

Research Paper



Application of mid-infrared spectroscopy for the quantitative and qualitative analysis of organic matter in Holocene sediment records

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Abstract

The organic matter composition of lake sediments influences important in-lake biogeochemical processes and stores information on environmental changes. Extracting this information is notoriously difficult because of the complexity of the organic matter matrix, which routinely imposes trade-offs between high temporal and analytical detail in the selection of methods of analysis. Here, we demonstrate the potential of diffuse reflectance Fourier transform infrared spectroscopy (DRIFTS) for achieving both of these objectives using untreated bulk samples from two Holocene lake-sediment cores from central Sweden. We develop quantitative models for sediment total organic carbon (TOC) with the same predictive abilities as models based on samples diluted with KBr and qualitatively characterize the organic matter using a spectra processing-pipeline combined with principal component analysis. In the qualitative analysis we identified four organic matter sub-fractions and the interpretation of these is supported and further advanced with molecular data from pyrolysis-gas chromatography/mass spectrometry (Py-GC/MS). Within these organic fractions, compound groups such as aromatics, lignin, aliphatics, proteins and polysaccharides were identified by means of DRIFTS and the analyses and processes outlined here enables rapid and detailed quantitative and qualitative analysis of sediment organic matter. The DRIFTS approach can be used as stand-alone method for OM characterization with high temporal resolution in Holocene sediment records. It may also function as a screening process for more specific analyses of sample subsets, such as when coupled with pyrolysis-GC/MS to further tease apart the OM composition, identify sources and determine degradation status.

Keywords

DRIFTS, Holocene, lake sediment, mid-infrared spectroscopy, organic matter quality

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Introduction

In the analysis of lake sediment as well as other environmental archives spanning hundreds to thousands of years or longer, we are often confronted with the fundamental challenge of balancing the practical costs of high sampling (i.e. temporal) resolution against the costs of high analytical detail. The first – high sampling resolution – is essential to determine the timing and rates of sedimentary changes, the second – high analytical detail – is essential for developing specific insights into compositional changes and biogeochemical processes, and the balance between the two is dictated by the study questions.

Few analytical techniques are able to provide both high sampling resolution and analytical detail. For the geochemical composition of the sediment, core scanning may offer such solution by providing both a very high resolution (0.2 mm) and semi-quantitative data (e.g. Kylander et al., 2011). For the organic fraction we still largely rely on analyses of discrete samples to obtain information on organic matter (OM) composition, using common methods ranging from inexpensive but less-detailed methods such as % loss on ignition, total organic carbon (TOC) and nitrogen (TON) analysis (Hu et al., 2001), and Rock-Eval pyrolysis (Wohlfarth et al., 2004) to time-consuming and costly, but more detailed lipid biomarker analysis (Holtvoeth et al., 2010), solid-state ¹³C nuclear

magnetic resonance spectroscopy (Orem et al., 1991) and mass spectrometry-based methods (Ninnes et al., 2017).

One analytical method that holds potential for both high sampling resolution and analytical detail of the OM fraction is midinfrared spectroscopy (MIR). Sample preparation for MIR is rapid; for attenuated total reflectance (ATR) spectroscopy sample preparation only involves drying and grinding samples before analysis and for diffuse reflectance infrared Fourier transform

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spectroscopy (DRIFTS) typically also dilution with a powdered alkali halide transparent in the infrared region, such as potassium bromide (KBr). The purpose of diluting samples with KBr is to reduce specular reflectance distortion, which otherwise may result in reststrahlen bands with regions of strong absorption and reflection, such as the band reversal common in untreated, Si-Orich samples <1200 cm⁻¹ (Nguyen et al., 1991). DRIFTS can, however, just like ATR spectroscopy, also be performed on untreated samples (Alaoui et al., 2011), which reduces the preparation time of DRIFTS further and does not consume sample material. With an analysis time of only a few minutes per sample, approximately 100-150 samples can be processed daily with ATR, and approximately 100 samples daily with DRIFTS if accounting for packing and emptying of sample cups and the analysis, and up to 300 samples daily if accounting only for the analysis. With minimal maintenance and consumable costs (e.g. liquid N coolant for DRFTS), high sampling resolution is certainly achievable.

High analytical detail can be achieved due to the highly resolved character of the spectral information (Kubo and Kadla, 2005). MIR collects information on the frequencies of molecular vibrations in samples over the spectral range ~4000 to 400 cm⁻¹ (2.5–25 μm). Frequencies of these vibrations are linked to specific molecular structures and can therefore be assigned to absorption bands of molecules from different functional groups or compound classes. Published literature on such spectra-structure correlations for band interpretation is widely available, spanning from individual compounds to various complex matrices, which allows for comprehensive information extraction. The amount of extractable detail can, however, be affected by the sample preparation approach. For organic samples, where specular reflectance is expected to be low or absent, the use of untreated samples may be advantageous because the higher sample concentration enhances relatively weak absorption bands (Nguyen et al., 1991). Additionally, MIR spectra are highly reproducible over time on the same instrument (Aryee et al., 2009; Hofko et al., 2017), and between similar instruments assuming appropriate sample and data handling (Hofko et al., 2018; Mirwald et al., 2022).

The main challenge with the MIR technique is linked to data interpretation, more specifically to discriminating between molecules with functional groups that absorb IR radiation at similar frequencies, so-called band overlap (Tinti et al., 2015). Band overlap occurs for all complex matrices and may prevent identification of individual compounds or even groups of compounds. A considerable portion of the MIR spectra complexity can be handled with multivariate chemometric analysis, such as partial least squares regression (PLSR) and principal component analysis (PCA) (e.g. Alaoui et al., 2011; Martinez Cortizas et al., 2021b; Pérez-Rodríguez et al., 2016; Rosén et al., 2010). These methods can resolve many issues linked to band overlap through identification of dependence, or patterns, among multiple variables. For example, Martinez Cortizas et al. (2021b) provide a detailed compositional analysis of a peat sequence using a semiautomated peak selection processing pipeline (Álvarez Fernández and Martínez Cortizas, 2020) coupled with PCA, and verify their analysis against independent molecular OM data obtained with pyrolysis-gas chromatography/mass spectrometry (Py-GC/ MS). For lake sediment, the most common application so far is multivariate prediction models for quantification of major and minor sediment components based on DRIFTS, including total organic and inorganic C, biogenic silica (bSi), charcoal and common minerals (Brisset et al., 2013; Cadd et al., 2020; Hahn et al., 2013, 2018; Meyer-Jacob et al., 2014; Rosén et al., 2010, 2011; Vogel et al., 2008). Only more recently has multivariate chemometric analysis been used for extraction of qualitative information on OM composition (Maxson et al., 2021; Pérez-Rodríguez et al., 2016).

Our objective here is to evaluate DRIFTS on untreated samples as a time- and cost-efficient method to facilitate high-resolution analyses of changes in OM composition in Holocene lake-sediment cores. To do so we address the following aims: (1) quantify the major organic sediment fraction, TOC. For this purpose we use surface sediment samples from a large number of lakes in northern Sweden to develop a general northern Swedish calibration model, subsequently used to predict TOC content in two Holocene sediment cores. One of these records is then used to develop a local calibration model and the performance of the two models compared and (2) determine the qualitative OM composition using wavenumber assignment based on known information about MIR spectra. To support our evaluation of the compositional information indicated by DRIFTS, we analyze the same two Holocene lake-sediment cores that we have previously studied using Py-GC/MS (Ninnes et al., 2017).

Materials and methods

Samples

The training set used to develop the general calibration model constitutes 93 lake surface sediment samples from a northern Swedish training set (Bigler and Hall, 2002; Rosén et al., 2011). These lakes are found at 169–1183 m a.s.l., with catchment vegetation ranging from boreal forest to alpine/tundra vegetation and even barren ground. Surface areas are all <20 ha and maximum depths range from 1.5 to 16 m (Bigler and Hall, 2002).

