

Supporting Information

Strong, machinable and insulating chitosan-urea aerogels: towards ambient pressure drying of biopolymer aerogel monoliths

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Figure S1. Thermal conductivity device

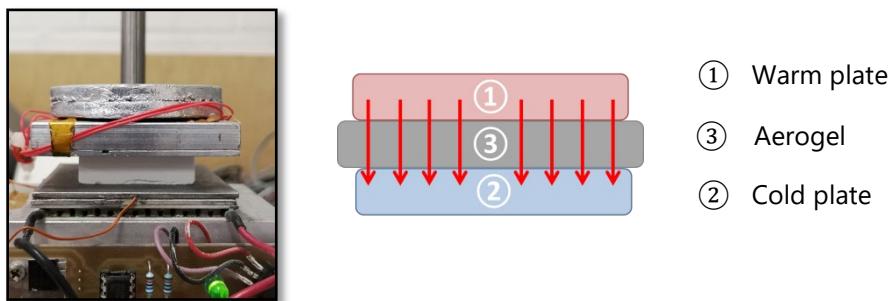


Figure S2. Humidity uptake

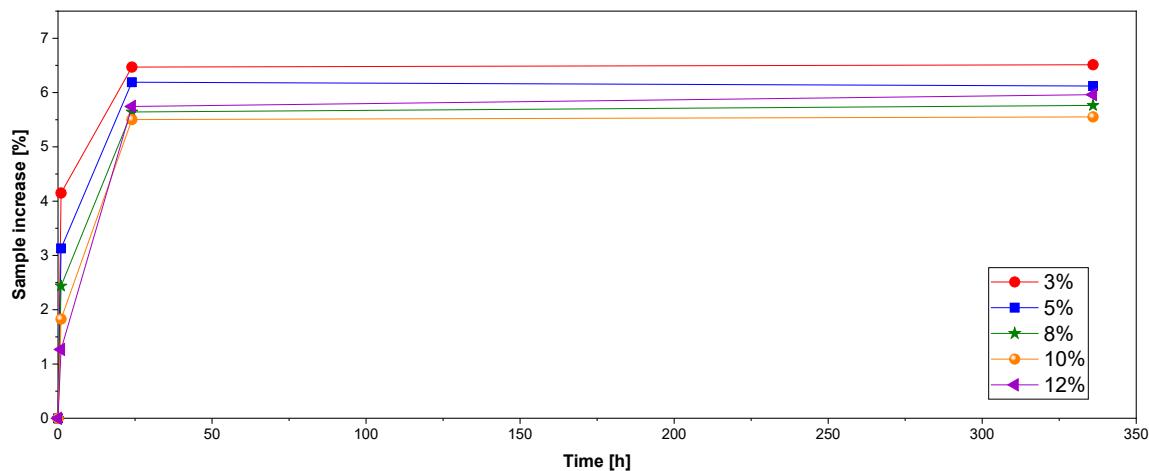


Figure S3. Nitrogen sorption isotherms and BJH pore sizes distributions for all chitosan concentrations

