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Molecular motor crossing the frontier of classical to quantum tunneling motion

Authors: Samuel Stolz^{1,2}, Oliver Gröning^{1*}, Jan Prinz^{1,2}, Harald Brune², Roland Widmer¹

5 Affiliations

¹Empa, Swiss Federal Laboratories for Materials Science and Technology, 8600 Dübendorf,

Switzerland

²Institute of Condensed Matter Physics, Station 3, EPFL, 1015 Lausanne, Switzerland

*Correspondence to: oliver.groening@empa.ch

Classification

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Probe Microscopy

Abstract

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The reliability by which molecular motor proteins convert undirected energy input into directed motion or transport has inspired the design of numerous artificial molecular motors. By applying scanning tunneling microscopy (STM), we have investigated an artificial molecular motor, which consists of a single acetylene (C₂H₂) rotor anchored to a chiral atomic cluster provided by a PdGa(111) surface that acts as stator. By breaking spatial inversion symmetry, the stator defines the unique sense of rotation. While thermally activated motion is non-directed, inelastic electron tunneling triggers rotations, where the degree of directionality depends on the magnitude of the STM bias voltage. Below 17 K and 30 mV bias voltage, a constant rotation frequency is observed which bears the fundamental characteristics of quantum tunneling. The concomitantly high directionality, exceeding 97%, implicates the combination of quantum and non-equilibrium processes in this regime, being the hallmark of macroscopic quantum tunneling.

The acetylene on PdGa(111) motor therefore pushes molecular machines to their extreme limits, not just in terms of size, but also regarding structural precision, degree of directionality, and crossover from classical motion to quantum tunneling. This ultra-small motor thus opens the possibility to investigate in-operando effects and origins of energy dissipation during tunneling events, and ultimately energy harvesting at the atomic scales.

Significance statement

Conversion of undirected energy inputs into directed motion on molecular scales is the basis for controlled movements in living organisms. In this context, fundamental insights can be obtained by investigating artificial molecular machines under well-defined conditions. We devised the currently smallest, atomically precise molecular machine, whose rotor (C₂H₂) consists of just 4

atoms and whose functioning we have tracked employing scanning tunneling microscopy (STM). Unlike all other surface reported anchored rotors, ours is characterized by an extremely high degree of directionality which is independent on STM-tip conditions or position, therefore solely defined by the chiral support. Owing to its ultra-small size, our rotor's operation crosses the well-established classical to a unanticipated quantum tunneling kinetic regime without loss in directionality.

Article

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In 1959 Richard Feynman envisioned downscaling of information storage and machines to atomic dimensions (1). Both visions were eventually realized, by writing information via positioning single atoms on a nickel surface in 1990 (2) and by devising the first artificial, light-driven molecular machine in 1999 (3). The latter has been inspired by molecular machines in biological systems (4, 5) and lead to the design of countless further molecular machines (6–12). However, most synthetic molecular machines, although driven by quantum processes, exhibit classical kinetics (13, 14), whereas operation by quantum tunneling motion is largely elusive. STM provides an ideal platform for investigating the dynamics of atoms and molecules on surfaces (10–12, 15–22). However, few studies were aimed at achieving controlled, STM tip position independent, directional motion requiring breaking of inversion symmetry, which is commonly achieved by adsorbing chiral molecules on achiral surfaces (10–12, 15). We reverse this concept by using the surface of non-centrosymmetric PdGa crystals as chiral stator. This relaxes the geometric constraints on the rotor molecule, and allows directed motion even for simple and symmetric molecules such as C₂H₂.