The two Holocene sediment cores come from lakes Dragsjön and Lång-Älgsjön; both small headwater lakes located in the southern boreal zone of central Sweden (60°23′60″N, 15°7′23″E; Figure S1). These neighboring lakes are situated ~430 m a.s.l. and both have surface areas of 5 ha. Dragsjön has a maximum depth of 6 m and a catchment to lake ratio of 22, while Lång-Älgsjön has a maximum depth of 12 m and catchment to lake ratio of 13. The surrounding vegetation is dominated by *Pinus sylvestris* and *Picea abies* forest on shallow, nutrient-poor glacial till. Both lakes are fringed by floating *Sphagnum* mats and in direct connection with extensive blanket peatlands. Dragsjön was sampled in 2014 and Lång-Älgsjön in 1998. Dating and initial age-depth modeling were done by Ninnes et al. (2017) and Meyer-Jacob et al. (2015) for Dragsjön and Lång-Älgsjön, respectively, and both age-depth models updated by Myrstener et al. (2021).

Additionally, five reference samples were analyzed for comparison with published spectra in the literature to verify that spectral features necessary for accurate compound assignment can be identified in the DRIFT spectra of untreated samples. These samples include three different types of organic materials (wood, peat, and green algae), bSi isolated from marine sediment and a Na-rich montmorillonite clay (Table S1).

DRIFT spectroscopy

Surface, core and reference samples were all scanned untreated (no KBr added) using a Bruker Vertex 70 equipped with a N-cooled Mercury–Cadmium–Telluride detector, a KBr beam splitter and a HTS-XT accessory unit. Each sample was scanned 128 times at a resolution of $4\,\mathrm{cm^{-1}}$ for wavenumbers $4000–400\,\mathrm{cm^{-1}}$. The analysis was performed in a temperature-controlled laboratory $(25.0\pm0.5^{\circ}\mathrm{C})$, in which the samples were allowed to equilibrate in a desiccator for at least 12h prior to the measurement. A gold plate was used as background and a total of 97 and 91 samples were analyzed for Dragsjön and Lång-Älgsjön sediment cores, respectively. Additionally, four of the reference samples were scanned both untreated and diluted with KBr (5.5 mg sample to 250 mg spectroscopic-grade KBr), and 61 samples from Dragsjön scanned diluted with the purpose of quantifying sediment bSi

content using the model by Meyer-Jacob et al. (2014). For all diluted samples KBr was used as background.

TOC content

For the surface samples from the Northern Swedish training set, TOC content was analyzed by Rosén et al. (2011), using a Perkin Elmer 2400 series II CHNS/O elemental analyzer operated in CHN mode. TOC for these samples ranges from 0.7% to 36.0% (dry mass), with a mean of $19\pm8\%$ (Table S2). The TOC contents for Dragsjön (n=58, TOC 2.7–33.6%) and Lång-Älgsjön (n=29, TOC 0.9–43.1%) were analyzed by Ninnes et al. (2017) and Meyer-Jacob et al. (2015), respectively, on a Flash 2000 Organic Elemental Analyzer (Thermo Fisher Scientific) at the Swedish University of Agricultural Sciences (SLU) in Umeå. Carbonates were not present in either data sets based on inspection of DRIFT spectra and samples therefore not pre-treated with acid before elemental analysis.

Py-GC/MS

The Py-GC/MS analyses of the two lake-sediment cores were performed by Meyer-Jacob et al. (2015) and Ninnes et al. (2017). Briefly, $200 \pm 20 \,\mu g$ of sediment was pyrolyzed in a Frontier-Labs PY-2020iD oven (450°C) connected to an Agilent 7890A-5975C GC/MS system. Peak integration and identification were performed using a data processing pipeline under the "R" computational environment (Tolu et al., 2015). Compounds were identified using the software "NIST MS Search version 2.0" containing the library "NIST/EPA/NIH 2011" and additional spectra from published studies (Tolu et al., 2015). Because the records were analyzed during different sessions, data are reported as relative abundances, that is, for each sample, the peak area of each identified compound was normalized against the summed peak area for of all identified compounds. Data was handled in this way because Py-GC/MS can be associated with strong instrument sensitivity drift with time, meaning that peak area of detected compounds for samples analyzed during different sessions cannot be directly compared. Further details about the method can be found in Tolu et al. (2015) and Ninnes et al. (2017). Data used in this study include 145 compounds, here grouped into 28 compound groups (Table 1), which have been identified in both lake-sediment cores by Ninnes et al. (2017), but here with the additional inclusion of the compound group fatty acids (Table S3).

Statistical analyzes

A first step in our analyses was to confirm that undiluted DRIFT spectroscopy can be used to estimate the TOC content in lake sediments. For this initial proof-of-concept we first developed a general calibration model using DRIFT spectra of untreated samples and the corresponding TOC concentrations in the 93-lake northern Swedish training set (Bigler and Hall, 2002; Rosén et al., 2010). By building a model on samples from a large number of lakes it is possible to compare the performance of our model to similar, existing models but which are based on spectra from samples diluted with KBr. For external validation, we applied the general northern Swedish model to the Holocene sediment cores from Dragsjön and Lång-Älgsjön. To demonstrate that our approach also performs well for site-specific conditions the Dragsjön core was then used to develop a local calibration model that was externally validated using the Lång-Älgsjön core.

Both TOC calibrations models were developed with PLSR in SIMCA 16.0 (Umetrics, Umeå, Sweden) using linearly baseline-corrected (3800–3750 and 2200–2210 cm⁻¹, Opus Interface) DRIFT spectra (3750–400 cm⁻¹). For the general northern Swedish model all data were centered. For the local model data were

Table 1. List of the 28 pyrolytic organic compounds groups and the individual organic compounds within each group (n = 144).

Acetic acid

Alkylfurans

3-furaldehyde

2-furaldehyde

Methyl-2-furaldehyde

Methyl-2-furaldehyde

Anhydrosugars

Levosugar (levogalactosan?)

Anhydrosugar

Levosugar (Levomannosan)

Levoglucosan

Dianhydrorhamnose

Levoglucosenone + Maltol

I,4:3,6-Dianhydro- α -d-glucopyranose

Other polysaccharides (O. polysaccharides)

I-hydroxy-2-propanone

3(2H)-furanone

2(3H)-furanone, 5-methyl

4-cyclopentene-1,3-dione

2-cyclopenten-I-one, 2-methyl-

2(5H)-furanone

Dihydro-methyl-furanone

4-hydroxy-5,6-dihydro(2H)-pyran-2-one

2-hydroxy-3-methyl-2-cyclopenten-I-one

4-hydroxy-2,5-dimethyl-3(2H)-furanone

2,3 dihydro-benzofuran

Acetamidosugars

3-acetamido-furan

3-acetamido methylfuran

Acetamide

Alkanenitriles

Alkanenitrile (C20)

Alkanenitrile (C22)

Alkanenitrile (C24)

Alkylamides

Alkylamide I

Alkylamide2

Alkylamide3

Alkylamide4

Alkylpyrrols

Pyrrole

Methyl-pyrrole

Methyl-pyrrole

2-formyl-pyrrole

2-acetyl-pyrrole

Diketopiperazines (DKPs)

Diketodipyrrole/ DKP Pro-Ala

DKP Pro-Val

DKP Leu-Pro

DKP Leu-Pro

DKP Leu-Pro

DKP Pro-Pro

DKP Pro-Lys-NH₃

DKP Pro-Phe

Other N compounds

2-methyl-pyridine

Pyridine

Maleimide

Succinimide

Benzeneacetonitrile

Benzenepropanenitrile

Indole

Methyl-indole

Oxazoline structure

9H-Pyrido[3,4-b]indole

Table I. (Continued)