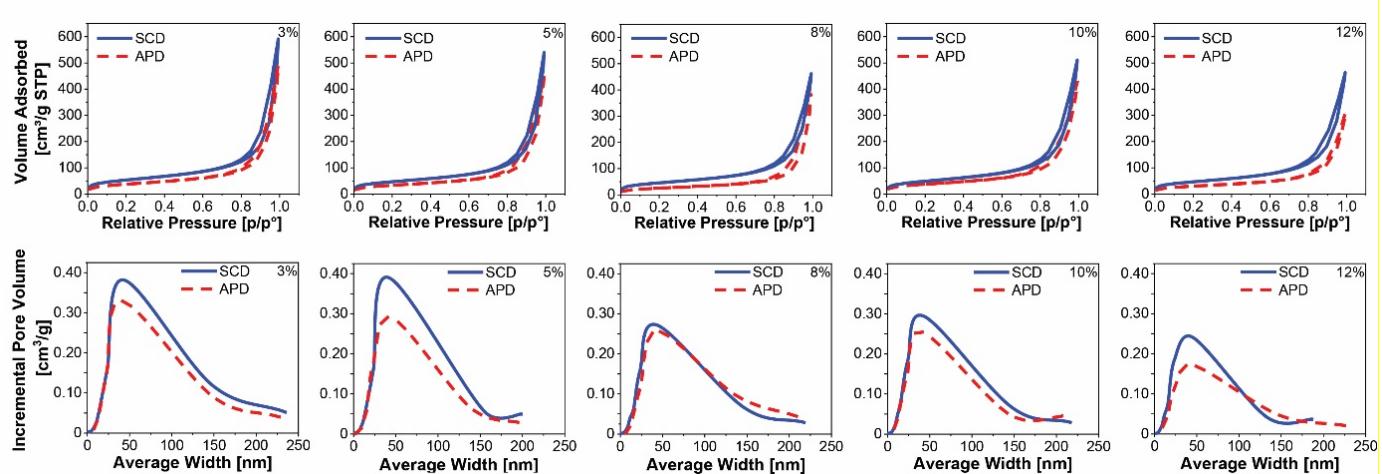


Figure S4. Burning test SCD 10% m/v chitosan aerogel

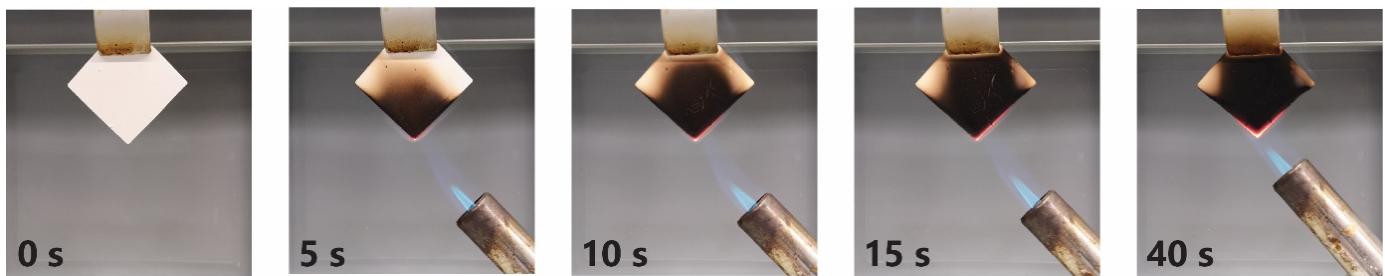


Figure S5. Simulated reaction between 100'000 amino groups and 833'333 urea molecules (equivalent to a urea:d-glucosamine ratio of 6, and a degree of deacetylation of 72%), assuming equal probabilities for the end-capping (ureido) and cross-linking (ureylene) reaction. Because of the large urea excess, probability for cross-linking (a reaction between a ureido group and an amino group) remains low and the final fraction of ureylene cross-linking is limited to 3%.