Starting point of our study is the creation of a well-defined chiral surface from a non-centrosymmetric single crystal, namely the intermetallic compound Palladium-Gallium with 1:1 stoichiometry (PdGa) exhibiting bulk terminated chiral surfaces (23). The chiral structure of some of these surfaces manifests itself in pronounced enantioselective adsorption properties (24). Here we choose the three-fold symmetric ($\overline{111}$) surface of the PdGa A enantiomorph (23). Under appropriate ultra-high vacuum preparation, it terminates by a layer containing 3 Pd atoms per trigonal surface unit cell ($a_0 = 6.95 \text{ Å}$) forming an equilateral triangle of 3.01 Å side length (Fig. S1 and

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ref. (25)). The local inversion symmetry of this Pd trimer is lifted by coordination of the 6 second layer Ga atoms and furthermore by 3 Pd atoms in the third layer (Fig. 1a-b). In the following we will denote this termination as Pd₃.

On Pd₃, acetylene molecules adsorb centered on the Pd trimers (26). When imaged by STM at 5 K, they appear as dumbbells with lobe-to-lobe separation of about 3 Å in three symmetrically equivalent 120°-rotated orientations (Fig. 1e-g) between which they switch quasi instantaneously (Fig. 1c-d). Acetylene molecules are firmly anchored to the trimer and usually dissociate before being dragged off the trimer by STM tip manipulation.

We have followed the rotation events by recording tunneling current time series $I_T(t)$ at a fixed tip position (Fig. 1h), in analogy to the STM investigation of the rotation of chiral butyl-methyl-sulphide on Cu(111) (10). In the latter case, a weak ($\leq 5\%$) asymmetry in the number of clockwise (CW) n_{CW} and counter-clockwise (CCW) n_{CCW} rotations was reported and tentatively attributed to chiral STM tips, as no correlation of the directionality with the molecule's enantiomeric form was found. The $I_T(t)$ of Fig. 1h, recorded over $\Delta t = 100 \ s$, exhibits cyclic jump sequences between 3 levels (... $R_A \rightarrow R_B \rightarrow B_C \rightarrow R_A$...) with $n_{CCW} = 23$ jumps in CCW direction and $n_{CW} = 0$ in CW, resulting in a frequency $f = \frac{n_{CCW} + n_{CW}}{\Delta t} = 0.23 \ Hz$ and perfect directionality $dir = 100\% * \frac{n_{CCW} - n_{CW}}{n_{CCW} + n_{CW}} = 100\%$. The movie SV1 shows a time lapse series of STM images evidencing the prevailing CCW rotation of the motor.

Analyzing the parametric dependence of the rotation frequency (Fig. 2a-c and S2), shows that this molecular motor operates in two distinct regimes; the *tunneling regime* (TR) where its rotation frequency v_T is independent of temperature T < 15 K, bias voltage $|V_G| < 30 mV$, and current $I_T < 200 pA$, and the *classical regime* (CR) where the frequency strongly depends on these

parameters. Even though all experimental data presented in Fig. 1 have been recorded in the TR, we first discuss the CR where C₂H₂ rotations can be selectively powered by thermal or electrical excitations. We find the temperature dependence of the rotation frequency at low bias (Fig. 2a) to follow an Arrhenius characteristics (solid line in Fig. 2a) $v(T) = v_T + v_A \exp(-\frac{\Delta E_B}{k_B T})$ (1), with $v_T = 4.5 \ Hz$, $v_A = 10^{8.7 \pm 2.0} \ Hz$ (attempt frequency) and $\Delta E_B = 27.5 \pm 7.1 \ meV$ (energy barrier for rotation). Above 30 mV the frequency increases exponentially with V_G , independent of polarity (Fig. 2b-c). Under the same conditions, but at constant bias voltage, the power-law dependence $v \propto I_T^n$ with $n \approx 1$ (Fig. 2d) identifies the electronically stimulated rotation as a single-electron process (27). As we will discuss later, the parametric dependence of the rotation frequency and directionality with T, V_G and I_T is very well reproduced by a Langevin kinetic model (solid lines in Fig. 2b-c).

