Effect of externally applied pressure on the magnetic behavior of $\text{Cu}_2\text{Te}_2\text{O}_5(\text{Br}_x\text{Cl}_{1-x})_2$

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The effect of externally applied pressure on the magnetic behavior of $\text{Cu}_2\text{Te}_2\text{O}_5(\text{Br}_x\text{Cl}_{1-x})_2$ with $x=0$, 0.73, and 1, is investigated by a combination of magnetic susceptibility, neutron diffraction, and neutron inelastic scattering measurements. The magnetic transition temperatures of the $x=0$ and 0.73 compositions are observed to increase linearly with increasing pressure at a rate of 0.23(2) and 0.04(1) K/kbar, respectively. However, the bromide shows contrasting behavior with a large suppression of the transition temperature under pressure, at a rate of $-0.95(9)$ K/kbar. In neutron inelastic scattering measurements of $\text{Cu}_2\text{Te}_2\text{O}_5\text{Br}_2$ under pressure only a small change to the ambient pressure magnetic excitations were observed. A peak in the density of states was seen to shift from ~ 5 meV in ambient pressure to ~ 6 meV under an applied pressure of 11.3 kbar, which was associated with an increase in the overall magnetic coupling strength.

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Considerable attention has been paid to frustrated quantum spin systems with reduced dimensionality in recent years due to their fascinating ground states and magnetic behavior.¹ The spin tetrahedral compounds $\text{Cu}_2\text{Te}_2\text{O}_5\text{X}_2$ ($\text{X}=\text{Br}, \text{Cl}$) are recent examples² of systems in which competing intratetrahedral and intertetrahedral interactions give rise to nontrivial ground state and dynamic magnetic behavior along with the possibility that they lie close to a quantum critical point.³

Early studies of $\text{Cu}_2\text{Te}_2\text{O}_5\text{X}_2$ revealed typical spin-gapped behavior in the magnetization and low lying singlet excitations in Raman spectroscopy.⁴ Below transition temperatures of $T_N^{\text{Br}} \sim 11.4$ K and $T_N^{\text{Cl}} \sim 18.2$ K (for $\text{X}=\text{Br}$ and Cl , respectively), the compounds form a complicated incommensurate magnetic structure.⁵⁻⁷ Inelastic neutron scattering⁸ measurements revealed further complexity in the dynamic behavior, with dispersive excitations associated with the incommensurate order present in both compounds. Theoretical work has investigated the effect of intertetrahedral coupling and the relative strengths of exchange interactions in this system.⁹⁻¹³ However, no theoretical treatment has satisfactorily explained the experimentally observed behavior, and the true nature of the underlying magnetic interactions remains unclear.

External pressure has been used as a tool to tune the underlying interactions in many magnetically ordered systems, and has proved to be an invaluable technique for investigating quantum critical points.¹⁴⁻¹⁶ Previous magnetic susceptibility measurements of polycrystalline $\text{Cu}_2\text{Te}_2\text{O}_5\text{Br}_2$ under

applied pressure have been reported by Kreitlow *et al.*¹⁹ T_N^{Br} is reported to quickly decrease under applied pressure and no longer be observable at 8.2 kbar; indicating that $\text{Cu}_2\text{Te}_2\text{O}_5\text{Br}_2$ lies close to a nonmagnetically ordered phase. In contrast, the temperature of the maxima in the susceptibility (T_{max}), which is associated with the overall magnetic coupling strength, is observed by Kreitlow *et al.* to increase with increasing applied pressure, by up to 25% under 8.2 kbar. A structural analysis of $\text{Cu}_2\text{Te}_2\text{O}_5\text{Br}_2$ under pressure using angle-dispersive x-ray powder diffraction has also been reported,²⁰ with the atomic positions refined in pressures up to 140 kbar. It is observed that the intertetrahedral Br-Br distance decreases with increasing pressure and the Cu-Br-Br path becomes slightly more linear, while the Cu-Cu distances increase under pressure. In this paper, the effect of externally applied pressure on the magnetic behavior of $\text{Cu}_2\text{Te}_2\text{O}_5(\text{Br}_x\text{Cl}_{1-x})_2$ is investigated through a combination of magnetic susceptibility, neutron diffraction and neutron inelastic scattering measurements under pressure.