4-formyl-syringol

```
n-alkenes
  n-C17:1
  n-C18:1
  n-C19:1
  n-C20:1
  n-C21:1
  n-C22:1
  n-C23:1
  n-C24: I
  n-C25:1
  n-C26:1
  n-C27:1
  n-C28:1
n-alkanes C17-19 (C17-19:0)
  n-C17:0
  n-C19:0
n-alkanes C18-28 (C18-28:0)
  n-C18:0
  n-C20:0
  n-C22:0
  n-C24:0
  n-C26:0
  n-C28:0
n-alkanes C21-25 (C21-25:0)
 n-C21:0
  n-C23:0
  n-C25:0
n-alkanes C27-33 (C27-33:0)
  n-C27:0
  n-C29:0
  n-C31:0
  n-C33:0
n-alkan-2-ones C17-19 (2K C17-19)
  n-C19
n-alkan-2-ones C21-25 (2K C21-25)
  n-C21
  n-C23
  n-C25
n-alkan-2-ones C27-33 (2K C27-33)
  n-C27
  n-C29
  n-C33
Fatty acids
  C14:0
  C15:0
  C16:0
  C22:0 methyl ester
  C24:0 methyl ester
Phenols
  Phenol
  2-methyl-phenol
  Acetophenone
  Dimethyl-phenol
  4-methyl-phenol
  Ethyl-phenol
4-isopropenylphenol
Lignin
  Guaiacol
  Methyl-guaiacol
  Ethyl-guaiacol
  4-vinyl-guaiacol
  Vanillin
  Syringol
  4-vinyl-syringol
```

```
Table I. (Continued)
```

```
Pristene
  Prist-I-ene
  Prist-2-ene
Phytadiene
  Phytadiene I
  Phytadiene 2
  Phytadiene 3
Steroids
  Cholest-2-ene
  Cholesta-4,6-dien-3-ol, (3β)-
  Cholesta-3,5-diene
  Stigmasta-5,22-dien-3-ol, acetate, (3β)-
  Stigmastan-3,5-diene
  Cholesta-3,5-dien-7-one
  Cholest-4-en-3-one or Stigmast-4-en-3-one
  Stigmasterol
  Sitosterol
  Stigmasta-3,5-dien-7-one
  Stigmast-4-en-3-one
Tocopherol
  γ-Tocopherol
  \alpha-Tocopherol
Hopanoids
  22,29,30-trisnorhop-17(21)-ene
  Urs-20-en-16-one
  Hop-17(21)-ene
  Diploptene
(Poly)aromatics
  Toluene
  Styrene
  2.3-dihydro-inden-I-one
  Benzene C9
  Benzene C12
  Benzene C14
  Retene
  Benzene C16
  Benzene C17
  Benzene C18
  Benzene C19
```

Further information on these pyrolytic compounds are given in Tolu et al. (2015) and Ninnes et al. (2017).

centered and standardized to unit variance (z-scored). The internal predictive performance of the models, that is, how well the DRIFT spectra predicts TOC content for the training sets, was evaluated with the coefficient of determination from seven-fold cross-validation (R^2_{CV}) and with the root mean squared error of cross-validation (RMSECV, in % TOC). The external predictive ability of the models, that is, how well the models predict TOC content when applied to samples not used in the model, but with known TOC contents, was evaluated with the coefficient of determination (R^2) and the root mean squared error of prediction (RMSEP, in % TOC).

For analysis of the qualitative OM sediment composition in the cores from Dragsjön and Lång-Älgsjön the main spectral peaks within the range 3750–400 cm⁻¹ were identified with the andurinha R application (Álvarez Fernández and Martínez Cortizas, 2020). This application provides the average, standard deviation and second derivative spectra, with the latter used for extraction of the main spectral peaks from baseline corrected spectra (3800–3750 and 2200-2210 cm⁻¹) for each sediment record. The main DRIFT spectral signals within the reduced spectra were extracted with PCA on standardized absorbance data in the R package "Psych" with a varimax rotation applied to maximize the variable loadings on the components. Principal component analysis was also applied

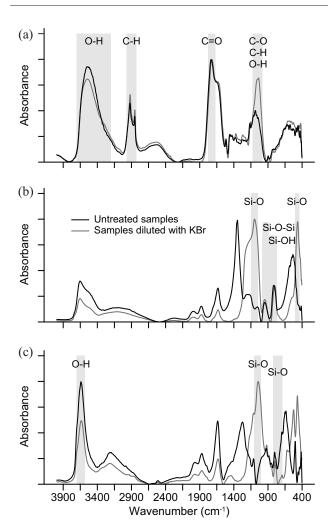


Figure 1. DRIFT absorbance spectra for reference samples (a) Sphagnum peat (CRM NJV 942), (b) biogenic silica, (c) Montmorillonite clay (SWy-2). Untreated samples are shown in black and samples diluted with KBr in gray. Data are baseline corrected (adaptive, coarseness 15, offset 0), smoothed (Savitsky–Golay, interval 10, polynomial order 3) and peak normalized (SpectraGryph v1.2.14; Menges 2016–2020) for visualization purposes. Vertical gray bars refer to absorbance bands mentioned in the text.

to the 28 pyrolytic compound groups to enable comparison between DRIFTS and pyrolytic OM composition. The comparison was done using bivariate correlation coefficients between respective component scores and passively plotting DRIFT spectra components in the pyrolytic PCA loading plots. All principal components retained for the spectral and pyrolytic PCAs have eigenvalues >1.

Results and discussion

Reference DRIFT spectra of untreated samples and samples diluted with KBr

DRIFT absorbance spectra for the untreated reference samples show good coherence with published spectra for similar types of samples but which have been diluted with KBr. Untreated spectra for decomposed *Sphagnum* peat (Figures 1a and S2) show characteristic bands for O–H stretching of phenolic and aliphatic groups (3600–3200 cm⁻¹); C–H stretching of aliphatics (2925–2850 cm⁻¹, Niemeyer et al., 1992); C=O stretching of esters, aldehydes, ketones and carboxylic acids (1750–1670 cm⁻¹, Marchessault, 1962), and C–O and O–H stretching of aliphatic alcohols and C–H deformation of aromatics (1155–1000 cm⁻¹, Orem et al., 1996; Pandey, 1999; Zaccheo et al., 2002). The only difference in

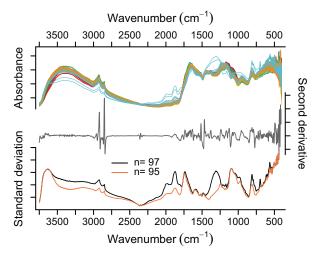


Figure 2. Relative absorbance, summed second derivative and standard deviation for lake sediment samples from Dragsjön. Standard deviation spectra are calculated for all samples (n=97) and with the basal samples (two light blue lines in top panel) removed (n=95).

the DRIFT spectra between the untreated and the diluted samples is the slightly lower intensity of the broad band at 1120–1000 cm⁻¹, but specular distortion effects are effectively absent. For bSi, the DRIFT spectra of untreated and diluted samples are near identical above 1500 cm⁻¹ and below this region characteristic bands linked to Si-OH stretching (947 cm⁻¹) and Si-O-Si stretching (~800 cm⁻¹, Farmer, 1974) are easily identified (Figures 1b and S5). The DRIFT spectra of the untreated sample deviate as expected in the 1400-1000 and 700-400 cm⁻¹ regions. Here, specular distortion in the form of band reversal (1270–1000 cm⁻¹ and 500-400 cm⁻¹) and Fresnel peaks (1350 and 530 cm⁻¹) mask the broad bands linked to Si-O stretching (1100 cm⁻¹) and Si-O bending (470 cm⁻¹), which are present in the DRIFT spectra of the sample diluted with KBr. The same distortion effects are present in the DRIFT spectra of untreated samples of the Na-rich montmorillonite clay, with band reversals at 1070, 530, and 470 cm⁻¹ and Fresnel peaks at 1270 and 650 cm⁻¹ (Figures 1c and S6). However, bands necessary for identification of silicate mineral phases remain intact, including bands linked to Si-O stretching (802/786 cm⁻¹ doublet) and bending (695 cm⁻¹) of quartz and feldspars (Farmer, 1974), and the strong band between 3700-3600 cm⁻¹ characteristic of structural O-H vibrations in clays (Madejova and Komadel, 2001). Absorbance and second derivative spectra, peak positions, peak assignments and references for peak assignments for all five reference samples are presented in figures S2-6.