Before we discuss the parametric dependence of the directionality, the influence of the STM tip, required for observing the motion, must be clarified. Particularly, we have to verify that breaking of the inversion symmetry due to the tip position (and possibly tip structure) in proximity to the motor does not prevail over the influence of the chiral substrate in determining the sense of rotation. To address this issue, we have measured 6400 constant current tip-height time series $z_T(t)$ on a grid of 80×80 equidistant points covering 1×1 nm^2 in the vicinity of single acetylene molecules in the TR. Analysis of all these $z_T(t)$ series reveals an intricate, regular pattern with alternating, highly directional ascending (red) and descending (blue) jump sequences (Fig. 2e). This pattern fully corroborates a tip position independent, unidirectional rotation of the molecule, which becomes apparent by modelling and mapping the position dependent jump sequence assuming a cyclic unidirectional CCW rotation of the molecule by 60° steps (Fig. S4-S7). After op-

timizing molecule configuration and tip-shape in the model, an excellent agreement of the simulated jump sequence map (Fig. 2f) with the experiment is found. Hence, we conclude that, regardless of the tip position, the jump sequences always correspond to CCW rotations. Furthermore, as witnessed from Fig. 2g, there is no pronounced dependence of v_T on the tip position, and all three rotational C₂H₂ configurations can be expected to be energetically equivalent, as derived from the residence time analysis in Figs. S8-S10. The three rotational states only become energetically non-degenerate if the tip is brought very close to the substrate, such that it singnificantly alters the surface ratchet potential (Fig. S3). Although we investigated hundreds of molecules with tens of different tip modifications, we never observed any systematic CW rotations in the TR or CR evidencing that solely the stator dictates the direction and directionality of the rotation. Evaluating 1792 rotation events ($n_{CCW} = 1771$ and $n_{CW} = 21$) in the TR, we determine a directionality $dir \ge 96.7\%$ with 2σ confidence. By matching the simulated jump sequence map to the experiment we identify the C₂H₂ rotation to be best described as a tumbling rotor, whose center of mass moves on a circle with radius $r = 0.5 \pm 0.1$ Å and a moment of inertia $I_{C_2H_2} = 5.62 * 10^{-46} \ kgm^2$ (Fig. 2h).

Having clarified the influence of the tip, we now turn to the discussion of the parametric dependencies of the directionality (Fig. 3a-d). The temperature dependence shows a rapid drop in directionality once thermally activated rotations start to contribute significantly. The solid line in Fig. 3a assumes that v_T exhibits 98% directionality, whereas the thermally activated jumps described by the Arrhenius equation (1) are purely random. These random thermal rotation events are expected because substrate, STM tip, and hence molecule are in thermal equilibrium and accordingly, unidirectional rotation (which reduces entropy) is forbidden by the 2^{nd} law of thermodynamics. At T = 5 K a decrease of directionality is also observed for bias voltages V_G beyond

 $\pm 35~meV$ (Fig. 3b). However, unlike thermal rotations, those induced by IET only become gradually non-directional. This is clearly observed in the regime where thermally and IET induced rotations coexist. As displayed in Fig. 3c, the voltage independent directionality of only 10% at T=19~K and $|V_G|<30~mV$, can be increased significantly at higher $|V_G|$ due to additional directed IET rotations. This increase is only effective in a narrow voltage window, above which the directionality rapidly decreases. By contrast, the I_T dependence of the directionality for a fixed voltage is weak (Fig. 3d), where the slight decrease with increasing current, i.e. frequency, is attributed to the detection of two shortly successive CCW rotations as a single erroneous CW one (solid lines in Fig. 3d). Hence we conclude that directionality stays above 95% for $|V_G|<40~mV$ even at high current.