Susceptibility measurements as a function of temperature were performed with a Quantum Design SQUID magnetometer. Hydrostatic external pressure was applied using an easyLab Technologies Mcell 10 pressure cell, using Sn as an *in situ* manometer. Neutron diffraction measurements were performed on a single crystal of $\text{Cu}_2\text{Te}_2\text{O}_5\text{Cl}_2$ with dimensions $5 \text{ mm} \times 2.5 \text{ mm} \times 2.0 \text{ mm}$, on the D10 diffractometer at the ILL. Hydrostatic pressure was applied to the sample using a CuBe clamp cell with Fluorinert pressure medium, and the *in situ* pressure was determined using a NaCl

FIG. 1. (Color online) The second derivative of the magnetic susceptibility ($d^2\chi/dT^2$) as a function of temperature for $\text{Cu}_2\text{Te}_2\text{O}_5\text{Br}_2$ under a number of externally applied pressures.

manometer.^{17,18} Neutron inelastic scattering measurements of polycrystalline $\text{Cu}_2\text{Te}_2\text{O}_5\text{Br}_2$ were performed on the direct chopper spectrometer HET at ISIS. A CuBe pressure cell was used and the pressure determined from diffraction measurements of a NaCl manometer on PRISMA at ISIS.

In ambient pressure magnetic susceptibility (χ) measurements of $\text{Cu}_2\text{Te}_2\text{O}_5\text{Br}_2$ as a function of temperature, χ is observed to have a maxima at $T_{\text{max}} \sim 30$ K, before dropping abruptly to a much reduced level at lower temperatures.⁴ The transition to long range magnetic order at $T_N^{\text{Br}} \sim 11.4$ K corresponds to a steplike feature in $d\chi/dT$, which in turn corresponds to a peak in $d^2\chi/dT^2$.

Figure 1 shows $d^2\chi/dT^2$ versus temperature for $\text{Cu}_2\text{Te}_2\text{O}_5\text{Br}_2$ in an applied magnetic field of 50 kOe as a function of temperature in a number of different applied pressures. Here, the transition temperature (corresponding to the maxima in the peak of $d^2\chi/dT^2$) is seen to decrease linearly in applied pressure, from ~ 11.5 K in 1.1 kbar, to ~ 6.1 K in 5.6 kbar. The peak width increases significantly with increasing applied pressure. By 7.6 kbar there is no longer a clearly defined peak that can be associated with the magnetic transition, although there is a small, broad peak centered at ~ 5 K. Unfortunately, below ~ 4 K, $d^2\chi/dT^2$ becomes rather noisy. This is believed to be because of problems with the background subtraction at low temperatures where the sample susceptibility is very low, and the signal from the pressure cell becomes large due to the presence of paramagnetic impurities in the cell. It is therefore difficult to ascertain whether or not the ordering temperature is suppressed toward $T=0$ K by further increasing the applied pressure. The pressure dependence of T_N^{Br} observed in this data differs somewhat from the results reported by Kreitlow *et al.*¹⁹ In their work, they report that $T_N^{\text{Br}} \sim 5$ K at a pressure of 3.5 kbar and as a consequence the pressure dependence is rather nonlinear. The large number of pressures measured in our work have allowed the observation of a more consistent, possibly linear pressure dependence up to at least 5 kbar. In contrast, the temperature at which the maxima in the susceptibility occurs is observed to shift linearly to higher tempera-

FIG. 2. The transition temperature T_N^{Br} and the temperature of the maxima in the susceptibility T_{max} of $\text{Cu}_2\text{Te}_2\text{O}_5\text{Br}_2$ as a function of pressure.

tures with increasing pressure, from ~ 29 K in 1.1 kbar, to ~ 39 K in 7.6 kbar, in good agreement with the work of Kreitlow *et al.*¹⁹ The pressure dependence of T_{max} is displayed in Fig. 2, along with that of T_N^{Br} . Linear fits give gradients of $-0.95(9)$ K/kbar and $1.42(6)$ K/kbar for T_N^{Br} and T_{max} , respectively.

Similar measurements have been performed with polycrystalline samples of $\text{Cu}_2\text{Te}_2\text{O}_5\text{Cl}_2$ and mixed composition $\text{Cu}_2\text{Te}_2\text{O}_5(\text{Br}_x\text{Cl}_{1-x})_2$ with $x=0.73$. Figures 3 and 4 show $d^2\chi/dT^2$ as a function of temperature for a number of different pressures for the chloride and $x=0.73$ sample, respectively. This data was taken in an applied field of 1 kOe for the chloride and 50 kOe for the $x=0.73$ sample. The transition temperatures (T_N^{Cl} and $T_N^{x=0.73}$) correspond to the minima in $d^2\chi/dT^2$. In the chloride, T_N^{Cl} is observed to increase in a linear fashion with increasing pressure, while for the $x=0.73$ $T_N^{x=0.73}$ barely changes with applied pressure, increasing by less than 1 K from ambient pressure to an applied pressure of 10.5 kbar.

