Supporting information

Electrostatic-driven gelation of colloidal nanocrystals

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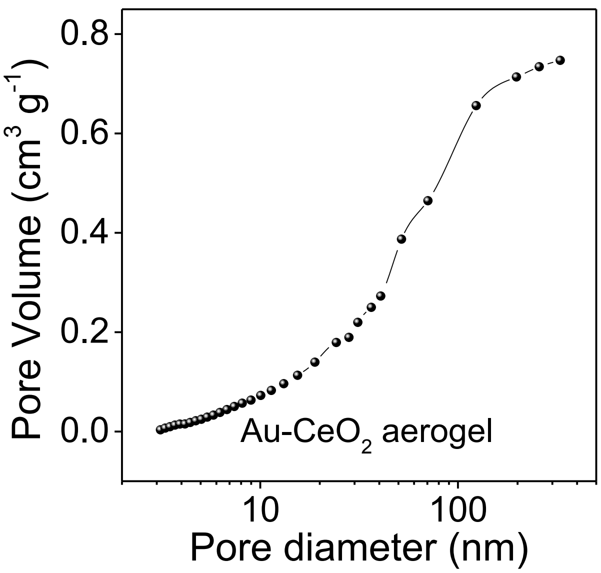

Figure S1. BJH Pore size distribution of an Au-CeO_2 aerogel.

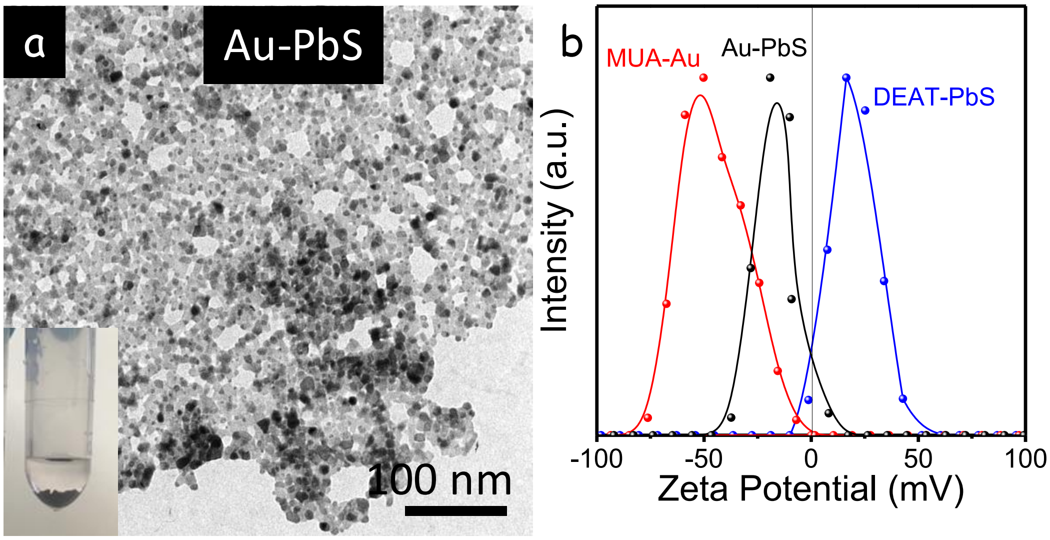

Figure S2. a) TEM image of the Au-PbS NCs gel obtained by mixing 100 µl of a 5 mg/mL MFA solution of DEAT-PbS NCs with 100 µl of a 5 mg/mL MFA solution of MUA-Au NCs. b) Zeta potential curves of the gel formed after mixing of NCs.

Table S1: Comparison of the catalytic activity of Au-CeO_2 aerogels against similar Au@CeO_2 systems catalysts reported in literature.

	Au loading (%wt)	T (°C)	CO / O_2 (v% / v%)	Specific Rate ($mmol_{CO}/g_{Au}/s$)	Source
Au-CeO_2 aerogels	13	50	1/10	0.0015	This work
Au@CeO_2	27.6	40	1/20	0.008	Liu et al.[1]
Au@h-CeO_2/SiO_2	0.74	40	1/6	0.151	Wu et al.[2]
Au@CeO_2	0.39	50	1.96/21	0.084	He et al.[3]
Au/CeO_2	2.68	75	1/10	0.006	Wang et al.[4]

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