The observation of directional motion triggered from a non-cyclic, directionless, and position independent energy input stemming from a single IET event, prompts us to apply a variant of the biased Brownian motion concept proposed by Astumian and Hänggi for modelling the underlying mechanism (28, 29). Our model of IET induced rotation assumes a static, periodic, but asymmetric potential $U(\phi)$ ($\phi = [0,2\pi]$, with $\frac{\pi}{3}$ periodicity), with the asymmetry of the potential, R_{asym} , defined in the inset in Fig. 3e and S11. A single IET event instantaneously excites the molecule from its ground state and its trajectory $\phi(t)$ is obtained from Langevin dynamics $I\ddot{\phi} = -\frac{\partial U(\phi)}{\partial \phi} - \lambda \dot{\phi}$, where I is the moment of inertia and λ the viscous dissipation coefficient (28, 29). Depending on R_{asym} and λ , two dissimilar minimum kinetic energies E_L and E_R are required to overcome the barrier to the left (i.e. CW) and to the right (i.e. CCW), respectively. These energies are the basis for describing frequency and directionality by the kinetic model (see SI).

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Matching this kinetic model to our experimental data in Fig. 2c and 3c, allows determination of the temperature dependent $E_L(T)$ and $E_R(T)$ which are represented by colored markers in Fig. 3e. From these values we deduce R_{asym} to be $1.25 < R_{asym} < 1.5$ assuming $\Delta E_B = 25 meV$. The reduction of the dissipation λ from about $1.6*10^{-33} \frac{kgm^2}{s}$ at 5 K to around $1.1*10^{-33} \frac{kgm^2}{s}$ at 20 K can be attributed to the less efficient coupling of the molecule to the substrate with increasing temperature.

Having successfully described the rich phenomenology of the over-the-barrier rotation processes in the CR, the unexpected, nearly perfect unidirectional rotation of C₂H₂ in the TR requires closer inspection. Tunneling, especially of hydrogen is a well-established phenomenon in chemistry (30) and surface science (19), and plays a crucial role in numerous biological processes like enzyme catalyzed reactions (31). The approximately exponential decrease of the tunneling rate with increasing mass, however, allows reasonably high tunneling rates of heavy atoms or molecules only for very small barrier heights and tunneling distances. Despite these restrictions, many tunneling transitions on surfaces involving heavy atoms like cobalt or small molecules are reported (15, 20, 21, 32).

In this respect, the tunneling of formaldehyde (CH₂O) between two adsorption configurations on Cu(110) reported by Chiang *et al.* is very close to the C₂H₂ rotation in terms of ΔE_B , moment of inertia, and rotation angle and thus yields comparable frequencies ν_T (32). In both cases ν_T is critically tip condition dependent and varies between 0.01 Hz and 0.1 Hz for CH₂O/Cu(110) and between 0.25 Hz and 5 Hz for C₂H₂/Pd₃ surface. Thus, to evidence the strong isotopic dependence and corroborating quantum tunneling, we have paid attention that the ν_T for C₂H₂, fully (C₂D₂), and partially deuterated acetylene (C₂DH) are determined consecutively on the same

sample with the same STM tip (Fig. S15). Figure 4a shows the resulting $I_T(t)$ sequences for C₂H₂, C₂DH and C₂D₂ which reveal ν_T ratios (with respect to C₂H₂) of 1: 0.56(11): 0.24(5) (C₂H₂:C₂DH:C₂D₂), which we consistently observe with different tips (Table ST3, Fig. S16). This strong relative reduction of ν_T is contrasted by the comparatively small relative change of moment of inertia 1: 1.08: 1.2 and thus indicative for quantum tunneling. Careful inspection of the $I_T(t)$ sequence of C₂DH with broken C₂-symmetry reveals that the rotation cycles through 6 rather than 3 current levels (Fig. 4b), which proves that a full acetylene rotation indeed requires six CCW 60°-rotations. Comparison of the experimentally determined ν_T ratios to the corresponding WKB tunneling frequency (SI) shows an excellent match for a barrier height of $\Delta E_B = 25 \ meV$ (Fig. 4d).

