaluation of MS2 spectra match of WWTP OZO samples to batch sample

a: 150310: 0.99(+++) a: 171121: 0.99(+++) 180905: 0(--)
b: 150414: 0.99(+++) b: 180418: 0.99(+++) 170420: 0.96(+++)
c: 181017: 1(+++)

MS2 spectra match sufficient: y

MS Spectra

Pos 450.2132 [m+H]⁺
Neg 448.1988 [m-H]⁻

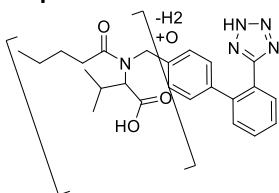
Formula

C₂₄H₂₇N₅O₄

Atomic modification

-H₂ + O

Proposed Structure



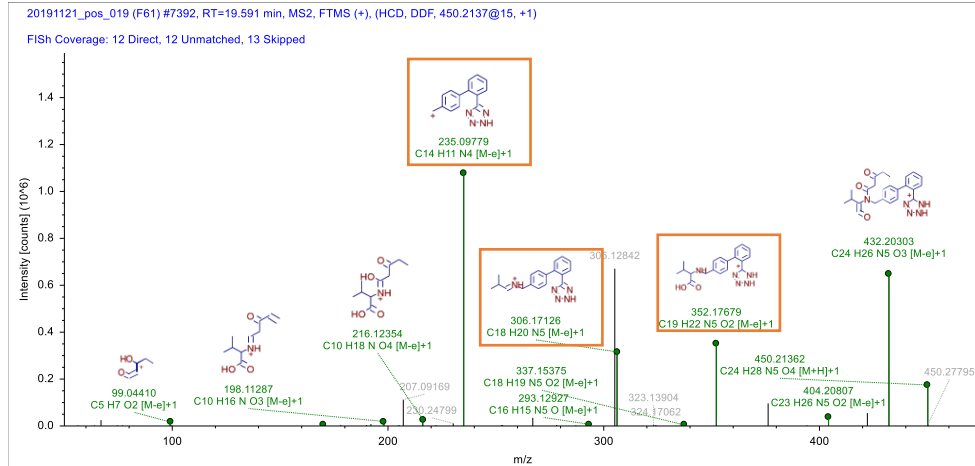
Confidence Level

Level 3

Massbank ID

ET404301

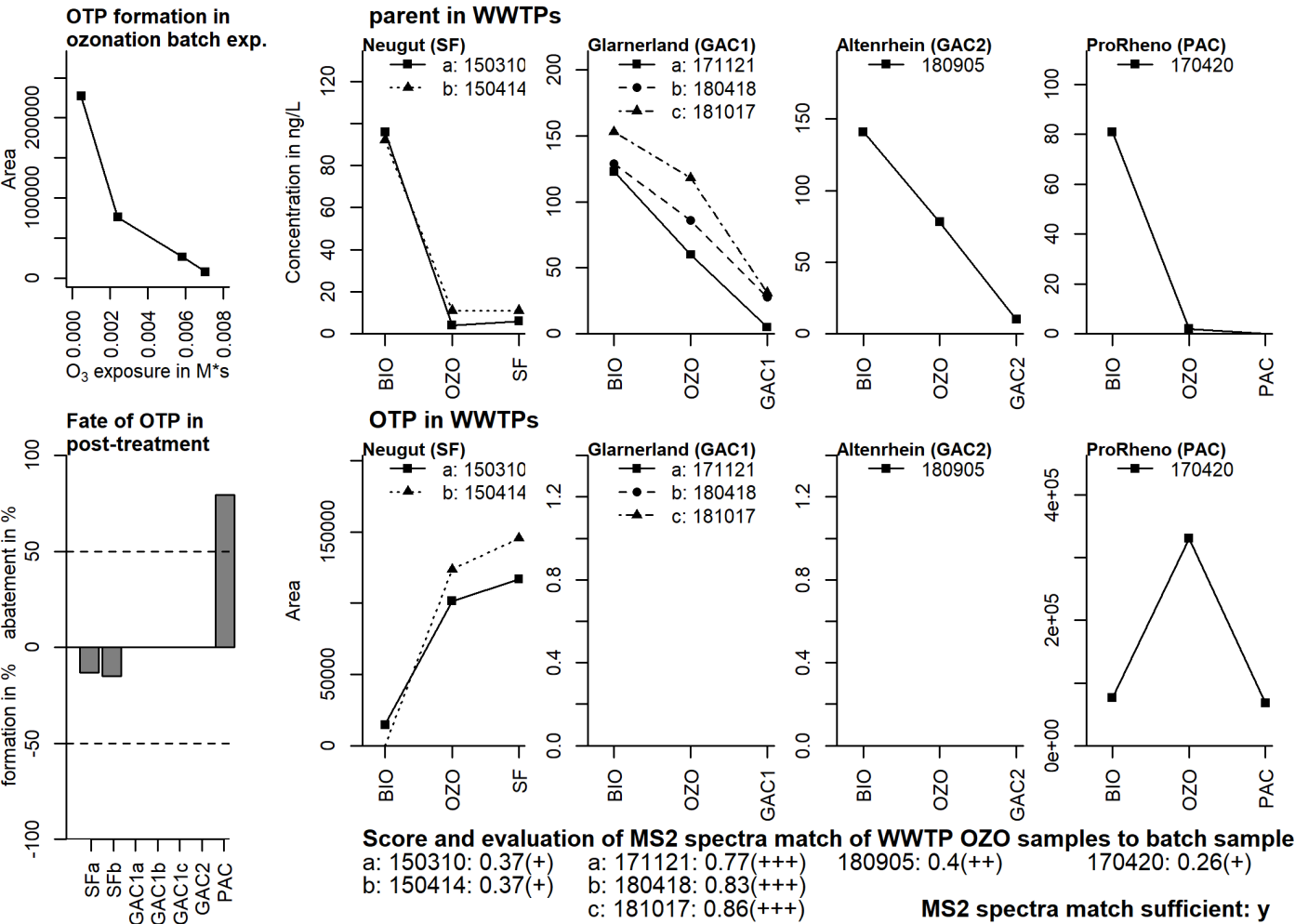
MS2 Spectra



Additional Evidence for Structure Interpretation

The MS² fragments at the nominal masses 352, 306 and 235 were also observed for the parent compound. They indicate that the structure drawn for the fragment 352 is the substructure of this OTP.

Fragment 235 indicates the modification(s) did not take place at the methylbiphenyltetrazole moiety. The modification -H₂ + O fits to the formation of a carbonyl group or an oxidation to a double bond with addition of an OH group, as suggested by Diehle et al. (2019). Diehle et al. (2019) detected two OTPs of valsartan with the same exact mass of 449. The exact type and location of the modification(s) remain unknown. The structures of the MS² fragments are drawn exemplarily.