DRIFT spectra of untreated lake-sediment core samples

Absorbance spectra for Dragsjön (Figure 2) and Lång-Älgsjön (Figure S7) are very similar and characterized by high absorbance in four broad wavenumber regions. The first region, 3600–3000 cm⁻¹, corresponds to O–H, N–H and C=C stretching of organics, such as polysaccharides, phenols, alcohols and proteins (Shurvell, 2006), as well as to O–H stretching in bSi, silicate minerals and freely adsorbed water (Farmer, 1974). The second region, 2950–2850 cm⁻¹, corresponds to C–H stretching of aliphatics and aromatics (Boeriu et al., 2004; Niemeyer et al., 1992), and the third region, 1700–1000 cm⁻¹, to vibrations of a range of functional groups including carbonyl C=O vibrations in organic acids and proteins; C=C, C=N and N–H vibrations of aromatics and proteins; C–O–C and C–OH of alcohols, ethers and sugars, and C–H deformation of aliphatics (Shurvell, 2006). The fourth

Table 2. Statistical performance of the general Northern Swedish and the local PLSR calibration models for sediment total organic carbon
(TOC) resulting from internal (calibration set samples) and external (validation set samples) validation.

	General model			Local model	
	Calibration set	External validation set		Calibration set	External validation set
		Dragsjön	Lång-Älgsjön	Dragsjön	Lång-Älgsjön
PLS components	5			3	
Samples (n)	93	58	29	58	29
Min (TOC %)	1.78	2.7	0.9	2.7	0.9
Max (TOC %)	36.0	33.6	43.1	33.6	43.1
Mean (TOC %)	19.9	28.0	29.9	28.0	29.9
SD (TOC %)	7.5	5.1	10.3	5.1	10.3
R^2_{CV}	0.96			0.96	
RMESCV (TOC %)	1.7			2.6	
RMSECV (% TOC gradient)	5.0			8.2	
R^2		0.96	0.97		0.98
RMESP		2.0	4.1		2.6
RMSEP (% TOC gradient)		6.4	9.7		6.2

region, 750–400 cm⁻¹, corresponds to vibrations of polysaccharides, aliphatics, and minerals such as silicates and iron oxides (Farmer, 1974; Shurvell, 2006). Each of the two Holocene sequences includes two basal samples characterized by postglacial, silty clay with low (\sim <5% TOC) organic contents. These samples have a peak at \sim 3620 cm⁻¹, linked to structural O–H vibrations in clay (Madejova and Komadel, 2001), Si–O overtone and combination bands at 2000–1800 cm⁻¹ (Nguyen et al., 1991), and characteristic Si–O bending and stretching peaks at 697, 772, 790, and 810 cm⁻¹ (Farmer, 1974). Standard deviation spectra calculated for all Dragsjön samples (n=97) and with the basal samples removed (n=95) clearly highlight these differences.

Sediment TOC quantification with DRIFT spectra from untreated samples

Our models for TOC quantification in lake sediments use the common DRIFTS technique coupled with PLSR, but contrast with previous work on lake sediments by using untreated samples for analysis. The general calibration for sediment TOC based on DRIFT spectra and TOC content for the northern Swedish training set (n=93) resulted in a 5-component model with an $R^2_{\rm CV}$ of 0.96 between the quantitative TOC measurements obtained with conventional elemental analysis and the PLSR-predicted TOC content, and a prediction error (RMSECV) of 1.7% (Table 2 and Figure 3). The first three latent components capture 89% of the variation in the spectral signal (DRIFT spectra, x-variables) and 94% in the TOC (y-variable). Loadings on components 1–3 expressed as weight vectors (w*c) show that wavenumbers 3600–2250, 1780–930, and 600–400 cm⁻¹ are strongly positively correlated to TOC.

The local calibration model based on DRIFT spectra and TOC content for Dragsjön (n=58) resulted in a 3-component model with an $R^2_{\rm CV}$ of 0.96 between the quantitative TOC measurements obtained with conventional elemental analysis and the PLSR-predicted TOC content, and a prediction error (RMSECV) of 2.6% (Table 2, Figure 3). The first two latent components capture 78% of the variation in the DRIFT spectra and 92% of the TOC. Loadings on components 1 and 2 show that wavenumbers 3500–3250, 3000–2235, 2214–2200, 1800–1640, and 1150–1026 cm⁻¹ are strongly positively correlated to TOC. Thus, for both models C–H stretching and overtones in aliphatics at 3000–2650 cm⁻¹; C=O stretching in carbonyl groups (1800–1640 cm⁻¹), and C–O and O–H stretching and deformation in polysaccharides, cellulose and lignin (1180–995 cm⁻¹) are important regions. The same three regions underpin calibration models for TOC and OM in soils and

lake sediments based on samples diluted with KBr (Janik and Skjemstad, 1995; Rosén et al., 2010, 2011; Viscarra Rossel et al., 2006; Vogel et al., 2008), with the 3000–2800 cm⁻¹ region particularly important for the quantification of OM, TOC and humic substances in estuarine sediments (Alaoui et al., 2011; Tremblay and Gagné, 2002).

When the general northern Swedish model is applied to the sediment sequences from Dragsjön (R^2 =0.96, RMSEP=6.4% of gradient) and Lång-Älgsjön (R²=0.97, RMSEP=9.7% of gradient) the downcore trends are captured well, but concentrations are slightly underestimated for most samples (Figure 3). These underestimations likely occur because the samples in the northern Swedish training set to some extent are qualitatively different from the sediments in Dragsjön and Lång-Älgsjön, despite the comparable quantitative TOC ranges. These differences are captured by the contribution of each wavenumber to the PLSR models. While the same general regions are important for both models, there are notable differences particularly within the 3500-3000, 2215–2070, 1600–1200 cm⁻¹ regions (Figure 3). Accordingly, the local model based on samples from Dragsjön more accurately predicts sediment TOC concentrations in Lång-Älgsjön ($R^2 = 0.98$, RMSEP=6.2), particularly for samples that fall within the range of concentrations that are well-represented in the calibration model (~23-34% TOC).

For both of our models, validation samples with measured TOC concentrations >34-36% are underestimated due to spectral saturation of these samples, which contributes to the higher RMSEP for Lång-Älgsjön than Dragsjön with the general northern Swedish model. The same issue was reported by Rosén et al. (2011) for samples diluted with KBr when TOC exceeded ~35%. When saturation occurs there is a deviation from the expected linear relationship between concentration and absorbance at the high end of concentrations, such that when the linear calibration model is applied to these samples, predicted TOC is underestimated. Here, dilution with alkali halides can be used to reduce band intensities and create models optimized for higher TOC ranges (Rosén et al., 2011). However, specular saturation will only be a problem for TOC analysis on untreated samples for lakes with highly organic sediment, such as those found in the upper part of the Lång-Älgsjön record, whereas most lake sediments will have TOC concentrations below 35%. For example, in the 93-lake northern Swedish training set only one lake exceeds 35% TOC, and in the 3164-lake training set spanning from polar desert to boreal forest environments used by Rosén et al. (2011), only ~1% of the lakes exceeded 35% TOC.