The temperature of the maxima in the susceptibility (T_{max}) was observed to increase with applied pressure in both samples, from ~ 23 K in ambient pressure to ~ 28 K in 10.0 kbar for the chloride and from 19 K in ambient pressure

FIG. 3. (Color online) The second derivative of the magnetic susceptibility ($d^2\chi/dT^2$) as a function of temperature for $\text{Cu}_2\text{Te}_2\text{O}_5\text{Cl}_2$ under different externally applied pressures.

FIG. 4. (Color online) The second derivative of the magnetic susceptibility ($d^2\chi/dT^2$) as a function of temperature for $\text{Cu}_2\text{Te}_2\text{O}_5(\text{Br}_x\text{Cl}_{1-x})_2$ with $x=0.73$, for different externally applied pressures.

to 30 K in 10.5 kbar for the mixed composition. Figure 5 plots the pressure dependence of T_N^{Cl} and T_{max} for the chloride, with linear fits giving gradients of 0.23(2) K/kbar and 0.52(3) K/kbar, respectively. Figure 6 shows the pressure dependence of $T_N^{x=0.73}$ and T_{max} for the $x=0.73$ compound, both of which show a linear relationship with gradients of 0.04(1) and 1.00(1) K/kbar, respectively.

Neither the transition temperature nor T_{max} respond as strongly to pressure as they do in the case of $\text{Cu}_2\text{Te}_2\text{O}_5\text{Br}_2$. T_{max} is almost three times as responsive to pressure in the case of $X=\text{Br}$ than $X=\text{Cl}$, and T_N is almost four times as responsive to pressure for $X=\text{Br}$ compared to $X=\text{Cl}$, and, moreover, the pressure dependence acts in the opposite sense for the two compounds. While for the bromide T_N decreases with increasing pressure, for the chloride T_N increases with increasing pressure, indicating that pressure has a significantly different effect on these two compounds. For the $x=0.73$ sample, T_{max} behaves under pressure in an intermediate manner to the two end compounds, while the small

FIG. 5. The transition temperature T_N^{Cl} and the temperature of the maxima in the susceptibility T_{max} of $\text{Cu}_2\text{Te}_2\text{O}_5\text{Cl}_2$ as a function of pressure.

FIG. 6. The transition temperature ($T_N^{x=0.73}$) and the temperature of the maxima in the susceptibility (T_{max}) of $\text{Cu}_2\text{Te}_2\text{O}_5(\text{Br}_x\text{Cl}_{1-x})_2$ with $x=0.73$ as a function of pressure.

increase in $T_N^{x=0.73}$ with pressure is closer to the behavior of the chloride.

In order to correlate the effect of pressure observed in these macroscopic magnetic susceptibility measurements with a microscopic view of the system, we have also performed neutron diffraction measurements of single crystal $\text{Cu}_2\text{Te}_2\text{O}_5\text{Cl}_2$ under an applied pressure of 4.5(3) kbar. The modulation vector of the incommensurate magnetic structure ($\mathbf{k}'_{\text{Cl}} \approx [-0.15, 0.42, 1/2]$, see Ref. 5) was not observed to change from that of the ambient pressure structure when under an applied pressure of 4.5 kbar. Figure 7 shows the integrated intensity of the $\mathbf{Q}=(0.56 \ 0.85 \ 0.5)$ magnetic reflection over the temperature range 2–18.5 K, in which the data taken in ambient pressure (filled circles) and 4.5 kbar (empty circles) have been normalized to the same intensity scale. The data shows a clearly resolved shift in the temperature dependence of the integrated intensity under an applied pressure. The intensity of the magnetic reflection in ambient pressure drops to $\sim 13\%$ of its value at 2 K by 17.75 K, and by 18 K there is no longer a discernible peak. In contrast, in the 4.5 kbar data, the intensity of the magnetic reflection at 18.5 K is still $\sim 18\%$ of its value at 2 K, although at higher temperatures no peak can be resolved above the background

FIG. 7. Normalized integrated intensity of the $\mathbf{Q}=(0.56 \ 0.845 \ 0.5)$ reflection as a function of temperature for $\text{Cu}_2\text{Te}_2\text{O}_5\text{Cl}_2$. The filled and empty circles represent the ambient pressure and 4.5 kbar data, respectively.