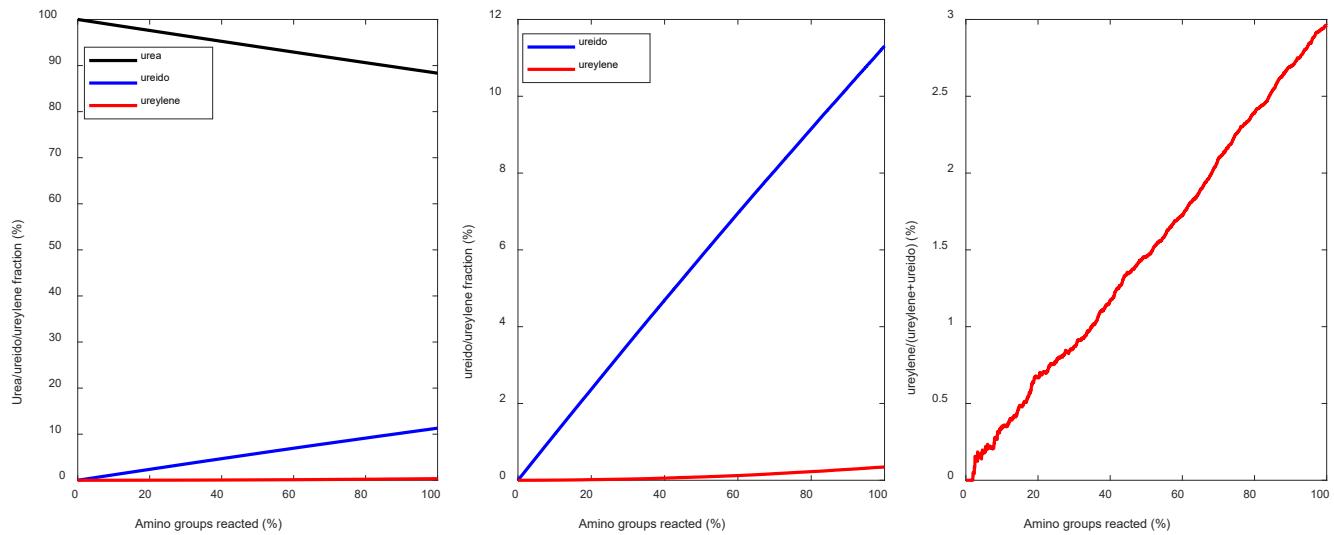


Table S1: Reagent concentrations

| | Chitosan [% m/v] | HCl [M] | HCl sol. [mL] | Chitosan [g] | Urea [g] | CTAC [mL] |
|------------------------|----------------------------|-------------------|-------------------------|------------------------|--------------------|---------------------|
| SCD Samples | 3 % | 0.34 M | 40 | 1.2 | 2.68 | 0.030 |
| | 5 % | 0.56 M | | 2.0 | 4.47 | |
| | 8 % | 0.89 M | | 3.2 | 7.16 | |
| | 10 % | 1.12 M | | 4 | 9 | |
| | 12 % | 1.34 M | | 4.8 | 10.73 | |
| APD Samples | 3 % | 0.34 M | 120 | 3.6 | 8.04 | 0.090 |
| | 5 % | 0.56 M | | 6 | 13.41 | |
| | 8 % | 0.89 M | | 9.6 | 21.48 | |
| | 10 % | 1.12 M | | 12 | 27 | |
| | 12 % | 1.34 M | | 14.4 | 32.19 | |

Table S2: Influence of chitosan:urea ratio on gelation and properties

| 10% (m/v) of chitosan in HCl aq. (1.12M) | | |
|--|---------------------------------------|---|
| Ratio [Chitosan:Urea] | Gelation after 24h at 80°C | λ [mW / (m·K)] |
| 1:0,5 | No | - |
| 1:1 | No | - |
| 1:2 | No | - |
| 1:3 | No | - |
| 1:4 | Yes | 28.1 ± 0.2 |
| 1:6 | Yes | 23.1 ± 0.3 |

Table S3: Temperature and time reaction influence

| 5% (m/v) of chitosan in HCl aq. (0.56 M) | | | |
|--|-------------|----------|------------------|
| Chitosan:Urea = 1:6 | | | |
| Reaction temperature and time = 80 °C, 24h | | | |
| T [°C] | Time [h] | Gelation | λ [mW /(m·K)] |
| 70 | 24 | No | - |
| 70 | 48 | No | - |
| 80 | 5 | Yes | 22.6 ± 0.1 |
| 80 | 12 | Yes | 22.8 ± 0.2 |
| 80 | 24 | Yes | 25.5 ± 0.2 |
| 80 | 48 | Yes | 23.5 ± 0.2 |
| 90 | 5 | Yes | 23.6 ± 0.1 |
| 90 | 12 | Yes | 24.4 ± 0.3 |
| 90 | 24 | Yes | 22.1 ± 0.1 |
| 90 | 48 | Yes | 25.4 ± 0.3 |