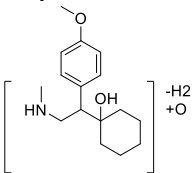


MS Spectra
Pos 278.1749 [m+H]⁺

Formula
C₁₆H₂₃NO₃

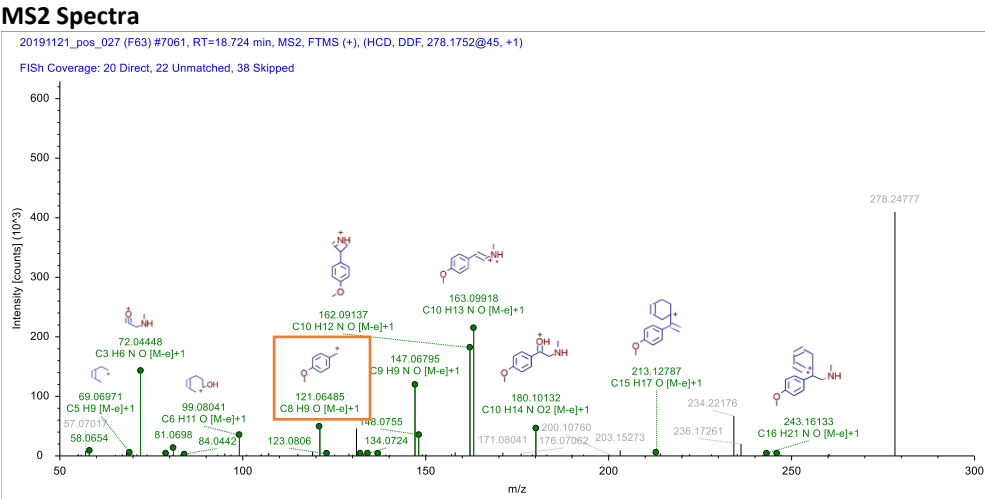
Atomic modification
+O -CH₄

Proposed Structure



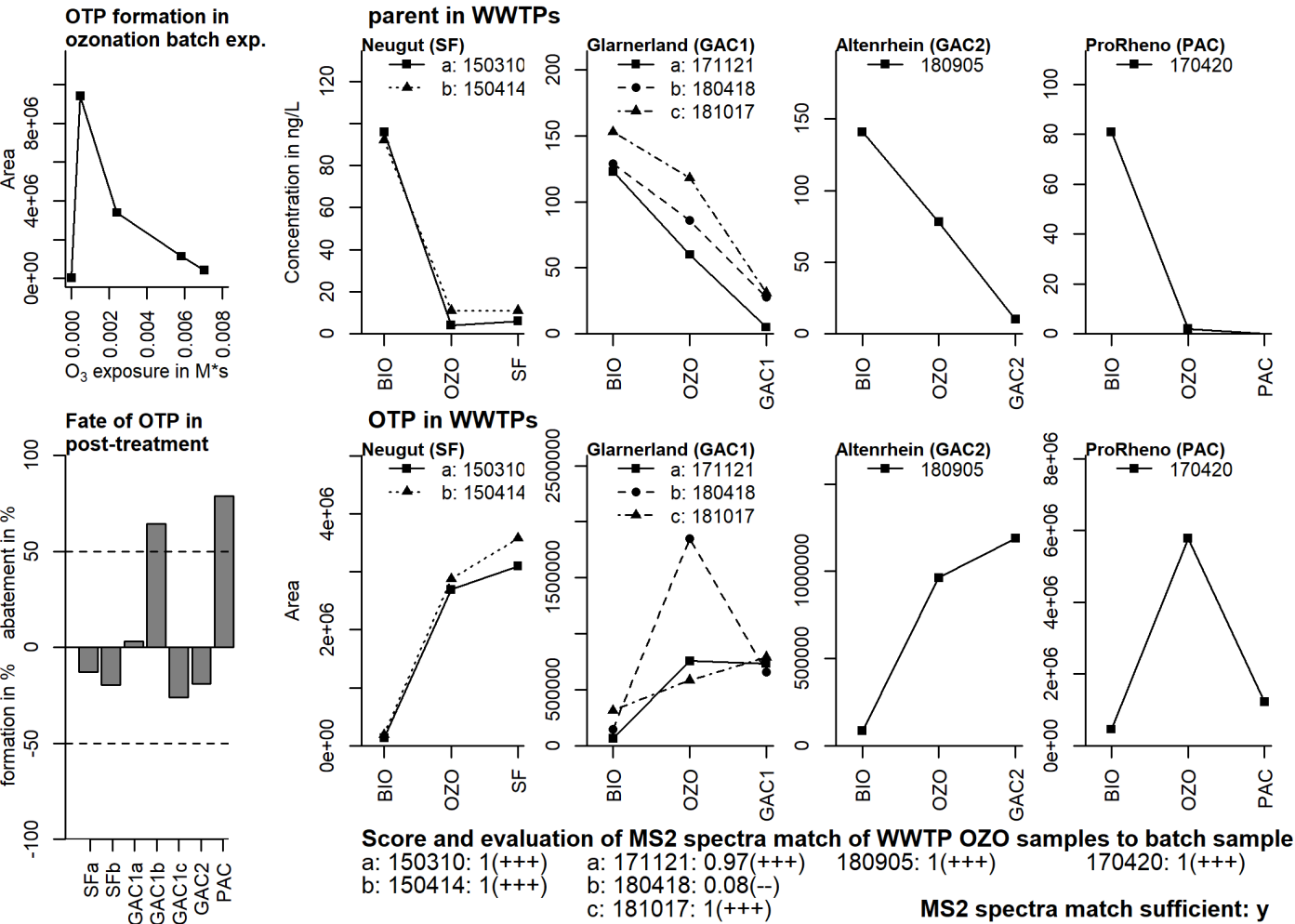
Confidence Level
Level 3

Massbank ID
ET404401



Additional Evidence for Structure Interpretation

The MS² fragments at the nominal mass 121 was also observed for the parent compound. It indicates that the structure drawn for the fragment 121 is a substructure of this OTP and that the methyl group of the methoxy moiety is conserved. *N*-desmethyl venlafexine was identified as minor OTP of venlafexine by Zucker et al. (2018). Therefore, is it likely that one methyl group was cleaved at the tertiary amine moiety. The remaining modification of -H₂ + O fits to the formation of a carbonyl moiety. Fragment 121 indicates the modification occurred at the methylaminomethylcyclohexanol moiety. However, there is no further indication for the exact type and location of the modification(s). The structures of the MS² fragments are drawn exemplarily.