Quantum tunneling rotations concomitant with high directionality of 97.7% allows for an estimation of the entropy change of a single tunneling rotation from the experimental CCW and CW rotation probabilities, given by $\Delta S = -k_B \ln(\frac{p_{pccW}}{p_{cW}}) \approx -k_B \ln(\frac{100}{1}) \approx -0.4 \frac{meV}{\kappa}$. This implies that the directional rotation in the TR must be a non-equilibrium process with energy dissipation $\Delta Q > 2$ meV at 5 K and $\Delta Q > 6$ meV at 15 K per rotation. As these values of ΔQ are of the order of the energy difference of two frustrated rotation modes of C_2H_2 (e.g. $\hbar\omega_{10} - \hbar\omega_{00} = 6.8$ meV; Fig. 4c, S14 and table ST2), one might assume that the required non-equilibrium tunneling proceeds via an excitation from the ground state to a bound rotational mode as proposed by Ch. Nacci et al (21). We estimate a maximum power dissipation of 100 meV/s per motor, assuming 10 Hz tunneling frequency as upper bound. On the other hand, the STM required for monitoring the rotation, locally dissipates at least 3×10^6 meV/s even at the lowest settings of 1

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pA tunneling current and 0.5 mV bias. We still observe the constant rotation frequency with persisting high directionality at such extreme settings. Therefore, the STM tip is presumably critical in driving the system out of equilibrium also in the regime of tunneling motion.

In conclusion, the highly directional tumbling rotation of C₂H₂ on the chiral PdGa{111}Pd₃ surface exhibits a rich phenomenology, most prominently characterized by an unprecedentedly high directionality and small motor size. Its rotor (C₂H₂) and stator (Pd₃-Ga₆-Pd₃ cluster) shown in Fig. 1a comprise just 16 atoms to form a unidirectional 6-state cyclic molecular motor (Fig. 4b) through all of which it cycles ceaselessly, powered exclusively by single electrons. This contrasts reported motors driven by light or chemical reactions, since for the former concerted thermal and light driven activation is required. The latter usually requires a cycling of the chemical environment to complete one cycle. In the classical regime, we could establish a Langevin kinetic model of the motion describing frequency and directionality with temperature, STM bias voltage and tunneling current. The model provides robust values for the rotational potential asymmetry R_{asym} and the temperature dependence of the viscous dissipation coefficient $\lambda(T)$ relating the operation of this molecular machine to atomic friction. The negative entropy change associated with the high rotation directionality, also observed in the tunneling regime, challenges the understanding of this simple cyclic machine in terms of dissipative quantum tunneling dynamics (33). In the future, it might be possible to convert energy via forced excitations, e.g. optical, by IET, into directional motion and thus investigating energy harvesting at the smallest possible length scale.

Materials and Methods

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All experiments were performed under ultra-high vacuum (UHV) conditions with a base pressure below 5×10^{-11} mbar using an Omicron low-temperature STM operated at 5 K. The measurements were performed with different tips including 80:20 Pt/Ir tip, Tungsten STM and Tungsten Q+ Sensor tips. We have found no systematic difference in the experimental results obtained with different tips. The PdGa crystal surface was prepared by repeated sputter and annealing cycles (sputtering: Ar+, 1 keV; annealing: 20 min at 870 K).

Before dosing C₂H₂, which was purchased from PanGas with a purity of 99.6%, the gas line was precleaned by purging with the gas or by freeze-thaw cycling (77 K). In case of C₂D₂, purchased from CDN isotopes with 99% purity with C₂DH being the main impurity, no precleaning was performed, because the gas was in a bottle with atmospheric pressure. Both gases were dosed by chamber backfilling through a leak valve at a pressure of 2x10⁻⁹ mbar. By removing the sample from the STM stage at 5 K exposing it to the acetylene outside the cryostat for a short time (generally 10-20 s) the most effective exposure conditions were achieved.