Table S4: HCl concentration influence

| Chitosan:Urea = 1:6 | | | | |
|--|-------------------|-----------|----------------|------------------|
| Reaction temperature and time = 80 °C, 24h | | | | |
| Protonation | Chitosan HW conc. | HCl conc. | Gelation | λ [mW /(m·K)] |
| 76.5 % | 3 % | 0.14 M | Yes | 37.9 ± 0.2 |
| 90 % | 3 % | 0.17 M | Yes | 36.1 ± 0.2 |
| 90 % | 10 % | 0.56 M | Yes | 40.1 ± 0.5 |
| 180 % | 10 % | 1.12 M | Reference: Yes | 22.4 ± 0.3 |

Table S5: Chitosan molecular weight influence

| Low molecular weight (LW), 50'000–190'000 g/mol; $\eta < 200 \text{ mPa}\cdot\text{s}$ (1% acetic acid, 20 °C) | |
|--|----------|
| Medium molecular weight (MW), 190'000–310'000 g/mol; $\eta = 200–400 \text{ mPa}\cdot\text{s}$ (1% acetic acid, 20 °C) | |
| High molecular weight (HW) = 310'000 – 375'000 g/mol; $\eta = >400 \text{ mPa}\cdot\text{s}$ (1% acetic acid, 20 °C) | |
| Chitosan:Urea = 1:6, Reaction temperature and time = 80 °C, 24h | |
| Molecular weight | Gelation |
| LW | No |
| MW | No |
| HW | Yes |

Table S6: Solubility of chitosan HW in different acids

| Dissolving temperature = 70 °C | | | | |
|--------------------------------|-------------------------------|------------|------------------|--|
| Chitosan % [m/v] | Acid | Acid conc. | Dissolution time | Observation |
| 10 % | Acetic acid | 99% | 2 h | No dissolution |
| 5 % | Acetic acid | 99% | 6 h | No dissolution |
| 5 % | Acetic acid in water | 50 % | 4 h | Very viscous gel, not usable |
| 10 % | Trichloroacetic acid | 99% | 6 h | No dissolution |
| 10 % | Trichloroacetic acid | 50% | 6 h | No dissolution |
| 10 % | Trifluoroacetic acid | 95% | 10 min | Dissolved but after aging got carbonized |
| 10 % | Trifluoroacetic acid in water | 50 % | 2h 30 min | Dissolved but after aging got carbonized |

Table S7: Peak positions, assignment and intensities of Solid-state CP-MAS ^{13}C NMR

| $\delta^{13}\text{C}$ [ppm] | # | Assignment | | Intensity ^a [%] | Intensity (total) ^{a,b} [%] |
|--------------------------------|------|------------------------|----------------|-------------------------------|---|
| | | Central band | ssb | | |
| 240.8 | 9* | | Ureido | 1 | |
| 214.4 | 7* | | Acetyl C=O | 5 | |
| 200.9 | 9* | | Ureido | 11 | |
| 174.5 | 7 | Acetyl C=O | | 11 | 22 |
| 161.2 | 9 | Ureido | | 51 | 77 |
| 144.7 | 1* | | Chitosan C1 | 3 | |
| 134.9 | 7* | | Acetyl C=O | 6 | |
| 123.4 | 4* | | Chitosan C4 | 4 | |
| 121.6 | 9* | | Ureido | 13 | |
| 115.2 | 3/5* | | Chitosan C3+C5 | 10 | |
| 104.6 | 1 | Chitosan C1 | | 97 | 100 |
| 96.5 | 2* | | Chitosan C2 | 13 | |
| 83.3 | 4 | Chitosan C4 | | 92 | 108 |
| 75.2 | 3/5 | Chitosan C3+C5 | | 206 | 225 |
| 61.2 | 6 | Chitosan C6 | | 87 | 91 |
| 56.1 | 2 | Chitosan C2 | | 103 | 120 |
| 44.3 | 4* | | Chitosan C4 | 12 | |
| 35.6 | 3/5* | | Chitosan C3+C5 | 10 | |
| 23.9 | 8 | Acetyl CH ₃ | | 23 | 23 |
| 21.1 | 6* | | Chitosan C6 | 4 | |
| 16.4 | 2* | | Chitosan C2 | 4 | |