FIG. 8. (Color online) $S(|\mathbf{Q}|, \hbar\omega)$ versus energy transfer for $\text{Cu}_2\text{Te}_2\text{O}_5\text{Br}_2$, summed over $0.5 \text{ \AA}^{-1} < |\mathbf{Q}| < 1.5 \text{ \AA}^{-1}$. Data for the sample in the pressure cell under 11.3 kbar is shown as circles, for the sample in the pressure cell at ambient pressure as triangles, and for the sample at ambient pressure and not in the pressure cell as squares.

level of scattering. The ambient pressure transition temperature of $\text{Cu}_2\text{Te}_2\text{O}_5\text{Cl}_2$ is $T_N^{\text{Cl}} = 18.2 \text{ K}$, while under a pressure it is observed to shift to $T_N^{\text{Cl}} \sim 19 \text{ K}$, in good agreement with the susceptibility measurements of $\text{Cu}_2\text{Te}_2\text{O}_5\text{Cl}_2$ presented above. Similar neutron diffraction measurements of $\text{Cu}_2\text{Te}_2\text{O}_5\text{Br}_2$ and the $x=0.73$ composition under pressure have not been possible to date due to the difficulty of growing large enough single crystals of these compounds.

The effect of applied pressure on the dynamic magnetic behavior of polycrystalline $\text{Cu}_2\text{Te}_2\text{O}_5\text{Br}_2$ has been investigated using neutron inelastic scattering (NIS) on HET at ISIS. An incident energy of $E_i = 18 \text{ meV}$ and chopper frequency of 150 Hz were used, giving an accessible $|\mathbf{Q}|$ range of ~ 0.5 to $1.5 \text{ \AA}^{-1}$ at an energy of 5 meV. Previous NIS measurements⁸ of $\text{Cu}_2\text{Te}_2\text{O}_5\text{Br}_2$ in ambient pressure have revealed the presence of a magnetic excitation with a flat component centered in energy at $\sim 5 \text{ meV}$, and a dispersive component centered at $|\mathbf{Q}| \sim 0.7 \text{ \AA}^{-1}$.

Figure 8 shows $S(|\mathbf{Q}|, \hbar\omega)$ versus $\hbar\omega$ summed over the $|\mathbf{Q}|$ range $0.5 \text{ \AA}^{-1} < |\mathbf{Q}| < 1.5 \text{ \AA}^{-1}$ for $\text{Cu}_2\text{Te}_2\text{O}_5\text{Br}_2$ at ambient pressure (triangles) and under an applied pressure of 11.3 kbar (circles) at $T=4 \text{ K}$. In both cases the background scattering from the pressure cell has been subtracted. The data shown as squares in Fig. 8 was taken at ambient pressure and outside of the pressure cell, such that the absorption was reduced and better statistics could be obtained in shorter counting times. A peak in the density of states is observed at $\sim 5 \text{ meV}$ in ambient pressure for the case in which the sample is in the pressure cell as well as the case in which it is not in the pressure cell. Indeed, $S(|\mathbf{Q}|, \hbar\omega)$ versus $\hbar\omega$ for both of the ambient pressure measurements are very similar, as expected, although there are small discrepancies, particularly at low energy transfer ($\sim 1-4 \text{ meV}$) $S(|\mathbf{Q}|, \hbar\omega)$, which may be an artifact of the background subtraction due to a possible overestimation in this region. However, the peak in the density of states for the sample under 11.3 kbar is at

FIG. 9. (Color online) $S(|\mathbf{Q}|, \hbar\omega)$ versus energy transfer for $\text{Cu}_2\text{Te}_2\text{O}_5\text{Br}_2$, in which the ambient pressure data has been subtracted from the 11.3 kbar data. The data has been summed over $0.5 \text{ \AA}^{-1} < |\mathbf{Q}| < 1.5 \text{ \AA}^{-1}$. The solid line is a guide to the eye.