References and Notes

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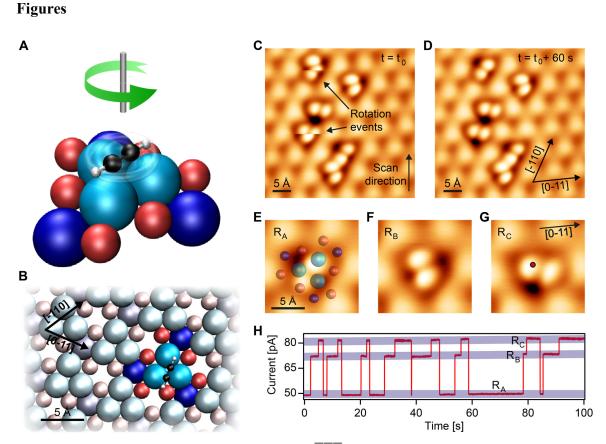


Figure 1 | Acetylene rotation on the PdGa:A(111)Pd3 surface

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a Sketch of the acetylene on Pd₃ motor. **b** Atomic structure of the PdGa:A($\overline{111}$)Pd₃ surfaces with the PdGa cluster acting as stator highlighted in saturated colors. The acetylene (C₂H₂) rotor is depicted in one (R_a) of its three equivalent adsorption configurations R_a, R_b, R_c. In **a** and **b**, the top-layered Pd trimers (z = 0) are depicted in bright blue, the second layer Ga trimers (z = -0.85 Å) in red and the third layered single Pd atoms (z = -1.61 Å) in dark blue. **c-g** Constant current STM images of C₂H₂ adsorbed on the Pd₃ surface (T = 5 K; $V_G = 10 \text{ mV}$; $I_T = 50 \text{ pA}$). In **c** two rotating molecules are pointed out, whereas in **d**, recorded 60 s after **c**, no molecular rotation is observed. **e-g** STM images of the same acetylene molecule in its three rotational configurations. In **e** the underlying PdGa stator structure is superposed. **h** Constant height $I_T(t)$ time

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series ($\Delta t = 100 \text{ s}$; $V_G = 25 \text{ mV}$; 1ms time resolution) measured at the relative position to the C₂H₂ indicated by the red marker in **g**.

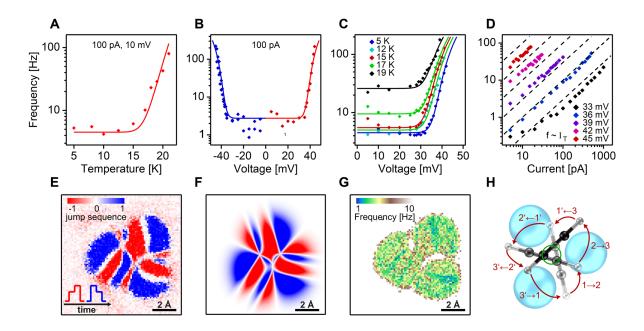
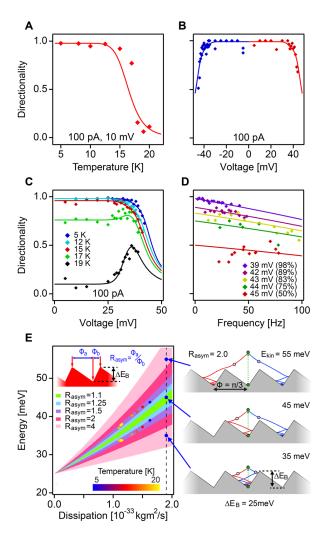