ssb: spinning sideband

^a Integrated intensity, as a percentage of the chitosan C1 total intensity

^b Sum of central band and spinning sidebands

Table S8: FTIR peak positions and assignments

| Urea | | Chitosan | | Aerogel: Chitosan-Urea | Assignment |
|-----------------------------------|----------------------------|-----------------------------------|-----------------------------------|---------------------------|---|
| Wavenumber [cm ⁻¹] | Assignment | Wavenumber [cm ⁻¹] | Wavenumber [cm ⁻¹] | | |
| 1000 | C-N stretching | 992 | | | |
| 1046 | NH ₂ rocking | 1030 | 1030 | | C-O stretching |
| | | 1065 | 1067 | | |
| | | | 1111 | | |
| 1152 | NH ₂ rocking | 1151 | 1156 | | C-O-C stretching |
| | | 1198 | 1203 | | |
| | | 1258 | 1263 | | OH bending |
| | | 1307 | 1309 | | C-N stretching Amide III |
| | | 1376 | 1376 | | CH bending CH ₃ deformation |
| 1457 | C-N stretching | 1422 | | | CH ₂ bending |
| | | | 1560 | | N-H bending Amide II |
| 1590 | NH ₂ scissoring | 1590 | | | NH ₂ scissoring |
| 1678 | C=O stretching | 1650 | 1650 | | C=O stretching Amide I |
| | | 2871 | 2877 | | |
| | | 2919 | 2941 | | C-H stretching |
| | | | 3118 | | |
| 3255 | | 3290 | 3282 | | N-H stretching |
| 3329 | N-H stretching | 3358 | 3361 | | O3-H stretching |
| 3429 | | | 3453 | | O6-H stretching |

Table S9. Theoretical relationships between surface area and particle size

| Single particle | | | 1 g chitosan @1.59 g/cm3 | | |
|-----------------|---------------------------------------|--------------------------------|--------------------------|----------------------|------------------------|
| diameter_p | volume_1p | area_1p | volume_1g | #particles_1g | surface area_1g |
| [m] | [m ³] | [m ²] | [m ³] | | [m ² /g] |
| 0.00000009 | 3.815E-22 | 2.54462E-14 | 6.28931E-07 | 1.64857E+15 | 41.9 |
| 0.00000005 | 6.5415E-23 | 7.85375E-15 | 6.28931E-07 | 9.61447E+15 | 75.5 |
| 0.00000002 | 4.18656E-24 | 1.2566E-15 | 6.28931E-07 | 1.50226E+17 | 188.8 |
| | = $(4/3)\pi^*(\text{diameter}_p/2)^3$ | = $\pi^*(\text{diameter}_p)^2$ | = $10^{-6}/1.59$ | =volume_1g/volume_1p | =#particles_1g*area_1p |
| | $V = 4/3\pi r^3$ | $A = 4\pi r^2$ | | | |