$\sim 6 \text{ meV}$, showing a shift of approximately 1 meV from the ambient pressure data. This shift is more clearly illustrated in Fig. 9, in which the ambient pressure data (taken with the sample in the pressure cell) is subtracted from the 11.3 kbar data. The experimental setup and instrumental configuration were identical in both cases, hence the subtraction gives the difference in $S(|\mathbf{Q}|, \hbar\omega)$ as a function of energy transfer between measuring under 11.3 kbar and in ambient pressure. The data shows an S-like feature (marked as a solid line in Fig. 9 as a guide to the eye), which corresponds to the shifting of the peak under pressure. The negative part of the S-shape feature indicates that there is less intensity in the 11.3 kbar data in the energy region $\sim 3-5.5 \text{ meV}$, in comparison with the ambient pressure data. Similarly, the positive part of the S-shape indicates that there is more intensity in the 11.3 kbar data in the region $\sim 5.5-8 \text{ meV}$ compared with the ambient pressure data. From Fig. 8, it appears that there may also be a small broadening of the peak as well as a shift of the center of the peak to higher energy when under applied pressure.

Pressure measurements often provide useful information about the underlying magnetic interactions in a system. As pressure is applied and the sample volume decreases one would expect, in general, the magnetic coupling strengths to increase due to the closer proximity of the atoms and resultant increased overlap of their orbitals. In particular, for $\text{Cu}_2\text{Te}_2\text{O}_5\text{Br}_2$ one may naively expect that the effect of pressure would be to push the magnetic behavior toward that of the $\text{Cu}_2\text{Te}_2\text{O}_5\text{Cl}_2$ compound, which has a 7% smaller volume. In this system there are both intratetrahedral and intertetrahedral competing interactions, as well as the presence of an antisymmetric DM interaction. Due to the competition of the different exchange paths, the magnetic ordering is likely to be very sensitive to the relative coupling strengths of the interactions present, which may respond to pressure in different manners. It is therefore important to closely consider the effect of pressure on the structure of the material alongside the effects observed in the magnetic behavior. The structural results of Wang *et al.*²⁰ for $\text{Cu}_2\text{Te}_2\text{O}_5\text{Br}_2$ suggest that the inter-tetrahedral exchange interactions J_a , J_b , and J_c may possibly increase under applied pressure, and the intra-tetrahedral exchange interactions J_1 and J_2 may possibly de-

crease under applied pressure (using the exchange interaction notation of Whangbo *et al.*¹¹).

First, consider $T_{\max}$, which is observed to increase linearly with increasing applied pressure for each of the samples. In general, $T_{\max}$ is associated with the overall coupling strength of the system, which therefore appears to increase under pressure for all compositions. However, it is not clear which exchange interactions determine the overall magnetic coupling in these materials. In the case of $\text{Cu}_2\text{Te}_2\text{O}_5\text{Br}_2$, the work of Wang *et al.*²⁰ suggests that the intertetrahedral coupling may increase with applied pressure, whereas the intratetrahedral coupling possibly decreases with applied pressure. This indicates that the overall coupling strength may perhaps be determined by the intertetrahedral coupling. In the case of $\text{Cu}_2\text{Te}_2\text{O}_5\text{Cl}_2$ and the doped $x=0.73$ composition there is no corresponding structural data, and so it is not known how the different exchange paths respond to pressure relative to each other.

Now consider the behavior of T_N under pressure for each of the compositions. In contrast to the chloride and $x=0.73$ sample, the transition temperature of the bromide decreases with increasing pressure, and the compound appears to move toward a nonmagnetic phase. However, it is not clear how the pressure dependence of T_N^{Br} develops beyond ~ 6 kbar, and whether or not it is completely suppressed at some higher pressure. It has previously been suggested that $\text{Cu}_2\text{Te}_2\text{O}_5\text{Br}_2$ and $\text{Cu}_2\text{Te}_2\text{O}_5\text{Cl}_2$ lie in the proximity of a quantum critical point.^{3,4} These results may therefore suggest that externally applied pressure has the effect of pushing the bromide closer to the quantum critical point, and pushing the chloride away from it. The $x=0.73$ composition, although somewhat intermediate in its behavior, seems to follow more closely the chloride and is pushed slightly further from the possible quantum critical point with a very small increase in $T_N^{x=0.73}$ under pressure.