Figure 2 | Parametric dependence of the rotation frequency and jump sequence

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a Rotation frequency dependence on temperature ($V_G=10~mV$; $I_T=100~pA$), **b** on bias voltage for both polarities (T=5~K; $I_T=100~pA$), **c** on bias voltage at various temperatures between 5 K and 19 K ($I_T=100~pA$), and **d** on tunneling current for different bias voltages between 33 mV and 45 mv at T=5~K. In **a-d**, the markers represent experimental data, while the solid lines are derived from the kinetic model (SI). **e** Constant current jump sequence map (js=3*) $\frac{n_{up}-n_{down}}{n_{up}+n_{down}}=sign(js)*|dir|$; $n_{up/down}$: number of jumps increasing/decreasing the tip height) generated from a 80x80 grid (1x1 nm²) of individual $z_T(t)$ spectra, each recorded for 4s (4000 points; $V_G=10~mV$; $I_T=100~pA$). **f** Simulated jump sequence map for a 100% CCW rotation based on the motion pattern shown in **h**. **g** Frequency map of the C₂H₂ rotation extracted from the same experimental $z_T(t)$ grid of **e**. **h** our best estimation of the tumbling acetylene rotation on Pd₃ for a full 360° rotation in six 60° steps indicated by tracking the motion of one H atom $(1\rightarrow2\rightarrow3\rightarrow1'\rightarrow2'\rightarrow3')$ with n and n' denoting indistinguishable C₂H₂ configurations) with the green circle indicating the motion of acetylene's center of mass.



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Figure 3 | Parametric dependence of the nanomotor's directionality

a Dependence of the directionality on temperature ($V_G = 10 \ mV$; $I_T = 100 \ pA$), **b** bias voltage for both polarities ($I_T = 100 \ pA$; $T = 5 \ K$), **c** bias voltage at various temperatures between 5 K and 19 K ($I_T = 100 \ pA$), and **d** rotation frequency controlled via varying I_T for several V_G . In **a**-**d** the markers represent experimental data, while the solid lines in **a**-**c** are derived from the kinetic model (SI). The solid lines in **d** show simulated dependencies of constant directionality (given in brackets) with frequency considering finite time resolution of the experiment (SI). **e** Schematic representation of the Langevin rotation dynamics derived for ratchet potentials with $\Delta E_B = 25 \ meV$. On the left, the range of transferred kinetic energy E_{kin} for directed motion, i.e.

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 $E_L < E_{kin} < E_R$, in dependence of energy dissipation is colored for several R_{asym} , as defined in the inset. The experimentally determined E_L and E_R are represented by two markers of the same color for several temperatures. On the right, the trajectories of the C_2H_2 60° rotation in a ratchet potential with $R_{asym} = 2.0$, $\lambda = 2 * 10^{-33} \frac{kgm^2}{s}$ and $\Delta E_B = 25$ meV are displayed as a function of E_{kin} . For $E_L < E_R < E_{kin}$ there is no unidirected motion, $E_L < E_{kin} < E_R$ results in directed motion by overcoming the steeper potential barrier, and $E_{kin} < E_L < E_R$ induces no rotation (from top to bottom).

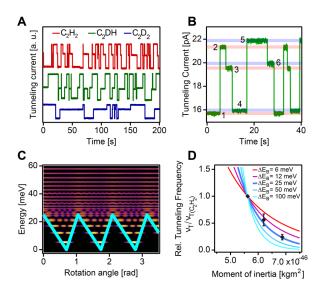


Figure 4 | Quantum tunneling rotation of acetylene

a $I_T(t)$ curves for C₂H₂, C₂DH, and C₂D₂, with a special focus on the six different current levels in a $I_T(t)$ curve of C₂DH in **b**. In **c** the ratchet potential is shown in turquois, based on which the C₂H₂ quantum states, energy levels and tunneling frequencies are determined. The color (black to yellow) represents the probability density of the quantum states. The dependence of ν_T in the WKB approximation with the moment of inertia, normalized to the ν_T at 5.62×10^{-46} kgm² (C₂H₂) is displayed as solid lines in **d** for several ΔE_B (see SI). The black markers represent the experi-

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Competing interests

The authors declare no competing interest.

Data and materials availability

The datasets generated and/or analyzed during the current study are available from the corresponding author on reasonable request.

The simulations used in the current study have been performed using a custom made code on the Wave Metrics IGOR Pro platform. Details of this code can be obtained from the corresponding author upon reasonable request.

List of Supplementary Materials

 $Table\ ST1-ST3$

 $Fig\;S1-S16$

Movie SV1