Table S10. Chitosan-urea aerogel properties

| Chit. | ρ [%] | Porosity | | | V_{pore} [cm ³ /g] | | | Ads. Des. [cm ³ /g] | | | $V_{\text{pore BH}}$ [cm ³ /g] | | | S_{BET} [m ² /g] | | | C - value | | | $D_{\text{pore BH}}$ Ads. Des. [nm] | | | λ [mW/(m·K)] | | | | |
|--------------------------|---------------|----------|-------|----|---|------|-----|--------------------------------------|----|-----|--|-----|-----|---|----|---|--------------|----|-----|--|-----|------|-------------------------|------|-----|------|------|
| | | Av. | SD | N | Av. | SD | N | Av. | SD | N | Av. | SD | N | Av. | SD | N | Av. | SD | Av. | SD | Av. | SD | | | | | |
| SCD | 3 | 0.094 | 0.018 | 5 | 91.8 | 1.7 | 9.9 | 1.9 | 3 | 0.8 | 0.1 | 0.8 | 0.1 | 165 | 20 | 3 | 144 | 9 | 245 | 68 | 3 | 22.5 | 2.1 | 21.3 | 1.8 | 24.1 | 0.9 |
| | 5 | 0.121 | 0.024 | 3 | 90.0 | 2.0 | 8.1 | 2.0 | 3 | 0.8 | 0.1 | 0.8 | 0.1 | 172 | 17 | 3 | 173 | 40 | 190 | 51 | 3 | 21.7 | 0.9 | 20.7 | 0.7 | 25.7 | 0.2 |
| | 8 | 0.145 | 0.021 | 4 | 88.0 | 1.7 | 6.4 | 1.0 | 4 | 0.7 | 0.0 | 0.7 | 0.0 | 159 | 3 | 4 | 140 | 13 | 160 | 22 | 4 | 20.8 | 0.9 | 20.1 | 0.8 | 27.8 | 1.9 |
| | 10 | 0.144 | 0.013 | 10 | 87.5 | 1.1 | 6.0 | 0.6 | 5 | 0.7 | 0.1 | 0.7 | 0.1 | 160 | 8 | 5 | 154 | 30 | 152 | 21 | 5 | 20.2 | 1.4 | 19.1 | 1.4 | 27.9 | 2.9 |
| | 12 | 0.152 | 0.011 | 4 | 87.3 | 0.9 | 5.9 | 0.5 | 3 | 0.7 | 0.0 | 0.7 | 0.0 | 167 | 7 | 3 | 124 | 4 | 141 | 10 | 3 | 19.4 | 0.5 | 18.4 | 0.4 | 26.2 | 3.8 |
| APD dried at RT | 3 | 0.305 | 0.138 | 4 | 79.4 | 9.5 | 4.9 | 2.2 | 3 | 0.9 | 0.1 | 0.9 | 0.1 | 150 | 12 | 3 | 119 | 12 | 131 | 51 | 3 | 25.3 | 0.8 | 23.0 | 0.5 | 37.5 | 0.0 |
| | 5 | 0.361 | 0.142 | 4 | 66.3 | 11.1 | 2.8 | 1.2 | 3 | 0.7 | 0.0 | 0.7 | 0.0 | 127 | 14 | 3 | 117 | 12 | 87 | 26 | 3 | 25.1 | 1.1 | 21.9 | 0.4 | 62.5 | 19.8 |
| | 8 | 0.423 | 0.138 | 3 | 65.1 | 11.3 | 2.7 | 1.2 | 3 | 0.5 | 0.1 | 0.5 | 0.1 | 83 | 16 | 3 | 147 | 31 | 149 | 95 | 3 | 26.8 | 2.2 | 23.7 | 0.2 | 68.1 | 25.1 |
| | 10 | 0.278 | 0.026 | 4 | 75.9 | 0.5 | 3.4 | 0.1 | 3 | 0.7 | 0.0 | 0.7 | 0.0 | 134 | 0 | 3 | 125 | 8 | 102 | 3 | 3 | 22.3 | 0.6 | 20.5 | 0.9 | 41.6 | 3.1 |
| | 12 | 0.225 | 0.005 | 3 | 81.4 | 0.4 | 4.4 | 0.1 | 3 | 0.5 | 0.1 | 0.5 | 0.1 | 106 | 20 | 3 | 156 | 51 | 173 | 35 | 3 | 20.3 | 1.2 | 20.0 | 0.9 | 36.3 | 0.3 |
| APD dried at 65 °C | 5 | 0.292 | 0.008 | 2 | 74.4 | 1.2 | 2.6 | 0.1 | 2 | 1.0 | 0.0 | 1.0 | 0.0 | 210 | 0 | 2 | 118 | 0 | 49 | 3 | 2 | 21.2 | 0.3 | 20.0 | 0.1 | 44.5 | 1.0 |
| | 8 | 0.304 | 0.000 | 2 | 76.3 | 0.0 | 2.9 | 0.0 | 2 | 0.4 | 0.4 | 0.8 | 0.0 | 174 | 2 | 2 | 135 | 4 | 65 | 0 | 2 | 21.2 | 0.7 | 19.9 | 0.8 | 55.6 | 4.0 |
| | 10 | 0.180 | 0.007 | 2 | 85.0 | 0.4 | 4.9 | 0.2 | 2 | 0.9 | 0.0 | 0.9 | 0.0 | 228 | 7 | 2 | 115 | 6 | 108 | 0 | 2 | 21.3 | 0.7 | 19.4 | 0.4 | 37.0 | 0.8 |
| | 12 | 0.167 | 0.006 | 2 | 86.2 | 0.4 | 5.3 | 0.2 | 2 | 1.0 | 0.0 | 1.0 | 0.0 | 227 | 5 | 2 | 139 | 16 | 94 | 6 | 2 | 19.2 | 0.1 | 18.0 | 0.0 | 30.9 | 1.0 |