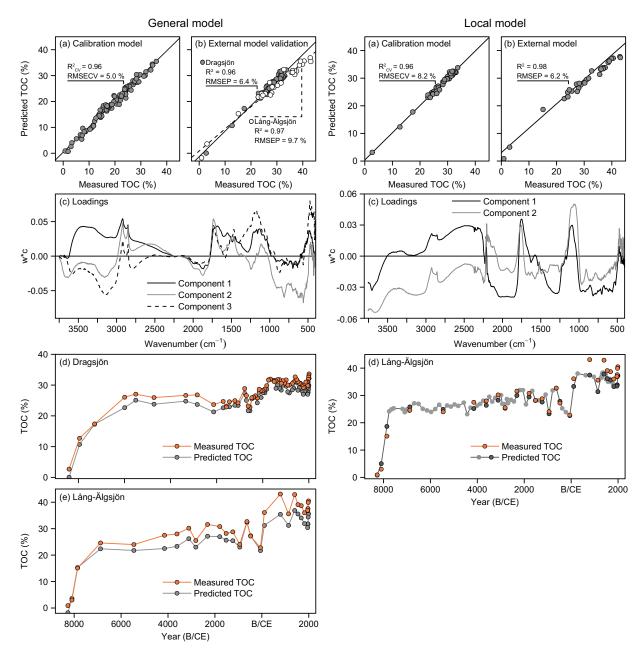


Figure 3. The general TOC calibration model (left) with measured sediment TOC concentrations versus DRIFTS predicted TOC concentrations for (a) the PLSR calibration model based on samples from the northern Swedish calibration training set (n=93); (b) the external model validation using samples from Dragsjön (n=58) and Lång-Älgsjön (n=29); (c) loadings on the first three components of the calibration model, expressed as weight vectors (w*c), where positive weight vectors indicate wavenumbers positively correlated to the TOC concentration and negative weight vectors wavenumbers negatively correlated to TOC concentration, down-core trends for measured and predicted TOC concentrations for the external validation for (d) Dragsjön, and (e) Lång-Älgsjön. The local TOC calibration model (right) with measured sediment TOC concentrations versus DRIFTS predicted TOC concentrations for (a) the PLSR calibration model based samples from Dragsjön (n=58); and (b) the external model validation using samples from Lång-Älgsjön (n=29); c) loadings on the first three components of the calibration model, expressed as weight vectors (w*c), where positive weight vectors indicate wavenumbers positively correlated to the TOC concentration and negative weight vectors wavenumbers negatively correlated to TOC concentration, and (d) down-core trends for measured and predicted TOC concentrations for Lång-Älgsjön, where orange and dark gray circles represent the validation samples (n=29) and light gray circles samples with model-predicted TOC only (n=62).

The performances of both models are nonetheless comparable to other calibration models that use DRIFT spectra, whether untreated estuarine sediment samples (Alaoui et al., 2011) or lake-sediment samples diluted with KBr (Vogel et al., 2008). The most comparable TOC ranges are those by Rosén et al. (2010, 2011), in which concentrations in lake sediment range from 0 to \sim 30–41% TOC, for a similar or higher number of samples (n=94–3164, $R^2_{\rm CV}$ =0.83–0.98, RMESCV gradient =6.1–7.7%). Our results verify that lake-sediment TOC concentrations can be estimated using DRIFT spectra of untreated samples but that the accuracy of estimated values may differ between general and more

lake-specific models. We expect that a more robust calibration model would result from a calibration data set that covers the entire qualitative sample range of the predictive data set, alternatively selecting a subset of samples for TOC analysis to develop a lake-specific calibration.

Qualitative sediment OM composition

Principal component analysis of lake sediment spectra. With help of the open-source *andurinha* R application, we identified 72 and 74 peaks from the second derivative of the DRIFT spectra for

Dragsjön and Lång-Älgsjön. Preliminary PCAs on the absorbance spectra for these peaks enabled identification of wavenumbers associated with each of the three main lake sediment fractions (OM, mineral matter and bSi). Groups of wavenumbers identified as capturing mineral matter (i.e. all wavenumbers with negative or positive loadings on a particular component) were subsequently removed in a couple of successive PCAs such that the final PCAs included only organic and bSi components. Minerogenic wavenumbers were removed to make the comparison with the organic pyrolytic data more intuitive and because approaches for qualitative and quantitative analysis of mineral matter with MIR has been done elsewhere (e.g. Hahn et al., 2018; Kanbar et al., 2021; Martínez Cortizas et al., 2021a). Their removal does not affect the order of the organic components nor which components the wavenumbers load on, with the exception of one wavenumber for Dragsjön and four for Lång-Älgsjön. Wavenumbers associated with the bSi fraction were kept in the final PCAs because these bSi components appeared to also carry partial polysaccharide signals. Removing them does not alter the PCA results for Dragsjön, suggesting only a small polysaccharide signal, but weakens the polysaccharide signal for Lång-Älgsjön, suggesting a larger partial signal. For reference purposes the PCAs performed on the initial 72 and 74 for peaks are shown in Figure S8. The final PCAs on wavenumbers associated with OM and bSi include 43 spectral peaks for Dragsjön and 55 peaks for Lång-Älgsjön (Figure 4).

For Dragsjön the first four components of the final PCA capture 97% of the total spectral variance. The first component (Cp1_{IR}), 51% spectral variance, Figure 4a, Table S4) has positive loadings for wavenumbers associated with O-H and N-H vibrations of polysaccharides, phenols and proteins (3552–3263 cm⁻¹) and C–H stretching of aliphatics and aromatics (2960–2850 cm⁻¹). Component 2 (Cp2_{IR}, 26% spectral variance) has positive loadings for wavenumbers associated with CO₂ (an impurity) or possibly polysaccharide overtones (2378 cm⁻¹, Coates, 2000), C=O stretching in organic acids (1768–1693 cm⁻¹), C=O stretching, N-H bending and C=C stretching of proteins and aliphatics (1678-1633, 1583-1150, and 1452 cm⁻¹); as well as C=C stretching and C-H deformation of lignin and aromatics (1662, 1608, 1516, and 1468 cm⁻¹) (López Pasquali and Herrera, 1997; Marchessault, 1962; Sigee et al., 2002). The third component (Cp3 $_{\rm IR}$, 11% spectral variance) has positive loadings for wavenumbers 1126 and 1080–1049 cm⁻¹, most likely associated with C-O, C-H and O-H vibrations in lignin and intact polysaccharides (Kubo and Kadla, 2005; Mathlouthi and Koenig, 1986; Sigee et al., 2002). Component 4 (Cp4_{IR}, 9% spectral variance), has positive loadings for wavenumbers 1205, 1167, 1161 cm⁻¹ and 492 cm⁻¹ (Table S4). These five wavenumbers are likely to represent Si-O stretching and bending in bSi from diatomaceous algae because bSi has two main bands centered on 1100 and $470\,\mathrm{cm^{-1}}$ (Gendron-Badou et al., 2003). It cannot, however, be excluded that at least part of this signal represents polysaccharides because the main band of C-O vibrations in polysaccharides overlaps with the main band in bSi, although typically centered lower, around 1050 cm⁻¹ (Sigee et al., 2002). A full list of vibrational modes, component loadings and communalities (% of variance explained for each component) for Dragsjön are given in Table S4.