The contrasting relative behavior of T_N and $T_{\max}$ may also give information about the underlying magnetic interactions in the three compounds. Firstly, in a low dimensional system the maxima in the magnetic susceptibility often gives a better indication of the underlying strength of the magnetic interactions than the actual transition temperature, which can reflect weaker “parasitic” interactions that finally lock in the three-dimensional order [examples of this are copper oxychlorides $\text{CaGdCuO}_3\text{Cl}$ and $\text{Ca}_4\text{R}_2\text{Cu}_3\text{O}_8\text{Cl}_4$ ($\text{R}=\text{Gd}, \text{Sm}$)²¹]. Indeed, for $\text{Cu}_2\text{Te}_2\text{O}_5(\text{Br}_x\text{Cl}_{1-x})_2$ the issue of the relative strengths of the possible interactions, and which determine the low dimensionality or which is dominant in driving the transition to 3D order, has yet to be established. However, it is clear that in the bromide the interaction driving the 3D order is not the same as that which can be thought of as mediating the overall (possibly low dimensional) coupling because $T_{\max}$ and T_N^{Br} act in the opposite sense under pressure. In the $\text{Cu}_2\text{Te}_2\text{O}_5\text{Cl}_2$ and $x=0.73$ samples, $T_{\max}$ and T_N both increase under pressure. This may perhaps indicate that the interactions driving the overall magnetic coupling strength also play a role in determining the magnetic ordering temperature. However, it may not be sufficient to think solely in terms of the relative strength of the inter and intratetrahedral interactions. Another means by which the magnetic transition in $\text{Cu}_2\text{Te}_2\text{O}_5\text{Br}_2$ could be suppressed is by an

increase in the frustration on the tetrahedra, or possibly a weakening of the DM interaction under pressure, as these are also parameters that may play an important part in the stabilization of a magnetically ordered state in this system.

The effect of externally applied pressure on the dynamic magnetic behavior of $\text{Cu}_2\text{Te}_2\text{O}_5\text{Br}_2$ appears to be rather more subtle than its effect on the magnetic transition temperature. The most notable change in the dynamics under an applied pressure of 11.3 kbar is a shift of the peak in the density of states of the magnetic excitations to 6 meV, from 5 meV in ambient pressure.

In the isolated tetrahedra model, the magnetic excitation present at 5 meV in ambient pressure measurements would correspond to a singlet-triplet spin-gap of $\Delta_{\text{Br}}=J_1=5$ meV ($J_1>J_2$). Under pressure, the increase of the peak in the density of states to 6 meV would require the intratetrahedral interaction J_1 to increase. However, the measurements of Wang *et al.*²⁰ suggest that under pressure the J_1 interaction may perhaps decrease. If, however, the system is considered to consist of isolated square planar units mediated by the intertetrahedral J_a and J_b interactions, then the peak energy in the density of states would be determined by the strength of J_a and J_b . The work of Wang *et al.*²⁰ indicates that these intertetrahedral J_a and J_b interactions do possibly increase under pressure, which could explain the small shift of the peak in the density of states to 6 meV. However, in a magnetically ordered system there must be some form of coupling between the Cu^{2+} clusters, whether they are tetrahedra or square planar units. Nevertheless, an increase in the overall magnetic coupling with pressure (as observed in the magnetic susceptibility measurements) is consistent with an increase in the energy scale of the magnetic excitations. Unfortunately the statistics of the NIS measurements under pressure are not sufficient to carry out a detailed analysis of the $|\mathbf{Q}|$ dependence in different regions of energy transfer. However, it does appear that the dispersive component of the magnetic excitation in $\text{Cu}_2\text{Te}_2\text{O}_5\text{Br}_2$ observed in ambient pressure is still present when a pressure of 11.3 kbar is applied. If the behavior of T_N^{Br} as a function of pressure displayed in Fig. 2 is extrapolated to 11.3 kbar, one would expect the transition temperature to be lower than 4 K, and possibly even suppressed close to $T=0$ K. Therefore the NIS measurement was most probably performed above the transition temperature, in which case it is proposed that the dispersive excitations are supported by low dimensional order or short range correlations.