Table S11

See separate Excel file.

Data sources: 1,2,11–17,3–10

- (1) Zhou, J.; Hsieh, Y.-L. Nanocellulose Aerogel-Based Porous Coaxial Fibers for Thermal Insulation. *Nano Energy* **2020**, 68 (November 2019), 104305. <https://doi.org/10.1016/j.nanoen.2019.104305>.
- (2) Wang, D.; Peng, H.; Yu, B.; Zhou, K.; Pan, H.; Zhang, L.; Li, M.; Liu, M.; Tian, A.; Fu, S. Biomimetic Structural Cellulose Nanofiber Aerogels with Exceptional Mechanical, Flame-Retardant and Thermal-Insulating Properties. *Chem. Eng. J.* **2020**, 389 (December 2019), 124449. <https://doi.org/10.1016/j.cej.2020.124449>.
- (3) Jin, H.; Zhou, X.; Xu, T.; Dai, C.; Gu, Y.; Yun, S.; Hu, T.; Guan, G.; Chen, J. Ultralight and Hydrophobic Palygorskite-Based Aerogels with Prominent Thermal Insulation and Flame Retardancy. *ACS Appl. Mater. Interfaces* **2020**, 12 (10), 11815–11824. <https://doi.org/10.1021/acsami.9b20923>.
- (4) Zhu, J.; Hu, J.; Jiang, C.; Liu, S.; Li, Y. Ultralight, Hydrophobic, Monolithic Konjac Glucomannan-Silica Composite Aerogel with Thermal Insulation and Mechanical Properties. *Carbohydr. Polym.* **2019**, 207, 246–255. <https://doi.org/10.1016/j.carbpol.2018.11.073>.
- (5) Zhou, S.; Apostolopoulou-Kalkavoura, V.; Tavares da Costa, M. V.; Bergström, L.; Strømme, M.; Xu, C. Elastic Aerogels of Cellulose Nanofibers@Metal–Organic Frameworks for Thermal Insulation and Fire Retardancy. *Nano-Micro Lett.* **2019**, 12 (1), 9. <https://doi.org/10.1007/s40820-019-0343-4>.
- (6) Zhang, S.; Huang, X.; Feng, J.; Qi, F.; E, D.; Jiang, Y.; Li, L.; Xiong, S.; Feng, J. Thermal Conductivities of Cellulose Diacetate Based Aerogels. *Cellulose* **2020**, 0123456789. <https://doi.org/10.1007/s10570-020-03084-y>.
- (7) Gupta, P.; Verma, C.; Maji, P. K. Flame Retardant and Thermally Insulating Clay Based Aerogel Facilitated by Cellulose Nanofibers. *J. Supercrit. Fluids* **2019**, 152, 104537. <https://doi.org/10.1016/j.supflu.2019.05.005>.
- (8) Song, J.; Chen, C.; Yang, Z.; Kuang, Y.; Li, T.; Li, Y.; Huang, H.; Kierzewski, I.; Liu, B.; He, S.; et al. Highly Compressible, Anisotropic Aerogel with Aligned Cellulose Nanofibers. *ACS Nano* **2018**, 12 (1), 140–147. <https://doi.org/10.1021/acsnano.7b04246>.