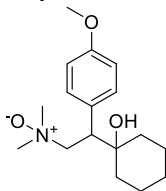


MS Spectra
Pos 294.2064 [m+H]⁺

Formula
C₁₇H₂₇NO₃

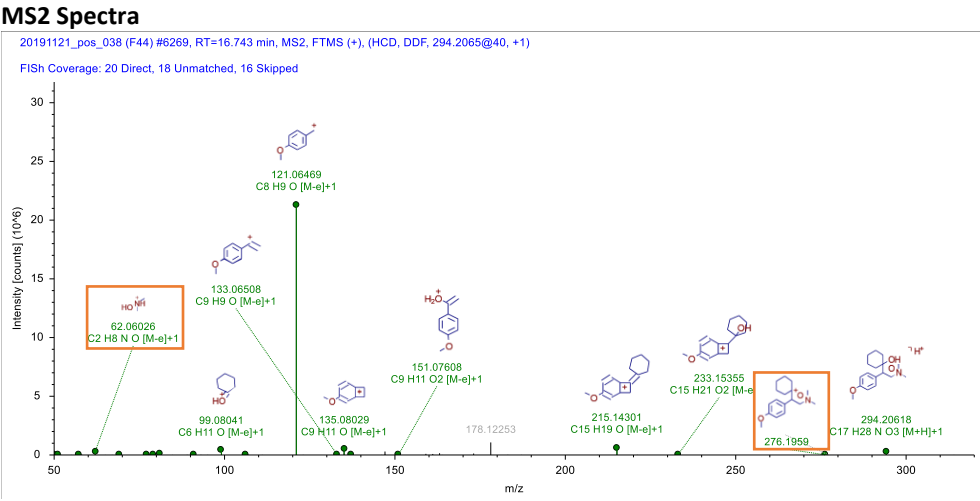
Atomic modification
+O

Proposed Structure



Confidence Level
Level 1

Massbank ID
ET404501



Additional Evidence for Structure Interpretation

The MS² fragments at the nominal masses 215, 135 and 121 were also observed for the parent compound. The MS² fragment 276 corresponds to the fragment 260 of the parent with an addition of an oxygen atom.

Tertiary amines are very reactive to form *N*-oxides during ozonation (von Sonntag & von Gunten 2012). Therefore, the formation of a *N*-oxide at the trimethylamine moiety is very likely. Fragment 62 indicates the formation of the *N*-oxide.

Venlafloxine *N*-oxide has been identified as major OTP of venlafloxine in different studies (Merel et al. 2017, Zucker et al. 2018, Lester et al. 2013, Lajeunesse et al. 2013). The MS² spectrum here matches the reference spectrum we recorded for venlafloxine *N*-oxide.