In conclusion, the effect of externally applied pressure on the magnetic transition temperature (T_N) of $\text{Cu}_2\text{Te}_2\text{O}_5(\text{Br}_x\text{Cl}_{1-x})_2$ with $x=0, 0.73$, and 1, has been studied using a combination of susceptibility measurements and neutron diffraction. T_N is observed to increase linearly with increasing pressure for the chloride, while it decreases rapidly toward $T=0$ K in the case of the bromide. For the mixed composition the behavior of T_N is somewhat $X=\text{Cl}$ -like, with a very small increase with increasing pressure. In all three compounds the temperature of the maxima in the susceptibility ($T_{\max}$), which is associated with the overall magnetic coupling strength, was observed to increase significantly with pressure. Neutron inelastic scattering (NIS) measurements of $\text{Cu}_2\text{Te}_2\text{O}_5\text{Br}_2$ revealed a small shift in the peak in

the magnetic density of states from ~ 5 meV in ambient pressure to ~ 6 meV under an applied pressure of 11.3 kbar, which is associated with the increase in the overall magnetic coupling strength. In addition, the dispersive component of the magnetic excitation is still present at 11.3 kbar, and may be supported by low dimensional short range order.

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- ¹*Magnetic Systems with Competing Interactions*, edited by H. T. Diep (World Scientific, Singapore, 1994).
- ²M. Johnsson, K. W. Tornroos, F. Mila, and P. Millet, *Chem. Mater.* **12**, 2853 (2000).
- ³C. Gros, P. Lemmens, M. Vojta, R. Valentí, K.-Y. Choi, H. Kageyama, Z. Hiroi, N. V. Mushnikov, T. Goto, M. Johnsson, and P. Millet, *Phys. Rev. B* **67**, 174405 (2003).
- ⁴P. Lemmens, K.-Y. Choi, E. E. Kaul, C. Geibel, K. Becker, W. Brenig, R. Valentí, C. Gros, M. Johnsson, P. Millet, and F. Mila, *Phys. Rev. Lett.* **87**, 227201 (2001).
- ⁵O. Zaharko, A. Daoud-Aladine, S. Streule, J. Mesot, P.-J. Brown, and H. Berger, *Phys. Rev. Lett.* **93**, 217206 (2004).
- ⁶O. Zaharko, H. M. Ronnow, A. Daoud-Aladine, S. Streule, F. Juranyi, J. Mesot, H. Berger, and P. J. Brown, *Fiz. Nizk. Temp.* **31**, N8/9 (2005).
- ⁷O. Zaharko, H. Ronnow, J. Mesot, S. J. Crowe, D. McK. Paul, P. J. Brown, A. Daoud-Aladine, A. Meents, A. Wagner, M. Prester, and H. Berger, cond-mat/0512617 (unpublished).
- ⁸S. J. Crowe, S. Majumdar, M. R. Lees, D. McK. Paul, R. I. Bewley, S. J. Levett, and C. Ritter, *Phys. Rev. B* **71**, 224430 (2005).
- ⁹R. Valentí, T. Saha-Dasgupta, C. Gros, and H. Rosner, *Phys. Rev. B* **67**, 245110 (2003).
- ¹⁰W. Brenig, *Phys. Rev. B* **67**, 064402 (2003).
- ¹¹M.-H. Whangbo, H.-J. Koo, D. Dai, and D. Jung, *Inorg. Chem.* **42**, 3898 (2003).
- ¹²J. Jensen, P. Lemmens, and C. Gros, *Europhys. Lett.* **64**, 689 (2003).
- ¹³V. N. Kotov, M. E. Zhitomirsky, M. Elhajal, and F. Mila, *Phys. Rev. B* **70**, 214401 (2004).
- ¹⁴C. Pfleiderer, G. J. McMullan, S. R. Julian, and G. G. Lonzarich, *Phys. Rev. B* **55**, 8330 (1997).
- ¹⁵M. Matsumoto, B. Normand, T. M. Rice, and M. Sigrist, *Phys. Rev. B* **69**, 054423 (2004).
- ¹⁶C. Pfleiderer and A. D. Huxley, *Phys. Rev. Lett.* **89**, 147005 (2002).
- ¹⁷D. L. Decker, *J. Appl. Phys.* **36**, 1 (1965).
- ¹⁸D. L. Decker, *J. Appl. Phys.* **42**, 8 (1971).
- ¹⁹J. Kreitlow, S. Sullow, D. Menzel, J. Schoenes, P. Lemmens, and M. Johnsson, *J. Magn. Magn. Mater.* **290-291**, 959 (2005).
- ²⁰X. Wang, I. Loa, K. Syassen, P. Lemmens, M. Hanfland, and M. Johnsson, *J. Phys.: Condens. Matter* **17**, S807 (2005).
- ²¹A. Sundaresan, A. Maignan, and B. Raveau, *Phys. Rev. B* **54**, 12576 (1996).