- (9) Fricke, M.; Weinrich, D.; Loelsberg, W.; Subrahmanyam, Ra.; Smirnova, I.; Gurikov, P. PROCESS FOR PRODUCING POROUS ALGINATE - BASED AEROGELS. US 2018 / 0258249 A1, 2018.
- (10) Nešić, A.; Gordić, M.; Davidović, S.; Radovanović, Ž.; Nedeljković, J.; Smirnova, I.; Gurikov, P. Pectin-Based Nanocomposite Aerogels for Potential Insulated Food Packaging Application. *Carbohydr. Polym.* **2018**, 195, 128–135. <https://doi.org/10.1016/j.carbpol.2018.04.076>.
- (11) Jiménez-Saelices, C.; Seantier, B.; Grohens, Y.; Capron, I. Thermal Superinsulating Materials Made from Nanofibrillated Cellulose-Stabilized Pickering Emulsions. *ACS Appl. Mater. Interfaces* **2018**, 10 (18), 16193–16202. <https://doi.org/10.1021/acsami.8b02418>.
- (12) Gupta, P.; Singh, B.; Agrawal, A. K.; Maji, P. K. Low Density and High Strength Nanofibrillated Cellulose Aerogel for Thermal Insulation Application. *Mater. Des.* **2018**, 158, 224–236. <https://doi.org/10.1016/j.matdes.2018.08.031>.
- (13) Grout, S.; Budtova, T. Thermal Conductivity/Structure Correlations in Thermal Super-Insulating Pectin Aerogels. *Carbohydr. Polym.* **2018**, 196, 73–81. <https://doi.org/10.1016/j.carbpol.2018.05.026>.
- (14) Ahmad, F.; Ulker, Z.; Erkey, C. A Novel Composite of Alginate Aerogel with PET Nonwoven with Enhanced Thermal Resistance. *J. Non. Cryst. Solids* **2018**, 491, 7–13. <https://doi.org/10.1016/j.jnoncrysol.2018.03.023>.
- (15) Plappert, S. F.; Nedelec, J. M.; Rennhofer, H.; Lichtenegger, H. C.; Liebner, F. W. Strain Hardening and Pore Size Harmonization by Uniaxial Densification: A Facile Approach toward Superinsulating Aerogels from Nematic Nanofibrillated 2,3-Dicarboxyl Cellulose. *Chem. Mater.* **2017**, 29 (16), 6630–6641. <https://doi.org/10.1021/acs.chemmater.7b00787>.
- (16) Fan, B.; Chen, S.; Yao, Q.; Sun, Q.; Jin, C. Fabrication of Cellulose Nanofiber/AIOOH Aerogel for Flame Retardant and Thermal Insulation. *Materials (Basel)*. **2017**, 10 (3), 311. <https://doi.org/10.3390/ma10030311>.
- (17) Takeshita, S.; Yoda, S. Chitosan Aerogels: Transparent, Flexible Thermal Insulators. *Chem. Mater.* **2015**, 27 (22), 7569–7572. <https://doi.org/10.1021/acs.chemmater.5b03610>.