For Lång-Älgsjön, the first four components capture 97% of the spectral variance. Component 1 (Cp1 $_{\rm IR}$, 46% spectral variance, Figure 4b, Table S5) has positive loadings associated with O–H and N–H vibrations (3606–3265 cm $^{-1}$), and C–H stretching (2690 and 2852 cm $^{-1}$). Wavenumbers with negative loadings are associated with vibrations of CO $_{\rm 2}$ or polysaccharide overtones (2351 cm $^{-1}$), C–H overtones and combination bands (1876 and 1863 cm $^{-1}$), C=O stretching in organic acids and C=C stretching and C–H deformation in lignin and aliphatics (1514–1381 cm $^{-1}$). The aliphatic wavenumber 2920 cm $^{-1}$ is found on component 2

(Cp2_{IR}, 26% spectral variance) together with CO₂/polysaccharide overtones (2376 cm⁻¹), C=O stretching in organic acids (1769–1693 cm⁻¹) and C=O and N-H vibrations in proteins (1680–1550 cm⁻¹). Component 3 (Cp3_{IR}, 18% spectral variance) has positive loadings for wavenumber regions 1234–1126 cm⁻¹ and 494–447 cm⁻¹. Similar to Cp4_{IR} for Dragsjön, this component likely primarily carries information associated with Si–O stretching and bending in bSi, because the main polysaccharide signal appears to be captured by the slightly lower wavenumbers (1086–1038 cm⁻¹, Mathlouthi and Koenig, 1986) on component 4 (Cp4_{IR}, 7% spectral variance), but may in part also carry a polysaccharide signal. A full list of vibrational modes, component loadings and communalities for Lång-Älgsjön are given in Table S5.

Comparison of spectral and pyrolytic OM composition

Dragsjön. Passive plotting of the four spectral components (Cp1– $4_{\rm IR}$) in the pyrolytic components 1 and 2 (Cp1– $2_{\rm PY}$) space supports the interpretation of Cp3 $_{\rm IR}$ (1126–1049 cm $^{-1}$) as capturing primarily polysaccharides. This is because Cp3 $_{\rm IR}$ plots with the indicators of intact OM in the bottom left quadrant (Figure 5a). These include anhydrosugars, which primarily are indicators of intact polysaccharides from plants (Pouwels et al., 1987), but also found in some algae (Piloni et al., 2021); 4-isopropenylphenol, a pyrolysis product of *Sphagnum* acid and sensitive to aerobic degradation (Schellekens et al., 2015); and the other polysaccharides from mixed origins (i.e. microbes, plants, algae; Nguyen et al., 2003; Pouwels et al., 1987).

The second spectral component, Cp2_{IR} (2378, 1768– $1452\,\mathrm{cm^{-1}}$), plots in the upper left quadrant, half-way between the fatty acids on the negative side of $Cp1_{PY}$, here dominated by the relatively labile, short chain fatty acids (C14-16:0, Table 2) that are typical of aquatic OM (Cranwell et al., 1987; Han et al., 2022; Jaffé et al., 2001), and a group of pyrolysis products of algal and bacterial origins in the top right quadrant on the positive side of Cp2_{pv}. The latter includes hopanoids, which derive from bacterial membrane (Ourisson et al., 1987); the chlorophyll-derived phytadienes; the protein and amino acid-derived diketopiperazines (Fabbri et al., 2012); the mixed origin steroids (Volkman, 1986) and the alkylamides, which are dehydration products of algal macromolecules (Derenne et al., 1993) and possibly also secondary pyrolysis products of amino acids and proteins (Nierop and van Bergen, 2002). This combination of compound groups has previously been interpreted as algal-derived OM in a boreal lake sediment sequence from south-western Sweden (Tolu et al., 2017) and the location of $Cp2_{IR}$ in the $Cp1-2_{PY}$ space agrees with the partial protein and aliphatic spectral signal captured by Cp2_{IR}.

The fourth spectral component, $Cp4_{IR}$, interpreted as mainly a bSi signal with a minor polysaccharide signal (1205–1161 and $492\,\mathrm{cm^{-1}}$), also plots in the upper left quadrant, partially capturing an algal and bacterial signal and partially a labile OM signal. This supports the bSi (i.e. algal) interpretation of $Cp4_{IR}$, but also the partial polysaccharide signal. The duality is also evident in down-core trends of $Cp4_{IR}$ component scores and DRIFTS-based, PLSR-modeled bSi content (Figure 6a). The trends are coherent until ~1000 CE, but after this point, bSi content deviates from the increasing trend of $Cp4_{IR}$, probably due to the increase in polysaccharides content. Finally, $Cp1_{IR}$, capturing O–H vibrations of polysaccharides and phenols, C–H stretching of aliphatics, and N–H stretching in proteins (3552–2850 cm⁻¹), plots in the upper center of the plot, close to the pyrolytic compound group phenols.

In the pyrolytic components 3 and 4 (Cp3– $4_{\rm PY}$) space, Cp1 $_{\rm IR}$ (3552–2850 cm $^{-1}$) plots on the positive side of Cp3 $_{\rm PY}$ where lignin, tocopherol, pristene, n-alkanes C21–25 and n-alkenes are found. Lignin include pyrolysis products of syringyl and guaiacyl methoxy-phenols and only derive from vascular plants

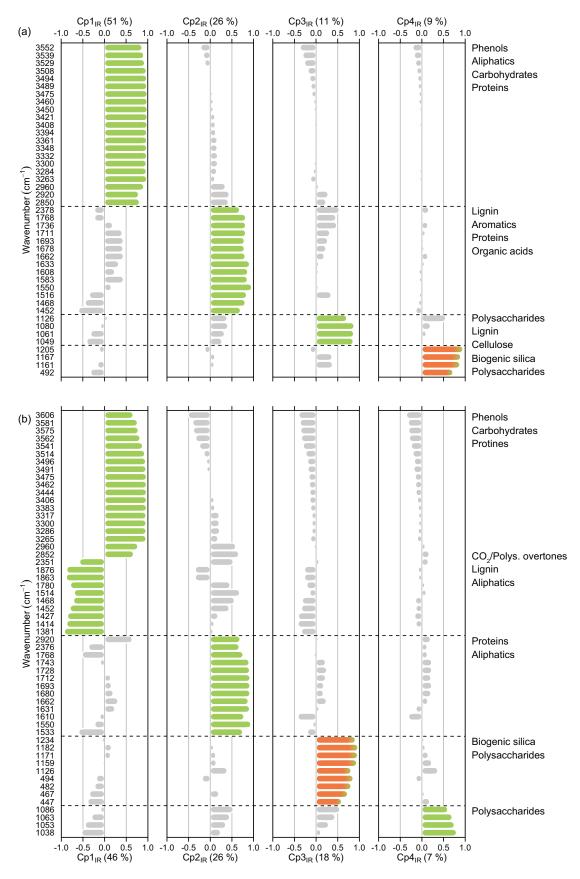


Figure 4. Principal component analyses of DRIFT spectra from untreated samples for (a) the 43 peaks capturing OM and bSi in Dragsjön, and (b) the absorbance spectra for the 55 peaks capturing OM and bSi in Lång-Älgsjön. Wavenumbers are ordered by component based on their strongest positive or negative loading, and colored by sediment fraction; organics (green), bSi (orange) and other variance allocation (light gray).

(Saiz-Jimenez and De Leeuw, 1986). Tocopherols are lipophilic antioxidants produced exclusively by photosynthetic plants, algae and some cyanobacteria (DellaPenna and Pogson, 2006), and pristene pyrolytic or degradation products of tocopherol, and

to a lesser extent chlorophyll (Goossens et al., 1984; Höld et al., 2001; Nguyen et al., 2003). The aliphatic *n*-alkanes C21–25 are found in both *Sphagnum* moss and in vascular plants (Bush and McInerney, 2013), but do also, along with the aliphatic *n*-alkenes,

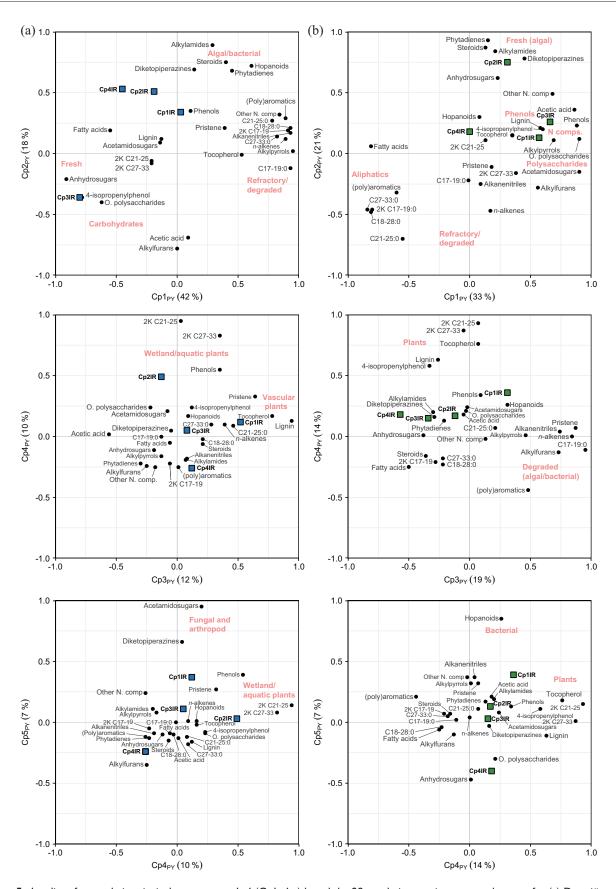


Figure 5. Loadings for pyrolytic principal components I–4 (Cp1–4_{PY}) based the 28 pyrolytic organic compound groups for (a) Dragsjön and (b) Lång-Älgsjön. Filled circles correspond to pyrolytic compound groups and filled squares to the passively added spectral components Cp1–4_{IR}. C17–19:0, *n*-alkanes C17–19; C18–28:0, *n*-alkanes C18–28; C21–25:0, *n*-alkanes C21–25; C27–33:0, *n*-alkanes C27–33; 2K C17–19, *n*-alkan-2-ones C17–19; 2K C21–25, *n*-alkan-2-ones C27–33, *n*-alkan-2-ones C27–33.

form during pyrolysis (Van Smeerdijk and Boon, 1987). These compounds indicate that $\mathrm{Cp1}_{\mathrm{IR}}$ captures information on aliphatics and phenolic compounds, as well as proteins, of vascular plant origin.

The spectral component Cp2_{IR} (1768–1452 cm⁻¹) plots half way between the wetland indicators *n*-alkan-2-ones C21–25 and C27–33, and phenols on the positive side of Cp4_{PY} and the compound groups other polysaccharides and acetamidosugars on the

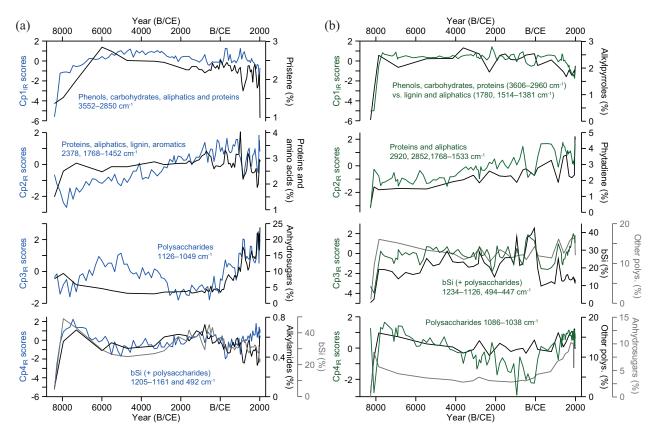


Figure 6. Down-core trends for spectral components CpI_{IR} to Cp4_{IR} (blue and green lines), selected pyrolytic compound groups and sediment bSi content for the past ~8500 years in (a) Dragsjön and (b) Lång-Älgsjön. bSi (%) data for Lång-Älgsjön are from Meyer-Jacob et al. (2015).

negative side of $\mathrm{Cp3}_{\mathrm{PY}}$. The aliphatic n-alkan-2-ones C21–25 and C27–33 are ubiquitous in lake and wetland environments as components of various aquatic and peat-forming plants (Nichols and Huang, 2007; Ortiz et al., 2011; Qu et al., 1999) and the phenols derive from lignin (Faix et al., 1991; Saiz-Jimenez and De Leeuw, 1986), cellulose (Pouwels et al., 1989), non-lignin plant phenols (Schellekens et al., 2009) and proteins (van Heemst et al., 1999). This location of $\mathrm{Cp2}_{\mathrm{IR}}$ strongly supports the mixed aliphatic, protein, lignin and aromatic signal captured by $\mathrm{Cp2}_{\mathrm{IR}}$.

Plotting of Cp1– $4_{\rm IR}$ in the pyrolytic components 4 and 5 (Cp4– $5_{\rm pY}$) space shows that Cp1 $_{\rm IR}$ plots in the upper center of the plot, the positive side of Cp5 $_{\rm pY}$. This component is driven by the chitin-derived acetamidosugars (Stankiewicz et al., 1996) and the protein and amino acid-derived diketopiperazines. In the plot space, however, Cp1 $_{\rm IR}$ is located closer to the phenols and pristene, which further supports that Cp1 $_{\rm IR}$ captures phenolic compounds, in addition to vascular plant-associated aliphatic and chlorophyll derivatives.

Lång-Älgsjön. Passive plotting of the four spectral components (Cp1–4 $_{IR}$) in the pyrolytic components 1 and 2 (Cp1–2 $_{\scriptscriptstyle \mathrm{PV}}$) space shows that Cp1_{IR}, capturing the relative proportions of phenols, polysaccharides and proteins at 3606–2960 cm⁻¹ versus lignin and aliphatics at 1780 and 1514–1381 cm⁻¹ plots on the positive side of Cp1_{pv} together with a range of polysaccharide, phenol and protein-derived compounds groups (Figure 5b). Cp3_{IR}, capturing mainly a bSi signal (1234–1126 and 494– 447 cm⁻¹), also plots with the pyrolytic compound groups to the right on Cp1_{py}. The opposite side of Cp1_{py} captures the fatty acids, (poly)aromatics and many of the aliphatic groups, which have in common very high relative abundances in the two silt and clay-rich basal samples. They likely represent refractory, or potentially shielded (Han et al., 2022), OM from the time of lake formation. The second spectral component, Cp2_{IR}, capturing aliphatics and proteins (2920-2852 and 1768-1533 cm⁻¹), plots with phytadiene, steroids, alkylamides and diketopiperazines,

indicative of algal OM, on the positive side of ${\rm Cp2_{PY}}$. Anhydrosugars also plot near these compound groups, as an indicator of intact OM, though a minor algal signal cannot be excluded. ${\rm Cp2_{IR}}$ plots closer to the algal OM compound groups for Lång-Älgsjön than for Dragsjön, probably because this spectral component includes predominantly vibrations of aliphatics and proteins, as opposed to also capturing vibrations of lignin and aromatics for Dragsjön.

In the pyrolytic components 3 and 4 (Cp3-4_{py}) space, only $Cp1_{IR}$ (3606–2960 cm⁻¹ vs 1780, 1514–1381 cm⁻¹) plots on the positive side of Cp3_{PY}, while the remaining three components plot on the negative side. The pyrolytic compound groups located on the positive side of Cp3_{py} likely represent degraded algal and bacterial OM. These groups include the n-alkanes C17–19, representing direct inputs from aquatic plants and bacteria (López-Días et al., 2013; Qu et al., 1999), but also microbial oxidation of *n*-alkanes and decarboxylation of fatty acids (Ambles et al., 1993; Püttmann and Bracke, 1995); pristene, which are pyrolytic or degradation products of primarily tocopherol; alkanenitrile, derived from pyrolysis of algal macromolecules; the small alkylfurans, which may be derived from plant polysaccharides (Marbot, 1997; Pouwels et al., 1987) and bacterial sugars (Buurman et al., 2005); and the protein-derived alkylpyrroles (Nguyen et al., 2003). Compounds found on the negative side of Cp3_{py} are instead the less-degraded compounds such as fatty acids, anhydrosugars, steroids and 4-isopropenylphenol. Along this gradient, Cp1_{IR} is located near the hopanoids and phenols, suggesting it carries a partial bacterial, more so than algal, signal, while the polysaccharide component, Cp4_{IR} (1086–1038 cm⁻¹), is located furthest left, near a cluster of compounds encompassing both intact polysaccharides (anhydrosugars) and algal OM indicators (alkylamides, diketopiperazines and phytadiene). Unlike for Dragsjön, none of the spectral components plot near the plant OM indicators, including *n*-alkan-2-ones C21–25 and C27–33, tocopherol, lignin and 4-isopropenylphenol, on Cp4_{pv}.

In the pyrolytic components 4 and 5 (Cp4– 5_{PY}) space, the bacterial OM indicator hopanoids is located toward the top of the plot on the positive side of Cp5 $_{PY}$, and pyrolysis products of intact polysaccharides are located toward the bottom, on the negative side. The spectral components are again found along a gradient, with Cp1 $_{IR}$ located the positive side of Cp5 $_{PY}$, half-way between the hopanoids and the plant OM on Cp4 $_{PY}$, and Cp4 $_{IR}$ down the bottom close to the other polysaccharides and the anhydrosugars. These patterns further support that Cp1 $_{IR}$ (3606–2960 cm $^{-1}$) represents a degraded OM fraction with a strong phenolic and proteinaceous signal, and the polysaccharides on Cp4 $_{IR}$ (1086–1038 cm $^{-1}$) the most-intact OM fraction in Lång-Älgsjön.

Temporal trends. The OM composition in these two lakesediment cores was previously divided into three major phases based on the pyrolytic organic compounds (Ninnes et al., 2017). To further validate our DRIFTS approach as a stand-alone method to characterize OM composition a brief comparison of temporal changes may be highly informative. The earliest of the three phases identified by Ninnes et al. (2017) is the post-glacial phase (>7900 BCE), which represents the time of lake formation. Mineral matter initially dominates the sediment but as the catchments become vegetated and soils stabilize (Myrstener et al., 2021), the relative abundances of most pyrolytic organic compound groups increase rapidly (Figure 6). This phase is equally apparent in the spectral data, with initially low scores for all organic components, corroborating the relatively low, but increasing OM content. The second phase spans most of the Holocene, until ~300 CE, and is characterized by stable or slowly changing relative abundances of pyrolytic organic compounds. This phase represents a climatically warm, but linearly cooling, phase of the Holocene (Seppä et al., 2009), where most trends for both pyrolytic organic compound groups and spectral components either show unchanging (Lång-Älgsjön Cp1_{IR}, and Cp3_{IR}), or slowly increasing (Cp2_{IR} both lakes) or decreasing trends (Lång-Älgsjön $\mathrm{Cp4}_{\mathrm{IR}}$). Slow, millennial-scale fluctuations are seen in Seen in $\text{Cp1}_{\text{IR}},\,\text{Cp3}_{\text{IR}}$ and Cp4_{IR} in Dragsjön. Minor fluctuations in the pyrolytic data during this phase are masked due to the lower sampling resolution in these sections of the cores, but the higher resolution spectral data corroborate that no rapid changes in OM composition occurred, with the exception for Cp3_{IR} (polysaccharides) in Dragsjön.

The third phase, spanning ~300 CE to present, is the most dynamic, hence also divided into three sub-phases. The first two are characterized by large increases in labile pyrolytic compound groups such as proteins and amino acids, anhydrosugars and 4-isopropenylphenol, but also in the aliphatic *n*-alkan-2-ones. These changes are probably linked to altered catchment inputs due to expansion of adjacent peatlands and/or the development of the floating, shoreline peat mats (Myrstener et al., 2021; Ninnes et al., 2017). Increased supply of OM from peat or *Sphagnum*, in combination with more rapid burial, would increase the proportion of polysaccharides in the sediment and also promote preservation of labile compounds (Ninnes et al., 2017). Spectral components that carry protein and polysaccharide signals similarly indicate large changes in OM composition during the first two sub-phases in both lakes.

During the third sub-phase, from 1970 CE to present, diagenetic processes are an important feature in the pyrolytic OM composition and possible to identify due to the high compositional detail provided by this technique, which allows for tracking of individual compounds (Ninnes et al., 2017). Diagenetic signals cannot be identified in the spectral components, nor in the spectra of individual wavenumbers of labile compounds or in ratios between labile and resistant compounds. This is possibly because the main wavenumbers for labile compounds, that is, polysaccharides, are not "clean," but also carry other types of information, mainly on Si–O vibrations.

Summary — Application of undiluted MIR spectroscopy for analysis of lake-sediment OM composition

Our spectral analyses of two lake-sediment sequences highlight the advantages and potential of MIR, whether DRIFTS as used in here or ATR, using untreated samples to extract quantitative and qualitative data on lake-sediment OM composition. Here we outline a strategy for applying DRIFTS to analyze lake-sediment cores simply using dried and ground samples, allowing for high sampling resolution and much greater detailing of OM composition than other rapid and low-cost methods (e.g. C and N elemental analyzes along with their isotopes).

First, as expected, we show that sediment TOC concentrations can be modeled either with a general calibration model using a wide variety of sediment samples or with a local model developed from a subset of samples from the sediment record(s) of interest. These TOC models based on the DRIFT spectra of untreated samples have the same predictive abilities as models based on samples diluted with KBr and can be applied as a first step to provide estimates and to identify subsets of samples to be analyzed for C content in order to develop a more precise local model. Furthermore, our DRIFT spectra and measured C concentrations could potentially be used as a foundation for developing calibration models by anyone using similar DRIFTS instrumentation. Previous work on DRIFT spectra from untreated soil, estuarine sediment and water samples show good results not only for C, but also for quantification of total N and other major components, including humic substances and specific compounds groups such as amino acids and amino sugars (e.g. Alaoui et al., 2011; Matamala et al., 2019; Reeves et al., 2001; Tremblay et al., 2011). Hence, there is great potential for development of quantitative calibrations based DRIFT spectra of untreated samples for broad scale characterization of lake sediments.

Second, and more importantly for our aim, using the opensource processing pipeline with peak selection and reduction based on second derivative spectra developed by Álvarez Fernández and Martínez Cortizas (2020) in combination with PCA, the DRIFT spectra can be used to establish qualitative compositional changes in OM, as well as the minerogenic and to a large extent biogenic sediment fractions. Within the organic fraction, compound groups such as aromatics, lignin, aliphatics, proteins and polysaccharides could readily be identified in our two Holocene sediment records. As such, MIR can be used as a rapid standalone method for OM characterization or function as a screening process for more-specific, time-intensive analyses, such as Py-GC/MS or GC/MS and LC/MS methods requiring sample extraction, but which can further tease apart the OM composition, and identify sources and degradation status. A next step can be to develop predictive models based on MIR and pyrolytic data for modeling of greater sample compound diversity to reveal more about temporal OM changes.

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Data availability

Spectroscopic data sets, including total organic carbon content, are available on the Figshare repository platform. Other supporting information can be found in the supplementary materials.

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Supplemental material

Supplemental material for this article is available online.

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